Remedial Investigation Report for the Cold Creek/LeMoyne Site Mobile County, Alabama

Draft (Revision # 2)

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Throughout this report, a number of chemical product names appear. The following list indicates those products which have registered tradenames:

Betasan
Imidan
Dyfonate
Crystex
Trithion
Eptam
Sutan
Vernam
Tillam
Ordram
Ro-neet

Site Background

Stauffer Chemical Company previously owned and operated a multi-product inorganic chemical manufacturing plant at LeMoyne, Alabama and an agricultural chemical facility at the adjacent Cold Creek site. The LeMoyne plant, purchased by Akzo Chemie America, Inc. in 1987, has been in operation since 1953. The Cold Creek plant has been in operation since 1966 and is currently owned by ICI Americas, Inc. Halby Chemical Company (later part of Witco, Inc.) also operated a small facility for a time on a leased section of the LeMoyne property.

Until 1973, industrial wastes from these operations were disposed in unlined disposal areas and, in the case of wastewater, to unlined ponds or, after treatment, by discharge to Cold Creek Swamp. Presumably as a result of these practices, a ground-water contamination problem developed. This was recognized in the early 1970's, and many improvements and waste-handling modifications were made. Lined ponds were installed, solid wastes were diverted for off-site treatment and/or disposal, and the existing disposal sites were cleaned, consolidated, and capped with impermeable liners and clay. The ground-water problem was addressed by installation of an intercept and treatment system. This latter work was conducted with the review of, and approval by, the Alabama Water Improvement Commission (AWIC), predecessor agency to the present Alabama Department of Environmental Management (ADEM).

In 1982, an assessment of the plant sites was made by the Alabama Department of Public Health in response to submissions made by Stauffer to the House Committee on Interstate Commerce ("the Eckhardt Survey"). At the request of the Alabama Department of Public Health, monitoring wells were installed around the three closed landfills. In spite of the previously identified ground-water problems already under remediation,

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data primarily from these monitoring wells were held by the Federal Environmental Protection Agency (EPA) to be the basis for inclusion of these facilities on the National Priorities List (NPL), which ranks hazardous waste disposal sites under provisions of the Comprehensive Environmental Response, Compensation and Liability Act of 1980 (CERCLA), commonly known as "Superfund".

Purpose of Investigation

Camp Dresser and McKee, Inc. (CDM), under contract to the EPA, conducted preliminary sampling at the site in May of 1985 and prepared a Work Plan which is the basis for this Remedial Investigation (RI). Based on the sampling and previous investigations of both the Cold Creek and LeMoyne sites, and offsite on Courtaulds North America's (CNA) property, CDM concluded that there was possible ground-water contamination (primarily mercury, carbon tetrachloride, carbon disulfide and thiocarbamates). Further, CDM suggested that some contaminants were moving offsite towards the CNA production wells. The major potential sources of contamination were considered to be the Cold Creek Swamp, unlined waste holding and treatment ponds, and the Cold Creek and LeMoyne landfills (see Figure ES-1).

For the purpose of the RI, the Cold Creek and LeMoyne properties are considered one site, as outlined in the Work Plan and agreed upon by the EPA. The purpose of this Remedial Investigation is to characterize the type and extent of contamination; to identify contamination sources, migration pathways, and the potential for adverse environmental impacts; and to provide a basis for evaluation of the most cost-effective remedial action alternatives.

The Cold Creek/LeMoyne site is located just off U.S. Highway 43, approximately 20 miles north of Mobile, Alabama (see Figure 1-1 in Section 1.2). The site is surrounded by several other chemical production plants, and the site area is very sparsely populated, the nearest community being Creola, 5

miles to the south. The Cold Creek Swamp lies between the plant sites and the Mobile River, which is approximately 1 1/2 miles to the east of the main facilities. The swamp flows northeast, then east, discharging to the Mobile River.

The Cold Creek plant manufactures proprietary herbicides and pesticides used in agricultural farming. Among the principal products made are Betasan, Imidan, Dyfonate and several thiocarbamates. The LeMoyne plant manufactures carbon disulfide, carbon tetrachloride, sulfuric acid, caustic, chlorine and Crystex (a proprietary sulfur compound).

Previous Remedial Activities

As mentioned above, after ground-water contamination was discovered in the early 1970's, investigations of potential sources were initiated and clean-up activities begun. Two unlined waste burial sites at Cold Creek were capped as was the LeMoyne landfill. The use of unlined wastewater treatment ponds was discontinued, and several were closed. New lined ponds were installed, and the treated wastewater was discharged to the Mobile River. Spill control and storm-water recycling and drainage controls were put in place. Low-lying plant areas adjacent to the unnamed stream feeding the Cold Creek Swamp were selectively backfilled with clean fill material to control flooding. A number of monitoring wells were drilled and ground-water analysis commenced.

Finally, after a hydrogeologic investigation by Ground Water Associates, Inc. (GWA), Stauffer, in 1980, installed a ground-water intercept and treatment system. This system, which has been operating since 1981, consists of three interceptor wells with a total design capacity of 1500 gallons per minute (gpm). The wells are situated along the southern property line and are located just downgradient of the inactive carbon tetrachloride (CTC) plant wastewater treatment (WWT) pond. Contaminated ground water is pumped to an air stripping/aeration pond and, following treatment, is discharged

to the Mobile River under provisions of an NPDES permit (see Appendix XXIII for a more detailed description of the ground-water intercept and treatment program). Since this installation, the ground water has steadily improved in quality.

The RI field investigation, as proposed in CDM's Work Plan, was carried out in total except for the conditional Phase II swamp sampling. Based on the initial swamp soil sampling and the ground-water sampling results, EPA decided to omit the Phase II sampling. A total of 311 samples were collected between May and August of 1986. Complete analytical results are included in Appendices I-1 through V-2, and summary tables are presented in Chapters 1 and 5.

Major Investigation Findings

The Cold Creek Swamp was sampled at 34 locations with 3-foot deep soil borings (see Drawing Number 1.3 in Appendix XVII for locations). The same technique was used at four locations in the LeMoyne Swamp (see Figure 5-3). composite soil samples were analyzed for thiocarbamates, chlorides and priority pollutants (Tables 5-7 and 5-8), and 31 samples for mercury only (Table 5-9). Except for mercury, no priority pollutants were found other than typical levels of some heavy metals commonly found in natural soils (see Appendix XVIII for comparison). Most thiocarbamates were found to be non detectable, with a few between 0.1 and 1.8 milligrams per kilogram (mg/kg, or parts per million, ppm). Mercury concentrations, as shown in Table 5-9 and Drawing 1.3 (Appendix XVII), indicated low to high (BMDL to 690 mg/kg) levels. No mercury was found in any of the ground-water samples indicating, as shown later, that mercury was not being transmitted from the swamp to adjacent underlying ground waters.

Fish samples were collected at five locations and analyzed for mercury. Levels ranged from 0.4 to 3.1 mg/kg whole fish. The species of fish collected are shown in Appendix XXI.

A total of twelve soil samples were taken around the three landfills (see Figure 5-1 and Tables 5-1, 5-2 and 5-3). No priority pollutants were found other than low parts per million (ppm) levels of a few heavy metals. A few samples showed above average values for antimony and mercury. The area around and under the Cold Creek landfills showed no detectable levels of site-specific (production-related) compounds with minor exceptions, the highest being 1.5 mg/kg molinate. Vanadium levels were typically 1.1 to 30 mg/kg, which are low compared to those found in natural soil (20 to 500 mg/kg; see Appendix XVIII). The synthetic membrane covering of each of the landfills was exposed and sampled. These were found to be sound with no apparent deterioration (see Appendix XVI).

Eighteen (18) soil borings were made around nine ponds (see Figure 5-2, Tables 5-4 and 5-5). Analysis of composite samples did not detect priority pollutants except for background levels of some heavy metals. A sample taken inside the closed Halby pond showed high levels of copper (442 mg/kg), zinc (1,170 mg/kg) and cyanide (240 mg/kg), but samples taken adjacent to the pond were at or below background levels for these compounds. Heavy metals were not found in the ground water. Site-specific compounds were not detected in soil samples with the exception of thiocyanate found in soil in and around the Halby pond and low levels of thiocarbamates under Cold Creek's closed neutralization pond. Priority pollutants were not detected in surface-water samples from two small unnamed tributaries to Cold Creek or in samples taken from three active ponds.

Ground-water samples were collected from 15 source wells and 36 area wells (see Figures 4-2 and 4-3 and Tables 5-11 and 5-12). Except for expected high levels of carbon disulfide (CS₂) and carbon tetrachloride (CTC) in wells 0-29 and 0-31, which are located just downgradient of the old CTC plant wastewater treatment pond (see Figure 5-6), all other samples showed essentially no detectable levels of priority pollutants. Three other wells in the immediate vicinity of the

old carbon tetrachloride plant WWT pond showed low levels (0.8 to 1.5 milligrams per liter, mg/l, which is equivalent to ppm) of CTC. All well samples analyzed for site-specific compounds showed non-detectable to very low levels, except for 6 mg/l thiocyanate in well 0-79, which is just downgradient of the Halby pond.

Conclusions

These ground-water results demonstrate conclusively that the mercury found in the swamp soil is insoluble and not leaching to the aquifer. Further, a review of well water results shown in Figures 5-6 and 5-7 shows clearly that the ground-water intercept system has been very effective in capturing CTC and CS₂.

The following major points can be made:

- Except for carbon tetrachloride found in source wells immediately downgradient of the old carbon tetrachloride plant WWT pond, essentially no priority pollutants were found in any ground-water samples.
- Although mercury was found in swamp soil samples, it is in an insoluble form as evidenced by its absence in ground water (see Appendix XXV).
- With two minor exceptions, all source wells sampled indicate thiocarbamates to be at very low levels (less than 0.06 milligrams per liter and most under 0.01 mg/l).
- All area wells south (immediately downgradient) of the property line contained less than 0.027 mg/l thiocarbamates, less than 0.046 mg/l CS₂, and less than 0.018 mg/l CTC. The one exception was NM-l, just downgradient of the LeMoyne landfill (one mile east of main facility), which contained 0.25 mg/l of CTC.

2. SITE FEATURES

2.1 Demography

As mentioned previously, the Cold Creek/LeMoyne site is situated just to the east of U.S. Highway 43 and approximately 20 miles north of Mobile, Alabama. Interstate Highway 65 is located 7 miles south of the site. The LeMoyne plant property, which is south of the Cold Creek plant, extends eastward to the Mobile River. The manufacturing facilities, however, are one and one-half miles from the river.

The Cold Creek/LeMoyne complex, which encompasses approximately 947 acres, is in an industrial area comprised mainly of other chemical production plants. Courtaulds North America, Inc. (CNA), which manufactures viscose rayon fiber, is located directly south and borders the LeMoyne plant. Shell Chemical Company (now duPont), an insecticide manufacturer, is south of CNA. Virginia Chemicals, Inc. is north of the Cold Creek plant, and M&T Chemicals is northwest. Alabama Power operates a coal-fired electrical generating station further north.

Very sparsely populated, rural communities are within a few miles of the site. The largest residential areas are Mt. Vernon, located approximately 8 miles to the north, and Creola, which is approximately 5 miles to the south (see Appendix VIII for map showing locations of nearby residences).

2.2 Land Use

The land in the immediate vicinity of the Cold Creek/LeMoyne complex is used almost entirely by industrial plants. In addition, some forest products are also harvested in the area. Because of the natural swamp, no farming is done in the area.

2.3 Natural Resources

The major natural resource in the site area is the Mobile River, which is located one and one-half miles to the east of the main plant facilities. At its closest point to the site, the river is approximately 500 feet wide and has a mean depth of 28 feet. Minimum flow, which is exceeded 99% of the time, is 4,800 cubic feet per second representing 3.1 billion gallons per day. Further information regarding the Mobile River can be found in the publication entitled "Water Resources of the Mobile Area, Alabama," which is included in Appendix XIV. The river, which flows south discharging into the Gulf of Mexico, is mainly used for barge transportation.

Some oil wells have been drilled 5 to 10 miles south of the site, but no oil has been discovered within the site area. Some cypress trees and pulpwood are harvested on the east side of the Mobile River.

2.4 Climatology

The climate in the site area is temperate, bordering on subtropical. The mean annual temperature is 67.6 degrees Farenheit. July is the hottest month, and January is the coldest month. Winters are usually short and mild with only a few days with temperatures below freezing. The summers are usually hot and humid. Average annual rainfall is 63.6 inches and is evenly distributed throughout the year. Average monthly temperature and rainfall data are presented in Table 2-1 (Hickman and Owens, 1978). Heavy rainfalls are common in the area, and a summary of data on rainfall intensity is given in Appendix IX (Stauffer interoffice correspondence, Bill Erdmann to Bill Cawthra dated 9/26/78).

TABLE 2-1

TEMPERATURE AND RAINFALL DATA

(recorded 1951-1975 at Mobile, Alabama,

	Average	Average	
Month	Temperature (OF)	Rainfall (in.)	
January	51.6	4.39	
February	54.0	5.13	
March	59.9	6.12	
April	67.9	5.18	
May	75.0	4.63	
June	80.4	5.34	
July	82.1	7.80	
August	81.7 -	6.89	
September	77.9	6.52	
October	. 68.9	2.51	
November	58.7	3.30	
December	53.4	5.77	
YEARLY	67.6	63.58	

1. INTRODUCTION

1.1 Objectives

Two adjacent sites in Mobile County, Alabama, formerly owned by Stauffer Chemical Company, were placed on the National Priorities List (NPL) by the Environmental Protection Agency (EPA) in 1982 under provisions of the Comprehensive Environmental Response, Compensation, and Liability Act of 1980 (CERCLA). As a result, Camp Dresser and McKee, Inc. (CDM), under contract to the EPA, conducted preliminary sampling at the site in May of 1985 and prepared a Work Plan, which is the basis for this Remedial Investigation (RI).

Based on the sampling results and previous investigations both onsite and offsite, CDM concluded that there was possible ground-water contamination, primarily mercury, carbon tetrachloride, carbon disulfide, and thiocarbamates, at the site. Furthermore, CDM suggested that some contaminants were moving offsite towards production wells at a facility south of the site. Possible major sources of contamination were identified. The purpose of this Remedial Investigation is to characterize the type and extent of contamination; to identify contamination sources, migration pathways and the potential for adverse environmental impacts; and to provide a basis for evaluation of the most cost-effective remedial action alternatives.

1.2 Site Background Information

1.2.1 Site Location and Description

Stauffer Chemical Company previously owned and operated two adjacent facilities in Mobile County, Alabama. The LeMoyne plant, purchased by Akzo Chemie America, Inc. in 1987, has been in operation since 1953 and manufactures multi-product inorganic chemicals, including carbon disulfide, carbon tetrachloride, sulfuric acid, chlorine and Crystex (a

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proprietary sulfur compound). The Cold Creek plant manufactures agricultural herbicides and pesticides, including several thiocarbamates. The Cold Creek plant has been in operation since 1966 and is currently owned by ICI Americas, Inc. (ICIA).

In the past, the Halby Chemical Company (HCC) leased a small parcel of land on the western portion of the site. Little is known of this operation, however, other than that waste products and effluents from the facility were held in a pond on the property. The pond has been closed and filled.

The Cold Creek/LeMoyne site ("the site") occupies 947 acres between the Mobile River and U.S. Highway 43, approximately 20 miles north of Mobile, Alabama (see Figure 1-1). The site is bounded by Virginia Chemicals, Inc. (VCI) to the north, Courtaulds North America (CNA), another chemical company, to the south, the Mobile River to the east, and Route 43 to the west. M&T Chemicals is located immediately to the west of Route 43. VCI produces sulfur dioxide, amines, sodium hydrosulfite, sodium bisulfite, and tetramethylfuran disulfide. CNA consists of a viscose rayon fiber processing plant and nylon spinning operation. M&T Chemicals is an organotin compound manufacturing plant.

The site is situated in a predominantly industrial area, but several sparsely populated rural communities are located within a few miles of the site. Maximum relief at the site is on the order of 30 feet. The Cold Creek Swamp lies between the plant facilities at the site and the Mobile River, and flows to the northeast and then east to discharge into the river. Surface—water drainage on site is generally toward the swamp or the river, and is governed by a drainage divide between the two. The Mobile River flows to the south and discharges into the Gulf of Mexico.

The majority of the chemical plants as well as the local communities in the area obtain water supplied from the water-table aquifer. As shown in Figure 1-2, the Cold Creek facility has one drinking-water well (CC-12) and one backup well

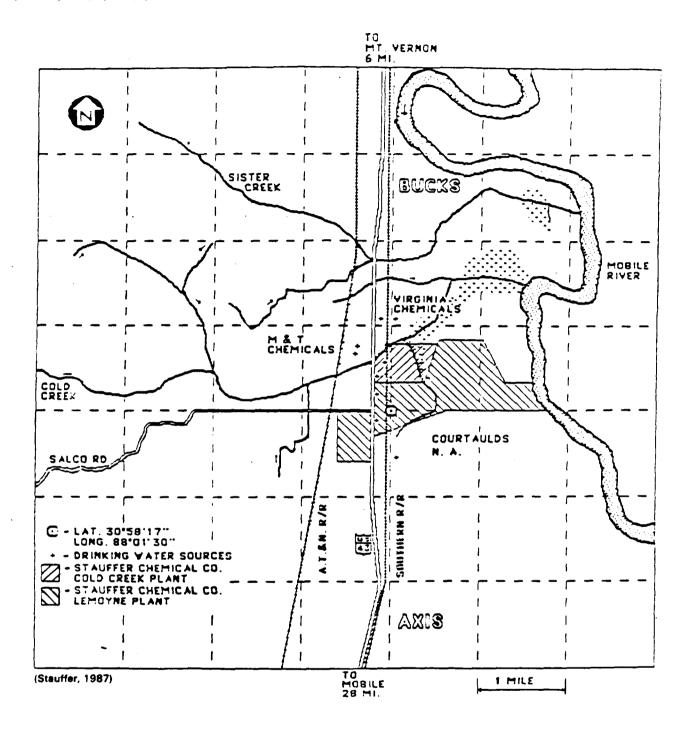


Figure 1-1 Approximate Location of the Cold Creek/LeMoyne Site Mobile County, Alabama

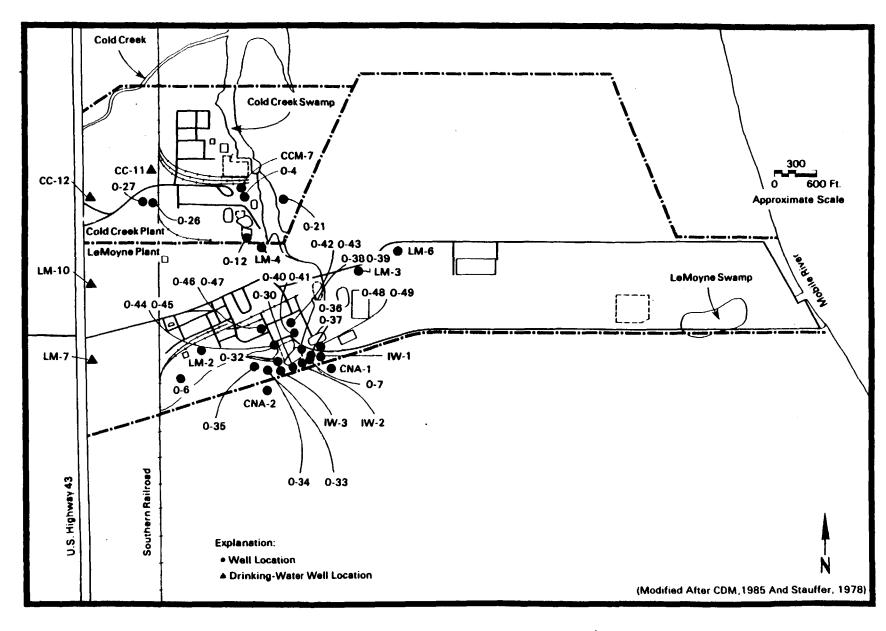


Figure 1-2 Site Area Well Locations

(CC-11), which serve 250 employees. The LeMoyne facility has two drinking-water wells (LM-7 and LM-10), which provide water for 230 employees. The CNA facility to the south has one drinking-water well (CNA-16) and a backup well (CNA-4), which provide drinking water to 750 employees. M&T Chemicals, located northwest of the site, also utilizes well water as a drinking-water source for its 200 employees. The total population served by industrial drinking-water supply wells within a two-mile radius of the site is 1,585 people.

As shown in Figure 1-3, a total of approximately 21 residential water wells are located within a two-mile radius of the site. Seventeen of these wells are located more than 1 1/2 miles west of the site. One well is located 2 miles due south of the site, and another is located 2 miles north of the site. The remaining 2 wells are located 1 mile southwest of the site. The total population served by residential drinking-water supply wells within a two-mile radius of the site, assuming 4 persons per home, is 84. The other residences located within two miles of the site are served by municipal water. The population within a two-mile radius of the site is not located downgradient to ground-water movement.

1.2.2 Site Use History

For approximately 20 years after initiation of operations at the Cold Creek/LeMoyne site, industrial waste products resulting from the Stauffer processes were disposed of in unlined surface disposal sites. From 1965 to 1974, solid waste from the LeMoyne plant was placed in an unlined landfill approximately one mile east of the plant. Between 11,000 and 12,000 tons of brine muds were placed in this landfill along with plant refuse, used samples, and some absorption oil (see Appendix XXIX for information contained in the Eckhardt Survey, which lists the substances stored in this landfill).

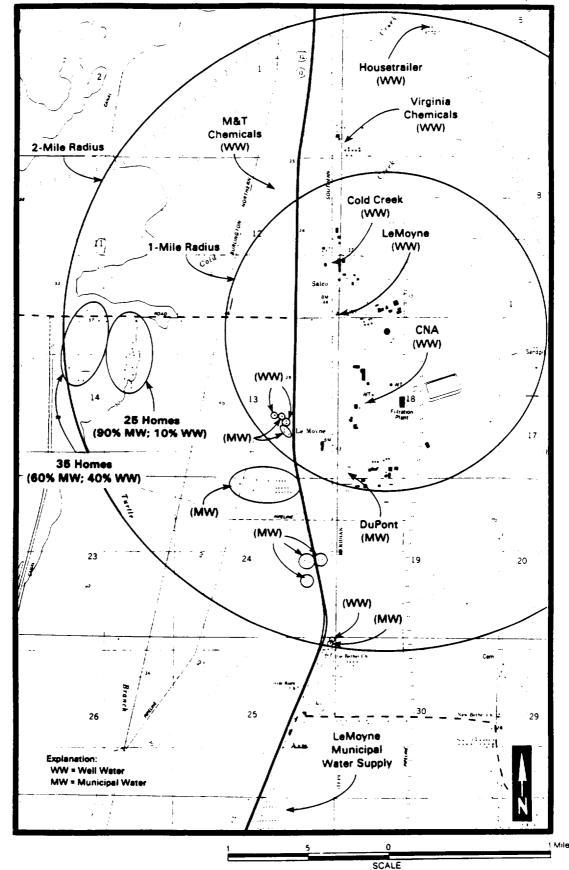


Figure 1-3 Location of Water Wells Within Two Miles of the Cold Creek/LeMoyne Site

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Two waste disposal sites, referred to as the north and south landfills, are located on the Cold Creek plant property. The exact quantity of material placed in the landfills is not known; however, sludges and solid wastes containing a variety of herbicides and pesticides are thought to have been buried in these two landfills. The LeMoyne and the two Cold Creek landfills were closed and capped with liners in 1974.

Wastewaters from the Stauffer processes were held in clay-lined lagoons and discharged to the Cold Creek Swamp. As shown in Figure ES-1, there are currently six closed or inactive wastewater ponds and seven active ponds. The seven active ponds, LeMoyne LeCreek, Cold Creek LeCreek, the new carbon tetrachloride plant wastewater treatment (WWT) pond, the ground-water treatment pond, the LeMoyne acid plant WWT (solids settling) pond, and the north and south chlorine plant wastewater check ponds, are all membrane-lined and monitored regularly. Of the six inactive wastewater treatment ponds, four (old carbon disulfide plant WWT pond, old chlorine plant WWT pond, Halby treatment pond, and Cold Creek old neutralization pond) are closed and covered. The old carbon tetrachloride plant WWT pond was lined and contains approximately 1900 cubic yards of sulfur sludge; it is inactive but not closed. The old brine mud pond is a lined pond used for storage of brine muds from the chlorine plant. It was originally a RCRA facility but has been delisted by the EPA; closure is awaiting State approval. Two ponds on the site are RCRA facilities (new brine mud pond and chlorine stormwater surge pond) and meet current RCRA standards. Three fire water storage and supply ponds are also located on the site.

Cold Creek Swamp received effluent from the LeMoyne and Cold Creek plants as well as from a previous tenant, the Halby Chemical Company (HCC). The effluent from the LeMoyne plant consisted of process waters from production units, containing 10 parts per million (ppm) of mercury. Neutralized waste brine from the Cold Creek plant was also discharged to the swamp during the late 1960's and early 1970's. The contribution from

HCC was assumed to have been thiocyanate- contaminated wastewater. The LeMoyne Swamp may have received surface-water runoff from the area of the LeMoyne landfill prior to its closure in 1974.

A small parcel of land on the western portion of the Cold Creek/LeMoyne site was leased from 1965 to 1979 to the Halby Chemical Company (HCC), as noted above. Witco, Inc. purchased the HCC facility in 1974, and continued to operate the plant until approximately 1979, when the buildings were razed. Although little is known of this operation, waste products and effluents were reported to have been discharged to the Cold Creek Swamp to the east and/or held in a pond on the property. The Halby pond has since been closed and filled.

1.2.3 Previous Investigative and Remedial Activities at the Site

During the late 1960's, Stauffer monitored their on-site water-supply wells and CNA monitored their production wells for chloride content. By 1971, ground-water monitoring results indicated that chloride concentrations in the wells were increasing, and the ground-water quality was deteriorating.

In October, 1972, an inventory shortage of approximately 10,000 gallons of carbon tetrachloride (CTC) was discovered at the LeMoyne facility storage area. All of the CTC is assumed to have seeped into the ground. Although the CTC storage tank was not diked, a short retainer wall separated it from a nearby fire pond, which, therefore, was unlikely to have been affected by the release (see Appendix XXII for maps of the LeMoyne plant which show the location of the CTC storage area and the spill area). None of the CTC could be recovered.

By early 1973, Stauffer determined that contaminants were entering the ground water beneath the plant sites, and an in-house task force was organized to evaluate the problem and develop recommendations for remedial action. Twenty-one ground-water monitoring wells were installed, and a

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swamp-sediment sampling program was initiated. The major contaminants found in the ground water were thiocarbamates, chlorides, and sulfur compounds. Mercury, most likely in the insoluble mercury sulfide form (See Appendix XXV), was found in the swamp sediment samples but not in the ground water. The major sources of these contaminants were determined to be Cold Creek Swamp, unlined wastewater holding and treatment ponds, unlined waste-disposal sites containing drummed and uncontained waste materials, process leaks and spills, and leachate from contaminated soils. The predominant areas of ground-water contamination were to the south and east of the site.

By late 1973, Stauffer had initiated contaminant source cleanup activities. Drummed waste debris and sludges in the disposal areas were consolidated and transported off site to permitted hazardous-waste landfills. Liquids were decanted and removed for treatment or off-site commercial disposal. The disposal areas were permanently closed and capped with an impermeable synthetic liner, clay, and soil. Ponds were dewatered, filled, and replaced with lined ponds. Pursuant to an NPDES permit, effluent flow from both facilities was treated and discharged to the Mobile River instead of the swamp. Lowlying areas were regraded with clean fill from construction activities in order to control local flooding, and a stormwater drainage control system was designed and constructed. These source control and cleanup activities were completed in 1976.

In addition to the remedial activities described above, an extensive ground-water monitoring program was initiated in December of 1973. Ground-water monitoring continued, and by late 1977, Stauffer concluded that ground-water quality along the southern property boundary of the LeMoyne facility was improving but at an unacceptable rate (see Appendix XX). In addition, CNA reported continued contamination of two of their wells, chloride and iron being the primary contaminants. In response to continuing ground-water quality deterioration, Stauffer installed seven new observation wells along the southern property line of the LeMoyne facility.

In December, 1977, leak detection systems for two treatment and storage ponds on site indicated leakage through pond liners. These two treatment and storage ponds were subsequently repaired. In 1978, Stauffer dewatered and capped an old mercury sulfide WWT pond and a WWT pond in the carbon disulfide plant area.

In 1978, Stauffer contracted with Ground Water Associates, Inc. (GWA) to perform a hydrogeological investigation of ground-water contamination at the LeMoyne facility. During September of that year, GWA collected water samples from 32 wells and analyzed the samples for carbon tetrachloride and carbon disulfide contamination. The results are shown in Figures 1-4 through 1-10, respectively. Based on the analytical results and the results of pump tests, it was concluded that three intercept wells, each pumping a maximum of 500 gallons per minute (gpm), would be required to intercept the bulk of the contaminated ground water. GWA used a numerical model to represent the aquifer system, including the proposed intercept wells. The resultant contour map of the regional piezometric surface is shown in Figure 1-11. concluded that "this map indicates that it is not likely that any contamination from the Stauffer property will migrate further than CNA's wells Nos. 5, 6 and 10. Some contamination [present prior to installation of the intercept wells] will probably reach these wells, however, and samples of their water should be periodically analyzed for traces of carbon tetrachloride and carbon disulfide".

The three interceptor wells were installed in late 1980 along with an air stripping treatment system approved by the Alabama Water Improvement Commission (AWIC), which is now the Alabama Department of Environmental Management (ADEM). Since that time, the levels of ground-water contamination have been substantially reduced, as shown in Figure 1-12, and treated effluent concentrations have continuously met discharge permit limits established by the ADEM (see Appendices XXIII and XXIV).

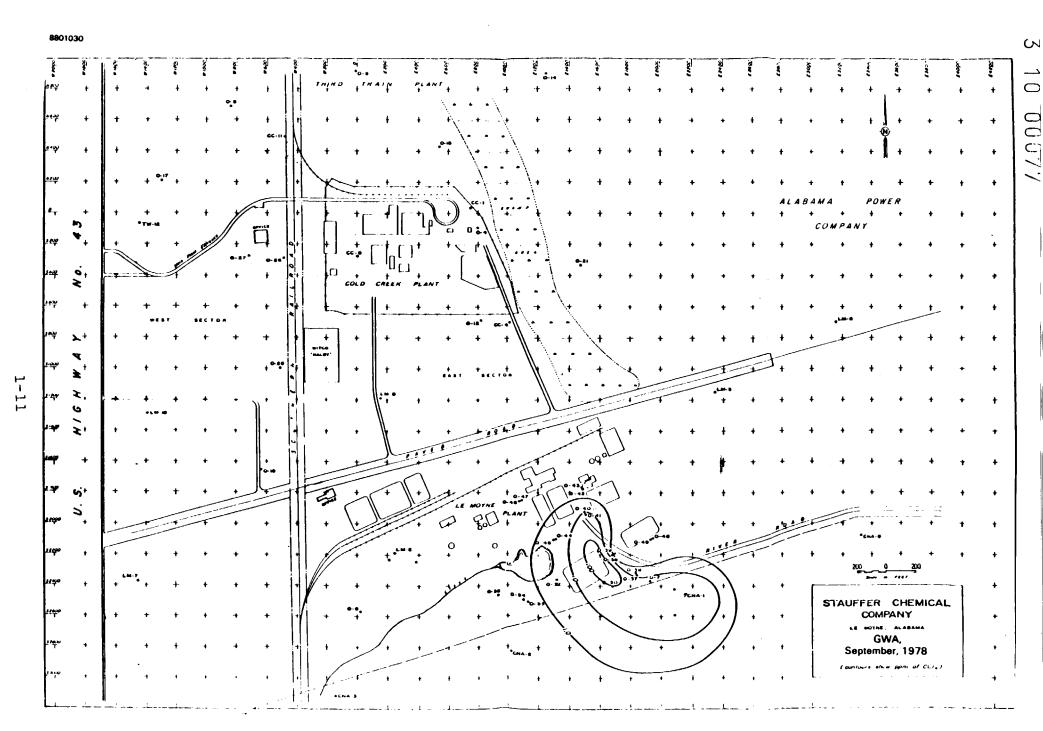


Figure 1-4 Concentration of Carbon Tetrachloride in Shallow Wells

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1-12

Figure 1-5 Concentration of Carbon Disulfide in Shallow Wells

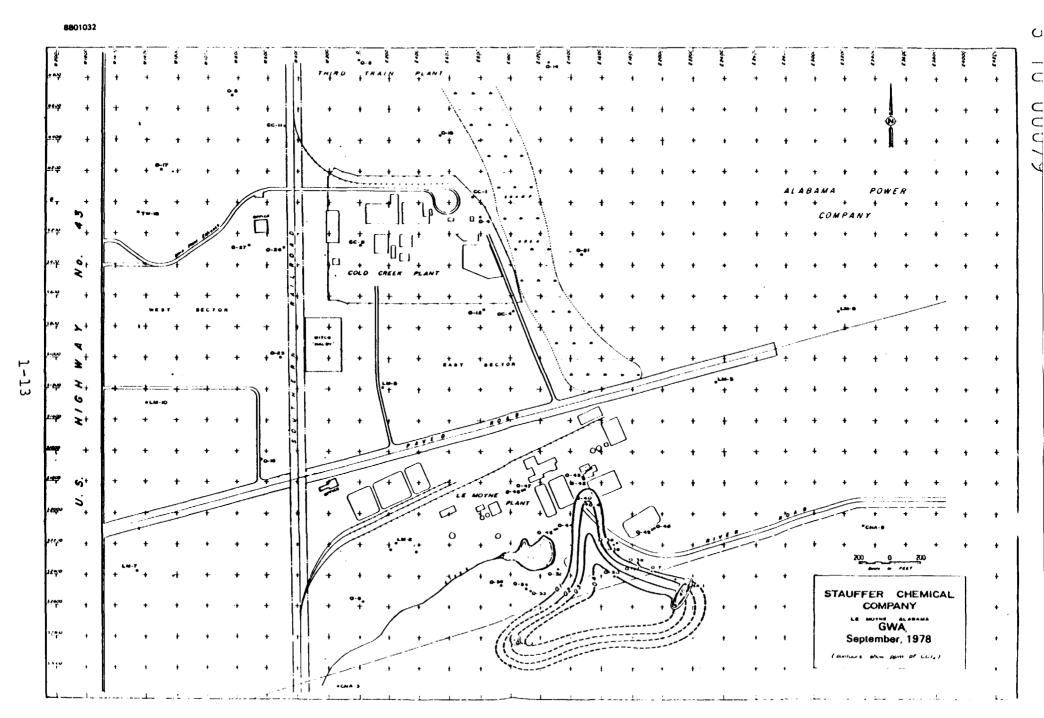


Figure 1-6 Concentration of Carbon Tetrachloride in Deep Wells

Figure 1-7 Concentration of Carbon Disulfide in Deep Wells

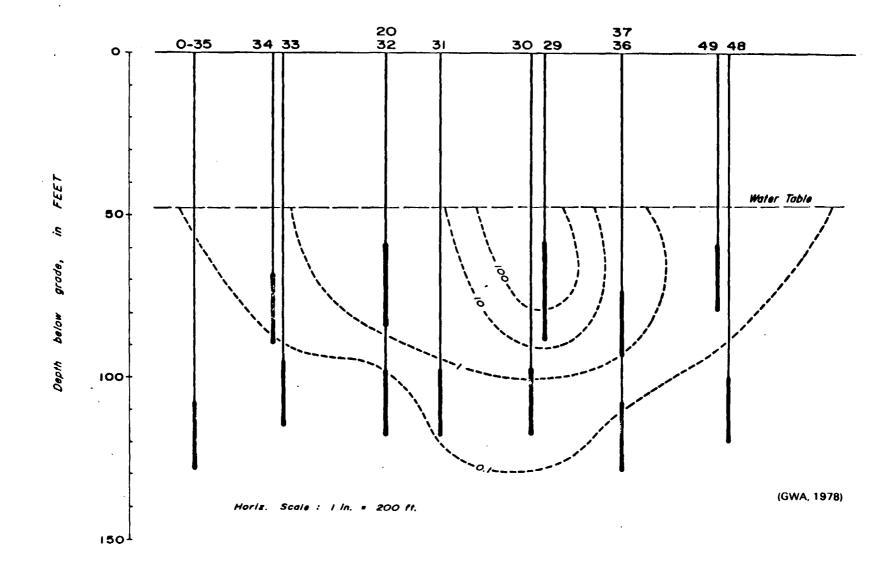


Figure 1-8 Vertical Cross Section of Carbon Tetrachloride Concentrations in East-West Plane

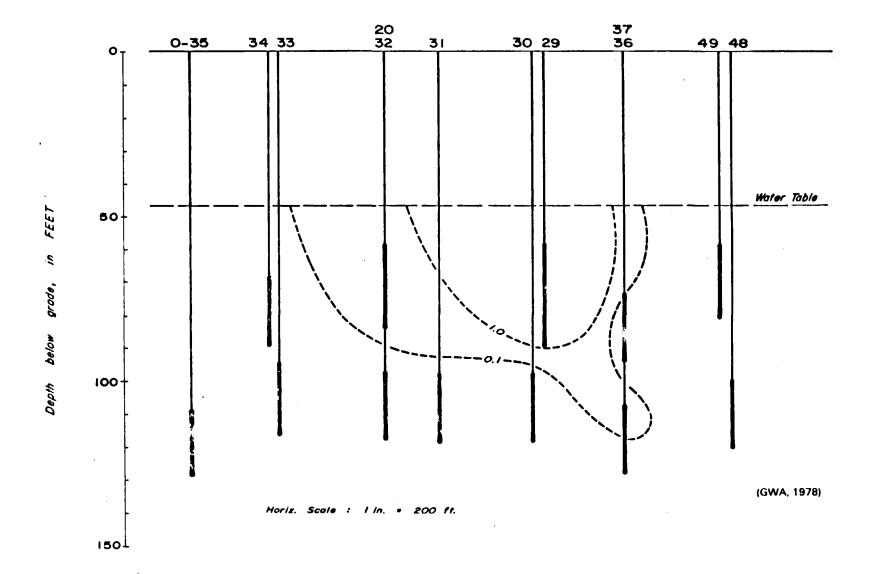


Figure 1-9 Vertical Cross Section of Carbon Disulfide Concentrations in East-West Plane

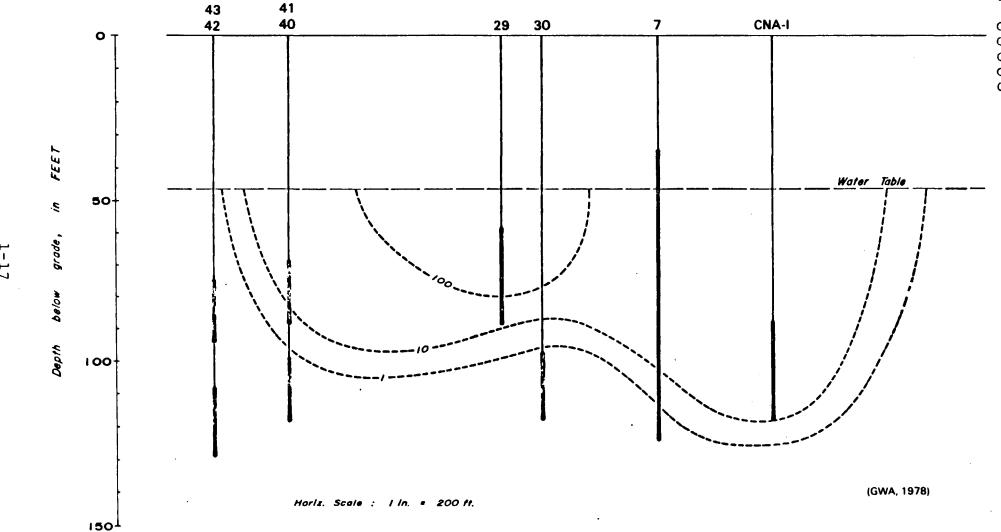


Figure 1 10 Vertical Cross Section of Carbon Tetrachloride Concentrations in North to South-Southeast Plane

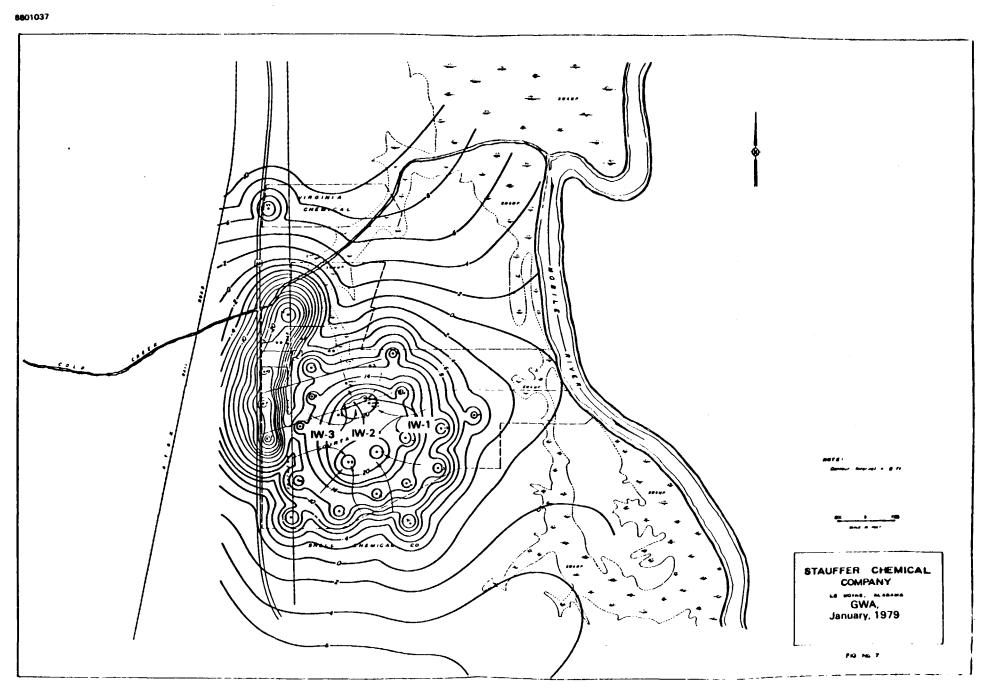


Figure 1-11 Contour Map of Piezometric Surface

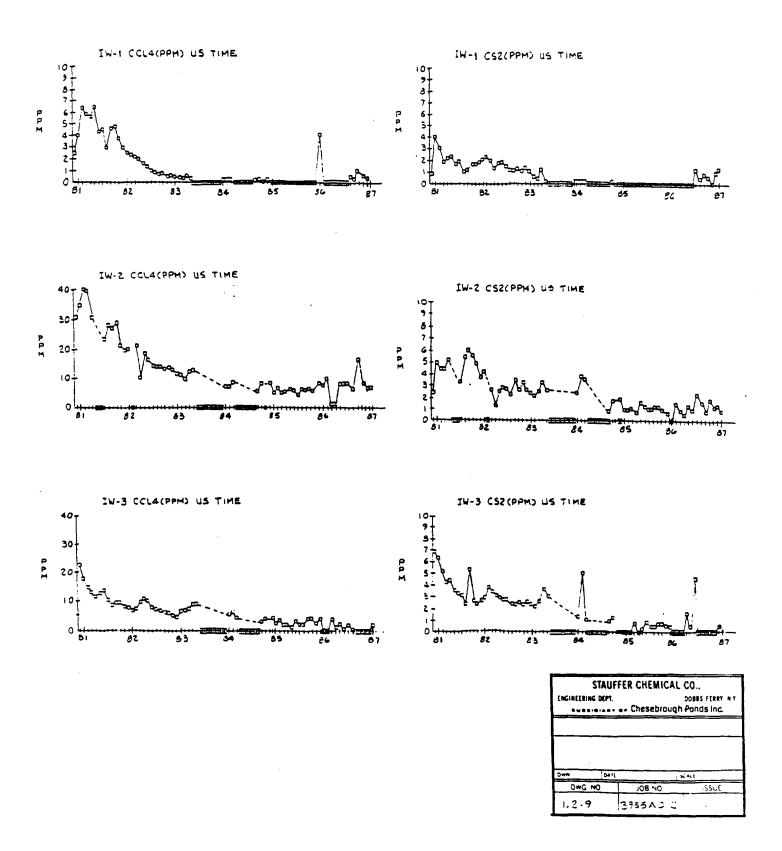


Figure 1-12 Carbon Tetrachloride and Carbon Disulfide Concentrations Versus Time at Ground-Water Intercept Wells

At the request of the Alabama Department of Public Health (ADPH), Stauffer installed seven new monitoring wells around the two closed Cold Creek landfills in 1980. Early in 1981, the leaking treatment and storage ponds at the LeMoyne facility were repaired.

In 1982, the ADPH, in conjunction with the U.S.
Environmental Protection Agency (EPA), conducted an on-site
assessment of the Cold Creek and LeMoyne facilities. Analyses
of ground-water samples collected during the assessment
identified ground-water contamination on the plant sites and at
the CNA well that was sampled. The primary contaminants
detected were metals, chlorides, carbon tetrachloride and
miscellaneous organic compounds. As a result, the Stauffer
sites were added to the National Priorities List (NPL) by the
EPA under the mandate of the Comprehensive Environmental
Response, Compensation, and Liability Act of 1980 (CERCLA),
also known as Superfund. The Cold Creek site was ranked number
204 and the LeMoyne site number 467 on the NPL. To date, there
has been no extraordinary public interest in the sites.

Camp Dresser and McKee, Inc. (CDM), under contract to the EPA, conducted preliminary sampling at the site in May of 1985 and prepared a Work Plan for performance of a Remedial Investigation/Feasibility Study (RI/FS) at the Cold Creek/LeMoyne site. Stauffer performed the RI in 1986 and submitted a draft RI Report in February of 1987. In response to review and comments by the EPA, Stauffer commenced revision of the report and submitted data dated July 1, 1987. ERT, A Resource Engineering Company (ERT), was retained by Stauffer to revise the RI Report in response to additional comments from the EPA. Report revisions and responses to EPA comments have been incorporated into this document.

1.3 Remedial Investigation Summary

1.3.1 RI Objectives

The overall objectives of this Remedial Investigation were to define all potential contamination sources and to identify the extent of any contamination migration and potential transport pathways. Specific areas to be investigated included the north and south Cold Creek landfills (see Figure 5-1), the LeMoyne landfill (see Figure 5-1), nine ponds and lagoons (see Figure 5-2), and the Cold Creek and LeMoyne Swamps (see Figures 5-3 and 5-4 and Drawing 1.3 in Appendix XVII). The extent of ground-water contamination was also determined. The primary objective was to obtain enough additional data to evaluate the success of past and current remedial actions and to determine the need for further measures.

1.3.2 Source and Area Characterization

In order to accomplish the objectives outlined above, the RI was divided into two major subtasks, source characterization and area characterization. Source characterization included the installation of 3-foot-deep soil borings at 34 locations in the Cold Creek Swamp and at four locations in the LeMoyne Swamp. Of these 38 composite samples, 7 were analyzed for thiocarbamates, chlorides, and priority pollutants, and 31 were analyzed for mercury only. A second, deeper sampling program for mercury in the swamp was considered if interpretation of the initial results indicated that it was necessary. Following receipt of Stauffer's report of November 12, 1986, which included supporting findings that the swamp recharges the ground water (see Appendix XX), EPA agreed that the Phase II sampling was not necessary.

Soil sampling was also performed under each of the nine ponds and three landfills on site. Soil borings were installed at 45-degree angles beneath the ponds and landfills and

advanced to 50 feet or until ground water was encountered. Composite samples were collected at 10-foot depth increments and analyzed for location-specific compounds and priority pollutants. A total of twelve soil borings were installed under the three landfills, and a total of eighteen soil borings were installed under the nine ponds. Water samples were collected from three active ponds and analyzed for priority pollutants. Samples of the liner membrane material covering the landfills were taken and tested for integrity. In addition, fish samples from five locations in the Cold Creek Swamp were collected and analyzed for mercury. Source characterization was completed with the installation and sampling of two new ground-water monitoring wells and sampling of thirteen existing wells. All fifteen well samples were analyzed for priority pollutants; three of the fifteen samples were also analyzed for site-specific compounds.

Area characterization involved sampling 36 site-area wells for location-specific compounds. Seven of the 36 samples were analyzed for priority pollutants. In addition, two surface-water samples and two soil samples were collected offsite to determine background concentrations of the compounds of concern.

All the samples were collected between May and August, 1986 by Stauffer's Research personnel working with a Stauffer geologist and representatives of the EPA and EPA's contractor. Sampling, field quality assurance, chain of custody procedures, and all analytical methodology were submitted to the EPA on August 9, 1985 in Stauffer's Sampling and Analysis Manual, which was approved for use by the EPA.

All laboratory analyses were performed by Environmental Testing and Certification, Inc. (ETC). The data were reviewed and compiled by Stauffer Research. The complete laboratory data packages are included in Appendices I through V. A discussion of sampling, analysis, quality assurance and quality control is presented in Appendix VI.

In the discussion of quality assurance in Appendix VI, it is concluded that four common laboratory contaminants, methylene chloride, toluene, bis (2-ethylhexyl) phthalate and di-n-butylphthalate, which were reported in many samples near or below their detection limit, should be considered invalid because they were found in one or more blanks at concentrations near or below the method detection limits. Also considered suspect are samples reportedly containing levels near the detection limits of chloroform, trichloroethylene, cyanide, copper and zinc because they were also found at similar levels in field blank samples. For these reasons, all analytical results considered invalid or suspect are reported as non detectable throughout this report. As previously mentioned, nevertheless, all of the analytical results are included in Appendices I through V.

Results of the soil, biota, surface-water, and ground-water sampling and analysis conducted during the RI are summarized in Tables 1-1 through 1-3. Only those compounds detected are included in these tables.

The following is a list of tests and/or samples either not taken or relocated, and the reason for the deletion or move (see Figures 5-1 and 5-2 for location).

- Phase II swamp sampling was not conducted because of decision by EPA to omit as unnecessary.
- Relocated soil boring MTP 2-S to southeast corner of covered pond because of proximity to high voltage power lines.
- 3. OCTC-1S moved to west end of WWT pond to minimize damage to dike.
- 4. Reversed locations of OCTC-2S and OCS-1S to avoid potential damage to pond banks and covers.

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CCS-2-1E CCS-2-1M CCS-3-1E CCS-4-2E CCS-4-2E CCS-4-3M CCS-4-3M CCS-4-3M CCS-4-3M CCS-5-3M CCS-5-3M CCS-5-3M CCS-5-3E CCS-6-1E CCS-6-1E CCS-6-1M	1. 8000 7. 3000 530. 0000. 29. 0000 58. 0000 1. 2000 0. 1400 15. 0000 12. 7000 9. 3000 5. 3000 1. 8000 4. 9000 6. 4000		:														•
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Fish Sample: 80-1 80-2 80-3 80-4 80-5	0.6300 3.1000 0.9100 1.1000 0.4200																
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NOTES:

All Concentrations are in ppm Average = Ci+...C2+../total no. of samples Frequency = Number of samples where compound was detected/Total number of samples analyzed. The average of the two samples were taken for duplicate samples

- Relocated NLF-1S to west side of north landfill to avoid high voltage power lines.
- 6. Relocated SLF-1S to east side of south landfill because pipe racks prevented access from south.
- Relocated SLF-2S to west side of south landfill because fire pump house blocked access from north.
- 8. CCAP-1S placed at southwest corner and oriented to northeast because of equipment access problem.
- 9. CCAP-2S shifted to east to cover other end of pond.
- 10. CCLP-2S moved to east end of north side oriented to south because of equipment access problems.
- 11. CCLP-1S shifted to south end of west side oriented to the east to cover other end of pond.
- 12. HTP-1S was stopped at 16.5 feet because recognizable waste material was encountered.
- 13. Additional boring, HTP-3S, was drilled approximately 20 feet east of northwest corner of former pond area to 45° angle oriented to south in order to obtain samples beneath pond.
- 14. Dye test was not necessary because the shallow wells, when installed, were dry.
- 15. Well 0-41, which is located approximately 10 feet south of well 0-40, was sampled instead of 0-40 because a crimp in the casing of 0-40 prevented this well from being sampled.

1.4 Report Overview

The balance of this Remedial Investigation Report generally follows the format as suggested by EPA in their guidance document dated June, 1985.

Chapter 1 provides information regarding site background, the nature and extent of contamination, previous investigative and remedial activities, and a summary of the RI site investigation program.

Chapter 2 provides information on population density and distribution, land use, natural resources, and climate.

Chapter 3 provides information on the types of wastes disposed of in the landfills, ponds and lagoons, and swamps on site, as well as the toxicity and environmental behavior of the waste components.

Chapter 4 describes the environmental setting of the site and its surroundings. The surface characteristics, including physiography, topography, and surface-water drainage, are described as are the subsurface features such as the soils, geology, and ground-water flow system.

Chapter 5 describes and presents summarized results of the RI sampling and analysis program, which involved source and area characterization. Soil, surface-water, ground-water, and biota results are presented. Complete laboratory analysis results can be found in Appendices I through V.

Chapter 6 deals with the public health and environmental concerns associated with the site. Contaminant migration pathways, potential receptors, and potential impacts on public health and the environment are discussed.

Chapter 7 describes the remedial action alternatives selected for preliminary assessment in the Feasibility Study and the remedial alternative screening process.

Finally, Chapter 8 provides a summary of the RI findings and conclusions.

As noted in the Draft RI Report, the Air Investigation (Chapter 6 of the Draft RI Report), was essentially eliminated because air testing was not as relevant and was not included in the EPA contracted Work Plan. For this reason, Chapter 6 of the Draft RI Report is not included in this document. In addition, Chapter 8 of the Draft Report, entitled Bench and Pilot Tests, has been included as an appendix to this report (Appendix XXIII). This appendix contains a discussion of the current ground-water intercept system, including operation of the interceptor wells and treatment ponds.

3.1 Waste Types

The site contains three known closed landfills, six closed or inactive wastewater ponds (five of which were investigated during the RI), seven active ponds (four of which were investigated during the RI), and two swamps. All of these areas are identified on the site map, Figure ES-1. Detailed graphic depictions of many of these areas are included in Appendix XXVI.

3.1.1 Landfills

The LeMoyne landfill contains approximately 11,000 to 12,000 tons of brine mud from chlor-alkali production along with absorption oil and plant refuse (see Appendix XXIX). The landfill was never used for burning materials, although trash fires occasionally broke out and spread beyond the landfill. The area adjacent to and east of the landfill was used to store acid brick and fire brick, blasting sand, and raw sulfur recovered from leaks which occurred along the sulfur transport line from the Mobile River to the plant. The sparseness of vegetation in the area east of the LeMoyne landfill is related to the presence of sandy soil in this area and, to a lesser extent, to the former storage of sulfur. It is unlikely that retort tower refuse is related to the lack of vegetation in this area, because retort tower refuse was stored within the boundaries of the landfill proper.

The Cold Creek plant landfills, designated north and south landfills, contain the following solid waste:

- Water treatment plant sludge,
- Used sandblast grit,
- Generator coke,
- Incinerator ash, and
- Filter aid waste.

The filter aid waste is the only material from the above list that would contain pesticides. Analytical data have shown that the total pesticide content typically found in this material is approximately 10 ppm. There could also be some solvents and other organic compounds in the filter aid waste.

Although drummed liquids were temporarily placed in the Cold Creek landfills, they were removed prior to closure. It is possible, however, that leakage or spillage of pesticides may have occurred during the period the drums were in the area.

These three landfills were closed in 1974. A 20 mil plastic membrane was placed on top and along the sides of the landfills to a depth just below the bottom of the buried waste. A 12-inch layer of compacted clay topped by clean soil was added (see Appendix XXVI for graphic depictions of the landfills). Section 5.1.1 describes the soil sampling around the landfills, and subsequent sections describe the ground-water sampling results.

3.1.2 Ponds and Lagoons

Three of the active ponds, LeMoyne LeCreek, Cold Creek LeCreek, and the new carbon tetrachloride plant WWT pond, are membrane lined and used for wastewater treatment under NPDES permits. The fourth active pond, the acid plant WWT pond, is also membrane lined. All four active ponds are monitored regularly. Of the six inactive wastewater treatment ponds, four (old carbon disulfide plant WWT pond, old chlorine plant WWT pond, Halby Pond, and Cold Creek's old neutralization pond) are closed and covered (see Appendix XXVI). The old chlorine plant WWT pond was clay lined and is now filled with excavation material from the plant and fill from the plant borrow pit. The old carbon tetrachloride plant WWT pond was lined and contains approximately 1900 cubic yards of sulfur sludge. is inactive but not closed. Section 5.1.2 describes the soil sampling around these ponds, and subsequent sections describe the surface-water and ground-water sampling program results.

3.1.3 Cold Creek and LeMoyne Swamps

The Cold Creek Swamp had received effluent from the LeMoyne and Cold Creek plants as well as from a plant owned by a previous tenant, the Halby Chemical Company (HCC). The effluent from the LeMoyne plant consisted of process waters from production units, containing 10 parts per million of mercury. Strong evidence indicates that the mercury present in the swamp is insoluble and immobile, and is therefore present as a sulfide. (See Appendix XXV for a detailed explanation of why the mercury in the swamp is in the form of mercury sulfide.) Neutralized waste brine from the Cold Creek plant was also discharged to the swamp during the late 1960's and early 1970's. The contribution from the HCC was presumably thiocyanate-contaminated wastewater.

The LeMoyne Swamp may have received surface-water runoff from the area of the LeMoyne landfill prior to its closure. Section 5.1.3 describes the soil sampling around these swamps, and subsequent sections describe the ground-water sampling results.

3.2 Waste Component Characteristics and Behavior

The major waste components found during the RI site investigation were mercury, carbon tetrachloride, carbon disulfide, cyanide (one isolated sample), thiocyanate, and six thiocarbamates (EPTC, butylate, vernolate, pebulate, molinate and cycloate). A few other heavy metals were found in soil samples, but their concentrations were in the range commonly found in natural soil or slightly higher. These data, as well as other laboratory analyses of environmental media, are contained in Tables 1-1, 1-2, and 1-3, which are included in Section 1.3.

The environmental behavior and toxicity assessment of each of the major waste components are described in the following sections. Appendix XIX provides some empirical toxicity

information for each major waste component; in addition the Endangerment Assessment (separate report) contains detailed toxicological profiles.

3.2.1 Mercury

3.2.1.1 Environmental Behavior

The major removal mechanism of mercury from natural water systems is adsorption onto the surfaces of particulate phases and subsequent settling to the bed sediment. According to the EPA (1979), the overwhelming majority of any dissolved mercury is removed in this manner within a relatively short time, usually in the immediate vicinity of the source.

Photolysis seems to be significant in the chemical speciation of mercury in the atmosphere and perhaps in the aquatic environment. Because of the limited amount of information available regarding this process, however, it is not clear what impact it may have on the overall fate of mercury in the aquatic environment. Because of its uniquely high vapor pressure relative to other metals, metallic mercury can enter the atmosphere from the aquatic environment in the form of several different gaseous compounds. This factor makes volatilization important for the aquatic fate of mercury. It is not clear what impact volatilization will have on the overall fate of mercury in the aquatic environment, however, because of the limited quantitative data available regarding the volatilization of mercury compounds from natural waters.

EPA's (1979) review of environmental studies and theoretical considerations indicates that mercury adsorption onto sediments is probably the most important process in determining the fate of mercury in the aquatic environment. Carr and Wilkness (1972) found that radioactive mercury, when added to natural samples, was rapidly apportioned out of the particulate phases with half-lives for adsorption ranging from less than one to fifty hours. Ramamoorthy and Rust (1976)

conducted laboratory studies on mercury sorption and found that sorption rates were highest in organic-rich sands, that sediment binding capacity was most closely related to organic content, and that mercury sorption was little affected by pH. In their study of mercury adsorption on and desorption from sediments, Reimers and Krenkel (1974) found that, at a constant pH, the adsorption of inorganic mercury is affected by aquatic chloride concentration, with the percent loss in capacity depending upon the constituents of the sediment. They also found that inorganic mercury is bound strongly enough by sediments to be transported by sedimentary mobilization.

Because of the concern regarding the danger to human health from eating mercury-contaminated fish, the bioaccumulation of mercury in the aquatic environment has been well studied. Methyl mercury is the form of mercury present in most fish tissue, and it is the most readily accumulated and retained form of mercury in aquatic biota. Methyl mercury is readily accumulated by fish both from their food and through water, and is very difficult to eliminate after entering the biological system. Most studies report that the half-life of methyl mercury in aquatic organisms is between one and three years.

As an element, mercury is not intrinsically altered by chemical reaction, but it does take part in biologically-mediated reactions which drastically alter its mobility and toxicity. According to the EPA (1979), virtually any mercurial compound can be microbially converted to methyl mercury upon entering an aqueous system. Conditions reported to enhance the methylation process include large amounts of available mercury, large numbers of bacteria, absence of strong complexing agents such as sulfide, circumneutral pH, high temperature, and a moderately aerobic environment. (As discussed in greater detail in Appendix XXV, conditions at the Cold Creek/LeMoyne site did not favor the formation of methyl mercury. Rather, conditions favored the precipitation of insoluble mercury sulfide because of the presence of the strong

complexing agent, sulfide.) Bacteria can act not only as mediators of methylation, but can also preferentially accumulate large amounts of mercury. Although sediment is probably the most important sink for mercury, methylation by bacteria could reduce the mercury content of overlying waters, resulting in the mobilization of inorganic mercury from the sediments. In summary, mercury can be metabolized by bacteria to methyl and dimethyl forms which are quite mobile in the environment.

3.2.1.2 Toxicity Assessment

Exposure to mercury in most forms is associated with a high degree of toxicity. Chronic exposure to elemental (metallic) mercury causes behavioral effects and other nervous system damage. Inorganic mercury salts do not generally reach the brain, but will produce kidney damage. Organic mercury compounds are toxic because they can cross the blood-brain barrier and produce nervous system disorders.

The metabolism, distribution, and excretion of mercury depends largely on its chemical form. Preferential deposition of elemental mercury vapor occurs in the lungs. Organic mercury is efficiently absorbed by the gastrointestinal tract based on its ability to traverse biological membranes. Both elemental and organic mercury break down in the body to the more toxic form, divalent mercury.

In a review of carcinogenic data for either inorganic mercury or methyl mercury, the EPA (1984) noted that none of the available data indicated "carcinogenic potential". The federal maximum contaminant level (MCL) is 0.002 mg/l.

3.2.2 Carbon Tetrachloride

3.2.2.1 Environmental Behavior

Volatilization is the major transport process for carbon tetrachloride in aquatic systems because of its high vapor pressure. There is little information specifically pertaining to adsorption of carbon tetrachloride onto sediments.

McConnell et al. (1975) found that coarse gravel had little adsorptive capacity for carbon tetrachloride, whereas sediments rich in organic detritus had a much higher adsorptive capacity. In general, however, there is no clear evidence of selective concentration of carbon tetrachloride in sediments.

Neely et al. (1974) have shown that bioaccumulation is directly related to the logarithm of the octanol/water partition coefficient (log K_{OW}) of the compound. The log K_{OW} of carbon tetrachloride is 2.64, indicating a tendency for this compound to bioaccumulate under conditions of constant exposure. Pearson and McConnell (1975), found that although carbon tetrachloride and other organochlorines examined are somewhat lipophilic and tend to be found at higher concentrations in fatty tissues, there is no evidence for the biomagnification of carbon tetrachloride or other short-chain aliphatics in the food chain. According to the EPA (1979), however, the difficulties associated with the analytical methods in the Pearson and McConnell (1975) study make estimates of bioaccumulation based on their experimental results somewhat unreliable.

There is little information specifically pertaining to biodegradation of carbon tetrachloride. Thom and Agg (1975) have included carbon tetrachloride on a list of synthetic organic chemicals which should be degradable by biological sewage treatment methods provided suitable acclimatization can be achieved. They note, however, that not many compounds on the list occur free in nature, and, therefore, it is unlikely that microorganisms already possess the ability to destroy them.

3.2.2.2 Toxicity Assessment

The toxicity of carbon tetrachloride can be accounted for because it is a fat-soluble, halogenated hydrocarbon.

Absorption of carbon tetrachloride through dermal and inhalation routes is greater than through ingestion routes.

Again, because of carbon tetrachloride's fat-soluble nature, it is readily absorbed through the skin. Distribution of carbon tetrachloride is body-wide with the highest concentrations found in fat tissues, liver, bone marrow, and blood (EPA, 1984a). Carbon tetrachloride is metabolized into chloroform and a trichloromethyl radical in the liver.

The EPA (1984b) noted that, although the majority of several mutaginecity assays were negative, there was evidence supporting the classification of carbon tetrachloride as a weak mutagen.

Carbon tetrachloride is classified as a group "B2" carcinogen in the EPA's weight-of-evidence ranking scheme (SPHEM, 1986). This category indicates that there is sufficient evidence for causing liver cancer in animals but that the corresponding human data is inadequate. The federal MCL, for an incremental increased lifetime cancer risk of 1 x 10⁻⁶, is 0.005 mg/l.

3.2.3 Carbon Disulfide

3.2.3.1 Environmental Behavior

Very little information is available regarding the transport and fate of carbon disulfide. Nevertheless, data regarding the physiochemical properties of carbon disulfide (Verschueren, 1983) can be used to qualitatively assess the processes that may be important in determining its fate. Carbon disulfide has a relatively low boiling point and high vapor pressure, indicating that it is very volatile. It has a specific gravity of 1.263 and is therefore more dense than

water. With a solubility of 2300 mg/l, it is very soluble in water from an environmental perspective. The octanol-water partition coefficient of carbon disulfide is in the range of from 69 to 145, indicating that it is not readily adsorbed to soils. Given this information, it is likely that volatilization and ground-water mobility are significant whereas adsorption is of lesser importance in terms of the transport of carbon disulfide. The significance of photolysis, hydrolysis, bioaccumulation, and biotransformation cannot be assessed with the information available from Verschueren (1983) regarding the chemical and physical properties of carbon disulfide.

3.2.3.2 Toxicity Assessment

Carbon disulfide is hazardous as a liquid and a vapor. Although inhalation is the major route of toxic exposure, exposure may be compounded by dermal absorption and ingestion. Because of its fat solubility, carbon disulfide is reported to be one of the strongest skin irritants known, causing third degree burns within minutes (Spyker et al., 1982). Skin absorption may cause localized degeneration of peripheral nerves, usually in the hands (Sittig, 1985). Inhalation of carbon disulfide vapor may also cause respiratory irritation and progress to bronchitis and emphysema. The primary manifestations of acute and chronic carbon disulfide exposure include pyscological, neurological, and cardiovascular disorders.

Carbon disulfide has been demonstrated to produce reproductive and teratogenic effects in animals when inhaled (Bariliak et al., 1975). Data on mutagenicity and carcinogenicity were not available.

3.2.4 Cyanides

3.2.4.1 Environmental Behavior

Cyanides are a diverse group of compounds whose fate in the aquatic environment varies widely. The cyanide ion (Cn) can react with a variety of metals to form insoluble metal cyanides. If the cyanide ion is present in excess, complex metallocyanides may be formed. These compounds are quite soluble and can be transported in solution. The fate of low molecular weight organic cyanides (nitriles) is expected to parallel the fate of hydrogen cyanide, a gas which may be destroyed by biodegradation or removed from solution by volatilization or adsorption.

The significance of photolysis on the aquatic fate of the cyanides has not been investigated fully, although it is possible that the photolysis of the metallocyanides could result in the release of cyanide ion (Broderius, 1977). EPA (1979) notes that this process could be important in aquatic environments downstream from metallocyanide discharges.

According to the EPA (1979), the hydrolysis of nitriles in the aquatic environment is slow in most cases, and is probably not competitive with other processes.

Cyanides are fairly mobile in the soil environment (Alesii and Fuller, 1976), indicating that adsorption is probably not a significant control on mobility in most aquatic environments where sorbents (e.g., clays, biological solids, sediments) are much less concentrated. Alesii and Fuller (1976) reported that cyanide mobility is least where soils have a low pH, high concentrations of free iron oxides and positively-charged particles such as the clay minerals kaolin, chlorite, and gibbsite. Cyanide mobility is greatest at high pH, high concentrations of free calcium carbonate (high negative charge), and low clay content. According to the EPA, biological solids sorb cyanides, but, as with the other sorbents, the amount bound is probably insignificant in comparison to the amounts volatilized or biodegraded.

Broderius (1973) reported the bioaccumulation of metal (copper and silver) cyanide complexes in fish. According to the EPA (1979), it is difficult to assess the environmental importance of metal cyanide bioaccumulation other than to note that the metal cyanides are generally less toxic than hydrogen cyanide. Nevertheless, bioaccumulation undoubtedly enhances the chronic toxic effects of the metal cyanides.

Although biodegradation of cyanides is known to occur in natural waters, the rates of cyanide biodegradation have not yet been ascertained. According to the EPA (1979), the importance of this process varies according to such factors as cyanide concentrations, pH, temperature, microbe concentration and acclimatization to cyanide, and the availability of nutrients.

3.2.4.2 Toxicity Assessment

The term "cyanides" encompasses those inorganic or organic compounds which contain the CN group. Examples include cyanide ions that form complexes with metals, cyanates that contain the OCN radical, alkyl cyanates that trimerize to cyanurates, nitriles, and cyanohydrins. The toxicity of many of these substances is related to subsequent release of hydrogen cyanide (HCN) or the CN radical. These components can be released as a result of photodecomposition, ionization, or dissociation (Dourdoroff et al., 1966; EPA, 1980).

Cyanides are readily absorbed through the lungs, gastrointestinal tract and skin. Death from acute cyanide poisoning is the result of "cytotoxic anoxia", or cellular asphyxiation. It is one of the most rapidly acting toxins known to humans (Gilman et al., 1980).

The detoxification of cyanide in humans is extremely efficient (Klaassen et al., 1986), and therefore, chronic toxic effects of cyanides are rare. Many chronic studies have been performed in rodents and dogs, all with negative findings (EPA, 1980; EPA, 1985). There are, however, conflicting data regarding the teratogenicity of cyanides.

The Carcinogen Assessment Group of the EPA has not evaluated the carcinogencity of cyanide because there is a lack of human and animal carcinogenicity data. The EPA, therefore, has designated it as a group "D" - Not Classified Chemical.

3.2.5 Thiocarbamates

3.2.5.1 Environmental Behavior

Very little information is available regarding the transport and fate of the individual thiocarbamate compounds, or the thiocarbamate compounds as a class. Nevertheless, data from the Handbook of Environmental Data on Organic Chemicals (1983) regarding the physiochemical properties of the thiocarbamates have been used in this report to qualitatively assess the processes that may be important in determining their fate. In addition, the limited amount of information available regarding the known properties of the individual compounds is reported here.

In general, the thiocarbamates have relatively low to moderate boiling points and moderate vapor pressures, indicating that they are somewhat volatile. Their aqueous solubilities range from 30 to 370 ppm, which indicates that they are soluble to very soluble in water. Given this information, it is likely that ground-water mobility is a significant factor in the fate and transport of the thiocarbamates. Because octanol-water partition coefficients were not available for these compounds, the importance of adsorption and bioaccumulation could not be assessed. In addition, the significance of photolysis, hydrolysis, and biotransformation cannot be assessed with the information available from Verschueren (1983).

Limited information regarding the fate of S-ethyl dipropylthiocarbamate (EPTC) and molinate was available in the Handbook of Environmental Data on Organic Chemicals (1983). Verschueren reports a 75% to 100% disappearance of EPTC from

soils in four weeks time as the result of "aquatic reactions". According to Verschueren, molinate added to tapwater decreased over a fourteen-day holding period to 40% based on recovery of isotopic carbon (14C). The loss of 14C was attributed primarily to volatilization. The five major organosoluble metabolites in the tapwater were molinate sulfoxide, 3- and 4-hydroxymolinate, 4-ketomolinate, and ketohexamethylene-imine. In addition, Verschueren (1983) reports that two kinds of microorganisms isolated from garden soils and rice-field drains degraded molinate completely into various hydroxy and oxidized products. Thus, it has been shown that volatilization and biodegradation are significant in terms of the fate and transport of at least one thiocarbamate.

3.2.5.2 Toxicity Assessment

The thiocarbamates are a class of compounds with multiple uses. The bioloical activity and corresponding toxicity of these compounds varies greatly and is contingent upon structure. In general, however, the thiocarbamates act upon specific biochemical components of the nervous system. They inhibit the neurotransmitter enzyme, cholinesterase. Lethal doses that effect 50% of the exposed animal population, referred to as the LD₅₀, range from 500 mg/kg to 4000 mg/kg (Gosselin et al., 1984). The wide range of reported LD₅₀'s may be due to the varying extent of cholinesterase inhibition by the different compounds.

Several chronic exposure studies (Stauffer Chemical data) using cycloate and molinate revealed that no neurotoxicity was observed at low doses in rodents or in higher doses in chickens. At mid-to-higher doses, chronic exposure to EPTC in rats revealed peripheral nerve and spinal cord damage. In a rat teratology study, no teratogenic effects were noted at any dosage for butylate, cycloate, EPTC, or molinate. Empirical data for the thiocarbamates are contained in Appendix XIX, and a review of the Stauffer Chemical studies and other pertinent literature is presented in the Endangerment Assessment.

3.2.6 Thiocyanate

3.2.6.1 Environmental Behavior

There is virtually no information available regarding the fate and transport of thiocyanates. According to the EPA (1979), thiocyanates (SCN radical) are formed from cyanides and sulfur-containing materials, and are more stable than cyanates (OCN radical). Solutions of thiocyanates form free hydrogen cyanide in acidic media.

3.2.6.2 Toxicity Assessment

The toxicity of the organic thiocyanates appears to be dependent upon the length of the aliphatic side chain. Methyl, ethyl and isopropyl thiocyanates and Lethane 384 are potent and rapidly acting poisons (Gosselin et al., 1984). The mean lethal dose of methyl thiocyanate fed to rats is less than 20 mg/kg (von Oettinger et al., 1936). Effects produced by these compounds include central nervous system depression with a transient period of respiratory stimulation. This may progress to death due to respiratory failure (Gosselin et al., 1984). It is known that liver enzymes liberate cyanide from these thiocyantes (von Oettinger et al., 1936), and this is probably the cause of poisoning. Butyl thiocyanate (and higher homologues) do not liberate significant amounts of cyanide in vivo (Ohkawa et al., 1973; von Oettingen et al., 1939). The mechanism for toxicity is not known for the higher thiocyanates. All thiocyanate derivatives can cause primary irritation to the skin and eyes. Undiluted solutions may produce severe cutaneous reactions (Cameron et al., 1939).

4. ENVIRONMENTAL SETTING

4.1 Physiography and Topography

The Cold Creek/LeMoyne site is located in the Piney Meadows physiographic province, which borders the Mobile River from central Mobile County down to and along the Gulf Coast of Alabama. In the site area, the province is twelve miles wide and is developed on late Pleistocene river terraces and present-day flood-plain deposits. These deposits are superimposed on, and entrenched into underlying Miocene deposits consisting of sands and clays, with the latter predominating toward the south in Mobile County.

Surface elevations range from highs of approximately 45 feet to lows of less than 10 feet in marshy areas within one-half mile of the Mobile River. Most of the area containing plant facilities (about 1 1/2 miles west of the river) is at an elevation of between 30 and 40 feet.

4.2 Surface Water

Natural drainage from several hundred acres of land, including the western part of the LeMoyne plant property, a portion of the north central part of the adjacent CNA property, and a part of the adjacent Route 43 right-of-way, forms an unnamed stream that flows in an easterly direction south of the LeMoyne plant area. This stream then turns northward and flows generally north-northwest through a 20-acre area of the Cold Creek Swamp. Flow from the marsh area joins Cold Creek, which flows northeasterly and then easterly to discharge into the Mobile River more than a mile upstream from the eastern extension of the LeMoyne plant property.

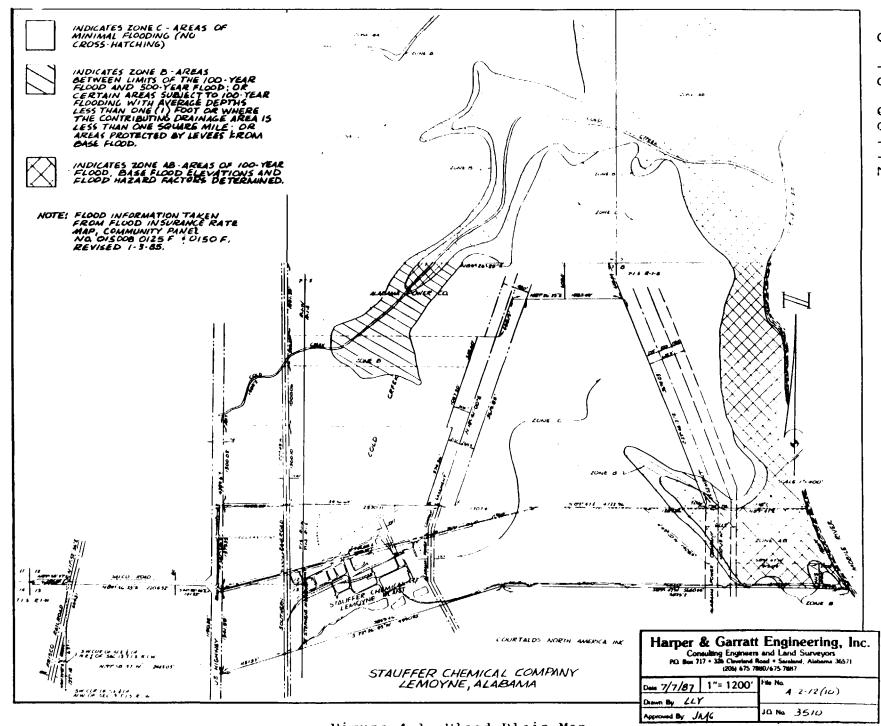
Generally, the potential for flooding in the site area is considered minimal. Although high-intensity rain storms (greater than two inches per hour) are not uncommon, they generally do not last for significant lengths of time. Based

on current flood insurance rate maps, the 100-year flood zone within the site area (Zone A8 in Figure 4-1) is confined to the eastern-most section of the LeMoyne plant property adjacent to the Mobile River. The approximate area covered by the 100-year flood plain in this part of the site is 55 acres. The zone of influence of the 500-year flood plain (Zone B) is only slightly larger (narrow band) than that of the 100-year flood plain. The closed LeMoyne landfill is approximately 500 feet west of and not within the flood plain.

An extension of the present flood plain also occurs north of the site along the lower portions of Cold Creek and portions of the Cold Creek Swamp. The combined 100- and 500-year flood-plain areas are indicated as an approximate 800-foot wide band along the section of Cold Creek which flows in a northeasterly direction, with the flood plain broadening where Cold Creek begins its east-southeasterly direction of flow. The flood plain falls within but does not completely encompass the Cold Creek Swamp (see Figure 10-1 in Appendix X) and does not extend southward toward the site along the swamp area which is associated with the unnamed stream. The majority of the site, i.e., all portions of the site and surrounding properties not within the 100- or 500-year flood areas, are classified as Zone C, or areas considered as having a minimal flooding potential.

4.3 Soils

The soils encountered at the site are loamy clays with loamy clayey subsoils. Locally, poorly drained areas of organic material and mucky clays occur. The major soil type identified by the Soil Conservation Service is the Izagora-Annemaine Association with gentle to moderate undulations.



4.4 Geology

4.4.1 Regional Geologic Setting

The site is physiographically located within the Southern Pine Hills Section (Piney Meadows subsection) of the East Gulf Coastal Plain Physiographic Province. The generalized geology of the coastal plain region includes several types of Mesozoic and Cenozoic-age sedimentary rocks that occur in narrow northwest-southeast-trending bands which dip gently southward at approximately 20 to 40 feet per mile. Within the Southern Pine Hills Section of the coastal plain, the underlying sedimentary units are overlain by Miocene estuarine deposits consisting of interbedded sands and clays, and in some areas the younger Pliocene Citronelle Formation which generally consists of sand and gravel (Geological Survey of Alabama, 1968, 1971). These deposits are in many areas overlain and incised by younger Pleistocene- and Holocene-age alluvial deposits, deposition occurring from long-term sedimentation from several north-south trending streams and rivers.

4.4.2 Site Geology

The site is underlain by low river terrace and alluvial deposits that are approximately 110 to 130 feet thick. These deposits thin to approximately 60 feet adjacent to the Mobile River, which is located approximately one and one-half miles east of the central plant area. The deposits consist of generally clean, unconsolidated, fine- to very coarse-grained sands that contain some interbedded, discontinuous clayey seams as well as some gravelly zones. Table 4-1 summarizes the stratigraphic column in the site area. The upper sands, varying in thickness from 0 to 50 feet, consist of fine- to medium-grained sands, fine-grained sandy silts, silty clays, and clays. The upper sands have moderate to low permeability. The lowermost sands, situated generally between 80 feet below

TABLE 4-1 STRATIGRAPHIC COLUMN

Range of <u>Thickness</u>	Range of	Bottom	Description
10-17	0	8-22	Red, yellow to brown stiff clay with basal sandy clay section pinching out locally.
0-35	10-15	11-74	Sand and clay interbeds grading laterally into sand.
14-34	8-63	30-63	Clean coarse sand with some clay interbeds.
18-45	30-74	63-102	Sand and gravel with lenses of sand or sand with some clay interbeds. Clay occurs interbedded with sand and gravel locally.
3-20	63-82	75-110	Gray sand and clay grading laterally into either sandy clay or sand.
1-23	75-110	75-115	Gray sand with some clay with lenses of sandy clay.
0-23	80-115	111-131	Sand and gravel with some clay interbeds.
	111-131		Blue clay.

After: ERT, 1985

ground surface and the base of the aquifer, contain the most highly permeable material. A very stiff, dense, bluish-gray clay, presumably of marine origin, underlies the alluvial deposits.

Numerous test borings and well installations have been completed within the study area for various purposes (see Figures 4-2 and 4-3, and Figure 1-2). Most borings have been terminated within a unit described as a "blue clay", which has been encountered in over 50 borings across the Cold Creek/LeMoyne site, as well as the adjacent CNA property to the This clay unit is encountered at depths varying from 60 to 130 feet below land surface, depending upon boring location, with the average depth of the clay layer found at approximately 115 to 130 feet below land surface. The clay layer is generally described as a medium-stiff to stiff blue or blue/gray clay and is apparently overlain in some areas by a thin layer of softer yellow/brown clay. Most borings have been terminated just within this clay layer; therefore, there is relatively little information regarding its vertical extent or thickness across the site. Logs for two test borings which extend to depths below the average depth of the clay layer indicate that the layer may be at least 20 feet thick (boring 0-9) and greater (boring 0-28, approximately 70 feet) in some areas.

Previous studies indicate that the clay unit dips very slightly to the southwest (Stilson, 1974). The depth at which the blue clay unit has been encountered in various borings across the site is indicated in Table 4-2. This table illustrates that the clay layer is present in several areas of the site and is therefore likely to be horizontally contiguous across the site. As indicated in the Work Plan, previous studies have also suggested that the clay unit is continuous across the site and serves to isolate the alluvial deposits from other stratigraphic units below. Graphic and descriptive logs for many of the test borings are included in Appendix XXVII. Permeability testing of a single sample of the clay, collected with a Shelby tube from boring 0-86 at a depth of 118

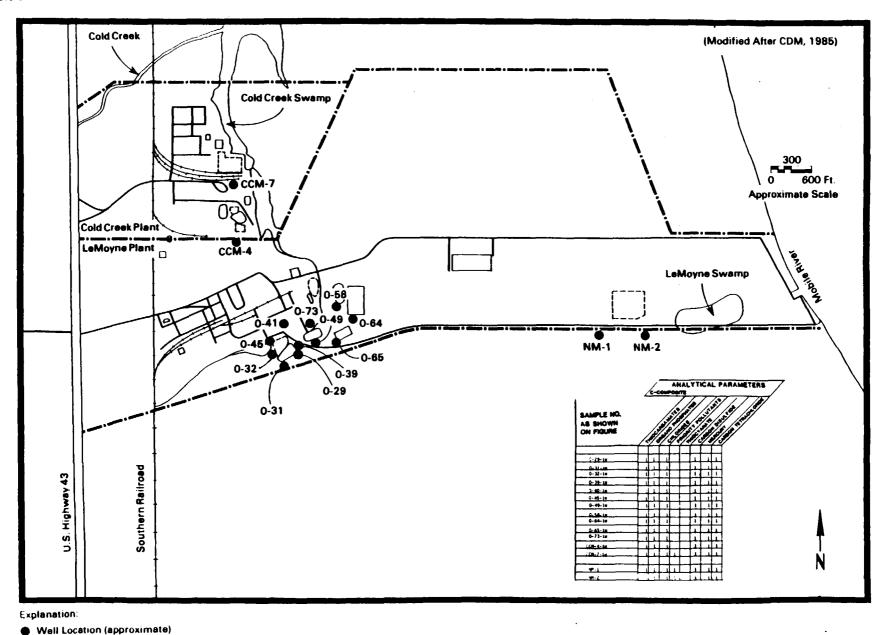


Figure 4 2 Source Well Sample Locations

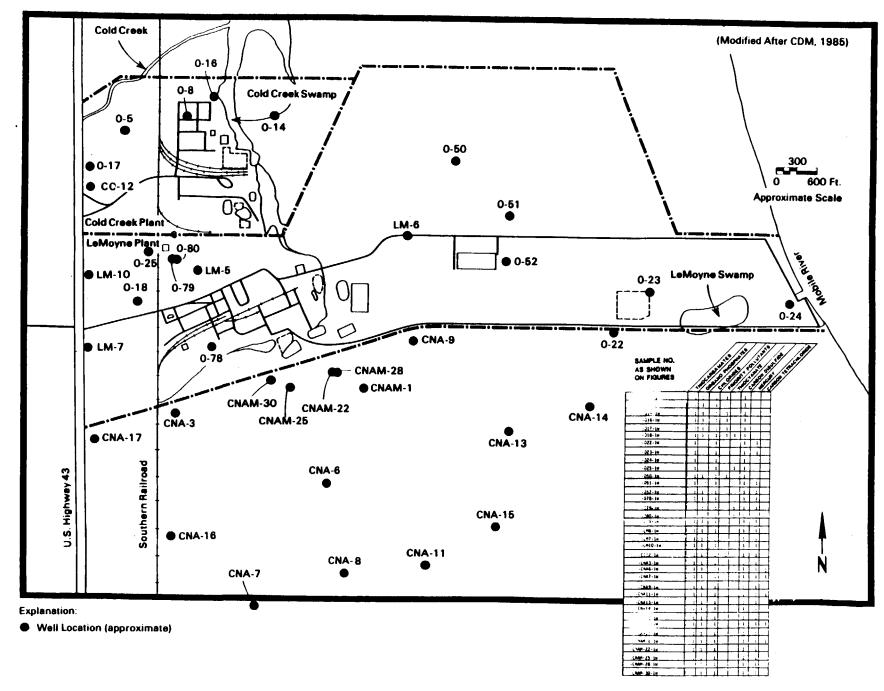


Figure 4-3 Area Well Sample Locations

TABLE 4-2

DEPTH OF BLUE CLAY LAYER

(see Appendix XXVII for well logs)

Boring Number	Depth Encountered (feet below ground surface)	Total Boring Depth (feet below ground surface)
0~5	118	120
0-6	132	136
0-8	121	122
0-14	115	120
0-21	117	120
0-22	100	105
0-23	76	80
0-24	60	63
0-59	117	123
0-62	116	120
0-68	118	119
0-70	117	118
TW-12	122	124.5
0-28	123	610
0-9	121.5	461
CNAM-23	117	119
CNAM-24	125	127
CNAM-25	130	131
CNAM-26	124	126
CNAM-27	131	132
CNAM-30	130	132
CNAM-32	121	121
CNAM-34	127	128.5
0-75	124	125
0-76	112	114

to 120 feet below land surface, yielded a permeability value of 4.4×10^{-8} cm/sec.

GWA (1978) identified an intermediate clay unit in the western sector of the property. They suggest that it becomes discontinuous, thinning to the east into intermittent lenses.

4.5 Hydrogeology

4.5.1 Regional Hydrogeologic Setting

There are two principal water-table aquifers in Mobile County. A major aquifer is located a few miles west of the site in the Miocene Uplands section of the county. The second aquifer is located within the Mobile River Valley, where the site is located. This aquifer is the principal source of water for users located within the Mobile River Valley. Wells in this aquifer typically yield 470 to 846 gallons per minute (gpm), with specific capacities of 6 to 73 gpm per foot of draw down (Riccio et al., 1973).

4.5.2 Site Hydrogeology

The Mobile River Valley water-table aquifer at the site is recharged through infiltration from the Cold Creek Swamp, the Mobile River, and rainfall. The background water quality is potable, with low total dissolved solids and iron. Prior to industrialization, ground-water flow was toward the Mobile River. The ground-water table varied from 0 to 20 feet below ground level depending on the topography. Presently, however, the direction of flow is toward the south-southeast, because of the local influence of pumpage at CNA and from Stauffer's interceptor wells (see Figure 4-4). The advent of industrialization and accompanying ground-water pumpage at the CNA plant site and surrounding area has resulted in a lowering of the water table and localized changes in the direction of ground-water flow, i.e., presently, ground-water generally flows away from the Mobile River (ERT, 1984).

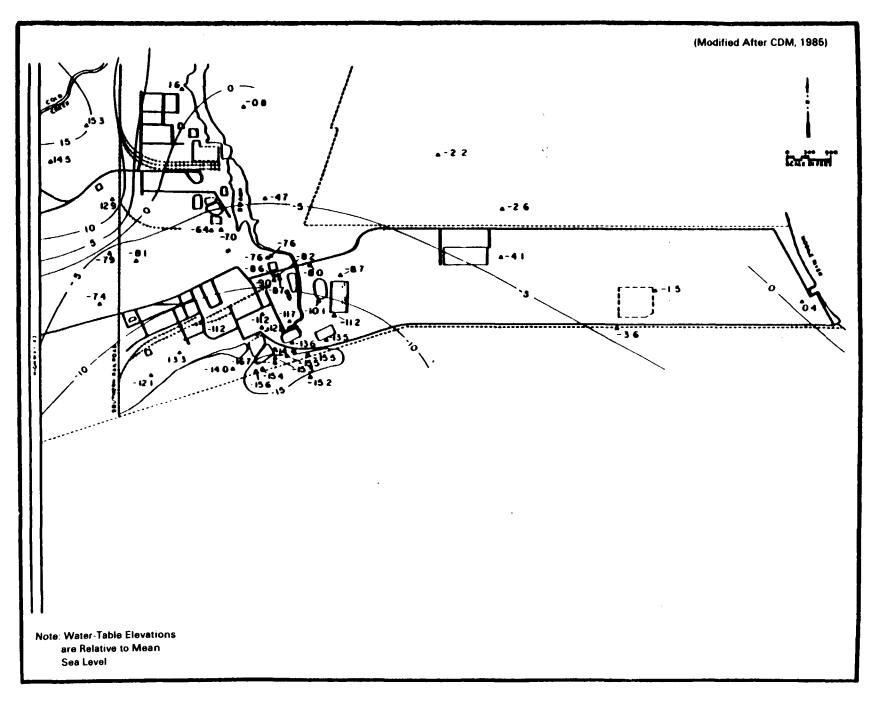


Figure 4-4 October, 1983 Water Table Elevations

The ground-water table within the site area is generally encountered at depths ranging from 25 to approximately 75 feet below ground surface. The actual water-table depth across the site area varies according to geographic location within the site area, as well as concurrent ground-water pumping at the time of measurement. Numerous ground-water supply wells and extraction/interceptor wells which pump several hundred gallons per minute are located in the site area. A recent ground-water level sampling program was completed at the Cold Creek/LeMoyne site and adjacent CNA plant property in October, 1987. data provide information on ground-water levels under near-stable withdrawal conditions typical of normal operations at the site. Water-level data for various wells are presented in Appendix XXVIII. These data indicate perceived normal water-table depths of approximately 40 feet below ground surface within the Cold Creek portion of the site and greater than 50 feet below ground surface within the LeMoyne plant portion of the site. Deeper water levels in the southern part of the site likely reflect higher rates of ground-water withdrawal in this area.

It is unlikely that buried wastes within any of the waste disposal areas, landfills or wastewater treatment ponds come into contact with the ground-water table under current site conditions. All of the waste disposal areas are less than 20 feet deep, and all of the landfills are less than 12 feet deep (see Appendix XXVI for graphic depictions of these areas), whereas depth to ground water varies from approximately 40 to 50 feet below ground surface within the site area. Depth to ground water under static and non-pumping conditions is not known; however, water level data in remote corners of the site area indicate water levels varying from approximately 24 to 40 feet below land surface. Therefore, even under non-pumping conditions, it is unlikely that buried wastes would come into contact with ground water.

Relatively few deep borings or well installations have been completed within the immediate site area that extend into stratigraphic units beneath the blue clay layer. Well

construction information provided in Appendix XII indicates a total of eight borings which extend to a significant depth beneath the clay layer. Three of these borings represent injection wells which were drilled to depths varying from 4330 to 4750 feet below ground surface, with screens set for injection purposes at varying depths greater than 3400 feet below ground surface. Monitoring wells installed in association with the injection wells (total of five indicated) have been installed with screens at depths varying between 207 and 1160 feet below ground surface. Specific well construction details are provided within Appendix XII, and information regarding screened intervals and intended use is presented in Table 4-3. Lithologic logs available for two of the borings, IM-1 and IM-2, indicate numerous alternating layers of clay with silty fine sand and, in one case, medium to coarse sand (IM-2), extending several hundred feet beneath the blue clay layer. Both logs (Appendix XII) indicate that clay is the dominant lithologic formation encountered with relatively thin layers of fine sand or silt to depths approaching 500 feet below ground surface. Ground-water usage within the site area is believed to be limited to the upper aquifer above the clay layer.

GWA (1979) conducted aquifer pumping tests with existing production wells LM-2 and CNA-1 (see Figure 1-11) to evaluate hydraulic responses and determine aquifer characteristics in the site area. Transmissivity from LM-2 testing was determined to be 93,123 gallons per day per foot (gpd/ft), and the storage coefficient was calculated to be 0.31. Transmissivity from CNA-1 testing was 85,232 gpd/ft, and the storage coefficient was 0.15. Based on an average saturated thickness of 77 feet, the average hydraulic conductivity was calculated to be 1,100 gallons per day per square foot (gpd/ft²).

To properly characterize the ground water in the vicinity of the LeMoyne landfill area, the Work Plan proposed the installation of two new monitoring wells downgradient of the landfill. These wells, NM-1 and NM-2, were installed to a

TABLE 4-3

DEEP BORING INFORMATION

(see Appendix XII for boring logs)

Well Number	Total Depth (feet below ground surface)	Screened Interval (feet below ground surface)	Intended Use
IM-1 (0-9)	461	207-227	Monitoring well for INJ-1; first aquifer beneath blue clay
IM-2 (0-28)	610	580-590	Monitoring well for INJ-2; aquifer at 560-610
IM-4	242	208-228	Monitoring well for INJ-3; aquifer at 208-228
IM-5	1204	1140-1160	Monitoring well for INJ-3; sand aquifer above bucatunna clay
IM-6	250	222-242	Monitoring well for INJ-2
INJ-1	4330	3402-3491*	Injection well; plugged and abandoned at 4184
INJ-2	4601	4458-4600*	Injection well
INJ-3	4750	4415-4515*	Injection well

^{*}Perforated zone.

depth of 62.5 feet and 37 feet, respectively. A report by Thompson Engineering Testing, Inc., which includes a description of their installation, log sheets, and locations, is included in Appendix VII. Included in Appendix XII is CDM's Figure A-1, showing typical well construction details of the existing source and area wells, and tables showing coordinate locations, elevations, depths and screened interval of these wells.

3 10 00125 5. SAMPLING AND ANALYSIS PROGRAM

As discussed in Section 1.3, the RI was divided into two major subtasks, source characterization and area characterization. The sampling and analysis programs used to characterize the possible sources and areas of contamination as well as the analytical results of these investigations were summarized in Section 1.3.2. The following sections provide a detailed description of the source and area characterization sampling and analysis programs, the results of these investigations, and conclusions regarding the nature and extent of contamination.

5.1 Source Characterization

5.1.1 Landfills

5.1.1.1 Program Description

Soil borings were installed at 45-degree angles beneath the three existing landfills on site and advanced to 50 feet or until ground water was encountered. Composite samples were collected at 10-foot depth increments and analyzed for location-specific compounds as well as priority pollutants and vanadium. A total of twelve soil borings were installed under the three landfills, as shown in Figure 5-1.

In addition to the soil boring program described above, samples of the DuPont 3110 plastic membrane that lines the top and sides of the three landfills were carefully exposed, sampled, and tested for integrity (see Appendix XVI).

5.1.1.2 Findings and Conclusions

Table 5-1 includes the results of the 12 composite soil samples that were collected from the three landfills and analyzed for priority pollutants and vanadium. Other than

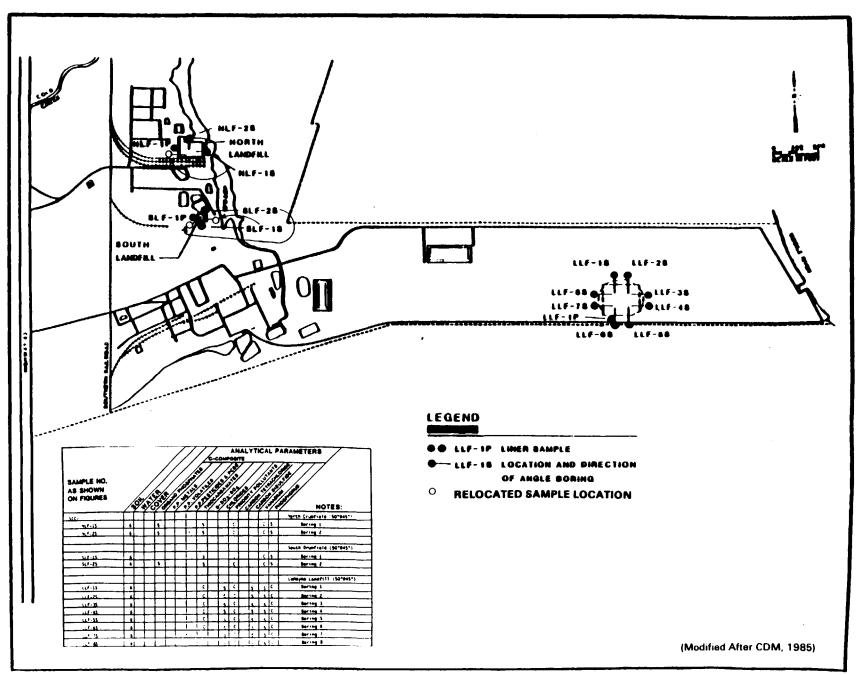


Figure 5-1 Landfill Sample Locations

TABLE NO. 5-1

COMPOSITE SOIL SAMPLES - LANDFILLS - PRIORITY POLLUTANTS IN Mg/Kg - INCLUDES VANADIUM

(Except for heavy metals, all other priority pollutants are ND)

(See Fig. 5-1 for location)

	NLF-1S	<u>NLF-25</u>	SLF-1S	SLF-2S	LLF-15	<u>LLF-28</u>	<u>LLF-38</u>	<u>LLP-45</u>	LLF-58	LLF-65	LLF-78	LLF-85
Ant imony	ND	, ND	ND	ND	24	7.5	23	33	16	11	BMDL	8.6
Arsenic	BMDL	1.0	1.0	3	ND	BMDL	ND	BMDL	1.0	BMDL	ND	BMDL
Beryllium	0.17	0.14	0.21	0.65	BMDL	BMDL	0.24	0.055	BMOL	BMDL	ND	BMDL
Chromium	6.5	14	16	31	46	5.3	15	22	7.8	7.1	3.8	5.2
Copper	2.5	2.9	4.5	7.7	48	7.2	23	35	26	19	15	32
Lead	BMDL	BMDL	4	BMDL	6	BMDL	BMDL	ND	BMDL	BMDL	BMDL.	BMDL
Mercury	BMDL	0.6	BMDL	BMDL	0.1	BMDL	0.1	ND	BMDL	,1.9	0.1	ND
Nickel	1.4	1.6	2.1	3.5	2.8	1.3	3.8	3.9	1.6	4.1	BMDL.	1.9
Selenium	ND	BMDL	0.5	0.8	ND	ND	ND	ND	ND	ND	ND	ND
Zinc	8.9	9.6	15	28	12	6.8	19	22	7.1	14	4.4	8.1
Vanadium	8.3	20	26	49	-	_	-	-	-	-	-	-
	Not	rth	S C	wth		Th	iocarbamate	s & phosph	orous in	all LLF (L	eMoyne Lan	df111)
		Cold Cree	k Landfill	L				Composite	samples	are ND		

*Possible False Positive

relatively low levels of heavy metals, which are in the average range for natural soils (see Appendix XVIII), no other priority pollutants were found.

Table 5-2 includes the results of site-specific compound analysis (organophosphates, thiocarbamates, and phosphorous) of soil samples collected from the north and south Cold Creek landfills. With a few minor exceptions, location-specific compound concentrations in the two Cold Creek landfills were not detected or were below the method detection limit (BMDL) for that compound. The highest compound concentration found in the Cold Creek landfills was 1.5 milligrams per kilogram (mg/kg, or parts per million (ppm)) of molinate, a thiocarbamate pesticide which is manufactured at the Cold Creek plant. As shown in Sections 5.1.4 and 5.2.2, thiocarbamates were found at very low levels in many of the source and area However, the mean molinate levels in all wells are below the proposed safe drinking water level of 0.2 mg/l (see Appendix XIX). Based on these findings, the Cold Creek landfills are not considered a significant source of contamination.

Results of the soil samples collected from the LeMoyne landfill are analyzed for chloride, vanadium, and carbon disulfide are included in Table 5-3. Carbon disulfide was not detected in any of the LeMoyne landfill soil samples. One relatively high level, 220 mg/kg, of chloride was detected, and vanadium was found at relatively low levels of from 1.1 to 30 mg/kg. The concentrations of both of these compounds are within the common range of natural soils (see Appendix XVIII). Thus, the LeMoyne landfill is not considered a source of contamination.

The test results of the landfill liner material that was analyzed by Matrecon, Inc. are included in Appendix XVI. Stauffer's materials specialist, Mr. L. Drake, has reviewed these test results and concluded that the membrane covers are still in excellent condition with a high life expectancy.

TABLE NO. 5-2 SOIL SAMPLES - COLD CREEK LANDFILLS - SITE SPECIFIC COMPOUNDS @ 10 FOOT DEPTH INTERVALS - Mg/kg (See Fig. 5-1 for location)

	NLF-15-10'	NLF-15-20'	NLF-15-30'	NLF-15-40'	NLF-18-50'	NLF-25-10'	0
							
							\sim
EPTC (Eptam)	BMDL	BMDL	BMDL	BMDL	ND	0.9	NLF-28-20', SLF-15-10', SLF-15-20', SLF-15-30',
Butylate (Sutan)	0.1	0.2	0.1	· ND	BMDL	BMDL	SLF-18-40', SLF-18-50', SLF-28-10', & SLF-28-20'
Vernolate (Verna	ura) BMADL	BMDL	BMOL.	ND	ND	0.1	are all ND in site specific compounds except
Pebulate (Tillam	n) ND	ND	ND	ND	ND	BMDL.	for 0.1 mg/kg Ordram in SLF-2S-20'
Molinate (Ordram	i) BMDL	BMDL	0.2	0.9	0.1	1.5	• •
Cycloate (Ro-nee	t) BMDL	ND	BMDL	ND	BMDL	0.1	
Bensulide (Betas	an) ND	0.5	ND	0.1	ND	ND	

TABLE 5-3 SOIL SAMPLES - LEMOYNE LANDFILL - VANADIUM & CHLORIDE IN Mg/Kg @ 10 POOT DEPTH INTERVALS (See Fig. 5-1 For location)

	<u>LLF-1S-10'</u>	<u> LLF-15-20'</u>	LLF-25-10'	LLF-25-20'	<u>LLF-35-10'</u>	LLF-35-20'	LLF-45-10'	LLF-45-20'	LLF-5S-10'	LLF-55-20'	LLF-55-30'
Vanadium Chloride	14 ND	1.6 ND	30 220	1.3 ND	9.7 ND	4.8 ND	24 ND	4.9 ND	3.6 ND	3.3 ND	1.9 57
Vanadium Chloride	<u>LLF-6S-10'</u> 6.1 ND	LLF-6S-20' 5.7 ND	LLF-6S-30' 1.8 57	LLF-7S-10' 3.5 ND	LLF-7s-20* 1.1 ND	LLF-75-30' 6.2 91	LLF-85-10' 2.1 ND	LLF-85-20' 4.9 ND	(Carbon di LLF sam	sulfide was ples)	ND in all

5.1.2.1 Program Description

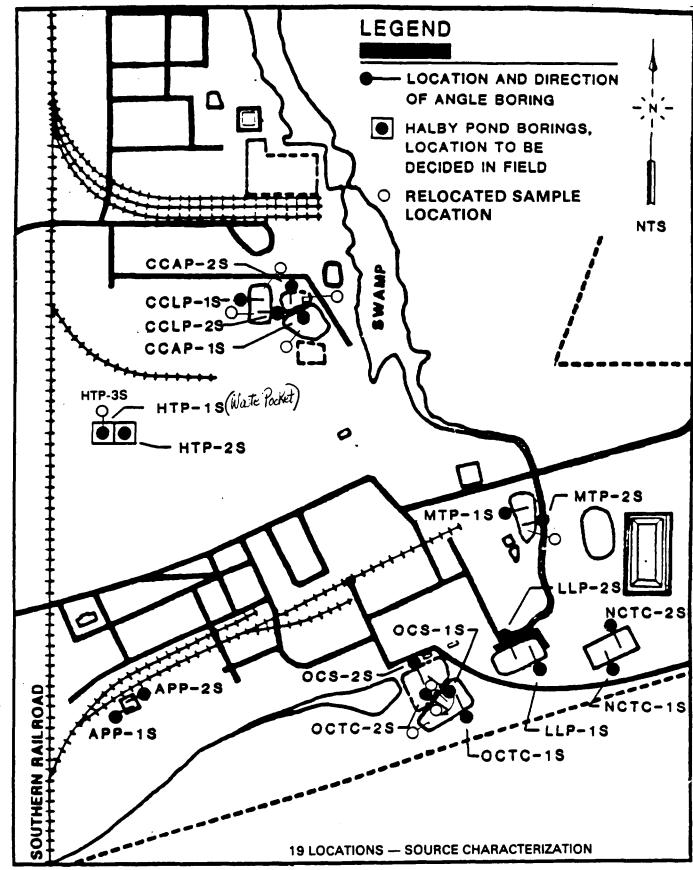
The soil sampling program for the ponds and lagoons was similar to that for the landfills: soil borings were installed at 45-degree angles beneath the nine ponds investigated on site and advanced to 50 feet or until ground water was encountered. Two soil borings were installed under each pond, creating a total of 18 soil borings, as shown in Figure 5-2. Where possible, six samples were collected from each boring. One composite sample was selected for priority pollutant analysis, and the other five were analyzed for compounds specific to the pond.

In addition to the soil samples described above, water samples from three active ponds were collected for priority pollutant analysis.

5.1.2.2 Findings and Conclusions

Table 5-4 includes the results of the composite soil samples that were collected from the nine ponds and lagoons and analyzed for priority pollutants. Except for low levels of heavy metals, which are in the average range for natural soils (see Appendix XVIII), the results showed no detectable levels of priority pollutants in any of the composite samples. One exception was sample HTP-1S, taken from the former Halby pond, which showed elevated levels of copper (440 mg/kg), zinc (1170 mg/kg), and cyanide (240 mg/kg). It is thought that this boring encountered a pocket of waste. A second boring within the pond and an angled boring beneath the pond both showed background levels of copper and zinc and no detectable levels of cyanide. Thus, except for a small pocket of waste in the Halby pond, the nine ponds showed no priority pollutant contamination.

Table 5-5 includes the results of composite soil samples from the nine ponds and lagoons that were analyzed for



(Modified After CDM, 1985)

.Figure 5-2 Pond and Lagoon Soil Sample Locations

TABLE NO. 5-4

COMPOSITE SOIL SAMPLES - PONDS & LAGOONS - PRIORITY POLLUTANTS IN mg/kg
(Except for heavy metals and some cyanide, all other priority pollutants are ND)

(See Fig. 5-2 for location)

CCAP-1S CCAP-2S CCLP-2S HTP-1S HTP-2S HTP-3S HTP-1S HTP-2S NCTC-1S NCTC-1S OCTC-1S OCTC-1S OCTC-2S OCS-1S OCS-2S APP-1S APP-2S LLP-1S LLP-2S Arsenic 2.0 1.0 2.0 2 BMDL. 4.0 BMDL. 1.0 7 1.0 2 AMDI. RMI)I. 1.0 BMDL. ND Beryllium 0.11 0.52 0.1 0.11 0.2 0.08 0.15 0.1 0.12 0.45 0.36 0.45 0.04 BMDL 0.13 0.15 BMDL. 0.08 0.16 Chromium 9.5 7.4 15 43 16 14 4.5 8.6 29 100 5.2 15 14 16 12 11 6.5 9.4 16 Соррег 3.8 2.3 4.2 2.7 5 5.4 442 5.1 4.3 2.3 3.3 4 4.6 18 3.4 15 10 3.8 2.2 Lead ND RAIDI. BMDL. BMDL. 22 RMDI. BMDL. 4.9 5.8 5.8 BMDL. BMDL. BMDL. BMDL. BMDL. 7.2 7.6 5.4 6.1 Mercury ND ND BMDL. 0.9 BMDL BMDL 24 BMDL. 0.3 0.4 BMDL BMDL BMDL. BMDL BMDL BMDL 1.4 BMDL BHDL. Nickel 1.6 1.6 3.6 3.7 18 6.3 3.3 BPDI. 1.6 1.8 8.2 1.2 2.8 2.6 8.4 1.2 1.2 1.7 2.6 Zinc 7.3 5.3 15 1170 37 21 23 9.7 12 15 21 6.1 11 15 14 8.4 10 11 20 Cadmium ND ND ND ND 1.6 ND ND ND ND ND ND MD ND ND ND ND ND ND ND Ant imony ND ND ND ND ND ND MD ND MD MD MD 9.7 31 32 33 NO ND ND ND 1.0* Cyanide ND ND ND 240 ND ND ND MD ND ND ND ND ND ND ND ND ---- Cold Creek Ponds --- Halby Pond LeMoyne Ponds

^{*}Questionable based on QA field blank analysis (see Appendix VI for explanation).

TABLE NO. 5-5 (1 of 2)

SOIL SAMPLES - POND'S & LAGOONS - SITE SPECIFIC COMPOUNDS IN Mg/Kg @ 10 FOOT DEPTH INTERVALS

(See Fig. 5-2 for location)

	CCAP-15-10'	CCAP-1S-20'	CCAP-18-30'	CCAP-15-40'	CCAP-15-50'	CCAP-2S-10'	CCAP-28 201	CCAP-2S-30'	CCAP-25-40*	CCAP-2S-50'
EPTC (Eptam)	0.3	25	1.2	0.9	ND	ND	1.6	BMDL	0.2	NÐ
Butylate (Sutan)	0.1	0.3	0.1	0.2	ND	ND	ND	ND	ND	ND
Vernolae (Vernam)	0.1	8.6	0.3	0.3	ND	ND	1.1	0.1	0.1	. ND
Pebulate (Tillam)	0.1	1.2	0.2	0.2	ND	ND	0.1	ND	ND	ND
Molinate (Ordram)	0.3	6.4	5	3.7	ND	ND	0.8	0.2	0.2	ND
Cycloate (Ro-neet)	0.2	2.7	0.1	0.5	ND	ND	0.2	ND	ND	ND
				(Cold Creek's o	old neutraliza	ition pond			
	CCLP-15-10'	CCLP-15-20'	CCLP-18-30'	CCLP-15-40'	CCLP-18-50'	CCLP-25-10'	CCLP-2S-20'	CCLP-2S-30'	CCLP-25-40'	CCLP-2S-50'
RPT C	ND	ND	ND	ND	ND	0.1	BMDL	ND	ND	ND
Butylate	ND	ND	ND	ND	ND	0.1	BMDL	ND	ND	ND
Vernolate	ND	ND	ND	ND	ND	0.1	BMDL	ND	ND	ND
Pebulate	ND	ND	ND	₩D	ND	0.1	0.2	ND	ND	ND
Molinate	ND	ND	ND	ND	ND	0.1	0.1	ND	ND	ND
Cycloate	ND	ND	ND	ND	ND	0.2	BMDL	ND	ND	ND
Cold Creek	's LeCreek w	AT PONG								
	HTP-15-10'	HTP-2S-10'	HTP-25-20'	HTP-25-30'	HTP-25-40'	HTP-3s-10'	HTP-35-20'	HTP-3S-30'	HTP-35-40'	
Thiocyanate	1,010	550	640	20	ND - Halby Pond	2,480	1,030	420	190	
	MTP-15-10'	MTP-15-20'	MTP-15-30'	MTP-2S-10'	MTP-25-20'	Old Chlorii	ne plant wut p	pond		
Chloride	650	1,200	680	ND	ND					
	NCTC-15-10'	NCTC-15-20'	NCTC-1S-30'	NCTC-15-401	NCTC-18-50'	NCTC-25-10'	NCTC-28-201	NCTC-28-30'	NCTC 25 40'	NCTC-25-50'
Carbon Disulfide Carbon Tetrachlori	ND .de ND	ND ND	ND ND	ND ND	ND ND	ND ND	ND ND	ND ND	ND ND	ND ND
	× - + .			New Carbon	tetrachloride	wwt pond	•		- · vi =	
	OCTC-15-10'	OCTC-15-20'	OCTC-15-30'	OCTC-1S-40'	OCTC-15-50'	OCTC 25-10'	OCTC - 28-20'	ocs-1s-10'	OCS-15-20'	OCS-15-30
Carbon Disulfide	ND	ND	ND	ND	ND	0.016	ND	0.002	ND	0.008
Carbon Tetrachlor	ide ND	ND	ND	0.001	0.001	ND	ND	ND	ND	ND
			Old carbor	i teliachloric	le w wt pond	. ·	-	- Old ca	rbon disulfide	e w wt pond

S

TABLE NO. 5-5 - continued (2 of 2)

	OCS-18-40'	OCS-1S-50'	OCS-2S-10'	ocs-25-20'	OCS-25-30'	OCS-25-40'	OCS-2S-50'			
Carbon Disulfide Carbon Tetrachlo		0.003 0.003	ND 0.685	0.0025 0.0066	0.005 0.008	0.005 0.0035	ND ND			
					Acid	Plant Pond				
	<u>APP-15-10'</u>	APP-15-20'	APP-15-30'	APP-15-40'	APP-18-50'	APP-2S-10'	APP-2S-20'	APP-25-30*	APP-2S-40'	APP-25-50'
Iron Sulfate	25,600 ND	4,270 ND	10,100 ND	2,810 210	1,780 570	24,500 ND	11,100 ND	4,540 ND	5,780 200	1,880 580
	LLP-1S-10'	LLP-15-20'	LLP-15-30'	LLP-15-40'	LLP-15-50'	LLP-25-10'	LLP-25-20'	LLP-25-30'	LLP-2S-40'	LLP-2S-50'
Chloride	ND	ND	ND	65	ND LeMovne	ND LeCreek wwt p	50 ond	ND	ND	240

site-specific compounds. Low to moderate (25 mg/kg maximum) levels of a few thiocarbamates were found at the 20-foot depth in the southwest corner of the Cold Creek old neutralization. pond. As shown in subsequent sections, the closest downgradient well, CCM-4, showed thiocarbamate levels ranging from below the mean detectable limit to 0.23 mg/l, whereas the next most immediate downgradient wells showed thiocarbamate levels of from below detection limits to 0.01 mg/l. Ultimately, the existing ground-water intercept system is capturing these trace amounts which probably originate from the Cold Creek old neutralization pond.

Discharge of treated water from the intercept wells to the Mobile River is currently governed by the provisions outlined within the existing NPDES permit for the facility. This permit indicates a current limitation or allowance for combined discharge of up to 3.3 lbs/day of thiocarbamates from the facilities. Single-day water analyses recorded monthly since April, 1984 from both the LeMoyne plant and the Cold Creek plant treatment systems indicate that in only one instance (July, 1985 - 4.10 lbs) has this discharge limitation not been met. The majority of the discharge data (see Table 24-1 in Appendix XXIV) indicate less than 1.00 lb/day of thiocarbamate within the discharge stream.

There is little information available regarding the environmental persistence, or fate and transport, of thiocarbamates as they are released to a surface-water body such as the Mobile River. Some information regarding thiocarbamates in general is reported within the Waste Component Characteristics and Behavior section of this report, Section 3.2.

High levels of thiocyanate (2480 mg/kg maximum) were found under the Halby pond, indicating localized contamination around the closed pond, as suggested above. As discussed later, a low level of thiocyanate was detected in ground water downgradient from the Halby pond, whereas no other site-specific compounds were detected. The Halby pond is therefore considered the

probable source of thiocyanate contamination found in the ground water.

High iron concentrations (25,600 mg/kg maximum) were found under the LeMoyne acid plant WWT pond. The only finding in the nearest downgradient well, CNA-3, was 0.0003 mg/l $\rm CS_2$. Because the acid plant pond has never received any $\rm CS_2$, it is not likely to be a source of $\rm CS_2$ contamination.

Results of the analysis for priority pollutants of water samples taken from three active ponds are included in Table 5-6. Except for very low levels of a few heavy metals, all the priority pollutants were essentially non detectable.

5.1.3 Swamps

5.1.3.1 Program Description

Three-foot deep soil borings were installed at 34 locations in the Cold Creek Swamp and at four locations in the LeMoyne Swamp, as shown in Figures 5-3 and 5-4, and in Drawing 1.3 in Appendix XVII. A split-spoon sampler, shelby tube, or acrylic tube was used, depending on soil conditions. Excess water was decanted, and the sample was mixed in a stainless steel tray. A subsample was then taken for the composite. Of the 38 composite samples collected, 3 from the Cold Creek Swamp and 4 from the LeMoyne Swamp were analyzed for thiocarbamates, chloride, and priority pollutants. The other 31 samples were analyzed for mercury only.

In addition, fish samples from 5 locations in the Cold Creek Swamp (see Drawing 1.3 in Appendix XVII) were collected and analyzed for mercury in accordance with the Work Plan and subsequent meetings with the EPA. Because of the very small size of the individual fish, an aggregate sample was taken and the analysis was performed on a homogenate, i.e., whole fish.

TABLE NO. 5-6

POND & SURFACE WATER - PRIORITY POLLUTANTS IN mg/l
(Any priority pollutants not shown are ND or BMDL)

	SW-01-1W	SW-11-1W	CCLP-1W	NCTC- LV	LLP- LV		
Carbon Tetrachloride	-	ND	0.0056	0.032	ND		
Mercury	0.0002	_	BMLD	0.0003	0.0008		
Nickel	ND	-	0.013	BMDL	0.022		
Zinc	0.31	_	0.16	0.012	0.023		
Cyanide	_	_	ND	ND	0.132		
Phenol	ND	ND	ND	0.007	ND		
Priority Pollutants							
Volatiles	_	All ND	_	-	-		
	Bac	kground ·	((Ref. to Fig. 5-2			
	(Ref. to	•	for location)				
	Pla. 5-5	for location)					

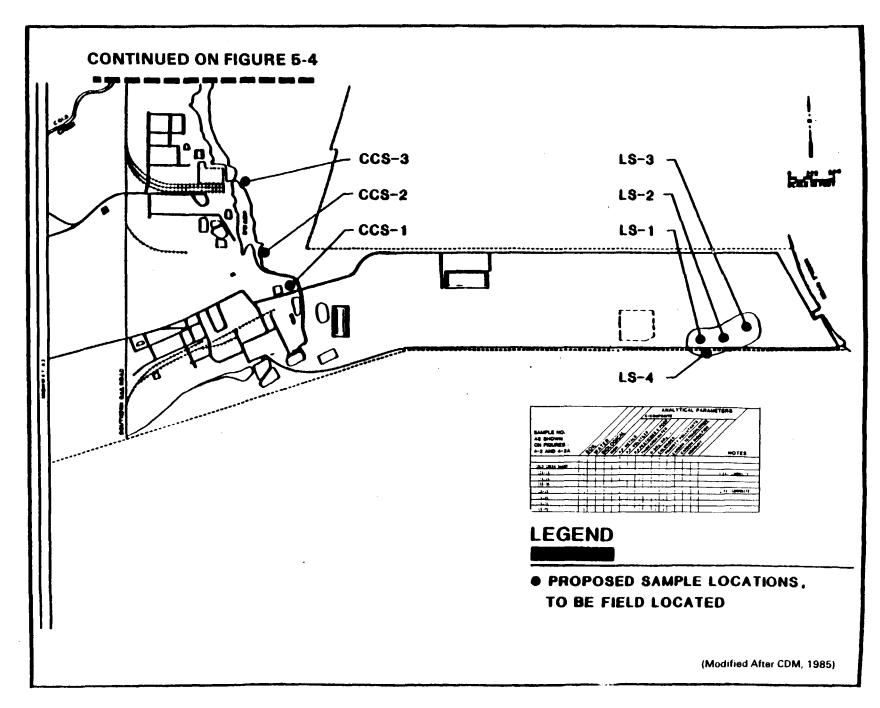


Figure 5-3 Swamp Sample Locations

Figure 5-4 Northern Swamp Sample Locations

Results of the seven swamp samples analyzed for thiocarbamates, chloride, and priority pollutants are included in Tables 5-7 and 5-8.

Among the priority pollutants, only heavy metals with levels typical of natural soils were found. Mercury was the only exception, with a maximum concentration of 300 mg/kg. These heavy metals are insoluble and effectively immobile because the analytical results of ground-water samples, reported later, do not indicate their presence. It is noteworthy that one soil sample, SW-Ol-1S, selected in the Work Plan as the background level for heavy metals, shows the same relative levels of metals as found in Cold Creek Swamp except for mercury. SW-Ol-1S was collected near LeMoyne's south property line next to the railroad tracks (see Figure 5-5). It also compares favorably with a road-side sample from U.S. Highway 231 (see ADEMS December 19, 1986 letter in Appendix XV).

Most thiocarbamates were not detected, although a few were present at low levels (1.8 mg/kg maximum). Analysis of downgradient ground water, however, as discussed in subsequent sections, did not detect any thiocarbamates. Therefore, the swamps are not considered sources of thiocarbamate contamination.

Chloride in the Cold Creek Swamp soil samples varied from not detected to 50 mg/kg, and in the LeMoyne Swamp samples from not detected to 190 mg/kg. These levels are within the average range for natural soils; therefore, the two swamps are not considered sources of chloride contamination.

Table 5-9 includes the mercury results of composite soil samples collected from the Cold Creek Swamp. Mercury was found in all but one of the 31 Cold Creek Swamp samples at concentrations of from 0.14 mg/kg to 690 mg/kg. Although the mercury levels are high, the mercury is probably present as the sulfide (as discussed in Appendix XXV) and is therefore immobile and insoluble. This is evidenced by the fact that

TABLE NO. 5-7

COMPOSITE SOIL SAMPLES - COLD CREEK & LEMOYNE SWAMP - PRIORITY POLLUTANTS IN Mg/kg

(All priority pollutants not shown are ND)

	LS-1S	<u>LS-2S</u>	<u>LS-3S</u>	LS-4S	<u>W-01-15</u>	CCS-1S	<u>ccs-2s</u>	CCS-3S
Arsenic	ND	ND	2	BMDL	5.0	5	5	5
Beryllium	0.4	0.11	0.71	0.65	0.53	0.46	0.31	0.81
Chromium	19	6.2	24	50 ·	88	130	140	180
Copper	6.4	1.9	10	8.3	68	35	14	34
Lead	15	BMDL	23	12	100	31	BMDL	26
Mercury	0.11	0.15	0.22	0.28	0.9	300	190	230
Nickel	2.1	BMDL	5.8	3.5	5.6	51	32	56
zinc	12	20	131	55	180	171	312	561
		LeMo	yne Swamp		Background Ref. to Fig. D-1 for location		Cold Creek Sw	amp

NOTE: Refer to Fig. 5-3, 5-4 and Dwg. No. 1.3 for location of Swamp Samples.

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	ccs-1s	CCS-2S	CCS-3S	<u>LS-1S</u>	<u>LS-2S</u>	<u>LS-35</u>	<u>LS-4S</u>
EPTC (Eptam)	0.1	0.2	1.0	ND	ND	ND	ND
Butylate (Sutan)	ND	0.3	1.8	ND	ND	ND	ND
Vernolate (Vernam)	ND	0.2	1.1	ND	ND	ND	ND
Pebulate (Tillam)	ND	ND	0.3	ND	ND	ND	ND
Molinate (Ordram)	0.1	0.4	0.9	ND	ND	ND	ND
Cycloate (Ro-neet)	ND	0.5	1.8	0.2	ND	ND	ND
Chloride	ND	50	50	190	ND	100	50
		Cold Creek			leMc	VD6	

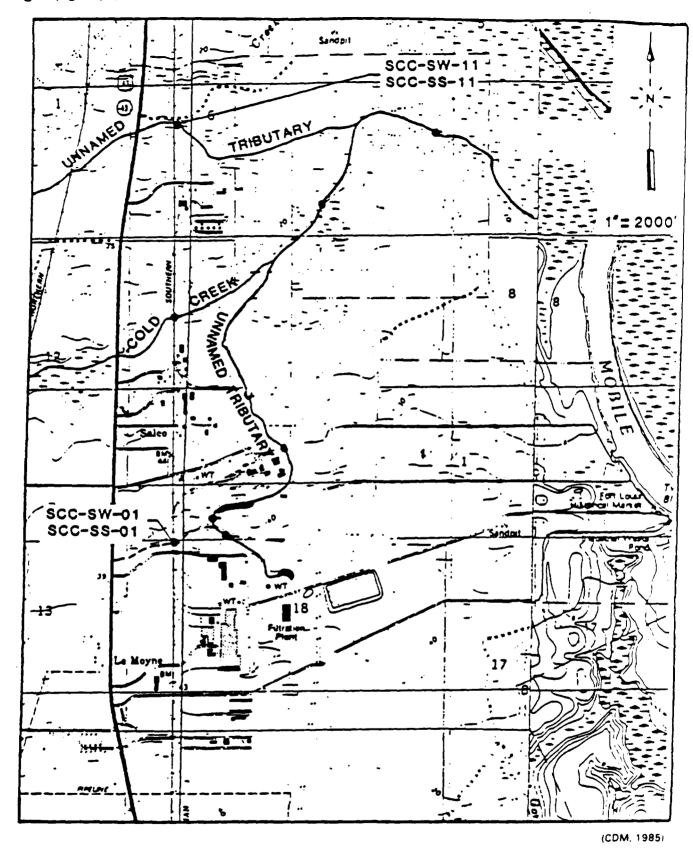


Figure 5-5 Sampling Points on Cold Creek and Tributaries

COMPOSITE SOIL SAMPLES	· COLD CREEK SWAMP -	MERCURY IN Mg/Kg (see Dwg.	1.3 for location)

	CCS-2-1E	CCS-2-1W	CCS-3-1W	CCS-3-1E	CCS-4-18	CCS-4-2B	CCS-4-3E	CCS-4-3W	CCS-4-2V	CCS-4-1W	<u>ب</u> ج
Mercury	1.8	7.3	690	29	58	1.2	2.0	BMDL	0.14	15	ဟ
	CCS-5-2W	CCS-5-1W	CCS-5-2E	CCS-5-1E	CCS-6-1E	CCS-6-1W	CCS-6-2W	CCS-7-3¥	CCS-7-2W	CCS-7-1W	
Mercury	12.7	9.3	5.3	1.8	4.9	6.0	5.6	0.9	22	7.7	
	CCS-7-18	CCS-7-2E	CCS-7-3E	CCS-7-4E	CCS-7-5E	CCS-7-6E	CCS-8-1W	CCS-8-1E	CCS-8-2E	CCS-8-3E CC	CS-8-4E
Mercury	103	35	49	25	10.5	17	2.1	8.3	2.2	1.7	7.0

mercury was not detected in downgradient ground water nor in ground water anywhere on the site. Therefore, the Cold Creek Swamp is not considered a source of mercury contamination to ground water.

Results of the analysis for mercury of fish collected from five locations in the Cold Creek Swamp are included in Table 5-10. Mercury levels for four of the sampling locations varied from 0.59 to 3.1 mg/kg of total fish. The background fish sample contained 0.42 mg of mercury/kg of total fish. the range may exceed the FDA established action level of 0.5 mg/kg for the edible portion of fish tissue, many of the fish were forage fish and unlikely to be eaten. The game fish found at the site were generally juveniles. Some of these may make their way to the Mobile River, and with lower ambient mercury concentrations, body burdens would decline over time because of growth and depuration. Human consumption of fish contaminated from the site, but caught in the river, is considered to be very unlikely. A more detailed discussion of mercury levels in fish, how mercury levels in the swamp may affect other biota, and other potential environmental impacts is included in the Endangerment Assessment, which is being submitted concurrently with this report.

5.1.4 Ground Water

5.1.4.1 Program Description

Two new monitoring wells (NM-1 and NM-2) were installed near the LeMoyne landfill (see Figure 4-2). A report by Thompson Engineering and Testing, Inc., which is included in Appendix VII, gives details on construction of the new wells. These wells and one existing well, CCM-7, were sampled for priority pollutants, thiocarbamates, organophosphates, chlorides, carbon tetrachloride, carbon disulfide and mercury in order to characterize possible source areas of contamination.

In addition, twelve other existing wells, 0-29, 0-31, 0-32, 0-39, 0-41, 0-45, 0-49, 0-58, 0-64, 0-65, 0-73 and CCM-4

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TABLE 5-10

FISH SAMPLES - COLD CREEK and LYMOYNE SWAMPS MERCURY in Mg/Kg TOTAL FISH

	Sampling Location								
	<u>BA-1</u>	<u>BA-2</u>	<u>BA-3</u>	BA-4	<u>BA-5</u>				
Mercury	0.59	3.1	0.91	1.1	0.42				
Concentration									
(mg/kg total fish)									

NOTE: The fish sampling locations are shown on Drawing 1.3 in Appendix XVII.

(see Figure 4-2) were sampled for location-specific compounds for evidence of pond leakage.

5.1.4.2 Findings and Conclusions

The results of the priority pollutant analysis of ground-water samples from the source wells are included in Table 5-11 along with the results for area wells, which are discussed in Section 5.2.2. Table 5-12 includes the results of location-specific compound analysis for the source wells and the area wells.

The priority pollutant and location-specific compound results for wells NM-1, NM-2, and CCM-7 are essentially non detectable except for chlorides, which are in the low to moderate range of from 22.1 to 232 mg/1.

As they have been in the past, high levels of carbon disulfide and carbon tetrachloride were observed in wells 0-29 and 0-31, which are less than 100 feet downgradient of the old carbon tetrachloride and old carbon disulfide/carbon tetrachloride plant wastewater treatment ponds. All other samples from the 12 wells sampled for evidence of pond leakage were low in almost all other compounds. Therefore, except for the old carbon tetrachloride and carbon disulfide/carbon tetrachloride plant wastewater treatment ponds, none of the other ponds on site represent sources of carbon tetrachloride or carbon disulfide contamination.

Carbon tetrachloride was measured in wells 0-39, 0-41, and 0-45 at levels ranging from 0.8 to 1.5 mg/l. These three wells are in the immediate vicinity of the old carbon tetrachloride plant WWT pond mentioned above. The levels of carbon tetrachloride and carbon disulfide in the source and area wells are not considered safe for ingestion by humans or other living organisms. However, because the existing withdrawal and treatment systems preclude access to these ground waters, there is virtually no risk to humans or other organisms. The ground-water intercept system, which has been operating since

TABLE NO. 5-11

GROUNDWATER - SOURCE & AREA WELLS - PRIORITY POLLUTANTS IN mg/l (Ref. to Fig. 4-2 & 4-3 for location)

(Any priority pollutants not shown are ND or BMDL) Note: Mercury was BMDL in all samples

	SCC-CC-12	SCC-017	<u>scc-05</u>	SCC-050	SCC-014	SCC-018	SCC-016	<u>ccn-7</u>	<u>NM-1</u>	<u>NM- 2</u>	All sample numbers have "-lw" as suffix
Phenol	0.006	· ND	ND	ND	ND	ND	ND	ND	ND	ND	
Arsenic	ND	ND	BMDL	ND	ND	0.011	ND	0.03	ND	ND	
Copper	BMDL	ND	ND	0.022	ND	ND	BMDL	ND	ND	ND	
Nickel	0.016	BMDL	BMDL	BMDL	0.014	0.013	BMOL	0.024	0.033	ND	
21nc	0.063	0.036	0.17	0.052	0.11	0.056	0.1	0.12	0.058	0.022	
Chloroform	-	_	-	-	~	_	-	_	0.051	0.009	

390

Chloride

110.9

35.2

59.8

36.3 130

367

64.2

61.5

169

159

Table No. 5-12 groundwater - source & area wells - site specific compounds in mg/1

Note: Fonofos (Dyfonate), Carbophenothion (Trithion). Phosmet (Imidan) & Bensulide (Betasan) are all ND (Ref. to Fig. 4-2 & 4-3 for location)

S

	500 AE	200 AA	500-014	000 016	600.017	000-010	ECC- 33	800-33	SCC-24	SCC-25	SCC-50	CCC-51	SCC-52	SCC-78	0
	SCC-05	SCC-08	SCC-014	SCC-016	SCC-017	SCC-018	SCC-22	SCC-23	SCC-24	SCC-25	2CC-20	SCC-51	SCC-32	SCC-78	
•															\circ
EPTC (Eptam)	ND	0.001	0.001	0.006	ND	ND	ND	ND.	ND	0.001	BMDL.	ND	ND	BMDL	CIT
Butylate (Sutan)	ND	BMDL	BMDL	0.001	ND	ND	ND	ND	ND	BMDL	BMDL	ND	ND	BMDL	0
Vernolate (Vernam)	ND	ND	BMDL	BMDL.	ND	ND	ND	ND	ND	BMDL	ND	ND	ND	BMDL	
Pebulate (Tillam)	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	BMDL	ND	ND	
Molinate (Ordram)	ND	0.002	0.002	0.001	ND	ND	ND	ND	ND	BMDL	BMDL	ND	ND	BMDL.	
Cycloate (Ro-neet)	ND	BMDL	BMDL	BMDL	ND	ND	ND	ND	ND	BMDL	ND	ND	ND	BMDL	
Mercury	ND	-	ND	ND	BMDL	BMD1.	. ND	ND	ND		ND	ND	ND	ND	
Carbon Disulfide	ND	ND	BMDL	BMDL	BMDL.	BMDL	. ND	ND	0.001	. ND	ND	ND	ND	0.0002	
Carbon Tetrachloride	-	-	-	-	-	-	_	-	-	-	-	-	ND	ND	
Thiocyanate	-	-	-	_	-	0.4		-	-	ND	~	-	-	_	
Chloride	9.7	41.3	16	26.4	4.8	5.3	123.	6 37.8	3,900	29.9	45	3.4	3.7	20.9	
										•					
	CCH-4	<u>CCM-7</u>	<u> MM-1</u>	<u>NM-2</u>	CNA-14	<u>CNA-15</u>	CNA-16	<u>CNA-17</u>	CNAM-1	CNAM-2	2 CNA	<u>4-25</u> . <u>C</u> 1	NAM- 28	CNAM-30	
EPTC (Eptam)	0.001	0.017	BMDL.	BMDL	ND	ND	ND	ND	ND	0.02	1 N	D	ND	0.007	
Butylate (Sutan)	0.003	0.013	ND	ND	ND	ND	ND	ND	ND	0.00	7 N	D	ND	0.002	
Vernolate (Vernam)	0.004	0.004	ND	ND	BMDL	ND	ND	ND	ND	0.00	7 N	D	ND	0.002	
Pebulat e (Tillam)	BMDL	BMDL	ND	ND	ND	ND	ND	ND	ND	₩D	N		ND	ND	
Molinate (Ordram)	0.231	0.024	BMDL	ND	ND	ND	ND	ND	ND	0.02		D	ND	0.026	
Cycloate (Ro-neet)	0.004	0.004	ND	ND	ND	ND	ND	ND	ND	0.00	6 N	D -	ND	0.001	
Mercury	ND	ND	ND	ND	BMDL	ND	ND	ND	ND	ND	N	_	ND	ИD	
Carbon Disulfide	0.0002	0.0005	ND	ND	0.002	0.005	ND	0.045	0.000		0	.001	ND	0.0006	
Carbon Tetrachloride	ND	ND	0.25	ND	-	-	ND	-	0.017			-	0.007	-	
Chlorid e	233	82.6	232	22.1	96.6	11.3	3	4.7	29.4	260	6	. 2	21.2	53.7	
	0-29	0-31	<u>o-32</u>	0-39	0-41	0-45	0-49	0-58	0-64	0-65	0-73				
EPTC (Eptam)	0.005	0.005	0.01	0.006	0.005	0.013	0.002	BMDL	0.004	ND	0.005				
Butylate (Sutan)	0.002	0.002	0.003	0.002	0.002	0.01	BMDL	ND	0.001	ND	0.001				
Vernolate (Vernam)	0.006	0.003	0.002	0.002	0.003	0.009	0.001	ND	0.002	ND	0.002				
Pebulate (Tillam)	0.002	BMDL	ND	BMDL	BMDL	0.001	ND	ND	BMDL	ND	BMDL				
Molinate (Ordram)	0.009	0.008	0.1	0.009	0.007	0.017	0.004	0.003	0.01	BMDL	0.007				
Cycloate(Ro-neet)	0.002	0.002	0.003	0.002	0.002	0.007	BMDL	BMDL	0.002	ND	0.002				
Mercury	ND	ND	ND	ND	ND	ND	ND	BMDL	ND	ND	ND				
Carbon Disulfide	22.2	55.6	ND	ND	ND	0.368	ND	ND	0.002	0.002	ND				
Carbon Tetrachloride	298	42.4	0.001	0.854	0.812	1.52	ND	0.001	0.001	0.001	0.002				
carpon retraction to	300	110.0	26.2	50.0		120	267				150				

TABLE 5-12 (cont'd)

	SCC-079	SCC-080	SCCLM-5	SCCLM-6	SCCLM-7	SCCLM-10	SCC-CC-12	CNA-3	<u>CNA-6</u>	CNA-7	CNA-9	<u>CNA-11</u>	CNW-13	
EPTC (Eptam)	ND	ND	0.057	0.005	ND	ND	ND	ND	ND	ND	0.002	ND	ND	
Butylate (Sutan)	ND	ND	ND	0.001	ND	ND	ND	ND	ND	ND	ND	ND	ND	
Vernolate (Vernam)	ND	ND	ND	0.002	ND	ND	ND	ND	ND	ND	ND	ND	ND	S
Pebulate (Tillam)	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	ND	· ND	ND	0,
Molinate (Ordram)	ND	ND	0.005	0.005	ND	ND	ND	ND	ND	ND	0.003	ND	ND	
Cycloate (Ro-neet)	ND	ND	ND	0.001	ND	ND	ND	ND	ND	ND	ND	ND	ND	
Mercury	ND	ND	ND	ND	ND	ND	BMDL	BMDL	ND	ND	ND	ND	ND	\circ
Carbon Disulfide	ND	ND	0.0008	0.232	ND	ND	ND	0.0003	0.006	ND	ND	0.0005	ND	
Carbon Tetrachloride	-	-	-	-	-	-	-		0.0009	0.0145	-	0.0007		0
Thiocyanate	6	ND	-	-	-	-	-	-	-	~	-	-	-	\circ

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1981, controls ground water in this area. Low levels of these compounds are evident at CNA wells south of the site property line probably because of migration prior to the installation of the intercept system (see Figure 5-7). The CNA system serves to effectively purge the area of any small residual quantities.

5.2 Area Characterization

5.2.1 Surface Water and Soils

5.2.1.1 Program Description

Two surface-water and two soil samples were collected off-site to determine background levels of the compounds of concern. As shown in Figure 5-5, a surface water sample was collected from, and a soil sample was collected adjacent to, two unnamed tributaries to Cold Creek. One tributary is located north of the Virginia Chemicals plant, which is north of Cold Creek, and the other is located approximately 100 feet north of the LeMoyne-CNA property line near the railroad tracks. As outlined in CDM's Work Plan, the soil and surface-water samples from the northern tributary were analyzed for priority pollutant volatile organic compounds, and the soil and surface-water samples from the southern tributary were analyzed for priority pollutant metals.

5.2.1.2 Findings and Conclusions

The results of the background surface-water and soil sample analyses are included in Appendix II and summarized in Tables 5-6 and 5-7. No priority pollutant volatile organic compounds were detected in the background soil or surface-water samples. Several heavy metals were found in the background soil sample taken near the railroad tracks. With the exception of mercury, however, all metals were found at levels within the common range for natural soils (see Appendix XVIII). These

results indicate that background mercury levels are higher than those in natural soils. Very low levels of mercury (0.0002 ppm) were found in the background surface-water sample analyzed for metals. The only other metal found was zinc, at a concentration of 0.31 ppm.

5.2.2 Ground Water

5.2.2.1 Program Description

In order to characterize the area ground water, 36 wells (one well, CNA-8, was inoperative) were sampled in addition to those mentioned in Section 5.1.4 above (see Figure 4-3). Seven of these well samples, SCC-05, SCC-014, SCC-016, SCC-017, SCC-018, SCC-050, and SCC-CC12, were analyzed for priority pollutants. The remaining well samples were sampled for location-specific compounds.

5.2.2.2 Findings and Conclusions

Table 5-11 includes the results of priority pollutant analysis of ground-water samples from area and source wells. As shown, no priority pollutants were detected except for low concentrations of heavy metals (0.01 to 0.17 mg/l).

The results of location-specific analysis, included in Table 5-12, show non-detectable to very low levels of almost all compounds. The two exceptions are 6 mg/l thiocyanate in SCC-079 and 0.23 mg/l carbon disulfide in SCC-LM6. SCC-079 is just downgradient of the closed Halby pond.

Chloride levels are low (3.4 to 123.6 mg/l) with the exception of well SCC-24, which is immediately adjacent to the salt barge unloading area at the Mobile River. Here, the chloride level of 3,900 mg/l is a highly localized condition attributed to the on-going surface handling of large volumes of rock salt.

These results clearly demonstrate that the heavy metals found in swamp soil samples are insoluble and immobile and have no effect on downgradient ground water.

Figure 5-6, carbon tetrachloride/carbon disulfide in LeMoyne source wells, and Figure 5-7, carbon tetrachloride/carbon disulfide in area wells downgradient from LeMoyne, are firm evidence of the effectiveness of Stauffer's existing interceptor well system. As concentrations of contaminants at the interceptor wells decrease (see Figure 1-12 for history since installation), the already low concentrations (less than 0.05 mg/l) of contaminants downgradient should eventually fall below detectable limits.

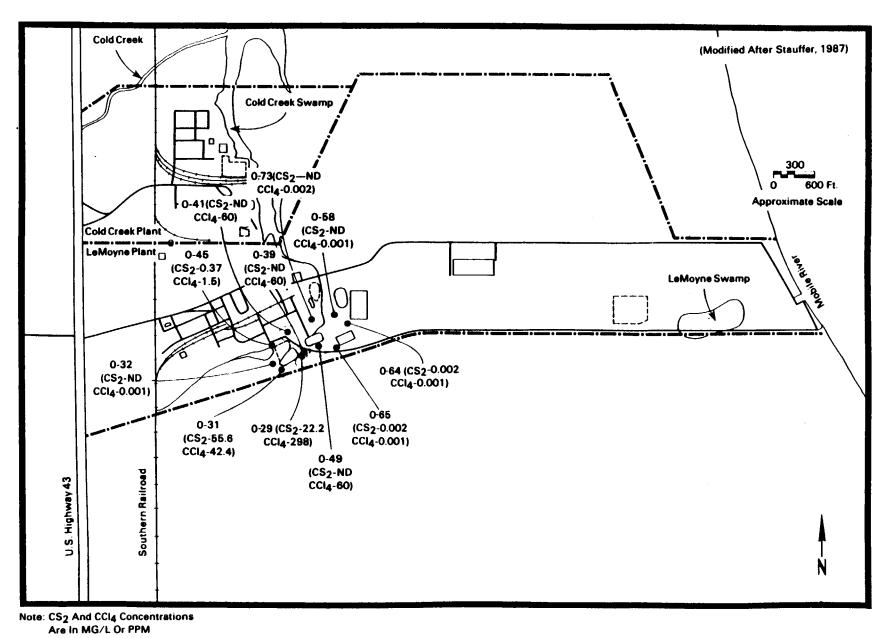
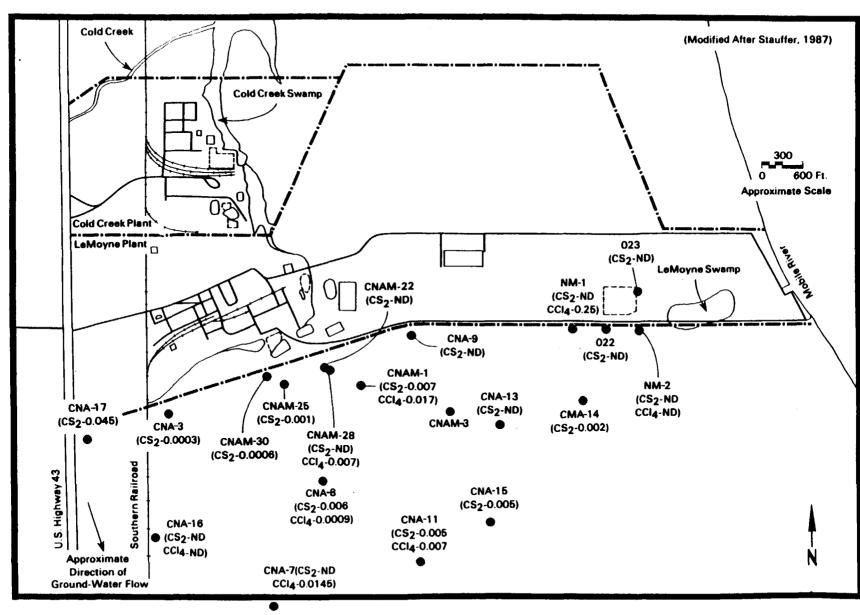


Figure 5-6 Carbon Tetrachloride and Carbon Disulfide Concentrations in LeMoyne Source Wells



Note: CS₂ And CCI₄ Concentrations Are In MG/L Or PPM

Figure 5-7 Carbon Tetrachloride and Carbon Disulfide Concentrations in Area Wells Downgradient from LeMoyne

6. PUBLIC HEALTH AND ENVIRONMENTAL CONCERNS

An Endangerment Assessment has been prepared for the Cold Creek/LeMoyne site by ERT (ERT, 1988). A summary of this work is presented in this section. The following discussion is divided into three parts: potential receptors, public health impacts, and environmental impacts. The first part identifies human and environmental (flora and fauna) receptors that are or may be exposed to site contamination, and the second and third parts discuss the impacts of site contaminants on human and environmental populations.

The purpose of an endangerment assessment is to determine if endangerment to human health or the environment exists as a result of a threatened or actual release of a hazardous substance (EPA, 1985). Usually, the endangerment assessment is the interpretive link between the remedial investigation and the feasibility study, and it is used to determine whether a remedial action is necessary and how quickly steps must be taken to protect potentially affected populations or environments.

6.1 Potential Receptors

As discussed previously in Section 1.2, the Cold Creek/LeMoyne site is located in an industrial area and is surrounded by several other large chemical production companies. Fewer than 10 residences are located within one mile of the site, and none of them are downgradient of the contaminated ground water at the site. The nearest population centers include Mt. Vernon (with a population of 1,038), which is located about 8 miles north of the site, and Creola (population of 673), which is located about 5 miles to the south (U.S. Department of Commerce, 1981).

The majority of the chemical plants as well as the local communities in the area obtain water from the water-table aquifer. As discussed earlier in Section 1.2, and shown in

Figure 1-2, the Cold Creek facility has one drinking-water well and one backup well, and the LeMoyne facility has two drinking-water wells. The CNA facility to the south has one drinking-water well and a backup well. These wells were sampled during the RI, and no contaminants were detected.

Both Cold Creek Swamp and LeMoyne Swamp represent the most important environmental receptors at the site. These swamps currently support a diverse variety of plants and animals, including the alligator, which is currently listed as a threatened species. The Mobile River, which forms the eastern boundary of the site, is also a potential environmental receptor.

6.2 Public Health Impacts

To properly assess the potential public health impacts from the Cold Creek/LeMoyne site, a characterization of the potential areas of contamination must be made. These include the nine ponds or lagoons, four of which are still active, the three closed landfills, and portions of the two swamps (Cold Creek and LeMoyne). Figure ES-1 shows the locations of these 14 potential source areas.

Based on the frequency of detection, the concentrations detected, and the toxicological properties of the contaminants which have been found at the site, the following compounds were selected as "representative" compounds. These are:

- carbon tetrachloride;
- carbon disulfide;
- cyanide;
- mercury;
- 6 thiocarbamates (including EPTC, butylate, cycloate, molinate, pebulate, and vernolate); and
- thiocyanate.

Information concerning safe drinking water levels for many of these contaminants is presented in Appendix XIX. In addition, available toxicity information for aquatic species is included in these reviews.

After identifying the potential receptors and the contaminants to which they may be exposed, it is necessary to determine the ways in which they may be exposed and the frequency and magnitude of the potential exposure. Human exposure to the contaminants identified at the Cold Creek/LeMoyne site can potentially occur directly through air, water or solid media (soils, sediments or sludges) or indirectly through the food chain; however, the most likely exposure pathways are as follows:

- Incidental ingestion of contaminated swamp sediments;
- dermal contact with contaminated swamp sediments;
- ingestion of contaminated fish; and,
- ingestion of contaminated ground water.

Exposure to contaminated swamp sediments and via ingestion of fish is probably infrequent because the site is located in an industrial area and people generally do not spend much time in wetland areas unless they are bird watchers or are on other kinds of nature walks. In addition, shoes and other articles of clothing will help to protect anyone from direct contact with the swamp sediments. Ingestion of contaminated fish is also expected to be infrequent because it is unlikely that anyone fishes in Cold Creek Swamp, given its proximity to industrial property, its restricted access, and the small size of the fish in the swamp.

No current risk appears to exist from exposure to contaminated ground water at the site, because none of the contaminants have been detected in any of the drinking-water wells in the immediate vicinity of the site. It is highly unlikely that future exposure to contaminated ground water would occur, because a permit is required for the installation

of a potable water well from the state of Alabama, and it is unlikely that a permit would be granted given the documented ground-water contamination at the site.

One way of assessing the risks is to quantify the potential for adverse health effects due to site-related chemical exposure. As described in more detail in the Endangerment Assessment (ERT, 1988), noncarcinogenic effects are assumed to have a threshold dose below which an adverse health effect will not occur. In order to determine the risks, the estimated intakes of indicator chemicals that may lead to noncarcinogenic effects are compared to acceptable daily intakes. This ratio is called the hazard index (HI). An HI of less than one (unity) results when estimated intake is less than acceptable intake, which indicates that levels of intake are lower than those expected to produce toxic effects.

As shown in Table 6-1, quantitative risk estimates were conducted for both adult workers and teenagers that could be exposed to contaminants at the Cold Creek/LeMoyne site. conservative assumptions, risks were calculated for hypothetical individuals who may come in contact with hazardous compounds via exposure to contaminated swamp sediments and ingestion of contaminated fish. Two scenarios were developed for each exposure route. The worst-case scenario assumes that an individual is exposed to the maximum concentration of the compound measured in that media (i.e., sediments or fish), and the realistic case uses average concentrations found at the site. For teenagers, it was assumed that they would be exposed to contaminated swamp sediments 6 times per year for a period of 5 years, and for adult workers it was assumed that they be exposed 12 times per year (once a month) for 30 years. determining risk from ingestion of contaminated fish, it was assumed that an individual would consume 6.5 grams of fish per day for a period of 30 years.

Results of the quantitative risk assessment for noncarcinogens show that individuals exposed to contaminants at the Cold Creek/LeMoyne site are not at risk, even if they are exposed to maximum concentrations. Even when the risks for

TABLE 6-1 SUMBLY OF QUANTIALINE RISK ESTIMATES FOR EXPOSURE TO CONTROLLERATS AT THE COLD CREEV LEMONES SITE, MOBILE COUNTY, REASONA.

Z.	Ordaic Bally Intersectable	Ingestion of D	Part Sediments Paulistic Com		Sediends Bedistic Case	Fish Co.	Concession Smallistic Case	Total Ha Horst Case	Total Hazard Index Case Malistic Case	
Carbon Tetrachloride Carbon Disulfide Cyanide Marcury		0.00E+00 6.00E+00 6.38E+03		9,00E + 80,00E + 80,0	444 333 333 333 333	7 74	2 3 37 2	0.00E-00 0.90E-00 6.80E-00	0.00£ 60 2.00£ 60 2.56 60	
Waty at a Waty and	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	4.7442元 経済経済経済 かかかかかれ			1220年128年128年128年128年128年128年128年128年128年128			4.7.7. 连续编辑编辑 企物中介令	246444 2572447 246444	
Totals		6. ME-03	A. 625-04	2. 2KE-07	4.376-48	6. 15E-42	3-33-7	4.66.4	2.506-02	
Orronic Mazard Index (noncarcinogens) for Tounagers of the Cold Dreat/Lathyne Site Incidental Ingestion of Bernal Recent Provided Incidents of Bernal Dreat Colone Selection of Semple Compound Intakeing/Ng/6697 (Mart Colon Mart Colone Servic Colone Service Colone Se	(noncarcinopana) for Commic Baily Intabuta/10/60/)	Total Paris	the Cold Death		Morrisis Missis Palistic Cas	Total Herst Case In	Nazard Indes Realistic Case			
Carbon Tetrachlorida Carbon Bisulfide Cyanide Marcury	10-388 10-388 11	4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4	0.0K:00	0.00€+00	G. 80£.00	6. 60E+60 6. 10E-64	A 00E 40			
Betylete Petylete Vernolete Petelete Polioste Optioste Piccyssate	25 25 25 25 25 25 25 25 25 25 25 25 25 2		まる は は は は に に に に に に に に に に に に に	によるによるのでは、	されなかから 体を記録法院 お言うである。	199119 高級政権発展 会会会会会	4::34: 石坑谷流道底 かわせかか			
Total		A. 11E-04	5.38-66	4.275-08	L 265-09	£.11E-04	1.3R-45			

each exposure route and each compound are summed, the total HI is still less than unity. However, care must be taken when interpreting summed HI's, because adding them assumes that their toxicological effects are additive, which may not be true.

The underlying assumption for assessing the risks from suspect or known human carcinogens is that there is no threshold for an adverse health effect. Carcinogenic risk is quantified by multiplying an EPA-derived cancer potency factor by the estimated intake (dose) to calculate cancer risk due to each site-related exposure. Carbon tetrachloride is the only indicator compound at the Cold Creek/LeMoyne site which is considered by the EPA to be a suspect human carcinogen. This compound was only detected in the ground water and in subsurface soil samples from beneath the wastewater treatment ponds. As discussed above, there is no current risk from exposure to contaminated ground water, and therefore, there is no current risk from carbon tetrachloride at this site.

6.3 Environmental Impacts

The most environmentally sensitive portion of the site is believed to be the Cold Creek Swamp. Sediment concentrations and fish body burdens of mercury are both elevated in this area. Potential exposure routes for aquatic and semi-aquatic organisms include water, sediment and food chain pathways, with the food chain exposure route predominating at higher trophic levels. The U.S. Fish and Wildlife Service (USFWS) considers the swamp to be a preferred habitat for the threatened American alligator (Alligator mississippiensis), and the potentially endangered Alabama Red-Bellied turtle (Pseudemys alabamensis).

Current data are not adequate to assess the exposed biota populations or estimate exposure via all pathways and thus to estimate potential risks to these populations. With the data at hand, the potential for adverse affects to sensitive birds and mammals, if these inhabit the wetland, cannot be ruled out.

7. DETERMINATION OF APPLICABLE REMEDIAL ACTION ALTERNATIVES

7.1 Description of Remedial Alternatives Screening Process

The principal objectives for any Remedial Investigation (RI) are to provide to the greatest extent possible a complete environmental characterization of the site and to determine the extent and type of environmental contamination which has occurred as a result of previous site activities. Information obtained during the RI is used to determine whether environmental contamination is confined to the site/source area, or whether an off-site release of contamination has occurred. The acquired information is used in the Feasibility Study (FS) process, the purpose of which is to identify these remedial technologies or combinations of remedial technologies (i.e., remedial alternatives) which are considered most cost-effective, technically feasible and reliable, and which adequately protect (or mitigate damage to) public health, welfare and the environment.

The Feasibility Study for the Cold Creek/LeMoyne site will be structured to follow as closely as possible the guidelines established by the EPA as documented in <u>Guidance on Feasibility Studies Under CERCLA</u> (June, 1985), and the new requirements of the Superfund Amendments and Reauthorization Act of 1986. A brief discussion of the Feasibility Study process follows.

The Feasibility Study will involve three steps: the development of remedial response objectives for the site; identification and initial screening of applicable remedial technologies and development of alternatives; and screening of remedial technologies and/or alternatives ultimately leading to recommendation of a proposed remediation program which is considered best suited for the site.

The remedial response objectives will be developed based on the findings of the RI and the site risk assessment. This

step generally involves the identification of those areas or portions of the site which require remediation and a determination of the extent of remediation to be attained.

The legally applicable or relevent and appropriate requirements (ARARS) of federal and state laws will represent institutional objectives for clean up of the site and form the basis for remediation goals. If there are no ARARS associated with a specific compound or medium, a health risk evaluation will provide the basis for remediation goals.

Based on the remediation goals developed for the Cold Creek/LeMoyne site, potential response actions will be identified for each area that may require remediation. The technology options associated with each general response action will be specified. In accordance with SARA, these options will include technologies that would provide a permanent solution and alternative treatment or resource recovery technologies. The identified technologies will be screened so that those technologies that are not applicable to site conditions or not technically feasible will be eliminated from further consideration.

Those technologies remaining after the preliminary screening will be used to develop remedial alternatives that are effective and technically feasible for the site.

In accordance with the Interim Guidance on Superfund Selection of Remedy (U.S. EPA, 1986), treatment alternatives will be developed that range from "an alternative that, to the degree possible, would eliminate the need for long term management (including monitoring) at the site to alternatives involving treatment that would reduce toxicity, mobility, or volume as their principal element. In addition to the range of treatment alternatives, a containment option involving little or no treatment and a no action alternative..." will be developed.

The potential alternatives will be screened in two phases to determine which should be retained for a detailed

evaluation. First, environmental and public health effects of each alternative will be evaluated. Those alternatives that would have adverse environmental or public health impacts will be eliminated from further consideration. The second phase will consist of an order of magnitude cost screening. The alternatives remaining after the initial screening will be grouped according to the level of public health and environmental protection and reliability associated with each. In accordance with SARA, a cost comparison will not be conducted between treatment and nontreatment alternatives. Those alternatives within each group which provide a commensurate level of protection, yet have costs an order of magnitude greater than the other alternatives in the same group, will be eliminated from further consideration.

Those alternatives remaining after the initial screening will undergo a detailed evaluation based on both non-cost and cost criteria. The non-cost evaluation will begin with an assessment of each alternative's feasibility and overall effectiveness. Other non-cost criteria will include: the criteria for selection of remedy in SARA § 121 (d); environmental and public health impacts; and the interplay between remedial technologies applied to different portions of the site.

The cost evaluation will involve a comparative analysis of the alternatives on the basis of present-worth costs for both capital expenditures, and operation and maintenance expenditures. In accordance with the current EPA costing guidance, the costs will be estimated within an accuracy range of -30 to +50 percent. While this range may appear large compared to typical construction estimates, there are many uncertainties and complexities associated with work involving the cleanup of hazardous wastes. As described for the initial screening evaluations, the cost comparison will be conducted between treatment alternatives only.

To summarize the data compiled during the detailed evaluation, pertinent information involving both cost and

non-cost criteria will be presented in tabular form. This will facilitate the overall evaluation and selection process by presenting the key differences between each of the alternatives in a clear and concise manner.

The Feasibility Study will conclude with recommendations for remedial action based upon the results of the detailed analysis of alternatives.

7.2 Remedial Alternatives Selected for Consideration

The Remedial Investigation of the Cold Creek/LeMoyne site has identified the areas on site which are considered as potential contaminant source areas requiring some form of remediation. These areas include the former Halby pond (Area I), the old carbon tetrachloride (CTC) plant wastewater treatment pond (Area II), the old carbon disulfide plant wastewater treatment pond (Area III), and the Cold Creek old neutralization pond (Area IV). Each of these areas presents a specific set of circumstances with regard to the type and extent of environmental contamination and the requirements for suitable remedial action.

The general response measures identified below are based on site conditions, information obtained during the Remedial Investigation, and the risk assessment. The following potential general response measures are identified:

No Action - A no-action response provides a baseline assessment for comparison with other alternatives that contain greater levels of response. An alternative involving no action may be considered appropriate when an alternative response action may cause a greater environmental or health danger than the no-action alternative itself. An evaluation of the no-action response is required as part of the feasibility study process. In some cases, the

no-action response may include some limited form of action, such as periodic sampling and analysis.

- Containment Containment measures include various technologies which contain and/or isolate the constituents of concern on site. These measures provide isolation and prevent direct exposure with or migration of contaminated media without disturbing or removing the waste from the site. Containment measures generally consist of measures which cover, seal, chemically stabilize or provide an effective barrier against specific areas of contamination.
- Withdrawal/Collection Collection of contaminated ground-water may be achieved via withdrawal techniques such as pumping or gravity drainage.
 Water treatment may be required in conjunction with withdrawal/collection actions to reduce contaminant levels in the extracted liquid, thereby allowing its discharge. Treatment techniques include chemical, biological, or physical removal systems. The existing extraction and treatment system will be evaluated and modifications proposed if appropriate.
- Removal Removal measures may be undertaken to remove contaminated media such as soils from the site. Implementation of a removal measure requires proper treatment and/or disposal of the soil, either on site or at an approved off-site waste disposal facility. Replacement of the media to restore the site is necessary with a removal action. Removal measures may be used to effectively remove the contaminated media from the site; however, a significant short term exposure hazard during remedial action implementation will occur.

Treatment - Various in-situ, or in-place, treatment methods will be considered which are effective in detoxifying the waste material without having to excavate it or remove it from the site.

Each of the general response measures will be considered for each of the individual areas of concern during the Feasibility Study. Technologies applicable to each response measure will be identified and screened as previously discussed. Remedial alternatives for the entire site will be developed from these technologies and evaluated. The remedial alternative recommended at the conclusion of the Feasibility Study will contain components which address each individual area of concern.

8. SUMMARY

As discussed in Section 1 of this report, the objectives of this Remedial Investigation were to characterize the nature and extent of contamination at the site and to identify contamination sources, pathways of migration, and potential adverse environmental impacts. In order to accomplish these objectives, a substantial amount of data pertaining to the site-area geology, hydrogeology, surface-water drainage as well as soil, surface-water, and ground-water quality has been collected and evaluated. The findings of the RI are summarized as follows.

Fourteen potential contaminant source areas on the site were investigated during the RI. These potential sources were initially identified in CDM's Work Plan and include the nine ponds or lagoons, four of which are active, three closed landfills, and the Cold Creek and LeMoyne Swamps (see Figure ES-1 for locations). A summary of the information presented in Section 5, which is a description of the sampling and analysis program, is included in Table 8-1. This table shows the major contaminants found in soil samples collected from beneath each potential source, the range of concentrations (minimum to maximum) of each major contaminant, and the levels of any contaminants in the nearest downgradient well samples. Based on the results presented in Section 5 and in Table 8-1, a summary of the findings and conclusions regarding the nature and extent of contamination at each of the potential source areas is presented as follows.

LeMoyne Landfill - Only background levels of heavy metals were found in any of the soil samples collected form the 8 borings installed beneath the LeMoyne landfill. The median level of the hazardous contaminant found (mercury) was less than 0.1 mg/kg, which is within the common range for natural soils and less than the level of mercury found in the

 ${\tt TABLE 8-1}$ POTENTIAL AREAS OF SOURCE CONTAMINATION AND DOWN GRADIENT GROUND WATER DATA

	Area	Potential Major Contaminant(s)*	Range mg/kg	Contaminants in Nearest Down Gradient Ground Water mg/l	Well Number (
l.	LeMoyne landfill	Mercury	ND to 1.9	0.25 Carbon Tetrachloride	NM-1
2.	Cold Creek (CC) North Landfill	Molinate Mercury	BMDL to 1.5 BMDL to 0.6	0.024 Molinate	сс и -7
3.	CC South Landfill	Molinate	ND to 0.1	0.23 Molinate	CCM-4
4.	CC LeCreek Pond	Thiocarbamates	ND to 0.2	0.23 Molinate	CCM- 4
5.	LeMoyne LeCreek Pond	Chloride	ND to 240	367 Chloride	0~49
6.	New Carbon Tetrachloride (CTC) Plant WWT Pond	None	-	169 Chloride	0-65
7.	CC Old Neutralization Pond	Thiocarbamates	ND to 25	0.23 Molinate	CCM- 4
8.	Halby Treatment Pond	Thiocyanate	ND to 2,480	6 Thiocyanate	0-79
9.	Old CTC Plant WWT Pond	Carbon Disulfide (CS2) Carbon Tetrachloride	ND to 0.016 ND to 0.001	298 Carbon Tetrachloride 55 Carbon Disulfide	0-29/0-31
10	. Old Carbon Disulfide Plant WWT Pond	CS2 CTC	ND to 0.008 ND to 0.685	298 Carbon Tetrachloride 55 Carbon Disulfide	0-29/0-31
11	. Old Chlorine Plant WWT Pond	Mercury	1.4 to 24	159 C1	0-73
12	. LeMoyne Acid Plant WWT Pond	Iron	2800 to 25,000	0.0003 Carbon Disulfide	CNA-3
13	. Cold Creek Swamp	Mercury	BMDL to 690	None	SCC-050
14	. LeMoyne Swamp	Mercury	0.11 to 0.28	0.002 Carbon Disulfide	CNA - 14

background soil sample. The corresponding downgradient well, NM-1, showed no mercury.

Carbon tetrachloride (CTC) was not found in the soil borings installed beneath the LeMoyne landfill, but it is seen at 0.25 mg/l in the ground water at well NM-1. CTC is known to be a localized ground-water contaminant that is extensively controlled in this area by the high withdrawal rate wells operated by Courtaulds North America (CNA) for industrial water supply. CNA usage dominates the local ground-water flow pattern, producing a gradient generally to the southwest with respect to well NM-1. To the south of NM-1, CNA withdrawl wells show levels of CTC ranging from non detectable to 0.0007 mg/l. If in the future, the CNA system were to be curtailed, modified or abandoned, there could be a need to install an additional interceptor system well to assure capture of the low levels of CTC in this area. However, in the opinion of Stauffer's staff geologist who has evaluated this data, the direction of ground-water flow in the absence of CNA pumping would tend to revert to the normal area flow pattern, which is toward the Mobile River.

The Safe Drinking Water Committee (National Academy of Science, <u>Drinking Water and Health</u>) determined that the estimates of lifetime risk for humans ingesting 1 X 10⁻³ mg carbon tetrachloride/l of water are 4.5-5.4 X 10⁻⁸ with 95% upper confidence limits of 1.0-1.1 X 10⁻⁷. Even though the water in well NM-l contained levels of carbon tetrachloride which exceeded 1 X 10⁻³ mg/l, there should be no risk to humans because this water is not anticipated to be ingested by humans or exposed to wildlife. In

fact, carbon tetrachloride levels were below 1 X $10^{-3}\ \text{mg/l}$ by the time the water reached well CNA-11.

- 2. Cold Creek North Landfill Most of the site-specific compounds analyzed for in the soil samples collected from this landfill were less than 1 mg/kg (see Table 5-2 in Section 5). The median level of molinate, the compound most frequently detected, was 0.1 mg/kg. Molinate was found in the downgradient well, CCM-7, at a level of 0.024 mg/l. This water is not available for ingestion by humans or other living organisms. Moreover, the mean molinate levels for all wells are below the proposed safe drinking water level of 0.2 mg/l (see Appendix XIX). Based on these low findings, this area is not considered a significant source of contamination.
- 3. Cold Creek South Landfill As with the North landfill, no priority pollutants were found in the South landfill other than heavy metals within the common range for natural soils. Moreover, no site-specific compounds were found in soil borings under this area. For these reasons, this area is not considered a contaminant source.
- 4. Cold Creek LeCreek Pond No priority pollutants except for heavy metals within the common range for natural soils were found. Moreover, except for two samples which contained thiocarbamates at or slightly above their limits of detection, no site-specific compounds were found in the soil samples collected from beneath the Cold Creek LeCreek pond. Based on these findings, this area is not considered a contaminant source.
- LeMoyne LeCreek Pond Of the 10 soil samples analyzed for site-specific compounds, one showed a

level of 240 mg/kg chloride, which is within the common range for natural soils. Two other soil samples contained low levels of chloride, and the remaining seven samples were non detectable for chloride. Because the heavy metals and chloride concentrations found beneath the LeMoyne LeCreek pond were within the common range for natural soils, this area is not considered a contaminant source.

- 6. New Carbon Tetrachloride Plant WWT Pond No site-specific compounds were detected, and no priority pollutants other than heavy metals within the average range for natural soils were found beneath this area. Therefore, the new carbon tetrachloride plant WWT pond is not considered a source of contamination.
- 7. Cold Creek Old Neutralization Pond - Seven of the 10 soil samples analyzed for site-specific compounds contained thiocarbamates. The median level for each of the six thiocarbamates analyzed for varied from non detectable to 0.25 mg/kg. However, the closest downgradient well, CCM-4, showed thiocarbamate levels ranging from below the mean detection limit to 0.23 mg/l. Whereas the data could indicate that this area is a probable source of thiocarbamate contamination, only one thiocarbamate, molinate, was found in the ground water at 0.23 mg/l. The other thiocarbamates were present at 0.004 mg/l or less. Furthermore, at wells further downgradient, molinate is non detectable or present at detection-limit levels. Thus, although the Cold Creek old neutralization pond may be a source of thiocarbamate contamination, ultimately, the existing ground-water intercept system is capturing these trace amounts.

The molinate level of 0.23 mg/l exceeds the proposed safe drinking water level of 0.2 mg/l. However, water containing 0.23 mg/l molinate is not anticipated to be ingested by humans or other living organisms. In fact, as mentioned above, molinate levels were much lower (0.01 mg/l to below detection limits) in the next most immediate downgradient wells.

- 8. Halby Treatment Pond - Soil sampling at the HCC site was made within the area formerly occupied by a small pond. One boring, HTP-1, encountered apparent waste material and was discontinued at that point. Analysis of this waste showed high levels of copper (440 mg/kg), zinc (1170 mg/kg) and cyanide (240 mg/kg). A second boring within the pond and an angled bore beneath the pond were both at background levels of copper and zinc and had no detectable levels of cyanide. Site-specific sampling detected elevated levels of thiocyanate (see Table 5-5) in all three borings. Sodium thiocyanate has a low acute toxicity (LD₅₀ for rats is 764 mg/kg's Sax, 1979). A low level of thiocyanate (6 mg/l) was detected in downgradient ground water at well 0-79. No other site-specific compounds were found in the ground water. Based on these findings, this area is presumed to be the probable source of thiocyanate found in ground water.
- 9. Old Carbon Tetrachloride Plant WWT Pond Although only low levels of carbon tetrachloride and carbon disulfide were found in soil samples under this pond, downgradient wells, 0-29 and 0-31, showed high levels of these two contaminants. These levels of carbon tetrachloride (298 mg/l) and carbon disulfide (55 mg/l) are not considered safe for ingestion by humans or other living organisms. However, because the

existing withdrawal and treatment systems preclude access to this ground water, there is virtually no risk to humans or other organisms. The ground-water intercept system, which has been operating since 1981, controls ground-water flow in this area. Low levels of these compounds are evident at CNA wells south of the site property line probably because of migration prior to the installation of the intercept system (see Figure 5-7). The CNA system serves to effectively purge the area of any small residual quantities.

- 10. Old Carbon Disulfide Plant WWT Pond This pond is similar to, and located adjacent to, the old carbon tetrachloride WWT pond. The discussion presented above in (9) regarding the old carbon tetrachloride WWT pond would apply to this pond as well.
- 11. Old Chlorine Plant WWT Pond Two soil samples from below the pond area showed moderate levels of mercury (1.4 and 24 mg/kg) and levels of chloride up to 1200 mg/kg. At the downgradient well, 0-73, mercury is non detectable and chloride is within normal limits at 159 mg/l. On this basis, the old chlorine plant WWT pond does not appear to represent a source of contamination to ground water, which is the only potential route of exposure.
- 12. LeMoyne Acid Plant WWT Pond Soil samples from beneath the pond contain only high levels of iron (up to 24,500 mg/kg). The only finding in the nearest downgradient well, CNA-3, was 0.0003 mg/l carbon disulfide (CS₂). The acid plant pond has no relationship to CS₂, and hence it would not appear to be a source of contamination.

13. Cold Creek Swamp - In addition to the high levels of mercury found in the swamp soil, low levels of thiocarbamates were also found (median level was 0.25 mg/kg). As discussed previously, the mercury present is probably in the sulfide form and is therefore immobile and insoluble. Analysis of downgradient ground water at well SCC-050 did not detect either mercury or thiocarbamates.

Furthermore, no mercury was found in ground water anywhere on the entire site. It is concluded, therefore, that the Cold Creek Swamp is not a source of contamination to ground water.

Fish samples were collected at five swamp locations. Because of the very small individual fish size, an aggregate sample was taken, and analysis was performed on a homogenate i.e., whole fish. range of mercury found was 0.59 mg/kg to 3.1 mg/kg for four locations, while a fifth point (BA-5) was considered a background sample and showed 0.42 mg/kg. Whereas the range may exceed the FDA established action level of 0.5 mg/kg for the edible portion of fish tissue, many of the fish were forage fish and unlikely to be eaten. The game fish found at the site were generally juveniles. Some of these may make their way to the Mobile River, and with lower ambient mercury concentrations, body burdens would decline over time because of growth and depuration. Human consumption of fish contaminated from the site but caught in the river is considered to be very unlikely.

14. LeMoyne Swamp - Soil/sediment samples were comparable to background levels for all heavy metals. The closest downgradient well, CNA-14, showed 0.002 mg/l of CS₂, but this may be attributed to contaminants present prior to closure of the LeMoyne landfill

3 10 00177 which is also upgradient to the well. Because the compounds detected in soil from beneath the LeMoyne landfill where all within the average range for natural soils, this area is not considered a contaminant source.

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APPENDIX I LIST OF PROPOSED SAMPLES AND LIST OF SAMPLES ACTUALLY COLLECTED

-LE 4-SAME LES - PENSE UNE ANALYTICAL PARAMETERS 10 00184 SAMPLE NO. AS SHOWN PIGLINES NO 4-2A NOTES: 20 CHEEL PARS :51-11 t to "Same or -53.25 ::U.B 1.1 1:11 -5.2.16 1 1 1 1.1 :3.1.1 ī, *** 1. 12 -ಚಿ--ಬ 33.4.I 111 ----~~.**~** 1 T₁ はよこに Т 1 (53-1-2[1 Τ 73-1-b -1.FI Til (13-4-12) 21-2 Τ ī 1 111 23.1.16 1. :3.7.8 13-7-X 23-7-4 111 : 11 :5-7-95 $\overline{1}$ 7.1 1:1 27.7.4 23-7-10 1:1 (0-1-2) 111 Ī -----III 3-4-16 5-4-X Tij īī TT :3.b.X 3-4-4 111 3-4- M TAI 131 OF STREET ES 84.1 M-5 44.1 1 Τ -14.5 1 | | 11| 111 1:1-1 -١١٠ المعط 1-41-2 1 1: \$0-11-10 ī 9-11-20 111 STANKS 20 LOUIS SALE 1:1:1 Ft. COPESITE 21-21 $\overline{1:1:1}$ 3.23 1:1: 11.3 1 -1 3-7 1 11 OF SHEET 7:45

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2	Area	8105	2		SOC-017		5/5/86	PD	Cl			CS2		Th	OP_	\circ
3	Area	8105	3		SOC-05		5/5/86		C1			CS2_		<u></u>	OP	
4	Area	8105	4		SOC-050		5/5/86		C)			CS2		Th	OP_	
5	Area	8106	5	Water	SOC-014		5/5/86		<u>Cl</u>			<u>CS2</u>		<u>Th</u>	OP_	. O
<u>6</u>	Area	8105			SOC-018		5/5/86	PP	<u>C1</u>	SCN		<u> </u>		<u></u>	<u> </u>	, C),
7	Area	8105			SCC-016		5/5/86		CJ CJ			<u>CS2</u>		<u>Th</u>	OP_	. —
8	Ponds	8105			<u>00H-7-1</u>	4-0	5/5/86	<u>DD</u>	<u>C1</u>			CS2	CTC	_ <u>Th</u>	OP _	- 🛇
9	Ponds	8105	9		004-7-1	dup 8 Spike 8	5/5/86 5/5/86	DIO	Cl Cl			CS2 CS2	CTC	<u>Th</u>	OP OP	· O
10	Ponds Ponds	8105	10	Water Water	00H-7-1	Blank	5/5/86	<u>DIO</u>	Cl	SCN		CS2	CTC	ThTh	OP	
11	Ponds Ponds	8105 8105	12	Water	CAP I-1	Blanke at		DID	C1	SCN		CS2	CTC	Th	OP.	
12 13	Ponds Ponds	8105	13	Water	M-1	DIMBA SI	5/5/86	DED	C1	<u> </u>		CS2	CTC	Th	OP OP	
14	Ponds	8105	14	Water	NH-2		5/5/86)DD	Cl			C92	CTC	Th	OP.	
15	Ponds	8105	15	Water	NCTC		5/5/86	1010			1			111		
16	Ponds	B105	16	Water	ILP		5/5/86	1010								
17 ·	Ponds	8105	17	Water	OCLP		5/5/86	pp								
18	Энавр	8105	18	Soil	SW-01-1	<u> </u>		pp metals								
19	Swamp	8105	19	Water	SW-11-1		5/5/86	DD VOS								
20	SHAMP	8105	20	Soil	SW-11-1		5/5/86	DD VOG								
21	SHEARD	8105	21	Water	SW-01-1	7	5/5/86	pp metals								
22 Batch #2	SHAMO	8105	22	Soil	LS-48		5/15/86	000	Cl					Th		
23	SHEAD	8105	23	Soil	LS-15		5/15/86	DD)	Cl				<u> </u>	Th	<u> </u>	
24	Знашр	8105	24	Soil	13-28		5/15/86		C)					Th		
25	SHEARING	8105	25	Soil	18-38		5/15/86		C1_					<u></u>		
26	SHAMO	8105	26	Soil	OCS-15		5/15/86		<u>cı</u>					Th_		
27	Sнашр	8105	27	Soil	003-19	<u>okuo 26</u>	5/15/86		<u></u>					<u>_Th</u>		
28	Энажо	8105	28	Soil	003-25		5/15/86	DD	C1					Th		
29	SHEARD	8105	29	Soil	003-2-1		5/15/86				He					
30	Sналер	8105	30	Soil	003-2-1	ł	5/15/86				<u>H4</u>					
31	Знавр	8105	31	Soil	<u> </u>		5/15/86		<u>cı</u>					Th		
32	SHALIND	8105	32	Soil	003-3-1		5/15/86				He					
33		8105	33	Soil	CC9-3-1		5/15/86				<u> </u>					
34	SHAMP	8105	34	Soil	~ ·	<u>blank</u>	5/15/86		Cl		11			Th		
35	SHAMP	8105	35	Soil	003-4-1		5/15/86				Her :					
36 37	SHEARD SHEARD	8105 8105	36 37	Soil Soil	OCS-4-2 OCS-4-3		5/15/86 5/15/86		_		Hg Hg					
38 Batch #3		8105	38	Soil	003-4-3		5/16/86				Ha .					
39 pau ch 73	Swamp Виамер	8105	39	Soil	003-4-2		5/16/86			_ 	Ha .					
40	Sмелир	8105	40	Soil	003-4-1		5/16/86				He					
41	Swamp Swamp	8105	41	Soil	OCS-5-2		5/16/86				He					
42	Swamp	8105	42	Soil	003-5-1		5/16/86				Hg					
43	SHAMD	8105	43	Soil	OCS-5-2		5/16/86				ila .					
44	SWAMP	8105	44	Soil	OCS-5-1		5/16/86				He					
45	Swamp	8105	45	Soil	CCS-6-1		5/16/86				Hg					
46	Swamp	8105	46	Soil	OCS-6-1		5/16/86				ila					
47	Swamp	8105	47	Soil	OCS-6-1		5/16/86				Hg					
48	Swamp	8105	48	Soil	OCS-6-2		5/16/86				ilg					
49	Swamp	8105	49	Soil	OCS-7-3	W	5/16/86				He					
		8105	50	Soil	OCS-7-2		5/16/86				Ha					

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51			8105	51	Soil	CCS-7-1W		5/16/		 		<u> </u>				
52			8105	52	Soil	<u>008-7-18</u>		5/16/				Ha				
53		Swamp	8105	53	Soil	008-7-21		5/16/				He				
54 55 i	Badah A	Swamp	8105	54 55	Soil	OCS-7-3E	Blank	5/16/ 5/20/				Ha Ha				
56	Batch 4	<u> Знамо</u> Sнамо	8105 8105	56	Soil Soil	CCS-7-48		5/20/				ilg				
57		<u>Знажр</u> Ѕнажр	8105	57	Soil	003-7-58		5/20/				Hg			•	
58		SHARD	8105	58	Soil	OCS-7-6		5/20/			•	He				
59		SHBIRD	8105	59	Soil	CC8-8-1W		5/20				He				
60		SHAMO	8105	60	Soil	CC3-8-1E		5/20				He				
61		SHAMO	8105	61	Soil	CCS-8-21		5/20				Har				
62		SHAMP	8105	62	Soil	CCS-8-3E		5/20				Hg				
63		Sнам р	8105	63	Soil	OCS-8-41		5/20	/86			He				
64	Batch 5	Area	8105	64	Water	SOC-022		5/21	/86	Cl		Ha	CS2_		Th	
65		Area	8105	65	Water	SCC-023		5/21/		C1		He	CS2		Th	
66		Area	8105	66	Water	SCC-051		5/21	/86	C1		Hg_	CS2_		Th	
<u>67 ·</u>		Area	8105	67	Water	SOC-24	 	5/21		Cl			CS2_		<u> </u>	
68		Area	8105	68	Water	SCC-25		5/21		<u>C1</u>	8CN		<u>CS2</u>		<u></u>	
69		Area	8105	69	Water	SCC-08		5/21		Cl			CS2		<u>Th</u>	<u>OP</u>
70		VLET	8105	<u>70 </u>	Water	90C-79		_5/21/		<u>ci</u>	SCN	<u> </u>	CS2_		<u></u>	<u>0P</u>
71		Area	8105	71	- HOLE	90C-080	- 11 - 20	<u>5/21</u>		<u>Cl</u>	9CN	HG	CS2		<u></u>	OP OP
72			8105	<u> 72</u>		8CC-079	Spike 70			<u>C1</u>	9CN	HO	<u> </u>		Th Th	OP OP
<u>73 </u>		Area	8105	73	Water	80C-079	<u>demo 70</u>	6/21		C)	9CN	HG	<u>CS2</u> CS2		Th	OP
75		_Area_	8105 8105	74 75	Water	SOC-079	Blanke st	5/21		Cl Cl	SCN SCN	<u>HO</u> HG	CS2		Th	OP
	Batch 6	Area Ponds	8105	76	Water		D19/E-3-3	5/28		Cl		HG	CS2	CTC	Th	OP OF
77	DE COL V	Ponda	8105	77	Water			5/28		Cl		HG	CS2	crc	Th	OP
78		Ponds	8105	78		0-65		5/28		Ci		HG	CS2	ETC	Th	OP
79		Ponds	8105	79		0-49		5/28		Cì		HG	CS2	CTC	Th	OP
80		Ponds	8105	80	Water	001-4		5/28		<u>cì</u>		HG	CS2	CTC	Th	OP
81		Ponds	8105	81	Water			5/28		Cl		HG	CS2	CTC	Th	OP
82		Ponds	8105	82		0-32		5/28		Cl	`	HG	CS2	CTC	Th	OP
83		Ponda	8105	83	Water	0-39		5/28	/86	Cl		HG	CS2	стс	Th	OP
84		Ponda	8105	84	Water	0-39	Spike 83			C1		HG	CS2	стс	Th	OP
85		Ponda	8105	85	Water	0-39	dup 83	5/28		C)		HG	CS2_	CTC	Th	OP
86		Area	8105	86		90C-52		5/28		C)		HG	CS2	***	Th	OP_
87		Агеа	8105	87		90C-78		5/28		Cl	·	HG	CS2	***	Th	OP_
88		Ponda	8105	88		0-39	<u>Blank</u>	5/28		Cl		HG	CS2	CTC	<u> </u>	OP
89		Ponds	8105	89	Water		Blank sp			Cl		HG	<u>CS2</u>	CTC	<u>Th</u>	OP_
90		Ponds	8105	90		0-11		5/28		<u></u>		<u>HG</u>	<u> </u>	CTC	<u>Th</u>	OP_
<u> </u>		Ponds	8105	91		<u>0-45</u>	0.114.55	5/28		<u>C1</u>		<u> </u>	<u>CS2</u>	CTC	<u>Th</u>	OP_
92		Ponda Ponda	8105	92		0-31	Split CD			<u>Cl</u>		HG	CS2	CTC	<u>Th</u>	OP OP
9 <u>3</u> 94			<u>8105</u>	93		0-29	Split Ch			<u>Cl</u>	-	ii <u>c</u>	<u>CS2</u>	CIC.	<u>Th</u>	OP OP
95			<u>8105</u> 8105	94 95	Water	0-29	COM Spike					HG	CS2	CTC	<u>Th</u>	OP OD
	Batch 7		8106			SCCLM-7	COM Blan					HG	<u>CS2</u>	<u>crc</u>	<u>Th</u>	OP
97	DE COLL	Area_	8106			SOCIM-1			/86	C)			CS2		<u> </u>	OP OP
98		Area	8106	7		SCOLM-5		6/3	/8 <u>6</u> /8 <u>6</u>	CI		<u> G</u>	CS2 CS2		<u>Th</u>	OP OP
99		Area	8106	—— -		SCC1M-6			/86	Cl Cl		HC	CS2		<u>Th</u>	OP OP
กัด		Area	8106			CNAM-30			/86	CI			CS2		Th_	OP

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Th(Thickarlametes), Of (Orkney those thetes), (Tr(Cartxx) Tetrachloride), pp((Priority Pollutants)

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Pords 8108 18 Soil CTC-1s Dup 1/25/86	
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Pords 8108 22 Soil OCTC-1s 30' 1725/86	333
Portal 8108	CS2
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Pords 8108	283
Pords 8108 25 2011 OCTC-2s 201 O	
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Pords 8108 28 8011 OCTC-2s 10' 7/25/86	
Ponda 8108 29 801 CCTC-2s 20' 7/25/86	697
Ponds 8108 31 301 0.33-18 0.000 7/25/86 1/	710 200
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APPENDIX II TOTAL PRIORITY POLLUTANT RESULTS

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CC/Lefloyne ... Sunap Phase I results, Anaples collected 5/11/86 - 5/20/86, Sanpine analyzed at Baviranae. ... : unting and Certification

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Sample I.D. : Description :	Approximate Detection	8105	8105	81 05 3	\$105	8105	\$105	8105
pescrpcion :	Limit	1 SCC-CC-12	2 300-017-19	3CC-05-1V	4 900-050	5 300-014	6 300-018	7 300-016
	Concentration		200-411-18	300-43-14	300-414	300-414	344-418	307-418
Compound	(ug/1)			•				
Volatile Compounds	_							
Acroleia	100	18	19	10	10	10	10	
Acrylomitrile	100	10	10	10	ЖD	M		10
Benzene		70	10	10	16	10	10 -	10
bis(Caloromethyl)ether		10	10 .	10	10	10	10	M
Brossfore		10	10		10	10	ND.	10
Carbon Tetrachloride			10	10	10	10		10 .
Chlorobenzene		18	100	10	10	10	10	4
Chlorodibromomethame	-	10		XD	10	10	10	
Caloroethane	•	10	10	10	10)(D)		10
2-Caloroethylvinyl ether		10.	1	10	*	10	10	
Calorofora		10	19	16	10	10	10	•
Dicklorobroscethane		10	79		10	16	10	10
Dichlorodifluoromethane		10	10	10 20	10 10		10 10	10 10
1,1-Dickloroeth ane 1,2-Dickloroeth ane		10 10	10 16	18				10
1,1-Dickloroethylene			5	20	5	2		7
1,2-Dichloropropane	5.0	-	-	10	<u>.</u>	10	3	10
cis-1,3-Dichloropropylene			<u>~</u>	ĩ	10	P		10
Bthylbeazene		2	9		- T	7	•	2
Sethyl brouide	10	20	2	n			-	10
Bethyl chloride		20	2	10	-	3		10
Nethylene chloride				20	2	10	ē	10
1,1,2,2-Tetrackloroethylene				10	10	10	•	10
	4.1	18	•		10			
Toluene		10	10	10	10	10	10	
1,2-frans-dichloroethylese		ND	10	10	10	10	10	10
		XD	10		10	10	10	AD.
1,1,2-Trichloroethane		10	18	10	10	10		10
- frichloroethylese		10				10	10	10
Trichlorofluoromethane	10	10		18	10	10	18	D
Vinyl chloride	10	100				10		10
trans-1,3-Dickloropropylene	10	10	10		10	10	10	10
Base/Neutrals & Pesticides								
Acenaphthene	2.2	10	10			10		10
Acesaphthylese	4.0	10	10			10		10
Anthracese	2.2	ND	10	10		10		10
Benzidine	51.0	ND	10	10		III	10	MD
Benzo(a)anthracene	9.0	ND	10	10	16	10	10	MD
Benzo(a)pyrene		NO	10	10	10	10)(D	MD
Benzo(b)fluoranthese		MD			10	10	10	10
Benso(ghi)perylese		ND	10	10	10	10	10	10
Benso(k) fluoranthese		ND	10	10	M	10	10	10
ors(2-Caloroethory) methane		ND .	10	10	10	10	16	10
bis(2-Caloroethyl)ether	6.6	ND	10	ID	10	M	10	10

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Cold Creek/LeHoyne RI/PS, Well Sumples (Source); Analysis by Bavironmental Testing and Certification

Sample 1.D. :	Approximate	\$018	8165	8105	20.00	\$018	\$018	\$105
Description :	Detection		~3	~	-	~	•	•
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Cosponed								
Endris eldebyte	e 11.0	9	9	9	9	9	S	ç
Hentacklor	r 2.2	g	9	9	•	? \$	1 9	? 5
Santachal as darked		? 5	? 5	: 5	1 9	? 9	? 9	? 9
niroda soverandan	•	2 (2 .	2	2	2	2	R
M2-1212	0.17 2	2	9	e	9	9	9	9
PCB-1254 41.0	4 41.0	9	9	9	Q	9	9	9
PCI-1221 41.0	1 41.0	9	9	9	9	9	9	£
PG8-1212 41.0	2 41.0	g	9	9	9	9	2	? 5
A 11 8151 PG	0 17 8	! 5	? \$? \$	2 \$? 9	2. 5	2 9
0.11 0.21 - 200 0 13 0.31 - 200	2.17	2 9	2 9	2 5	2 5	2 9	2 9	2 9
0.14 3401-007	2.5	2 9	2 9	2 9	2 9	2 4	2 9	2 9
	0.1.	2 9	₹ !	2 !	2 9	₹ 9	₹!	2 !
Toraphene II.0	0.11.0	9	9	9	9	2	9	9 1
Acids	1							
2-Chlorophenol	1 2.8	g	9	9	•	9	9	£
2.4-Dichlorophenol	1.1.1	2	9	9	#	9	9	9
2.4-Dimethylphenol		g	g	9	•		9	9
4.6-Dinitro-o-greeni	•	9	9	: 9	2 9	2 9	9	2 5
2. Laliaitmohan		? S	9	9	•	2 9	1	9
		2 1	1	2 9	1	2 9	2 9	R 9
		2	•	2	R	2	2	2
4-Nitroplesol	1. 2.8 1. 2.8	9	9	9	£	9	9	9
P-Caloro-a-cresol	1 3.4	9	9	9	9	9	9	9
Pentach lorophenol		9	9	g	9	9	9	9
Pheno!		5.5	9	9	9	9	9	9
2, 4, 5-Trichlorophenol	-	9	9	9		9	9	9
Metals & Cynaide	,							
Astison	Astisony Variable	8 0, (18	8 (13	8 , (19	81) 'Q	8 1)(1 8	8. 8.	8 1>' 9
Arnenia	Armenic Variable	9 0.00	9.43	BDC. (10	10		11.0	-
and I was	Aprelline Veriable	5	5	5	5	5	•	
Devel 190								

9	夂	9	£	9	오	9		Cymide, Total	
100.0	56.0	110.0	3.3	170.0	36.9	63.0		Zibc	
BOL, (S	.	8	BBDL, (S	6 , 6	8 , (5	8 , (S	Variable	Pallim	
300 C, (8. (9	Ð, 3.4	2004 , (3.4	₹. 9.	BEC., (3.4	BEDC, (8.4		Silver	
8	ë,	6	36 , 3	8 ,	€	9 , 6		Seleniu	
30, (11	13.0	14.0	300 , (11	HOL, CI		16.0		Nicke!	
7.), G	300 , (2, C.2	8 ,(2	?`,' Q	300°, <.2	BEDC. <.2		Mercury	
19 , (5	₽, <\$	18) 'Q	BDC, (58	5) (2	9 , (S	350 °, (2 1		THE STATE OF THE S	
	6 ,0	8 ,0	ä	€	8 1>' 9	810, (18		Copper	
£, (3	BEOL, CIS	£, €	8 0,0	€ , C	£), (9	61) (13		Chrosius Chrosius	
₿,(4.1	6 , (2.3	6 , €.3	BDL, (4.3	ĕ , (4.3	6 , (2.3	B , (4.3		Cadaite	
D , <. 57	6 ,<.57	10 , (57	6 , <.57	10 , (57	6 , (.57	10 , (57		Beryllium	
9T), (11.0	8 , 9	8 0,0		91) '9	9 , (10		Arsenic	
\$? ?	2 2 9	8 0,0	8. 2. 9	\$,°	2, G	8, C		Astisony	

3 10 00203

Sample [.D. : Description :	Approximate Detection	\$105 1	8105 2	8105 3	8105 4	8105 5	\$105 6	8105 7
peacrpctom .	Linit	3CC-CC-12	3CC-017-1W	3CC-05-1V		3CC-014	SCC-018	SCC-016
	Concentration	200 00-12			•	000-011	400-414	000-414
Compound	(u /1)							
bis(2-Chloroisopropyl)ether	6.6	10	10	10	MD	10	MD	ND.
bis(2-Sthylheryl)phthalate		10	19	MD	BNDL	BADC	14.1	SHOL
4-Bromophenyi phenyi phthalate		10	10		MD	MD.	MD	XD
Butyl bensyl pathalate		ND.	10	10	10	10	BIOL	10
2-Caloronaphthalene		10		18	10	10	10	
4-Chlorophenyl phenyl ether		70		10	10	10		10
Cirysene		10	10	10	10	78 18	10	10
Dibenso(a, h) anthracese		10 - 10	-	10 10	10		10	10 10
1,2-Dicklorobensene 1,3-Dicklorobensene		. 20		10	10	10	10	2
1,4-Dicklorobenzene		2	-	2	10	. 20	õ	10
3.3'-Dichlorobenzidine		2	-	2		2	ē	20
Diethyl phthalate			10	10	10	10	n n	10
Disethyl pathalate		10		10	10	10	10	10
Di-a-butyl pathalate		16	10	10	10	ND	M	MD
2,4-Dimitrotoluene		10	18	10	10	10	10	M
2,6-Dinitrotoluene		10	*	10	10	10	XD	10
Di-d-octyl phthalate		18		10	BEDL	MOL	BHDL	
1,2-Diphenylhydrasine				10		M	10	10
Pluorantiese	2.5	M		10	10		16	10
fluorese		10	10	10	10		10	10
Bezach lorobensene	-	10		10	M	M	10	MD.
Berach lorobutadiene		10			MD		10	10
Bezachlorocyclopentadiene		10	10	10	10	D		10
Serach loroethane		10		10	10			10
Indeno(1,2,3-c,c)pyrene		10		10			10	
Isophorone				10	X 0	10	**	10
Maphthalene		10		10		10	10 10	10 10
Witrobensene		XD	7	10	10	10	10	10
N-Hitrocodinethylanine		10		10	10	10	Ď	10
M-Witrosodi-e-propylamine M-Witrosodiphesylamine)D	2	10 10	10	7	10	10
Phenanthrene		10		10	10	7	-	10
Pyrene		100	1		10	2		10
1,2,4-Tricklorobensene		20	2	2	10	7	X	
Aldria			=	ñ	10	7	*	
Alpha-88C		2	2	10	10	10	2	10
Beta-BBC		-		20	10	20	-	10
Games-69C		16	ě	16	10	10	10	10
Delta-BBC		10		10	10	10	10	
Calordane	11.0	10		10	10	IO	ND	10
4,41-907	3.2	ND.		MD	ND.	X	10	MD
4,4'-008	6.4	10	10	100	ND	10	10	10
4,4'-000		10	10	ND:	MD	MD	ND	M
Dieldrin		颖		ND	NO.	10	ND	MD
Endosulfam I		10		XD	ND	10	10	10
Endosulfan II		10	10	MD	ND	10	10	NO.
Endosulfan sulfate		10	10	10	X 0	ND 10	10	10
- Bodria	11.0	10	10)(D)	MD	MD	Ю	10

Cold Creek/LeHoyne RI/FS; Ponds, Lagoous, and Landfills; Sampled 6/15/86 to 8/6/86; Analysis by Environmental Testing and Certification

Sample I.D. : Description : Compound	Detection Limit Concentration (ug/l)	8105 8 CCH-7-1	\$105 9 CCB-7-1 dup \$	8105 11 CCM-7-1 blank	8105 13 NH-1	\$105 14 MI-2	8105 15 NCTC	\$105 - 16 LLP	8105 17 CCLP
Volatile Compounds									
Acroleia	100	10		100	ND .	10	. 10	10	10
Acryloaitrile	100		10	10	10		10	10	
Benzene	4.4		10	10	10	10	10	10	10
bis(Chloromethyl)ether	10	10	10	10	MD	10	10	M	10
Bromoform	4.1	ND.	10		ND ,	110	10	ND	10
Carbon Tetrachloride	2.1)D	10	10	199	10	12.3	10	5.57
Chlorobenzene	6.0	10	10	10	10	10	MD.	M	10
Chlorodibromomethane	3.1	10	10	MD.	MD	M	M	XD	ND
Chloroethane	10	10	XD		ND	10	ND	10	M
2-Chloroethylvinyl ether	10	10	M	10	10	10	10	10	10
Aloroform (Aloroform	1.5	ND	10	10	51.5	9.04	XD	10	ND.
Dichlorobromomethane	2.2	MD		10	XD	10	10	10	
Dicklorodifluoromethame	10	16	10	10	10	10	ND	10	NB .
1,1-Dickloroethane	4.7	10	10	10	10	10	10	10	
1,2-Dickloroethane	2.8	10	10		10	10	10	X 0	XD
1,1-Dichloroethylene	2.8	10	10)(D	10	10) ()	10
1,2-Dickloropropane	6.0	10			10	10	10	10	10
cis-1,3-Dickloropropylene	5.0	10		· 10	10	10	10	18	
Sthyl bensene	1.2	10	×	10	10)D	10	
Methyl brouide	10	10	10	10	10	10	10	10	10
Hethyl chloride	10	10	10	10	10	10	10)D	1
Bethylene chloride	2.8	10	10	584	10	10	1	70	
1,1,2,2-Tetrachloroethylene	6.9	10	10	1	10	10 10	10 10	16 18	-
Tetrackloroethylene	4.1		18		10	8.47 ~			-
- Toluene	6.0		10	BIOL	10				3
1,2-frame-dickloroethylene	1.6		10	10	10	10		70 10	1 0
1,1,1-frickloroethane	3.8	10	10	D	M	10	10		10
1,1,2-Trickloroethase	5.0	10	10		10	70		10	10
Trickloroethylene Tricklorofluoromethane	1.9	10	10	10	10			*	70 M
Vinyl chloride	10 10		2	1	5		10	10	~
trans-1,1-Dichloropropylene	10	10	10		10	10	10		ď
Base/Neutrals & Pesticides									
Acenaph these	1. 9	10	•	100	10	ND	10	10	10
Aceasphthylene	3.5	10	10				70	2	-
Accesses Anthracese	1.9	2	10	2	10		20	-	<u> </u>
Benridine	1.3 44	10	10	-	XD	2	20	20	10
Benzo(a)anthracene	1.8	10	10	-	10	XD		10	10
Benzo(a)pyrese	2.5	XD	ñ	18	100	10	10	10	ND
Beazo(b) fluoranthene		10	10	100	10	ND.	10	10	WD
Benzo(ghi)perylene	4.1	10	10	ĬĎ.	MD	10	MD	ND .	10
Benzo(k) fluoranthese	3.5	100 100)(D	10	10	ND	ND	10	ND
-Chloroethory sethane	5.3	10	10)ID	MD.	MD	100	10
s(2-Caloroethyl)ether	5.1	10	Ď	~	ID	10	10	10	ND
bis(2-Caloroisopropyl)ether	5.7	10	XD	<u>.</u>	10	10	10	10	MD.
bis(2-Sthylberyl)phthalate	10	NO	10	10	BEDL	BMDL	10	MD	SHOL
<u></u>				. –			•		

ND, Not Detected BMDL, Detected, but below stated IPA method detection limit

Cold Creek/LeHoyne RI/PS; Ponds, Lagoons, and Landfills; Sampled 6/15/86 to 8/6/85; Analysis by Ravironmental Testing and Certification

Sample I.D. :		8105	8105	8105	1105	8105	\$105	\$105	8195
rescrition:	Detection	1	3	11	13	14	15	16	17
	Linit	CCH-7-1		CCH-1-L	唯-1	18H-2	NCTC	Ш	œ
	Concentration		dup \$	blank					
Compound	$\{ag/1\}$								
4-Bromophenyl phenyl phthaiste	1.9	10	MO	ND	MO	160	MD	XD	10
Sutyl bensyl pathalate	10	10	MO	10	10	NO.	10		16
2-Chloronaphthalene	1.9	XD	10	XD	10	19	10	10	
4-Chlorophenyl phenyl ether	4.2	WD.	10	10	10	10	10	18	10
Carysene	2.5	10	10	10	10	18	10	•	- 10
Dibeaso(a, h)anthracese	16	XD	10	· 10		10	10		18
1,2-Dicklorobensene	1.9	10	10	10	M	10			
1,3-Dichlorobensene	1.9	10	10	10	_ 10	10	, 10		
1,4-Dichlorobeasene	4.4	10	10	10	··)(0	10	10		10
3,3'-Dichlorobensidine	17	10	10	10	10	10	10	10	
Diethyl phthalate	10	10	10	10	M	10			
Dimethyl pathalate	10	10	10	ND.		10	•	10	10
Di-a-batyl pathalate	10	16	10	MD	10	10	10	10	
2,4-Dimitrotoluese	5.7	10	10	10	10	10		10	10
2,6-Dimitrotoluene	1.9	10	10	10	10	10	10	10	10
Di-a-octyl phthalate	19	10	10)(D	10	10	10		10
1,2-Diphenylhydrasine	10		10	10	10	10		10	10
· Fluoranthese	2.2 1. 9	10	2	, 10 10	10 10	10	10	19	_ B
Fluorese Bexach lorobenzene	1. 3 1. 9	10 10	16		10		18	20	
Bezach lorobetadiene	,,3 , 5	20	10	10	10	10	D	1	
	. . 10	_	12		10	18	10	7	
.chlorocyclopeatadiene Sexachloroethane	1.6	15) 16)		9	10		5	-	1
Indeno(1,2,3-c,c)pyrene	4.7	10		2	1		-		-
Isophorose	1.1	10	-	10	10	12		1	
Maphthalese	1.5	5	5		10				
Vi trobensene	1.9	2	-	20	5	5		ã	2
N-Hitrosodisethylanise	10	•					n	2	
N-Mitrosodi-a-propylamine	19	10	10	16	10	•		•	10
N-Mitrosodiphenylamine	1.9	10	10	10		10		1	10
Phenanthrene	5.4	10	10	10	10	10	10	10	
Pyrese	1.5	10			10	10			
1,2,4-Trichlorobensene	1.9	10	19	10			10	10	10
Aldria	1.9	10	10	10	10	10	10		10
Alpha-88C			10	10	10			10	
Seta-SEC	4.4	I	M	10	10	10			10
Gassa-BBC		10	10	10	ND	10	10	10	
Delta-BBC			18	M	10				
Calordane	10	10	10	10	100	10		10	10
4,4'-0 0T		10	100	10	10	10	10	18	
4,4'-008	5.6	10	10	MD	10	10	10	*	
4,41-000	4.7	10	10	MD	10	10	10	10	10
Dieldrin		10	10	ND .	ND	10	10	10	
Endosulfan I		10	10	ND ND	10	ND VD	10		
Endomilian II	10	10 10	10	10	10	10	10 10	10 18	
Badosulfaa sulfate Badria	5.6 10	18) 10)	10) 10)	16) 16)	100 100	10 10	10	10	10
	10	10	110 110	NO NO	10 10	10	10 10	19	10
Endrin aldehyde Heptachlor	1.5	10	16)(I)	10		10		
•)Ø	10	10	X 0)D	10	10	9
Reptachlor epozida	4.4		mŲ.	*	=	, W	N	AU	-

3 10 00206
Cold Creek/LeMayne BI/FS; Poods, Lagoons, and Landfills; Sampled 6/15/86 to 8/6/86; Analysis by Environmental Testing and Certification

Sample [.D. :		\$105	8105	8105	6105	8105	8105	8105	8105
Descrption :	Detection	1	\$	11	13	14	15	16	17
	<u>Lisi</u> t	021-1-1		CCH-1-1	16 -1	第-5	NCTC	UP	and the second
	Concentration		dup \$	blank					
Compound	(04/1)								
PCB-1242	38	10	10	10	10	16	10	10	10
PCB-1254	36	10	10	10	10	10	10	10	10
PCB-1221	36	10	10	10	10	M		10	
PCB-1232	36				100	M		10	10
PCB-1248	36	XD	10		10		10		
PCB-1250	35	10					10	10	
PCS-1016	36			10	10	ND.	10	YO	
Toraphene	10	M	₽.	10	10	MD.	10	Æ	MD
Acids									
2-Chlorophenol	- 1.3	10	10	10	10	XD	10	76	10
2,4-Dichlorophenol	2.7	10	18	10	MD	ND	10	10	10
2,4-Dimethylphenol	2.7	ND	10	10	10	10	10	10	10
4,6-Dimitro-o-cresol	24	10	10		10		MD.		10
2,4-Dinitrophenol	42	10			10		10	10	
2-Hitrophenol	3.6	10	10		10	10	10	10	
4-Hitrophenol	2.4		10	10	10	Ð	10		
p-Caloro-a-cresol	3.0	10	10	10	6	18	10	10	
Pentach lorophenol	3.6						ND.	10	
Phenol	1.5	16			10	10	7.52		
4,4,6-Tricklorophenol	2.7			10	10	10	10	10	10
Hetals & Cyanide		-							
Antinony	199	10	10		10	10	10	*	
Arsenic	26	39	25	10,(10		M	10, (50	10, (50	
Beryllium	1	10, (.57	10, (.57	10, (.57	10	10		10	
Cadmium	4	10					10	10	10
Chronium	20	HOL	10			10	MD	10	
Copper	10	HDL, (18	10, (18	10, (18			10		BEOL
iend	58	100				19	į III	10	
Hercary	0.2		10	10			0.30	0.20	
Victel	28	24	MDL , (11	₩,<11	13 ,		BADL	22	13
Seleni un	5	10			10		ID, <25		10
Silver	8	10	300L	10			10		
Thallien	5	10	10, (25	10			10 , (25	10, (25	10
Ziac		128		16	54	22	12	23	160
Cyanide, Total	25	10	10		dizzi	10	MD	132	
			Bottle	broke in	ck i preset				

Cold Creek/LeHoyne RI/PS; Ponds, Lagoons, and Landfills; Sampled 6/15/86 to 8/6/86; Analysis by Environmental Testing and Cartification

Sample I.D. : Description : Compound	Detection Limit Concentration (NE/NE)	\$106 27 APP-15 split EPA	8106 35 APP-23	8106 42 blank	8106 43 APP-25 dup 35	8106 44 HTP-13	\$106 52 MTP-23	8106 55 NCTC-18	61 61 ECTC-25	67 67 8106	\$106 74 blank	\$106 ?6 UP-22	\$106 17 LLP-22 dup 16
Volatile Compounds													
Acroleia	25000	10		10	10	10	M	10		10			10
Acrylomitrile	25000	10	100	18	18	10	10	ND.		18		10	
Beasene	1100	10	10	10	10		10	10				18	
bis(Chlorocethyl)ether	2500		10	10	10		18	10					
Bronoforn	1200	10	10		10		10	10			10	10	10
Carbon Tetrackloride	708	10	M		1	10		10	10		10	10	10
Calorobeasene	1500	18	10	10	10	10	10	XD	•	10	10	10	10
Calorodibrososethane	780				1	10	10	10			10		
Chloroethane	2500			100	10		10	10				10	D
2-Caloroethylvinyl ether	2500	ID		M	10		10				D	10	
Caloroform	100	10	10	XD	10		10	10					D
Dicklorobromomethane	550	10	10	M	10		10				10	D	
Dichlorodifluoromethane	2500	10	10	10	10			10			10	10	10
1,1-Dichloroethame	1200		10	10			10	10	1	*	18)D	
1,2-Dichloroethane	700	10	10			10	10	10			10	10	
1,1-Nichloroethylene	100	10	10		10		7	10	Ĭ		D	•	•
1,2-Dickloropropene	1500			1	10		10	10		1	10		D
cis-1,1-Dichloropropylene	1300	10	10	XD	10	10	10	10			10	XD	10
Rthyl bensene	1800	10	10	10	10			10		10		D	
Methyl bromide	2500	10	10	10		10	19	•			10	10	
Hethyl chloride	2500	19	10		10	10	10				1		•
Methylene chloride	700	10	10			10	1	3060	2390	3146	D	6320	
1,1,2,2-Tetrackloroethylene	1700	10		10		10	19	10	1			id Va	D
Tetrackloroethylene	1000	10	10	10	10	10 10	10 10	16 16		B	ID ID	10 10	1
To luene	1500		18		_		10	10	2	7	10	10	
1,2-frans-dickloroethylene	400	D		10	10	_		10	1			10	_
1,1,1-frickloroethane		10		10			10 16	1	-	10	10	10	
1,1,2-frichloroethane		10		10	18		10	7	-		10		
Trickloroethylene				10		=		-	-	=			
fricklorofluorosethase	_		10			=		-		-		10 10	
Vinyl chloride trans-1,3-Dichloropropyless		10	10		9	5		7	5	10		10	
Base/Neutrals & Pesticides													
Acenaph these	1900		•	10	12	10	10	10		19	10	D	•
Acenaphthylene		2	7	- 18	-	10	10) 0	5	10		D	•
Anthracese		5	X		2)D		10		10	10	<u>.</u>	•
an caracese Bensidise		10	10	10	10	10	2	100		10	10	10	9
Benzo(a)anthracene		10	10	2	7D	10	XD	10	10	10	18	10	1
Benzo(a)pyrese		XD	10	10) 10	10	10	10		10	ND	10	•
Benzo(b) fluoranthem			10	10	10	10	10	10	10	10	10	10	18
Benso(ghi)perylene		10	10	10	10	ND	10	ND	10	10	10	10	
Benso(k) fluoranthene		10	10	1		ND	10	10		10	10	10	
bi-12-Gloroethory) methane		10	10	1	10		10	10		10	10		
s(2-Chloroethyl)ether		10	10	2	20		10	10			Ñ	. 10	•
bis(2-Gloroisopropyl)ether		10	10	10	10	10	10	10	1	10	10	10	
bis(2-Sthylheryl)phthalate)D	10	10	10	10	10	ND		10	10	10	
and a sealthers		~		_	-	_	-		-				-

Cold Creek/LeHoyne RI/FS; Ponds, Lagoons, and Landfills; Sampled 6/15/86 to 8/6/86; Analysis by Environmental Testing and Certification

Sample I.D. :		8106	8106	8106	8106	8106	\$106	\$106	8106	8106	8106	\$106	8106
Descrption :	Detection	27	25	42	43	44	52	5\$	61	67	74	76	77
·	Lizit	APP-15	APP-25		AP7-23	HTT-13	117-23	NCTC-13	ETC-25	LP-13		W-2	TD-2
	Concentration	split PA		blank	dup 15						blank		dup 16
Compound	(us/ks)	•											
4.Onemakanal akanal aktholata	1900	10	100	-	18	10	MD.	120	10			-	_
4-Bromophenyi phenyi phthalate Butyi bensyi phthalate	9900	10	XD XD	100 140	16 10	10	10	XD		160 160	10 10	10	10
2-Chloronaphthalene	1900	2	10	10		-	20	10	10	10 10	1D	10	
4-Calorophenyi phenyi ether	1200	20	10	10		10	2	2	10	20	10	D	10
Carriene	2500	10	7	10	=		10	10	3	18	10	Đ	Ñ
Dibeaso(s, h) anthracene	9906	16	10	Ď	10	10	10	Ĩ	- -	10	10	10	-
1,2-Dick lorobenzene	1900	, i	X 0	10			10	10	10	10	D	10	7
1,3-Dicklorobensene	1900	10	10	8			10	10	-	3	<u> </u>	10	X 0
1,4-Dichlorobenzene		10	10		10	D	D	10	•	10	10	D	Ď
3,3'-Dichlorobenzidine	16000)D	10	10	10	10	10		100	10	10	10	10
Diethyl phthalate	9900	10	10	10	•	10		10	<u> </u>)D	XD	10	10
Dimethyl pathalate		10	10	*		10		10	10	10			<u>~</u>
Di-n-butyl phthalate	3900		10)D	D	10	10	10	D)D	10	D	10
2,4-Dinitrotoluene	5600	10	10	10	10	10	10	ND.	10		10	10	10
2,5-Dinitrotoluese		10	XD.	10	10	10	10	10	10	ND	16	10	XD
Di-m-octyl pathalate	• • • •	10	10	10	10	10	10	10	10	ND.	ND	10	
1,2-Diphenylhydragine		10	10	10	•	10	15	16	15	10	10	10	10
Pluoranthene		10		10	7	10	10		10	10	10	10	20
Pluorene		18	10	10	10		10	1	10	10	•		10
Hexach lorobensene			10	10	10		10	10	10	120	18	10	70
Serachlorobutadiene		. 🕦	10	10	10			10		10	10	Đ	20
exachlorocyclopentadiene		10	XD	10		10	10	10	10	10	10	10	ND
Berack loroethane		M	ND:	10	10	10	10	10	10	10	ND.	18	AD.
Indeno(1,2,3-c,c)pyrene	4700	10)ID	10	10	10	10	10	10)(D		10	MD
Isophorone		10	10	10	XD	10	10	10	10	YD		D	YD.
Kaphthalene		10	10	ď			10	10	10	10	18	B	10
Ni trobensene		18	10		10	10		10	10	10	10	10	18
N-Witrosodinethylamine			10	10	10	<u> </u>		10		10	10	10	16
W-Witrosodi-n-propylamine		10	10	18	10	10		10	ND	10	18	10	10
M-Witrogodiphenylamine		10	10	10			10	10	ND.	70	10	10	NO
Phenanthrene	5300	10		10	10					M	10	10	10
Pyrene	1906	10	10	10	10	10	MD	18		MD			10
1,2,4-Trichlorobenzene	1900		10	10	10		10	10	10	10	10		10
Aldria	1900		10	10	10	10	10	M		XD			
Alpha-BBC	9906	18	10	10	10	10	10	10	10	10	10	D	10
Beta-890	4400	10	10	MD	XD	10	10	10	10	10	10	10	10
Gamma-BBC	9900	10	10	10	10	10	18	10	18	ND.	10	10	10
Delta-BBC		10	10	10	10	10	10	10	10	10	10	10	10
Chlordane		10	10		10		10	10	10	10	10		
4,4'-DOT		18	10	10	10	×	10	XD.	D	10	10		10
4,4'-DOE		10	M	10	10	XD		10	10	10	10	10	ID
4,4'-000		10	MD	10	10	10	10	10	10	10	ND	10	10
Dieldr <u>ia</u>		MD.	10	10	10	ID	M	ND	ND.	10	10	10	10
Endosul fan -I		10	M	10	10	10	ND.	MD	MD	MD	顺	10	10
Radosulfan [I		ND .	ND	M	10	XD	M	MD	MD	10	MD	10	10
Endosulfan sulfate		MD	10	10		10	10	10	ND	110	10	10	10
Radria		10	10	10	10	10	10	10	ND.	10	10	XO	MD
Endrin aldehyde		18	10	10	10	10	10	10	10	100	10	•	10
Heptachlor		NO	10	10	10	10	10	XD	10	ND	10	10	XD
Heptachlor epozide	2200	ND.	10	18	10		10	XD .	ND	ND	ND	10	ND.

3 10 00209

Cold Creek/LeMoyne RI/FS; Ponds, Lagoous, and Landfills; Sampled \$/15/85 to \$/5/85; Analysis by Environmental Testing and Certification

Sample I.D. : Description :	Detection Limit Concentration	8106 27 APP-15 split SPA	\$106 35 APP-28	8106 42 blank	8106 43 APP-23 dup 35	8106 44 MTT-18	\$106 52 HTT-25	8106 55 NCTC-18	6106 61 MCTC-28	8106 67 LLP-18	8106 74 blank	8106 76 112-23	\$106 17 U2-23 dap 16	
Compound	(14/4)													
PC9-1242	36000	10	10	10		18	10	10	10	10	10	•		
PC9-1254	36000	10		10	10		10		10	10	M			
PC3-1221	36000	10			*	10	10	16		-				
PC9-1232	35000	16			10	19	10	10	10					
PCB-1248 PCB-1260	36000 36000	1) "	3		10	10	10	10	10	10			3 3	
PCB-1016	36000	2	5		7		-		3	5			10	
Tozaphene	3300	10	7	Ď	10	•	10	10	D	10	5		10	
kci ės														
A (\$1,)			_		_	_				_		_	_	
2-Chlorophenol 2,4-Dichlorophenol	3300 2700	10	10	10		3	7	7	70 70))	3	7	1D 10	
2,4-Dimethylphenol	2700	7	-	*			2		7	-	7		20	
4,5-Dimitro-o-cressl	*24000	2	5	10		õ	3	=		5	3	D	10	
2,4-Diaitrophenol	12000	1		10		1	10	1	10	1	3	1	7	
2-Hitrophenol	1600	18		10		10	16		10	18				
4-Hitrophenal	2400	18	10	1	10	10		10	10	10				
p-Caloro-e-cresol	3000	10	10	10	10	18	10	10	10	10			10	
Pentach lorophenoi	3600	10			10	10	10	10	18	10			10	
Phonol	1500	18				10		10						
2,4,6-Trichlorophenel	2700	*						1					10	
Metals & Cynnide														
Antimony	12000	10		10	10		10	10	10	10		10	10	
Armaic	1000	1000	2000		1000			1000	MOL	HOL		MOL		
Beryllium	42	200	160		160	13	MDL , (7)		150	11			270	
Cadmium	420	10000		10		10	10	1	10000					
Chronium	[300	12000	11000		9906 3700	4500	\$600	23000	100000	(500 2200		3406	12006 3100	
Copper	1000	1200 '	1806			2300	3300	1000	4600			2700		
Local Sercery	4600 180	7200 200L	7606 MBL		7900 100	1900 1400	5000 24000	5000 Jane	MOL.	5400 3400		6100 HDL	300 5300	
Fichel	726	1200	1200	386	1300	MAC	1500	1800	8200	1700		2500	2500	
Selenium	580			19	10	1		MDL.		•		MOL	10	
filver	1600	19	•	Ñ	10	Ĩ	19	10	10		1		Đ	
Thellium	500	500		1		•			10				HOL	
Zine	600	14000	1400	5700	9100	9700	12000	15000	21000	10000	1500	11000	12000	
Cyanide, Total	500	18		10		*	100	10	10	10	1900	•		

Cold Creek/Leffoyne EI/FS; Ponds, Lagoous, and Landfills; Sampled 6/15/86 to 8/6/86; Analysis by Environmental Testing and Certification

Sample I.S. : Descrption :	Detection Limit	\$106 \$3 LLF-19	\$106 \$6 ULP-15	8106 89 LLF-22	810 6 94	8106 95 [J.P-38	\$106 98 ULF-48	8106 1 U.P-55	\$106 5	8106 10 113-13	#106 14 LLF-#8
	Concentra		dep 83	- W	blaak			- A	- 44	122 - 10	-
Compound	(ug/kg)				٠		•				
Volatile Compounds											
Acroleia	25000	XD	*	10	10	10	10	•		•	10
Acrylonitrile	25000	10	10	10	10	10					10
Beareas		10			10		10			10	10
bis(Chloromethyl)ether		10			10						MD
Brosofore		XD		10	10	10		10			ND
Carbon Tetrachloride		X	10		10	10	*				10
Calorobeaseae		XD	M	10		×					HD:
Calorodibromomethane		MD	10	10	10	10	10				10
Caloroethane		M	10	. 10	10	10	X	10			
2-Caloroethylvinyl ether		MD	10	10	10	M	XD	10	10		10
Calorofora		MD	10		M D	10	10			, 🗩	10
Dicklorobromomethane		10	10	10	10	XD	10	10		10	10
Dicklorodifluoromethane		10		M	10	10	MD.				10
1,1-Dichlorosthase		100		10	10	10	10	10			10
1,2-Dickloroethane		18		10	10	XD					10
1,1-Dichloroethylene		10		10		10	10				MD
1,2-Dickloropropase						10	10				10
cis-1,3-Dichloropropylene		M				M			10		10
lthy i beasens						10			10		10
Hethyl browide		10				M					
Hethyl chloride										10	
Bethylene chloride			10	MOL			970		143	1520	
1,1,2,2-fetrachloroethylene		10	10	10	10	10	10				10
fetrackloroethylene		•	10		10		*			М	10
Toluene	••••	10	10		XD		10		10	5460	
1,2-frans-dichloroethylene		10	10	10			10	10	10	10	M
1,1,1-Trickloroethane		10	10			10	10				10
1,1,2-Trichloroethane		10	10	*	10	10	10		10		
frichleroethylene	180	10	Ð	10	10	10	10	10	10		
fricklore (lapromethene		10		10	**	10	10		10		
Vinyl chloride		10	10		10	10	10	10	10		D
trans-1,3-Dickloropropylene	2500	10	10	10		10	10	10	10		
Base/Neutrals & Pesticides	_										
Accessibilitiese	1900	10	10	10	10	10	10				10
Aceasphthylene		10		19	16	10	18	•			
Anthraceae		10	10	10		10	10	10	10	1	10
Bensidise	44000		10	10	10	10	180	10	10		10
Seaso(a)aathraceae	1700	10	19	10	10	10	10		10		10
Benso(a)pyrene	2500	ND .			10	10			10		10
Beass(b) fluorenthese		10	ID			M		10			10
Beaso(di)perylese		10	10	10	10	10	10	10	10		
Beaso(t) (luorantheme	3500	M	10	10	10	10					
s(2-Chloroethory) methans	5200	M	m	10	10	MD	10				1
bis(2-Caloroethyl)ether	5600	M		10		10	10	10			
bis(2-Caloroisopropyl)ether		10	10	10	10	MD	10		10	10	10
bis(2-Sthylberyl)phtheiste	9900	XD	10		10	10	10	10			

Beptachlor epoxide	Beptacilor	Indria aldelyie	bdru	Indocalian miliate	Indocalina II	Indocalian I	Dieldria	(, 1, -)00	-,-;- - -	1,1'-197	Calordae	Delta-mer	Canal-MC	3eca-180	Alpho-887	Aldria	1,2,4-Trichlerobensese	Pyrese	Phenathrene	Hitrogodiphenylamine	I-Hitrosodi-a-propylaniae	1-Ritrocodinethylaniae	SEPTEMBER 11	mph Calles	sec. particular de la companya de la	2007(1,6,5-C,C)press	Tainn/1 ? 1 - claran			Served lambered see	Farnat Ambus and		I) moranthan	1.2-Diphenylhydracine	Di-m-octyl mathalata	2.6-Dinitroteluene	2. 4-Dinitrotolpene	Di-m-intyl phthalate	Dimethy shthelate	Diethyl phthalate	3.3" -Dick lorobensidine	1.4-Dichlorobenzene	l, 3-Dichlorobensene	1,2-Dichlorobensese	Dibeaso(a, h)anthracese	OLT THE OLD TH	4-Calorophenyi phenyi ether	2-Chloromaphthalese	Butyi benzyi phthalate	f-Bromophenyi phenyi phthaiste		Cornound		reactificant .		
111	8	3	33	250	33	3 3 3	23 8	108	8	8	9 9	<u>=</u>	3	\$	3	ž	Ž	<u> </u>	\$	3	ğ	ä	Por	Ī						Š	Ē	Ē S		Š	Š	<u> </u>	558	ğ	3	2	: 200	\$	1308		9900	8	2 28	1906	9 9 6	1906	<u> </u>		Concentration	Light		
8	5	5	5	5	5	5	5	5	•	5	5	5	8	•	5	5	5	5	5	5	5			i a	j	j E	5 E	5 (5 1	5 Q	5 i	5 1	5 i	5 (5 i	5 i	5 i	5	5	5 i	5	5	5	5	5	5	8	5	5	5	-	-		_		
8	5	5	3	8	5	5	5	5	8	5	8	5	8	5	8	5		-	5	•	5	8			j E	5 1	5 (5 1	5 1	5 (5 (5 1	5 i		5	5 i	5	5	5	B i	5		=	*	5	-	*	5	5	5		4				Ē
*	5	5	*		5	5	5	5	•	•	5	*	5	*	5	=	•	•	•	-	5	•	•		•	•	5 1	5 1	5 (•	, ,	5 (5 i			5 i	5 i		5 1	5	5	5	5	5	•	Ţ	*	•	*	•			1		2 5	
8	5	5	8	8	8	5	5	5	5	5	5	5	•	8	•	•	•	•	-	*	5	8			į	į	5 2	5 t	5 2	5 l	5 1	5 ł	5 (5 (5 (5 i	5 7	5 (5	5	5	5	5	5	8	5	5	5	5	5		i		2	2	2
5	5	5	5	=	5	•	•	•	•	•	•	-	•	•	•	•	•	•	•	•	•	•			,	5 4	,	5 1	5 1	5 1	5 t	5 1	5 (5 (5	3 (5 i	5	5	5	5	5	•	•	•	=	5	5	5	5			1	11-67 11-67	s ;	<u>=</u>
5	5	5	5	8	5	5	5	5	5	5	5	5	5	5	5	5	=	=	5	•	-	•	=	5 W	5 a	•	•	•	•	5 1	5 1	5 1	5 (5 1	5 (5 i	5 i	5 1	5	5	5	5	5	•	5	5	*	5	5	5						
8	5	•	5	8	5	8	5	5	•	5	#	5	5	•	5	•	•	•	5	8	*	#				1	j	5 1	5 1	5 1	5 1	5 1	5 (5 (B (5 (5 (5 (5	5 i	ð	6	8	•		5	5	5	5	5			1	- ET-	- {	2
5	5	5	5	6	5	5	5	5	5	5	5	5	5	5	Ę	5	•	5	=	- -	*	5		.		,	5 T	5 1	S E	5 I	5 (5 6	5 (5 (5 (5 (5 1	5 (5 i	5 i	5	5	5	8	•	5	5	5	5	5			1		~ }	
5	8	5	5	8	5	5	5	5	5	5	8	5	5	8	8	5		8	5	5	8	8		i #	.		5 2	5 T	5 8	5 t	5 1	5 t	5 (y (5 i	p (5 !	B (5 1	5	5	5	5	5	5	5	5	8	5	5			1	E-1	=	
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Cold Creek/LeMoyne RI/PS; Ponds, Lagoons, and Landfills; Sampled 6/15/86 to 8/6/86; Analymis by Environmental Terting and Certification

5 10 00212

Limit Limi	Sample 1.0. :	Petertine	<u> </u>	8 *	¥ :	\$100	<u>8</u>	¥ 2	81 -	25	<u> </u>	8 -
Fig. 7.5		Verection Limit Concentrat	21-73 s	2 H	E S	ž ja	E a		- 2 2	E-a		= 7
70-124 15000 10 10 10 10 10 10 10 10 10 10 10 10		(#E/EE)		; ;		ļ						
Chical 15000 10	PCB-1242	3800	9	9	9	· 9	9	9	9	9	9	•
79-1212 15000	PCS-1254	3600	9	9	9	9	•	<u> </u>	œ	9	皇	•
Third 15000 No. 100 No	121-DA	36000	9	9	9	9	9	9	9	9	9	•
Chicket 15000 150	22-122	3600	9	R	9	9	9	•	9	9	9	<u>.</u>
25-154 15000 NB	-12-12-12-12-12-12-12-12-12-12-12-12-12-	3600	9	오	•	9	£	9	2	9	9	9
The color of the c	PCS-1256	1800	9 !	2	9	2	9		9	9	•	Q
Present 1700 150	Portificate	9009 9800	A 9	2	9 9	9 9	9 9	9 9	9 9	9 9	9 9	9 9
Present 1100 180	ide											
phenol 1100 NB												
pleco 2700 180	2-Calorophenol	800	9	9	9	9	9	•	9	9	9	9
Check 1700 180 1	(-Dick lorophesol	2700	9	9	9	9	9	g	•	9	9	9
phenol 1500 NB	(-Dimthelphano)	3 5	9	1 5	9	2 5	? \$		1 9	2 6) 5	1
Principal (1000 Mar 198 Mar 19	Torond (Practice)	3 3	2 9	2 9	2 9	2 9	2 9	• 1	2 9	a 9	P 9	a 9
Principal 15000 MB	Unitro-o-cresol	20017	2 9	2 9	R 9	2 9	2 9	n 9	R !	9 9		9 (
Spice 1500 190 1	, 4-maicroplesol	2002	R 9		2 !	9 (9	Q !	2	e	2
Copyest Copy	2-fitroppesol	899	A	e	9	오	9	9	9	9		9
Penal 1500 180 180 180 180 180 180 180 180 180 1	· 4-Witrophesol	8	9	9	=	9	9	#	•	9	_	•
Phenol 1500 150	Cloro-cresol	2000	2	2	9	9	9	•	•	9	•	
Phenol 1500 180 180 180 180 180 180 180 180 180 1	etach [orroheno]	388	9	g	9	9	•	•	•	g	g	•
Spiesol 2700 Mg	Nesso]	1500	9	9	9	9	9		9	9		
trimory 12000 21000 22000 7500 10 21000 110000 110000 110000 110000 110000 110000 110000 110000 110000 110000 1100	-Tricklorophesol	2728			9	9	9		9	9	9	. 🕿
Astimory 12009 24008 25009 1504 10 21009 15009 15009 15009 15000 15009 1	betals & Cruide											
Lesenic 1000 MB MB MBC, ND MBC, 1000	Antianay	12000	24000	90007	- 20	9	2000	2001	1600	90	Ž	3
Beryllium 42 NBBL 54 155 NBBL NBBL <th< td=""><td>Lraenic</td><td>900</td><td>9</td><td>•</td><td>2</td><td>2</td><td>9</td><td>Ä</td><td>8</td><td></td><td>9</td><td>,</td></th<>	Lraenic	900	9	•	2	2	9	Ä	8		9	,
Cadmium 120 H00 1500 51000 51000 120	Beryllius	2		.	2	9	240	Ħ		5	=	
Chronium 1300 (45000 51000 5300 BBCL, 15000 22000 75000 13000 15000 Copper 1000 (45000 61000 7200 15000 22000 1500	Cadmium	97	•	g	9	ğ		•	•	9	"	
Copper 1000 45000 61000 17200 15000 21000 15000	Omerine	8	TRABE		3	į	3	1		2	, E	
Lead 4500 6000 4500 100	Tours.					1 2 E						
Nickel 720 100 1		Wy	5	7								
Nickel 720 2300 4500 1300 3800 3500 1500 4100 3800 3500		<u> </u>	<u> </u>	2		ş	2			3	!	}
Selection 500 MBCL MBC	i afoir	<u> </u>		! !	<u> </u>	2 }		į	1	8 8		, 3
Silver 1500 No. 100 No		9			3	9		•	•		5	<u> </u>
Tailine 500 886, 10 86, 10 10 1000 1200 1100 1400 1400 1500 1500 1400 1400 1500 15	Si lan	931	•	•	•	: S	9	1 9	? 9	5	•	2 9
Line 500 1200 1400 1500 1500 1200 1400 1400 1500 1500 1400 1400 1400 14		8	· }	1 5	ì	1 5	1 9	9	2 9	1 9	2 6	ł s
1		3 5		2	3	2	. <u>\$</u>		2	2	3	, 5
(Sptas) 1 100 100 100 100 100 100 100 100 100		3	9	•	,	•	<u> </u>	,	•	•	•	} •
C (Sptas) 1 150 150 150 150 150 150 150 150 150 1	chartee	ţ	ì	R	R	1	2	•	2	2	•	R
C (Sptas) 1 100 100 100 100 100 100 100 100 100		(24/24)										
(Sutan) 1 50 50 50 50 50 50 50 50 50 50 50 50 50	Total (Batan)	-	•	9	•	9	•	9	S	5	•	5
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ple L.D. : -wescrption :	Detection Limit Concentrat		6106 50 OCAP-13 Bup	\$105 97 6 insk	\$110 1 CCAP-25	6116 † 177-23	1110 L 107-23	Silo Ii Blank	8116 16 UTP-18	8[]6 } TP-32	61(6 26 217-35 Bup
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APPENDIX III SITE SPECIFIC ANALYTE RESULTS

Mercury found (ng/kg) (Detection Limit 0.1 ng/kg)	Ē	Mercury found (ng/hg) Selection Limit 0.1 ng/hg)	Ş.	Heroury found (mg/hg) (Detection Limit 0.1 mg/hg)	Call	Hercary found (ag/kg) (betaction Limit 8.1 ag/kg)	\$	lecury (Oloride	Ei ocel Lasseus	BPC (bytas) betylate (brtas) benylate (bernas) benslate (billias) behslate (billias) bilaste (briras) Opticate (bi-Bect)	Biogramates	Copped	Sample 1.3. : Secription :
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APPENDIX IV FIELD QUALITY ASSURANCE SPIKE RESULTS

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Sample 1.B. :	Mescrytton :		Cospound	Action Contract Contr	hints beautional tables of a	d-Branchest should skill the	Intel bears shihalate 11.0	2-Calorenaphthalene 2.	4-Chlorophenyl phenyl ether	Chrysene 2.1	Dibeato(a,b)anthracene	1, 2-91ch lorobensene	1,3-Bichlorobencae	1.1'-Bichlorobearidine 19.6	Diethyl pathalate 11.	Disettyl phibalate 11.	Di-n-butyl patholate 11.	2,4-Diaitrotoluene	Z,6-Dimitrotolucae	1 2-m-octyl pathante 11.	Placenthese	And	Betach lar obestene	Reserved to colonical seas	Bezachlaracyclopeatadicae	ktuch jerrethae	Indexo(1,3,3-c,c)pmene	lange one	Mitrabeateae	Hitragodinetty lange 11.	Witresedi -t-propylanine 11.	M-Hitrocodiphenylasine 2.1	Pesasthrese	Pyrene	1,6,4-1010E010E01CBC	Aleka-MC	Petr-MC 5.	Game-1900 11.0	De Lu-BEC	Clordase 11.	100-10's	100 M	100 - 1.1 1.15 - 10	Endosulfas (11.6	Industrian 11.0	Indosulfan sulfate 6.4	Endrin 11.0

Cold Gren,

Description :	Approximate Detection	8105 11	8105 8	1105	- 4		0105 10	0105 10	8105 10		8105 12	8105 12
	Linit	CCH-1-1	00 1 -1-1	001-1-1	Spile		001-1-L	001-1-L	CON-1-1	Spike	001-1-1	CCN-7-1 Blank
	Concentration	lint		geli	Coac	•	Spike Found	Spi b e Hel	Spike 3 Recovers	Ceac.	Binak Spike	Spibe
Compound	(m/1)						7488		y secondil			aproc 1 Becovery
Andrin aldebyde	11.4	10					•					. mw.,
Beptach lot		5	5	-	101	1.0	71.1	71.2	121	107.0	49.1	51%
Reptachlor epoxide		5	-	5	-			•		-		•••
PCB-1242		-	-	-			-	ñ			10	
PCB-1254		-	-	-			-	-			-	
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PCD-1232		3	5	5			5	-		_	5	
PCB-1248		-	-	2	_		5	-		_	-	
PCB-1260			-	3	_		-	-				
1C3-1016		-	-	-	-		-	5			5	
Totaliese		-	5	5	_		-	- =		_		
lattheene	11.4	-	-	-	Ţ.		_	_			-	**********
								Average	ie		Average	213
Acids								*********	-			***
2-Chiorophenol	3.8	10			100	6.6	103.0	103.0	975	166.0	101.0	951
2,4-Dichlorophenol	3.1	10	10		100	5.8	111.0	111.0	1655	106.0	105.0	100%
2,1-Dinethylphesol	3.1	10	10		100	1.1	41.1	41.1	395	106.0	87.2	123
4,6-Dinitro-o-cresol	28.0	ii)			\$2	1.1	\$25.0	525.0	1005	\$26.0	546.0	1942
2,4-Dinitrophesol	(8.6)	10			310	5.6	331.0	331.6	105E	316.4	335.0	106X
2-Hitrophesol	4.1		. 🖿		100	5.8	30.7	30.1	363	106.0	92.1	873
4-Hitrophenol	1.8			10	521	6.6	344.0	144.0	27%	526.0	220.0	123
p-Chloro-a-cresol	3.4				52/	6.0	\$12.0	512.0	975	525.0	432.0	12%
Pentach lorophenol	4.1	10			52	6.8	121.0	424.6	112	526.0	441.0	845
Piesel	1.7				10	6.0	59. 1	59.1		166.0	\$6.0	232
1,4,6-Tricklerophenol	3.1	ID			310	6.0	275.0	275.0) 87L	316.0	259.0	851
									***************************************			************
								pactate	36 6		Average	141
Metals & Cynaide	_											
	Variable	10,(196	m ,(190	10 ,(196			189L, (199				MOL, (196	•
	Variable	ID, < 10	31.1	35.0	•	1.1	130.0	102.0		18.0	M. 0	
Beryllius		10 , (.57	m, c.51	10 , (.\$7	• •	1.0	101.0	101.0	•	101.0	100.0	
·	Variable	15 , (4.3	m , (1.1	MDL, (1.3	-	5.6	91.0	\$6.0		95.0	97.4	•
	Variable	10 , (19	MBL, (19	10 ,(19		5.8	110.0	110.0		115.0	110.0	
•••	Variable	10 , (11	100L, (18	10,01	-	4.5	8.8	96.6		94.0	91.0	
	Variable	m, (\$\$	10,64	10,658	-		MDL, (\$8	100L, (56		103.0	-	(60%
	Variable	10 , (.2	D , (.)	10, (.2	•-	6.0	\$3.0	93.6	•	106.0	110.0	
	Variable	m, (I)	21.0	mol,(11	_	2.0	100.0	83.0		92.0	97.0	•
	Variable	10,(\$	₽,6	10 , (\$	••	4.0	19.0	79.0	•	0. M)	91.0	
	Variable	10, (1.4	m , (1.1	NDL, (6.4	• •	1.1	63.6	63.0		[40.6	66.0	
	Variable	10 , (\$	■,4	10 ,45		3.0	6.0	85.0		133.0	%.0	
	Variable	15.9	120.0	45.0	•	1.0)70.8	90.1		131.0	150.0 580.0	
Cynnide, Total	t)		I		47	1.1	502.0	502.6	1051	(78.0	388.8	1171
								åvera t e			Average	332

	' le l.b. : .cacrption :	Approximate Betection Limit Concentration	8105 1 CCH-7-1 Blank	8185 8 CCH-7-1	8165 5 CCN-7-1 Bup		Spike Conc.	8185 18 CCH-7-1 Spike	8105 18 CCH-7-1 Spite	8185 18 CCN-T-1 Spike	Spike Coac.	8165 12 CCH-7-1 Black	8105 12 CCH-7-1 Blook
	Coapound	(mg/1)				•	••••	found	Bet	1 Secovery		Spike	Spike S Becovery
	Thiocarbanates					ing.							,
	SPTC (Eptem)		119	17.0	19.0	18.0	10.0	33.0	15.0	1503	10.0	8.6	861
	Butylate (Sutan)	- 1	11)	12.0	14.0	12.5	1.1	26.0	12.5	1375	9.1	1.1	1092
	Verselate (Versas)	1	118	1.1	1.0	3.9	15.0	26.6	16.1	1975	15.0	15.0	100%
	Pebulate (Tillan)	ı	110	BUBL	BUBL	1.1	11.6	12.0	12.0	163%	11.6	12.0	1035
	Holisate (Ordran)	1	25	24.0	25.0	24.5	1.0	38.6	13.5	130%	1.1	5.1	993
	Cycloate (Ro-Heet)	1	20	1.1	4.1	4.8	15.4	21.0	17.0	1105	15.4	16.0	1042
	Organophosphates												
	Fonofos (Dyfonate)		10	20	100	0.0	15.5	16.0	16.0	1035	15.5	16.6	1035
	Carbophenothion (Trithion)	1	113	110		1.1	10.5	1.1	1.1	143	10.5	15.0	1433
	Phospet (Inidan)	1	113	100	**	1.1	18.1	100	4.0	85	18.7	2.1	113
	Bensulide (Betassa)	1	110	8H9L	MOL	8.6	15.1	20.0	20.0	1325	15.1	15.0	1265
i										***********			**********
į	•								Average	1085		Lvetage	983
İ	Miscellaneous	_											
į	Hercury	1.1	10	110	11)		106.0	93.0	93.6	185	106.0	110.0	1043
i	Carbon dinulfide	1.1	1.4	0.5	4.4		27.0	42.4	42.6	1 1575	27.0	33.4	1243
	Carbon tetrachloride	0.2	. 10, (.2	10,(.2	10,(.1		34.0	61.8	61.6	1995	34.6	11.0	1298
		Coac. is ag/L			•								
i	Thiocyanate	0.4	MD, C.4	•	•		•	•	•	•	120.0	130.0	1885
1	Chloride	1.0	113	86.6	101.0		63.0	194.6	105.0	1678	16.0	16.3	1145

Sescrition:	Approximate Betection Limit Concentration	8105 74 8CC-079 blook	8105 16 8CC-019	8185 73 8CC-879 dape 18	Spike Conc.	8185 12 9CC-879 apike 10	8106 12 9CC-879 spike 18	8105 72 8CC-075 spike 70	Spite Conc.	\$165 75 SCC-079 blank spike	8105 15 8CC-019 black opite
Coapound	(uf/1)	71020				found	Bet	I Becovery		Found	1 Becovery
Thiocarbanates											
BPTC (Sptan)	1				10.€	8.2	8.2	825	10.0	8.1	875
Butylate (Satan)	1	10	10	M	9.1	1.0	1.0	995	1.1		1012
Vernolate (Vernon)	1		10	10	15.0	14.6	14.4		15.0		1001
Pebalate (Tillaa)	1	10	10		11.6	11.4	11.0		11.6		95%
Heliaate (Ordras)	l		III		1.1	1.5	1.5		9.8	1.1	1015
Cycloate (Be-Beet)	1	-	10	**	15.4	16.0	16.0	1043	15.4	17.0	110%
Organophosphates											
Fonofos (Byfonste)	1		100		15.5	16.0	16.0	1035	15.5	17.0	1105
Carbophenothion (Trithion)	1		*	10	10.5	14.8	14.0	1332	10.5	14.0	1332
Phosmet (Inidan)	l l		10	100	18.7	2.0	2.0	113	18.7	3.5	193
Benoulide (Betasan)	l	110	10	100	15.1	12.0	22.0	1463	15.1	17.0	1133
• • • •								**********			**********
							Average	961		Average	975
Bi scel laneous											
Hercary	0.2	100			105.0	10.0	10.0	675	105.0	10.0	161
Carbon disulfide	1.8	10 , (.2	10, (.)	m,<.2	27.0	11.0	38.6	1442	27.0	33.0	1225
Carbon tetrachloride	: 0.2 Come. im mg/L	-	•	•	•	-	•		-	-	•
Thiocyanate		ID , (.4	6.0	6.5	128.0	160.0	154.6	120%	129.0	150.0	1163
Chloride		,\.\	16.4	17.4	26.4	117.1	100.0		10.0		1153

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. ple 1.D. :	Approximate	¥	3	*		=		#		=	¥ .
secrption :	Detection Limit	2 1 2 2 2 2	= 1	21 - 7E	Spile	= 1	= 1	= 1	aliq.	≈ 91 · 18 °	≈ 10
Compensal	Concentration {ug/l}	1		Ē	<u>.</u>		_	Spile Lecovery		Spike	apite
Micerbastes											194934
BPTC (Betan)	_		•	•	=	=	=	#: #:	=	=:-	101
Butriate (Sutas)	_	=	=	2		=	=	1111	=	=	1171
Vernolate (Vernas)		=	=	=	. S.	=	=	.	12.0	≟	111
Pebulate (Tilles)	-	=	=	*	9.11	=	=	171	11.6	12.0	201
Belingte (Ordres)	_	=	=	2	-	=	=	101	-	=	22:1
Cycloste (Bo-Beet)	_	=	=		<u></u>	- E	13.0	=	15.4	1.	<u>=</u>
Organophospietes											
Posefor (Byforate)	_		•	=	15.5	<u>:</u>	<u>.</u>	101	15.5	1.0	101
Carbonhesothien (Trithien)	_	=	=	=	2.5		-:	Ħ	11.5	=	£
Phasact (Inidae)	_	=	=	=	T. II	-:	=	Ħ	. E.	=	E
Densulide (Detages)	_	=	•	=	12.1	==	Ξ	132	18.1	==	1111
							**	11 11 11 14 14 14 14 14 14 14 14 14 14 1			***************************************
							Arerage	¥4.		Average	Ħ
Histellaneus											
	. :	1 (1)	1).	E),C	18.0	120.0	130.0	201	105.0	150.0	101
Carbon dissifide	=		 2	3 .0	H.0	Ξ.	==	1912	11.0	Ξ	1111
Carbon tetrachloride		C), 2	3.5	:) 'g	3.0	33.0	2.	55	X.8	22.5	194
	Conc. is ng/L										
Thiscysaste	-:	•	•	•			•		•	•	•
Chleride	-:	=	::	::	=	=	Ξ:	3) E	3.5	Ξ	18 2

sescrition :	Approximate Detection Limit	8185 83 0-39	8105 85 0-39		Spike	8105 86 0-39	8105 12 0-35	8105 72 0-39	Spite	0106 09 0-35	8186 89 0-39	8105 94 0-25
Cooperad	Concentration (ug/1)		Dop 81		Cenc.	spike 83 Found	spike 83 Het	apibe 83 1 Recovery	Comc.	Black Spike Found	Black Spike I Recovery	Blank Spite by CDH Spite Conc. unknown Found
Thiocarbanates				årg.								
BPTC (Sptan)	٠,	6.1	5. 1	5.6	10.0	20.0	14.4	1445	10.0	11.0	1105	
Butylate (Sutan)	1	2.1	1.1	2.3	9.1	16.0	13.0	1513	9.1	11.0	1215	11)
Verselate (Versau)	1	1.5	1.1	1.2	15.0	22.0	18.8	1252	15.0	16.0	1873	
Pebulate (Tillan)	i	DUDL	NOL	0.0	11.6	15.0	15.0	1295	11.6	12.0	1632	10
Molinate (Ordran)	1	1.4	1.6	1.5	1.1	24.0	15.5	158%	1.1	1.1	911	E
Cycloate (Bo-Beet)	1	2.6	1.9	2.3	15.4	22.0	19.4	1285	15.4	15.0	915	10
Organophosphates												
Ponofos (Byfonate)	. 1			1.1	15.1	22.0	22.0	1425	15.5	17.0	1105	10
Carbophenothies (Trithies)	1	10		0.0	10.6	22.0	22.0	2161	10.5	1.6	821	110
Phonnet (Inidan)	i	10		0.0	18.1		1.0	20	18.7	2.1	14%	TD
Bengulide (Betagan)	i	DEDL		0.0	15.1	21.0	21.0	1395	15.1	16.0	1065	10

							Average	1332		Average	94%	
Hiscal lancous												
bercury	_			1.1	105.	110.0	110.0	1055	105.0	100.0	951	5.0
Carbon disulfide	1.1	ID, (30	10,46	1.1	27.0	B 10,46	0.6	95	21.0	12.1		ND, (.3
Carbon tetrachloride		854.8	1310.0	1002.0	34.0		228.4	6715	34.0	31.2	921	22.0
	Cooc. in ag/L											
Thiocyanate		•	•	-	•		•	•	•	-		-
Chloride	1.0	19.1	64.8	60.3	10.0	137.0	16.1	1105	61.0	17.2	1155	•

APPENDIX V FIELD QUALITY ASSURANCE BLANK RESULTS

Lettings 11/15; field blank Besits ; Analysis by Orvironmental Texting and Certification

8106 8100 8100 8100 8100 8110 84 15 15 84 71 97 N	blank blank blank Blank Blank. Blank	Condensed Field Blank Brentts for Total Priority Pollutant Analysis	ach or le	bal ball ball ball ball ball			
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<u>=</u> =	3			Belon	2312	22	*******
Sample [.D. : Descrytion :			ł	Volatile Composeds	Alersten Styles Alerie Palese Trichlerethyles	Base/Bettals & Betlicides bis(2-ft)ylkey/lydbalste Bi-e-bety/ pibalste	Metals & Cymide Armelic Beryllius Cadaius Caronius Caronius Caper Lead Berour Bideri Bilber Ellier

C-14 Greek/Lelloyne E1/75; Field Blank Bewits for Site Specific Parameters ; Analysis by Movincamental

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APPENDIX VI SAMPLING PROTOCOL, ANALYSIS AND QUALITY ASSURANCE

APPENDIX VI

Sampling:

Protocol

All sampling activities were carried out according to the protocols in the Sampling and Analysis Manual prepared by Stauffer Chemical Company specifically for this investigation, "Remedial Investigation/Peasibility Study Cold Creek/LeMoyne Site". The manual details the basic sampling principles used throughout the investigation, such as: the sample collection procedures, the sampling devices, sampling containers, preservation, holding times, and decontamination procedures before and during sampling. It also details the required quality assurance and quality control procedures, such as: frequency of collection of field blanks, field spikes, field duplicates and the chain of custody procedures used.

Sampling Plan

The proposed sampling and analysis for this investigation were detailed in the final work plan prepared by Camp Dresser and McKee, Inc. (CIM). In the work plan this portion of the investigation was divided into the source characterization and an area characterization. The source characterization was further subdivided into three categories: 1) Cold Creek/ LeMoyne Swamps, 2) Ponds, Lagoons and Area Wells, and 3) Landfills. The samples to be collected for each of these facets of the investigation are outlined in six tables in the Mork Plan. These tables have been reproduced and included as Appendix I-1.

Actual Samples Collected

Samples were collected from the beginning of May 1986, through the beginning of August 1986. Samples were not collected in the order outlined in the sampling plan, but were collected in batches in which samples requiring similar analyses were grouped together. This was done to maximize the efficiency of the use of field quality assurance samples. A chronological list of the samples collected is given in Appendix I-2. In these tables the field notebook sample reference number, the sample identification from the sampling plan; the date the samples were shipped for analysis, and the analytes determined are listed.

All the sampling was done according to the requirements of the Mork Plan excluding the exceptions noted for the samples listed below.

- N4-1 Cyanide was not determined in this sample because the sample bottle broke in shipment.
- CVA-8 This Courtaulds' production well was not sampled because it was inoperative due to a collapse.
- HIP-ls This bore hole was not continued past ten feet because of concern that lower samples would be contaminated from the sludge encountered nearer the surface. Instead, a 45° angle bore hole was sampled adjacent to HTP-is. This extra hore hole was labeled HTP-3s.
- sc-52. The contract laboratory analyzed these two samples for carbon tetrachloride SCC-78 in addition to carbon disulfide. The results for carbon tetrachloride, not detected, were reported although the analysis was not requested.
- 94-01-1w. The Mork Plan indicated that all four of these samples were water samples. SH-01-2w. Discussion with Camp Dresser and McKee at the time of sampling clarified

SH-11-1w, that they were intended to be one water and one surface sediment at both SH-11-2v SN-01 and SN-11.

. Bore Holes Many of the bore holes had to be relocated due to access problems.

Analysis

All analyses were done by Environmental Testing and Cartification, Inc. Analytical methods used are summarized in table 1. Where available, methods were from SM-846, EPA 600 Methods, "Standard Methods for the Examination of Mater and Mastewater", or from other EPA referenced sources. Two groups of site specific parameters were not done by methods from the sources listed above. The thiocarbamates and organophosphates were analyzed by a Stauffer developed procedure because no other documented procedure was available. Carbon disulfide and carbon tetrachloride were analyzed using a modification of EPA Method 8240 (GC/MS) which utilized selected ion monitoring. This was done to provide lower detection limits than the standard method allows.

The results of analysis were reported by ETC to Stauffer with full documentation for each sample, with a separate report in the form of a booklet for each individual sample. The results reported by ETC include documentation of the chain of custody, internal laboratory quality assurance results, and copies of chromatograms and other chromatographic data. This documentation is on file at ETC and at ETC.

The results provided by ETC have been compiled and are in Appendices II through V. Appendix II contains the results for the total priority pollutant analyses and a few site specific analyte results. Appendix III contains the remainder of the site specific results. Appendices IV and V contain the results for the field quality assurance spikes and blanks, respectively. In the actual analysis, there were two variations from the methodology required in the stated procedures.

In some cases, the priority pollutant GC/MS analyses were done by variations of the SM-846 methods which were more similar to the Contract Laboratory Procedures than SM-846. The variations from SM-846 were to use a larger sample weight, or to concentrate the methylene chloride extract further than required by the SM-846 methodology. Each of these variations results in lower detection limits than obtained using strictly SM-846 methodology. These variations in methodology are noted in the tables of results by the different detection limits indicated for the GC/MS work.

For three site specific categories of analytes (thiocarbamates, organophosphates, and elemental phosphorus) the required holding times were exceeded due to problems which required reanalysis of retained samples. Field quality assurance spike results indicated that this may have caused low recoveries for one of the organophosphate posticides. This is discussed further in the quality assurance section.

Quality Assurance

The general quality assurance procedures used for this project are described in the Sampling and Analysis manual. In the manual, the equipment cleaning procedures, sampling procedures, preservation procedures and frequency of field quality assurance samples are given. The laboratory quality assurance procedures used by Environmental Testing and Cartification are indicated in the complete reports provided by ETC.

The general frequency of quality assurance samples in the field and in the laboratory were the same. For every batch of approximately fifteen samples, there was a field and laboratory blank, a field and laboratory duplicate and a field and laboratory blank spike and sample spike. For soil samples, field spikes were cuitted. All of the field quality assurance samples were blind to the laboratory.

Field Quality Assurance Spikes

The results for the field quality assurance spikes are given in Appendix IV. There was one set of field quality assurance spikes for total priority pollutant analysis. A total of 57 analytes were added in the blank spike and sample spike. The overall recoveries for the two spiked samples were very good. There were some analytes with low recoveries, but in all cases these were for compounds known to be unstable in water or unstable to the conditions used in the analysis. The compounds known to be unstable are indicated in the October 26, 1984 Federal Register in which the EPA established acceptable quality assurance spike recoveries under the Clean Water Act.

There were 4 complete sets of quality assurance spikes for the site-specific parameters. The analytes spiked depended on the particular analyses required for the batch of samples being analyzed. In general, the recoveries for the site-specific spikes were very good. There was one case of no recovery for carbon disulfide in a sample spike. This was caused by interference from the high concentration of carbon tetrachloride, 1310 ppb, in this sample relative to the concentration of carbon disulfide spiked, 27 ppb.

Recovery was poor for Imidan for all 8 of the spiked samples. Recovery ranged from 0% to 19% with an average recovery of 10%. This poor recovery was probably contributed to by the fact that the samples for organophosphates, thiocarbanates and phosphorus were analyzed after Stauffer's recommended holding time of 7 days. It is not known what type of recovery could be expected if the holding time had been met. Hydrolytic stability data from Stauffer's Western Research Center shows Imidan to have a half life of only 1 day at 20°C and a pH of 7.

Field Quality Assurance Blanks

One field quality assurance blank sample was analyzed with every batch of approximately 15 samples. The quality assurance blank results are in Appendices II and III along with the actual sample results. The blank results are also summarized in Appendix V.

There were a total of 11 field blanks for total priority pollutant analysis. Four common laboratory contaminants (methylene chloride, toluene, bis(ethylhexyl)phthalate, and Di-n-butylphthalate) were found in one or more blanks at concentrations near or below the method detection limits. These four compounds were reported in the actual samples at a similar frequency and concentration. Any reported detections of these compounds at low levels should be considered invalid. It is recognized by the EPA in the Contract Laboratory Procedures menual that contamination by these common laboratory contaminants is unavoidable.

Methylene chloride was found in the water field blank and field blank spike at an elevated level, about 500 ppb. This was found to be due to contamination of a bottle in which the sample was prepared. This contamination of the field blank and blank spike had no impact on the actual samples results.

Two other organic compounds were detected in the field blanks; chloroform at a level below the method detection limit in one sample and trichloroethylene near the detection limit in another sample. These detections of miscellaneous compounds near or below the method detection limits support the view that any detections near the method detection limit should be considered suspect and any detections below the method detection limit should be considered invalid.

There was one detection of cyanide at 1.9 ppm, about 4 times the detection limit, in one of the 11 field blanks. There were two actual samples in which cyanide was detected at 4 to 6 times the detection limit. These detections may be questioned based on the quality assurance blank results.

There were two priority pollutant metals for which detections were reported at levels above the method detection limits in one or more of the 11 field blanks. Copper was reported in 3 solid matrix blanks at 4 ppm, 9 ppm and 15 ppm. Zinc was reported in all the blanks at from 1.5 ppm to 8.8 ppm. Based on these blank results, a reported concentration of less than about 10 ppm should be considered a probable blank level for these metals.

Fourteen field blanks were submitted for site-specific analyses. Carbon disulfide and iron were the only analytes detected in any of the blanks. Carbon disulfide was detected in two of seven samples at 0.4 ppb, twice the detection limit. Iron was detected at 100 ppm in the one field blank submitted. This reported concentration of iron is insignificant compared to the percent levels found in the actual samples.

Field Duplicates

The results for the field duplicates were acceptable. Most of the results for duplicates agreed within a factor of two. Except for one duplicate pair, any apparent disagreements were for detections near the method detection limits. In one case, carbon tetrachloride in sample CCS-2s, there was an order of magnitude different between the duplicates. This is attributed to the difficulty with providing good homogenization of a soil sample whilstill maintaining the integrity of the sample for volatiles analysis.

3 10 00248

Cold Creek/Leftsyne EL/FS Sampling and Assignis Requirements

أعمة	Tes					Preservative		
			Methods -			All soils cool 4oC, Aquacous below:		
Category	Sub-Category				Solding fine			
		Voter	Soil	Technique	(Yeeks)			
Priority Polli					_			
	Extractables	3510/8270	3540/2870	CC/E	1	Ph 6-9, .0085 thiosulfate if rasidaal C12, cool		
	Volatiles	5030/8240	5030/8240	CC/188	2	0.008% thiosulfate if residual Cl2, cool 4oC		
	Cymaides (Technicon II G Netals	P) 9010	9010	Colorimetric	1 -	Made pibl2, 0.6g ascorbic acid if raidual Cl2, o Filter through 0.45 on filter, ESO2 to pil (2		
	Antimony	6010	£010	ICAP	24	ritter directs v.45 as litter, mas to parts		
	Armic	7060	1060	Parmace	24	•		
	leryllien .	6010	6010	ICAP	24	_		
	Cadaine	6010	6010	ICL	24			
	Oronion	6010	6010	ICLP	24	•		
	Copper	6010	6010	ICT	24			
	Local	6010	6010	ICAP	26	_		
	lle rours	7470	7471	CAV	•	•		
	Heroury in Fish	(1)	(1)	CTAA	ì	Presse with dry lice		
	Mickel .	6010	6010	ICAP	24	•		
	Selector	7740	7740	Paraece	24			
	Silver	6010	6010	ICLE	24	•		
	Tailiu	1840	7240	Parmece	24	•		
-	Line	6010	6010	ICLP	24	•		
!aneous :	Mars e							
: 45504	Iros	•.	6010	ICAP	24			
	Vanadiua	•	6010	ICAP	24	•		
	· C		****	•	•,			
Miocarbances	and Organophosphates	SCC-10C-275b	900-EBC-215e	CC/MPD	1	Ph 5-9, .0085 thiosulfate if residual Cl2, cool		
Coloride		125.3	325.3	Colorimetric	4	coel 4oC		
Carbon Disulfic	de and Carbon Tetrachloride	\$638/CC/NE-STR	5030/CC/NE-STE	CC/18	t	8.008% thiosaifate if residual Cl2, cool 400		
Thiocyanate		•	412.k(2)	Colorimetric	ż	•		
Sulfide		•	376.2(3)	Colorimetric	4	•		
Sulfite		•	377.1(3)	fitrimtric	4	•		
Selfate		•	375.3(3)	Graviantric	4	•		
Elemental Phos	-hama	_	(4)	CC/FF9	1	•		
PICEMENT UNDE	hav. an	•	(*)	W/ 117		-		

1; 37-446 unless noted

(1) Interim Methods for the Sampling and Analysis of Priority Pollstants in Sediment and Pish Tissue U.S.S.P.A. MREL, August 1977, revised October 1988

Standard Nothods for the Erminetics of Sater and Sastomter, 15th M.

- (3) Bethole for Chemical Analysis of Sater and Sactor, EP1-500/4-79-429, revised Sarch 1963
- (4) "Birect determination of elemental phosphorus by gas-liquid chromatography" 2.F. Addison and 2.S. Achon, Journal of Chromatography, 47 (1970) 421-425

APPENDIX VII THOMPSON ENGINEERING TESTING INC., REPORT ON NEW WELLS

THOMPSON ENGINEERING TESTING, INC. 2

Geotechnical, Materials and Environmental Engineers + Laboratories

Thompson 3707 Cartage Hill Road • 205/666-2443

Nondestructive Testing and Examination 4234 Halls Mill Road • 205/666-1435

P 0 Orawer 9637

Mobile Alabama 36691

May 12, 1986

Stauffer Chemical Company P.O. Box 100 Axis, Alabama 36505

Main Office and Laboratories

Attention: Mr. William P. Stilson

Groundwater Monitoring Well Installation

Lemoyne Plant

Gentlemen:

Thompson Engineering Testing, Inc. has completed the installation of groundwater monitoring wells at the site of the Lemoyne Plant, Stauffer Chemical Company, Axis, Alabama. Authorization to proceed with the installations was issued by Mr. William Stilson on April 18, 1986 following site reconnaissance by our engineering staff. Reference is made to our proposed schedule of fees dated March 3, 1986, and Stauffer Chemical Company purchase order No. 25-384442 dated April 22, 1986.

This report presents well installation procedures, groundwater monitoring well construction details, and results of well completion. Groundwater monitoring well installation was directed on-site by Mr. William Stilson, Geologist for Stauffer Chemical Company.

Two (2) groundwater monitoring wells were installed at the subject site at locations specified by Mr. Stilson between 37.0 and 62.5 feet below existing ground surface. Each well borehole was performed using the mud-rotary drilling technique and BW drill rod with adapted 9.0-inch and 7.88-inch diameter three-winged drag bit. Slow advance of the drill stem enabled visual classification of soil material and approximate determination of depth at which the underlying soil strata were encountered.

Confirmation soil samples were obtained within the well-screen interval and in natural sands using an 18-inch by 2-inch 0.0. split-barrel sampler driven by a 140-pound hammer falling a standard height of 30-inches. The number of blow counts recorded for the final 12-inches of split-barrel penetration are reported as the standard penetration resistance and provide an indication of cohesionless soil relative density and cohesive soil Weight-of-Hammer (WOH) and Weight-of-Rod (WOR) are consistency. reported in lieu of blow counts where soils sampled are very loose or soft.



Page Two Stauffer Chemical Company May 12, 1986

Following borehole completion, the pre-assembled well casing was lowered into the borehole and subsequently flushed with pumped clean water to displace downhole drilling fluid.

Well casings consisted of 4-inch I.D., Schedule 40 PVC having "Tri-Loc" flush threaded couple joints manufactured by Brainard-Screen intervals have 0.010-inch width machined slots spaced on 0.25 inch centers resulting in an intake area of 5.16 square inches per linear foot and rated flow capacity of 1.60 GPM per linear foot under ideal conditions. The annular space between well-screen sections and borehole were sand-packed commercially available well-rounded and uniformly graded (20/40) silica sand where borehole sidewalls remained stable. The particle size distribution curve for the sand-pack material is provided in the attached information. The remainder of the borehole annulus was completed with a cement-bentonite grout Bentonite Pellets, 1/2-inch in tremied from bottom to top. diameter, were used to create an impermeable seal within the annulus and minimum of 24-inches above the screen interval. A 4foot square timber form was placed about the well riser for emplacement of the concrete collar.

At the request of our staff, a 100 CFM air compressor and the "Air-Lift" method was used at an attempt to evacuate and develop the groundwater monitoring wells. However, due to the semi-submerged construction of the well screen interval, this method was unsuccessful.

As an alternate method of well development, a swab surging tool and electric submersible pump were used. Continued use of the surge agitation-withdrawal technique for 14.5 hours resulted in clear water discharge and continuous pumping rate of 5 GPM in Well No. NM-1. Using a similar technique in Well No. NM-2 for 7.6 hours, a clear water discharge and continuous pumping rate of 8 GPM was achieved. Based upon well recovery rates and computation of permeability of natural sands per U.S. Department of The Navy, Naval Facilities Engineering Command, 1974, values greater than 1.65 X 10-4 cm/sec compare favorably with estimates for the soils encountered and having particle size distributions as provided.

It is noted that during the surge agitation-withdrawal technique on NM-1, electrical connections to submersible pump became detached from their bindings and allowed an unknown amount of pump motor lubricating oil to discharge into the well water. Upon its recognition, no further development was performed until a replacement motor was obtained. Our staff is confident that leakage was minimal, and assurance was made by our field crew to evacuate the well volume prior to additional surge agitation. Our firm retains a sample of the pump-motor lubricating oil for future reference should the analyses of water require comparison.

3 10 00252

Page Three Stauffer Chemical Company May 12, 1986

Thompson Engineering Testing, Inc. has appreciated the opportunity to provide Stauffer Chemical Company with our groundwater well installation services. Should there by any questions regarding the information presented herein, please do not hesitate to contact us.

Sincerely,

THOMPSON ENGINEERING TESTING, INC.

E. Fletcher Thompson Geotechnical Engineer

EIT No. 8830

EFT/cw



10 00253 THOMPSON ENGINEERING TESTING, INC. ENGINEERS TESTING LABORATORIES BILOXI, MISSISSIPPI

GROUNDWATER WELL LOG

CLIENT: Stauffer Chemical Company

GROUND ELEVATION: 44.5

-46.91°

COM SING ELAN.

JOB NO.:

E86-059

PROJECT Lemoyne Plant

DATE DRILLED: 4/23/86

GR. WATER DEPTH: 48.01

(Below Ground Surfa

JORING NO .: LOCATION: \$, 1984,53' E, 5769,92' TYPE BORING. ASMT 0-1586 WELL BLOWS PER FT.

	IN FEET	LOG	BAMPL	DESCRIPTION	NO.	MO		10	GRAF	PH 10 40 54	INSTALLATION DETAIL
Ī	=										
	0 —			Medium brown SANO Medium tan and gray SANO							Concrete
	5 —			Fine tan SILTY CLAY							Cement/ Bentonite Grout
1	10 —			Medium red SANO w/traces of fine CLAY	•						J J Gradi
1	15 —	. · · . · · . · ·		Medium tan SANO	:						
	20 —										
1	25		-	Fine orange SAND							
1	30 =										Bentonite Seel
	35		-	Fine tan SILTY SAND							
	40 _										20/40 Sand-Pack and
	45 -										Collapsed Netural Sands
!	50 -					Hend Drive	6				Sends



25

30

35

50.

10 00254

THOMPSON ENGINEERING TESTING, INC.

ENGINEERS TESTING LABORATORIES

BILOXI, MISSISSIPPI MOBILE ALABAMA GROUNDWATER WELL LOG GROUND ELEVATION: +44.5 CLIENT: Stauffer Chemical Company Casing Elevation: -46.91' DATUM: M.S.L. PROJECT: Lemoyne Plant JOB NO.: E86-069 DATE DRILLED: 4/23/86 GR. WATER DEPTH: 48,01 (Below Ground Sur LOCATION S. 1984.53' E. 5769.92' JORING NO .: NH-1 (Cont.) TYPE BORING: ASTM 0-1586 DEPTH BLOWS PER FT. WELL NO. INSTALLATION 106 DESCRIPTION 1M MO GRAPH DETAIL FEET 20 30 40 50 Fine tan SILTY SAND FIRM fine tan and yellow 20/40 SILTY SAND Sand-Pack 27 and Collapsed FIRM fine tan SILTY SAND Netural Sends 8.T. @ 62.5" 10 15 20



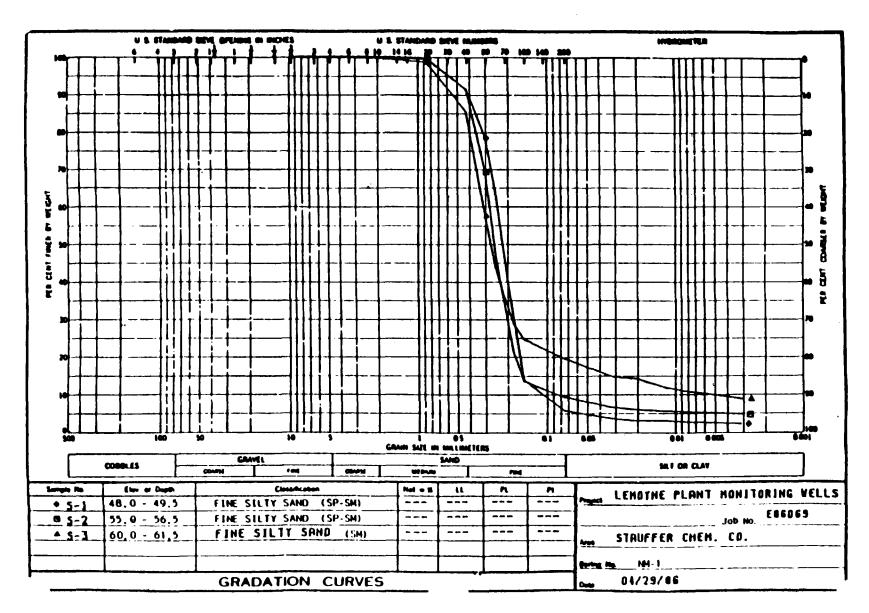
THOMPSON ENGINEERING TESTING, INC. ENGINEERS TESTING LABORATORIES BLOXI, MISSISSIPPI

MOBILE, ALABAMA

~ ~~~ ~

GROUNDWATER WELL LOG

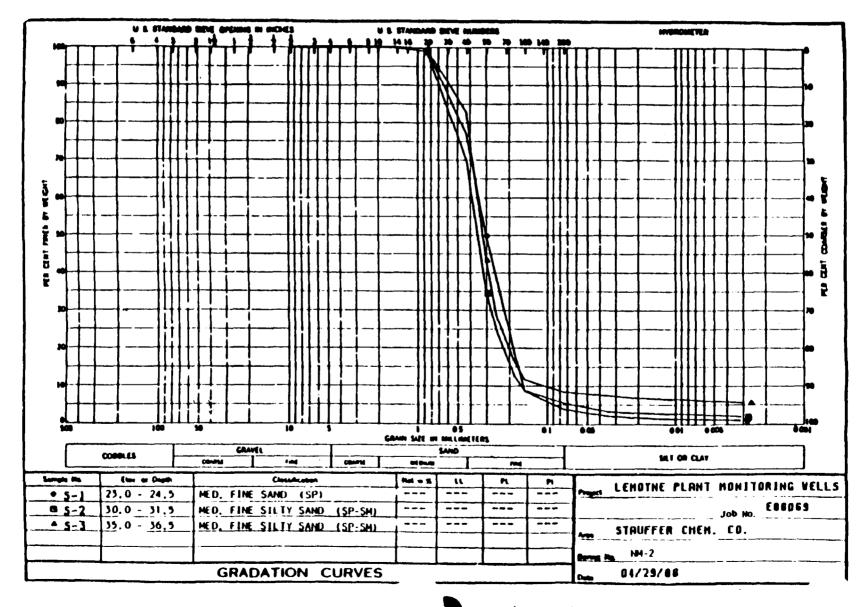
Stauffer Chemical Company GROUND ELEVATION. +21 CLIENT: Casing Elevation: +24,91' - 27, 4' CLAIC. PROJECT Lemoyne Plant DATUM: M.S.L. JOB NO .: E86-069 DATE DRILLED: 4/23/86 GR. WATER DEPTH: 24.31 (Below Ground St. LOCATION: S. 1988,72' E. 6663,76' JORING NO .: M4-2 TYPE BORING ASTM 0-1586 BLOWS PER FT. WELL DEPTH NO. SAMPLE INSTALLATION LOG DESCRIPTION IN NO GRAPH DETAIL FEET . 10 20 30 44 50 0 MANUS MALEONIN STREET SAND Concrete Medium tan and orange SAND Coment / Bentonite Grout 10 Bentonite 15 20 -20/40 FIRM fine white and tan SAND Sand-Pack to SILTY SAND 11 and 25 Collepsed Netural Sands 30 2 28 35 B.T. # 37.0' 50 .



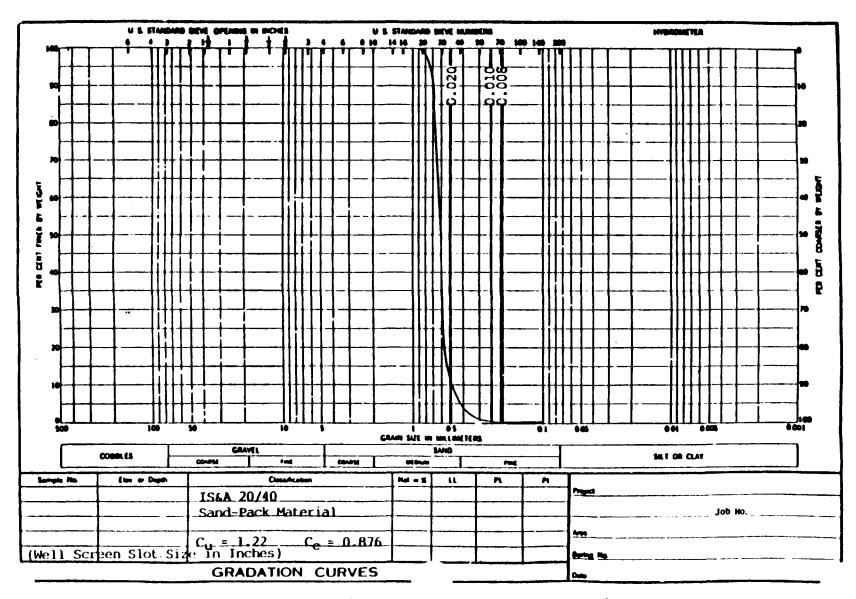
TEIP

THOMPSON ENGINEERING TESTING INC.

CHEMICAL MATERIALS AND GEOTECHNICAL LABORATORIES

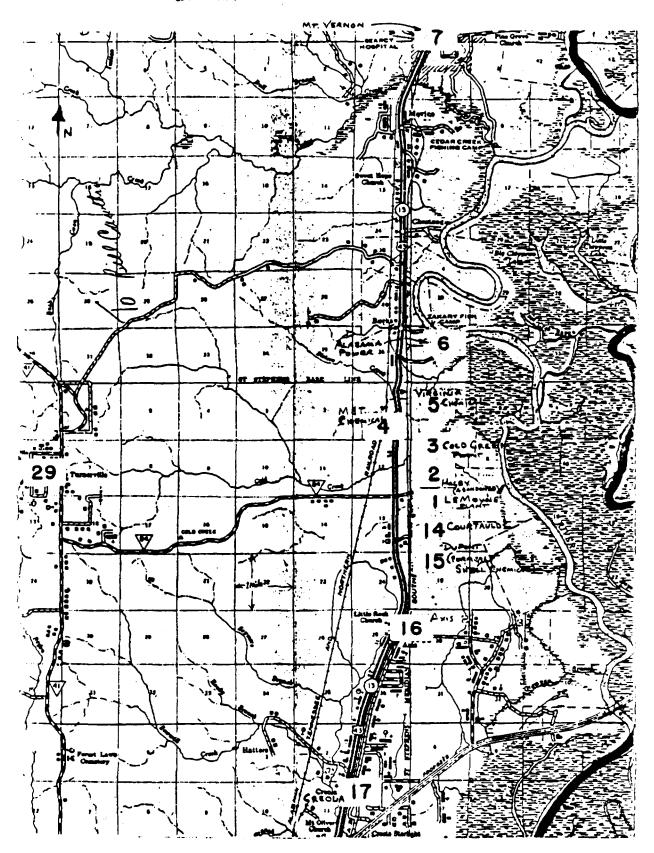






APPENDIX VIII POPULATION DENSITY DISTRIBUTION MAP

POPULATION DENSITY - DISTRIBUTION



APPENDIX IX RAINFALL INTENSITY DATA, MEMO 9/26/78

APPEND X . IX

Stauffer

INTER-OFFICE CORRESPONDENCE

4 (19)

Cold Creek

Dobbs Ferry

Bill Erdmann

9/26/78

Bill Cawthra

Rainfall (INTENSITY)

The following information on high rainfall was obtained from the U. S. Department of Commerce Weather Bureau. Local climatological data for Mobile, Alabama, Bates Field.

Rainfall exceeded 2.5" in 24 hour period in last 20 years (27) twenty-seven times.

Rainfall exceeded 3" in 24 hour period in last 20 years (26) twenty-six times.

Rainfall exceeded 4" in 24 hour period in last 20 years (9) nine times.

Rainfall exceeded 6" in 24 hour period in last 20 years (4) four times.

Rainfall exceeded 3" or more (39) thirty-nine times in past 20 years.

Highest rainfall in a one hour period recorded 2.99".

Rainfall of over 6" in past 20 years:

June 1961 6.08" Sep. 1974 6.17" Sep. 1967 6.58" Nov. 1975 7.01"

Recorded rainfall of over 2" in one hour for past 20 years:

Mar. 1958 2.11" Aug. 1963 2.06" May 1960 2.75" Apr. 1964 2.99" Mar. 1961 2.10" Nov. 1975 2.03"

Other hourly rainfall:

4.37" in 3 hours June 1961.

2.83" in 2 hours May 1978.

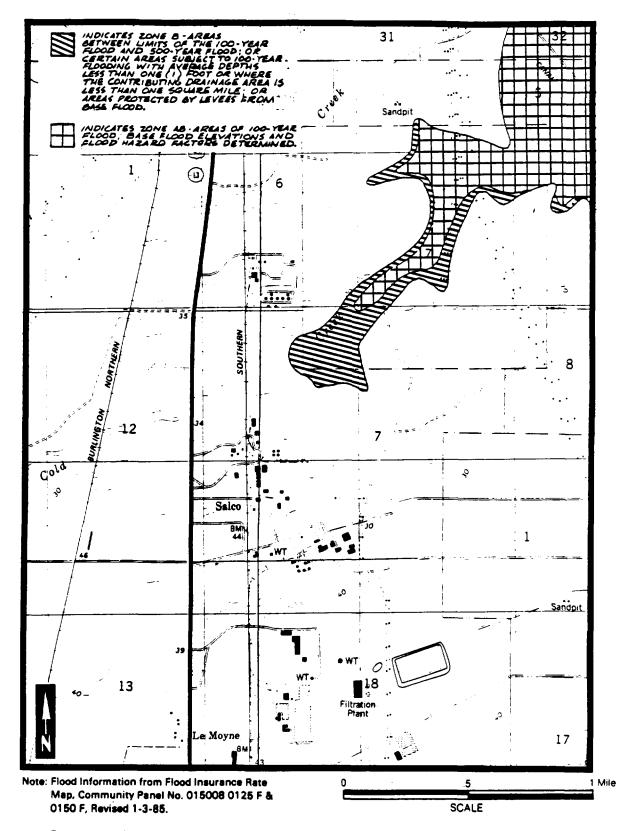
This information was taken from official records established at a location approximately 40 miles from the plant site and due to the weather pattern for this area may or may not represent a true picture of this location.

Attached is a full copy of records for the past 20 years.

SAFEGUARD COMPANY INFORMATION

Bill

APPENDIX X SWAMP AREA ENCOMPASSED BY FLOOD PLAIN

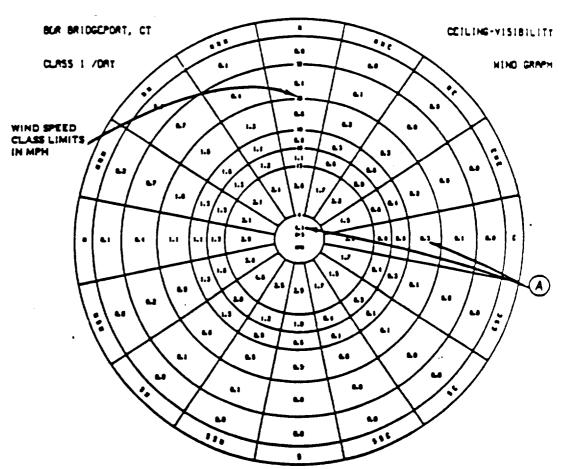


Topographic Information Taken From USGS Topo Map, 1982:

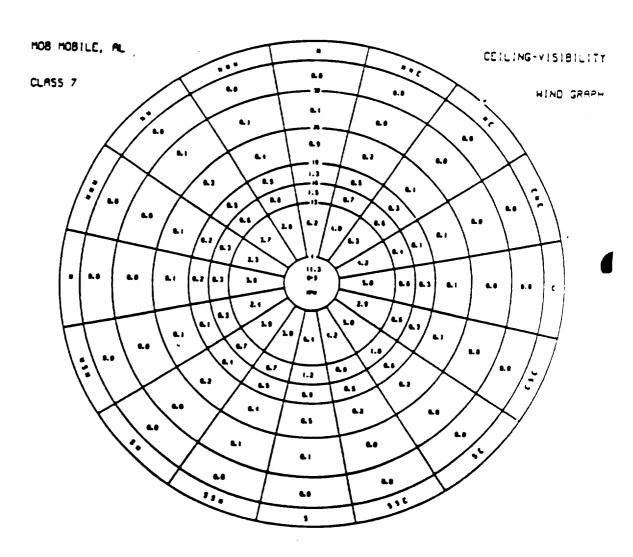
Figure 10-1 Swamp Area Encompassed by Flood Plain

APPENDIX XI WIND GRAPHS

Sample



A-MUMBER OF DAY (NIGHT) DESERVATIONS IN A GIVEN CEILING-VISIBILITY CLASS, WIND SPEED CLASS AND DIRECTION DIVIDED BY THE TOTAL NUMBER OF DAY (NIGHT) DESERVATIONS IN CEILING-VISIBILITY CLASSES 1, 3-4, ALL WIND SPEED CLASSES AND DIRECTIONS TIMES 100.

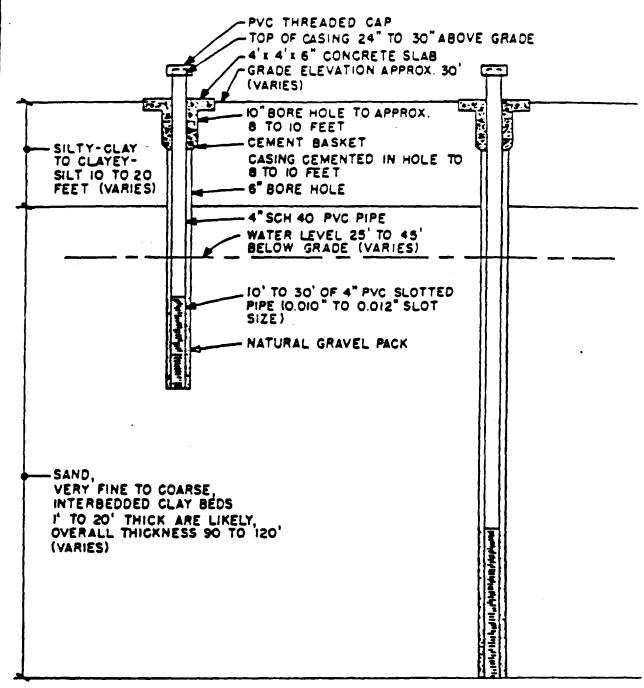


78

APPENDIX XII WELL CONSTRUCTION DATA, EXISTING WELLS

SHALLOW

DEEP



CLAY, DENSE, VERY STIFF

TYPICAL OBSERVATION WELL DESIGN
STAUFFER CHEMICAL COMPANY
MOBILE COUNTY, ALABAMA

FIGURE

A-1

SCoordinates not Established by James M. Garratt & Assoc., Inc. (consulting engineers & land surveyors). Indicates Never Elevation Data Based on Corrected (Adjusted) Base Elevation that is 0.2 ft Lover Than Older Elevation Data

MPS 4/29/85 Page 1 of 17

MELL MABER	STAUFFER COORDINATES	DATE COPLETED	TYPE OF MELL CONSTRUCTION		CASING OR HEAS. PT.	TOTAL DEPTH OF HOLE	DEPTH OF SCREENED INTERVAL BELOW GRADE	COMMENTS
	\$ 5.1940. E.1510.	Prior to	4"0.0. Steel	(G) 33.	34,68	7		*Old Hell - No construction details available. Data approximate.
0-2*	\$.2394.7 E.1558.1	7 Frior to 197)	40.0. Steel	33.76	34.97	7	7	*Old Well - No construction details available. Data approximate.
.0-3*	8.2577.40 E.1173.02	7 Prior to 1973	7º PVC	37.16	39.44		100-105 (Estimated)	*Did Hell - Construction details approximate. 5' long 2° PVC acreen.
6-1	\$ 8. 65.91 E. 794.05	84-05-73	J* PVC	31.61	32.14*	140	18-118	*Reference Pt. on top of broken casing. Has been broken further and covered with Dresser coupling and fittings.
U -5	W. 677.8 W. 023.7	04-06-73	PVC	32.60	34.92	120	97-117* 30-74*	*3" slotted PVC pipe **9-5" long sections of 4" ribbed PVC well screen (1.0. approximately 3-1/2")
0-6	8.2574.9 8.6019.6	84-89-73	PVC	41.16	42.96	136	52-126*	*15-5" Long Sections of 4" ribbed PVC well screen (I.D. Approximately 3-1/2"). Hell sampler(7) James In well screen cannot get below about 70 ft.
.8-7	5.2372.7 E.1922.6	04-10-73	PVC	39.69	41.42	134	35-45* 50-65 70-85 90-105 110-125	*14 - 5" long sections of 4" ribbed PVC well screen (1.D, approx. 3-1/2").
.0-0	N. 902.70 W. 14.00	05-02-73	PVC	32.66	34.73	133	100-115	*)-5" Long Sections of 4" ribbed PVC well screen (1.D. approximately 3-1/2")
g_gs Renumbered IM-1 by Cold Creek in 1983		05-07-73	6" Steel cemented in 12" hole to 120' 4" steel and 4" PVC	}	34.6	स्त	201-227	"Mell dies not monitor amifer(s) above "Blue Clay" installed as a monitor well for injection well INU-1, to monitor first amifer below blue clay. 20' of 4" dia. slotted PVC pipe on 206 of 4" galvanized steel pipe.
0:10*	S. N. 141.96 E. 750.85	05-09-71	I" PVC	30.70	33.00		30-60	Well was in the way of construction activity. Go

NPS 4/29/85 Page 2 of 17

		4	"Tup of PVC casing bent 6 crimped by tractor lawn	State of the state	Plugged and abandoned by Stauffer Construction in	1974 or 1975.	14-5' long sections of 4" ribbed PVC vell screen	(1.6. approx. 1-1/2")		*Coordinates approximate. Nell uss in the usy of	construction activity. Plunged and absentoned by	Cold Creek Plant in 1981.	"Conc. slab govered by several feet of fill. Casing	was extended.		Thotton of well approis. 12' higher than planned.	The of casing chancel. New elevation is alven	Old elevat fon was 45.27	Coordinates & elevation data Aproximate. Hell was	in the way of construction activity. Plugged and	There is a second of the second secon	Casho pulled agart to Jan. 1977.	Peplarment well 0.308 drilled under plant maerylain	Hell was to duplicate hell 0-20, but m amstruction	octalis wire recorder.	_
OSCALOS	INTERNAL	BELLIN CAND	X-2	97-0	;	,	-109			24-11			11-11			99-91	SIL-1's	•	P.1-93		11-11	;	Approx 55	to 057	411.44	21110
TOPAL	8		3	15	}		22			2			Q.			8	146		121		1	}	2	•	V	?
CO TOP OF	CASING OR	200	11.75	10.1			31.80			28.70			75.00			38.95	45.174		32.54		16.79		X.42		W 72	
DE CONC.	10 ens	3	3	10.00			Bolt set	de le el	2.5	18.87			25.35			36.30	42.56		(C) 79.6		35.06		34.49		71.17	
TIPE CF	TION OF	AND INC. I IUM	Ě	- MC	•		- PC			- MC			ž			- PK	to PVC		¥		- NC		MC.		JA DA	!
				85-16-73			05-11-71			06-75-75			06-26-73			06-77-73	. (4-94-30		07-02-73		10-10		14-1-10		47-01-71	
	STAIFFER	IL IT	Z. 078.16	\$ 5. 409.89	C. 666.11		M. 922.67	E. 1240.20		S H. 590	c. 635		2. X . X	E. 1/7.11		N. 200.3	\$.1655.1	w. 612.7	S 5.1210 •	K.11.2	5.7510	E.1397	5.7515	E. 1385	8. 771.79	
	7 E			613				Penathered	1964 app	0-13•			-1-		16 18 PO	0-17	91-0		- 13-		0-70		0-20A*	-	0.71	

SObordinates not Established by James M. Gerratt & Asmo., Inc. (consulting engineers & land marveyors). Indicates Never Elevation Data Based on Operacted (Adjusted) Base Elevation that is 0.2 ft Lower Than Older Elevation Data

Schoolinates not Established by James M. Garrett & Assoc., Inc. (confluiting engineers & land surveyors). Indicates Never Elevation Data Based on Oprrected (Adjusted) Base Elevation that is 0.2 ft Lower Than Older Elevation Data

		SD Gentle	*Doordinates approximate. Botton of casing at	agreementely 19'. 20' of blank pipe below acreen		Until 3/8/18. Then 13.90 (ron 3/8/18 until April- May 1981.			1-5' long sections of I' ribbed PMC well screen.	Well does not mailtor aquifer(s) above "Blue Clay". Installed as a mailtor well for injection well INJ-2, to monitor aquifer at arout \$60 to 610°. Mas 10° 55 well acreen, well is under alloht ressure - about	15 paig at autier. Well is epugged with valve and pressure game.					i i i i i i i i i i i i i i i i i i i
DEPTH OF	STICDED	DELCH CRATE	+-	,	0-0		104-178	%-% 110-130	119-134	0%: 0 5 X		21-15	111-16	98-116	111-96	
TOTAL	E a	704	2		G		Ē	Ē	E	019		\$	34	974		
CLEVATION		25.5	11.67		11.94 *		17.74	17.13	48.87	Neasur Ing Point 34.28		35.64	15.96	18:57	15.66	
ELEMPTON		(3) 7000	13.33		13.63		14.60	45.23	45.06	(6) 31.3		33.52	F. F.	X.3	17.99	-11.0
		10 L	1				- MC	2 2	to MC o		steel to 580'	₩) M	to PVC	t" PVC	17.50
	1	CITALITE	11-10-11		07-11-73		10-27-73	16-27-73	07-08-74	03-25-75		06-30-77	07-30-77	17-25-77	08 07-11	- FI - 16 - 18 X
	. STREETS	COMPINATES	8 W.1489.68	C.6777.79	6 N. 1651. 27* E. 6521. 10		S. 999.3 W. 504.3	S. 296.6 W. 405.1	S. 274.5 W. 710.0	S N. 859.67 E. 315.00		S. 2346. 3 E. 1660. 2	5.2416.4 E.1677.0	\$.7407.9 E.1498.8	5.1111.2	0 3000
		MARKE	0-23				0-25	6-36 Renumbered (CO+111by COP 1984	0-27	0-20* tementered 18-2 by Cold Oreek	(D6) u)	S-0	Q(- Q	11-0	21-0	TIL Y

NES 4/29/85 Paye 4 of 17

			_			_											•	
		COMPONE						Hop of casing changed, hey elevation is given.	old elevation was 12.81. Ref. old base.									
SCHM OF	IMERMY.	BELOW CINDE	169-129	104-124	178-16	K	\$11.56	51-52	\$F.113	69-69	109-179	36-24	99-119	П-И	106-176	16-16	100-120	\$6.78
POTAL DEFIN	8	HOLE	223	F	¥	:	133	26	120	=	F	*	1135	16	131	47	111	81.3
DE TOP OF	CASING OR	HEAS. PT.	39.55	1 25.00	77 87	:	11.51	JI.77e	M.07	15.44	37.00	40.64	35.44	05.49	18.39	19.78	11.65	11.19
ELEVAPION OF COME.	SLAB CR		17.62	M. 37	77 77		20.45	10.11	33.65	32.61	38.59	19.00	33.16	33.05	11.11	19.07	(0, 10.7	(G) 10.5
TIPE OF	PETT.	CASTRACTION	¥	to MC	- 100		to put	to PAC	JA NC	t. MC	to MC	1. PC	1. MC	1º MC	1" PVC	- Se Pac	1 PK	+ PK
	CATE		//	96-11-90	06-22-38		16-02-90	94-12-90	96-92-90	97-92-90	21-52-90	94-95-90	07-13-76	07-14-70	07-19-78	07-26-76	- 92-33-90	66-23-36
	STAUFFER		E. 929.9	5.7333.3	5.7111.6	E.1036.2	S. 2200.0	S. 270 J. Z E. 1690. 1	S. 1935.4 E. 1499.5	5.1947.0	5.1767.8	5.1770.2	5.2106.2	5.2110.4	5.1060.2	5 1601 3	S, 2105. 6 E. 1941. 8	S. 2111.7 E. 1910.0
	1 ELL.			97-9	6-33		3	£-9.	97-0	17-0	20-0	6-43	0-44	\$7-0	91-0	19-0	07.0	67-0

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				TOWNSTAN	The state of the			
7	Structure		*	0° co.	00 100 OF	ATAIL.	SCHEDED	
KPEER	COORDINATES	CONFICTION	COSTRUCTION	CHOC (C)	CASING OR	ò	INTERNAL	
S -0	N. 326.2	84-92-90	JA PC	(C) 38.5	11.12	1	105-115	CUMBALS
New Merce	6.3034.9				_			
(S) C(S)								
In 1984	•							
0-51	S. 154.6 E. 4665.2	06-71-79	24.0	(C) 45.1	47.37	F	211-28	
0-52	S. 1001.6 E. 4660.1	64-24-30	4. P.C	(C) 45.2	47.51	1	311-36	
26-2	S.1117.1 E.2160.2	11-36-79	To MC	33.78	18.61	175	164-123	*KOM monitor will upgradient of brine and pond SDE-761
245-0	S.1140.6 E.2155.3	11-27-19	P MC	35.40	35.40	101	16-19	* W.M. monitor will upgradient of brine and pond SOP-161
0-55.	S.1145.1 E.2140.8	11-11-19	4. PC	11.11	34.39	33	95-94	*ROM monitor well upgradient of bulne and pond Sep-201
9 95-0	5.1622.4 E.2205.6	6/-92-11	P.C	28.66	10.66	116.5	101-111	*ROM monitor well downgradient of brine and pond
0-574	5.1627.4 E.2273.3	11-28-79	4. MC	27.55	19.11	2	76-31	SEL-201 SER-201 SER-20
-9C-D	S.1621.0 E.2259.0	11-29-79	4. PVC	27.09	29.10	3	19-15	eRCM monitor well dompratient of brine and part
•65~0·	5.1361.5 E.2561.5	10-01-0	- PK	15.59	14.11	£ .	107-117	Min amiliar will uppression of brine and pour SEP-NO
.09-0	S. 1761.5 E. 2550.5	10-01-10	1. PVC	18.50	37.78	36	11-11	ACM monitor well upstalient of brine and join 509-30?
.19-0	S. 1761.5 E. 2510.5	10-11-(0	• PVC	JS.3I	17.78	2	12-15	TREM munitor well upgralient of beine mail juni sip.
.0-62	5.1746.0	19-21-10	4 PVC	11.0	75.86	ř	10 TILL	RIM amitter well dougrafient of luine mal in.
-12 0	7190 0	- In CITY	-		-	•	-	,

SChordinates not Patablished by James M. Garratt & Assoc., Inc. (consulting engineers & Land surveyors). Indicates Newer Elevation Data Based on Objected (Adjusted) Sase
Elevation that is 0.2 Pt Lower Than Older Elevation Data

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§Chordinates not fatabilished by James H. Gerratt & Assoc., Inc. (consulting engineers & land murveyors) . Indicates Newer Elevation Data Based on 'Corrected (Adjusted) Base
Elevation that is 0.2 Ft Lower Than Older Elevation Data

Siveneco	** August 1978, remarkered as 0-70 in April 1985	***Top of casing changed 4/12/84 - new elevation is given. Top of casing was broken again in Narch 1985	*Drilled as IM-2 in Aujust 1980. Renumbered as 0-79 in April 1985.	"Drilled as 184-1 in August 1960. Penumbered as 0-60 in April 1985.
CONTROL SCHEDING OF SCHEDING INTERNAL BELTH GAADE	M-134		87-B3	911-96
DEPTH OF HOLE	13		2	715
OF CINC. OF TOP OF DEPTH SCHEDED SLAB OR CASING ON OF HOLE OF TOP OF CEPTH SCHEDED OF HITSWALL GRADE (G) NEAS. PT. HOLE BELTA GRAD	44. [744		47.33	44.06
ELEVATION OF CONC. SILME OR CANDE (C)	3 . :∓		41.92	9. Ib
TIPE OF MELL. CONSTRUCTION	JW 44		3	4, PC
DATE	91-11-90		01-12-90	08-20-80
STAUFFER COORDINATES	S. 2246. 4 E. 302. 9		S.1095.61 W. 205.18	5.1095.61 W. 189.67
WELL,	Old Namber	1388.	.0-790 Old Namber was 101-2	.0-60° Old Maker

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. Land surveyo	
SCOOKINGATES NOT Extablished by James W. Garrakt & Assoc., Inc. (consulting engineers & land surveyors) .Indicates Newer Elevation Data Based on Corrected (Adjusted) Base .Indicates Newer Elevation Data Based on Corrected (Adjusted) Base .Indicates Newer Elevation Data Based on Corrected (Adjusted) Base	•
, Inc. (consulted) hase	
gratt 6 Assoc. gracted (Adjust Elevation Data	
by James H. G. La Based On Ob ex Then Older	
SCOOKINGARES FOR Extablished by James N. Garratt & Assoc., Inc. (indicates Newer Elevation Data Based on Operacted (Adjusted) Base, in the fact than older Elevation Data	
Coorindates no Indicates Nevel	
. .	

		COMPUTS												ברופנים תומני הבון	PLISTED under well 0-20			staff does not amittor amifers above blue clay. In-	at 11-1 as a menitor well fix injection well INJ-1 to	amplier amplier at 300 to 330 feet.			The plus clay in-	Well does not monitor applicately save one of the total	stalled as a minimum strain on the latter and the latter and the tarter and above the tarter of the latter and	allahi messure at surtare. Is equipped with valve and				
DEPTH OF	SCHEDED	BELCH GINDE	11-12		55-65 55-65	111-11		84-18	94-99		36 -10	69-65					94-99		277-102					1140-1160			•			
TOTAL	DEPTA		Τ		S			2	1			F					2		22				_	1704						<u>-</u> -
ELEVATION	OF 100 OF	CASING OF	100	_	33.07	70 60	24.72	46.29	28.45		70.04	13.33					15.59		16.31					15.53	•				_	_
ELEVATION 1	or conc.		2 2 2		33.00		Ŗ	17.91			18.71	24.86					34 14		13.23	_				11.24						_
	9 mm		COSTRUCTION	<u> </u>	To Mc		2	To MC.		2	Je PVC	10 100	į.				200	£ •	10 1/4 0.D.	steel cementer	In 15° hole to	137'. 4" PWC	1n 9 7/4	10 1/4 0 0	steel comenter	in 15" hale to	115. 1 1/2	0.0. steel	Cemented in 9	Pole to T.
+		DATE	COLUMN	08-22-80	08-23-80		09-22-90	01-11-10		08-22-80	01-31-10							69-87-10	11-11-11	; ;				10 /1	(9-91-90					
		STAUFFER	P		2 200			E. 690. 70				E. 492.60	H.041.64	E. 823.	E.577.72	S H. 899.67	E. 315.00	5.001.07	27. 34	2.43.7					5.652.1)					
			HAPBER	140.	1	- -	148	1700		. 100.	8		. S			14-71	1	(-WI.		<u> </u>					S-HI		_			

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OF CONC. SLAB OR	2 20 20 20 20	3	DEPTH OF	
		DEPTH	SCHEDED	
CONSTRUCTION CANDE (G)	MEAS. THE	ğ	DELCH CANDE	STAGE CO.
77.52	20.00	330	m-m	shall dose not monitor aquifer(s) above "blue clay"
nteel committed in 15° hole to	·			installed as a monitor well for injection well INU-2.
115'. 4" PVC	<u>. –</u>			
				*See listing under (C-1)
under OC-12.				
			•	
Double cased well. See under CC-12.				

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WELL NAMED IN	STALFFER COORDINATES	DATE	TYPE OF MELL CONSTRUCTION	OF CONC. SLAB OR GRADE (G)	OF TOP OF CASING OR HEAS, PT.	TOTAL DEPTH OF HOLE	DEPHI OF SCHEDHED INTERVAL BELOW GRADE	COMMENTS .
.OWI-I*	5.2190.6 E.2592.7	1941	4° PVC	35.26	37.75	Approx.	60-70 (Estimated)	*Monitor Hell II drilled by Vester Thompson (?) for Courtaulds. Renumbered ONN-1 and measurement made by
.OWI-7*	5.2241.3 . 6.2871.0	1983	1º Puc	43.04	45.45	Арргол . 76	(Est insted)	NPS February-May, 1984. *Nonitor Nell 12 drilled by Wester Thompson (?) for Courtaulds. Remumbered CNAM-2 and measurement made by NPS February-May, 1984.
.OW-3*	5.3150.6 E.3861.3	1983	I FIC	46.15	40.01	Хрргон. 70	60-70 (Estimated)	*Monitor Well 1 drilled by Vester Thompson (7) for Courtaulds. Renumbered OWH-) and measurement made by MPS Pubruary-May, 1984. Discovered that there is no bottom cap in this well - or screen is broken.
CNAH-218 MAS CNA-1-751 MA.	S.2429.4 E.2002.6	06-19-76	4ª MC	(c) X.1	34.00	130	166-126	*Drilled as offset well for OW-1 for pumping test in August 1978. Renumbered as CNAM-21 in July 1984.
.CHAH-224 MAS CHA-1-150' SH.	5.2600.7 E.2002.1	08-31-18	4" PVC.	(G) 30.0 Old base. S. 39.57	रा.भ	136	J06-126	*Drilled as offset well for OW-1 for pumping test in August 1978. Renumbered as OWH-22 in July 1984.
.OWF23*	\$,2311.15 E,2365.96	02-09-44	1º PVC	14.06	36.34	113	67-117	*Coordinates ware for stake pribr to drilling. Drilled as M-).
.OM-24* (n-4)	5.7609.84 E.1395.10	02-07-44	1" PVC	36.60	38.54	127	55-125	*Coordinates were for stake prior to drilling. Drilled as M-4.
.CWH-25*	S. 2756.57 E. 1450, 19	02-01-04	Sa BAC	41.25	43.16	131	69-179	*Coordinates were for stake prior to drilling. Drilled as M-5.
(NAH-26" (H 6)	5.7500.40 E.1724.49	02-00-04	I' PVC	16.61	30.75	178	61-01 and 91-121	*Coordinates were for stake prior to drilling, Orilled as M-6.
(H-7)	5.2609.49 E.1701.41	01-06-81	1º PVC	14.11	15.69	132	91-131	Coordinates were for stake prior to drilling, Orilled as M-7,
(H 8)	5.1600.16 E.2097.04	03-10-01	1 PVC	79.48	11.61	71	19-69	*Opendinates were for stake prior to drilling

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Elevation that is 0.2 Pt Lower Than Older Elevation Data

		7	_	-	-			
	STIGORIO	2	"Coordinates are he at	Orilled as #-17.	Coordinates were for state prior to drilling.	Coordinates were for state mily to deill	Modelingles are for each	Orilled as M-14
SLAR OR CASING OR OFTING OF INTERNAL	DELCIA CINDE	100.5-120.5	W.5 76.5-16.5		711-16	74-94	126.5 107.127	
OF THE SECOND	HOLE		27.5			2	7.8.5	
ELEVATION ELEVATION YORK, OF COPTING OR CASING OR CASING OR	16.99 (G) MEAS. PT. HOLE					11.72		
OF CONC. SLAB OR	10 M		15.45 17.61	15.66		42.43	43.47	
THE OF THE LAND			34. M C	to MC		¥	- PC	
DATE COPPLE	19-10-10			19-50-10	19, 20, 10		19-90-10	
STAUFFER	\$.2736.74	6.1036/4	E. 1564. 41	5. 2550. 46	5.786.81	E. 1885.45	5. 2479. 00 E. 1890. 40	
MPOER	-04-10-	- II-	(4-12)	- CAN- 32	- TANCO	(H-1))	(H-14)	

5Chordinates not Established by James N. Garratt & Assoc., Inc. (consulting engineers & land surveyors). Indicates Newer Elevation Data Based on Corrected (Adjusted) Base
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MELL MAPRIER EM-15	STAUFFER COORDINATES \$ \$.1234 E.1445.5	DATE COMPLETED	TYPE OP MELL CONSTRUCTION Double cased well by Layne. 24" outer casing 16" inner casing 12" screen.	ELEVATION OF CONC. SLAB OR GRADE (G) (G) 30 Conc. pump foundation - just above grade.	ELEVATION OF TOP OF CASING OR NEAS. PT.	TOTAL DEPTH OF HOLE	DEITH OF SCHEENED INTERVAL BELTH GRADE	CONTROL OF THE PROPERTY OF T
.LA-2	5.2197.5 E.242.3	07-08-52	Double cased well by Layne. As Above.	46.23	46.70 South 1° dia. pipe within discharge head.	132	105-125	"Hell was recompleted in 1975. 25" long 10" dia. stain- less steel well screen was set inside the original 12" well screen.
DF-34	5.1160.9 E.2345.1	01-06-56	Double cased, well by Layne. As above,			132	,	*Well abandoned in 1973 or before. Pump and motor have been removed. Steel plate covers top of casing.
DE-11	5 S.1016.24 E. 660.2	Spring 1965	Double cased well by Layne. As above.		26.16 Top of 4" pipe welded into steel cover plate		,	*Hell abandoned - never in service. Pump and motor were pulled and installed in well No. LM-5 in June 1965.
.ta-5*	S.1199.5 E. 150.4	06-15-65	Double cased well by Layne. As above,	corner	14.05 S.E. corner of cast steel mole plate,	tup of	94-124 Below top of pump founda- tion (approx- imated in Aug. 1983).	*Mell was recompleted in August 1983, after well was pumping gravel pack. Installed 30° of 8° dia. stainless steel well screen inside the original 12° well screen.

SChordinates <u>not</u> Established by Garratt Engineers .Indicates Never Elevation Data Based on Corrected (Adjusted) Base Elevation that is 0.2 Ft Lover Than Older Elevation Data NPS 4/29/85 Page 13 of 17

MELL. MURBER	STAUFFER COOPDINATES	DATE COMPLETED	TYPE OF MELL CONST.	ELEVATION OF CONC. SLAD OR GRADE (G)	ELEVATION OF TOP OF CASING OR NEAS. PT.	TOTAL DEPTH OP HOLE	DEPTH OP SCREENED INTERVAL BELOW GRADE	COPPENTS
LH-6	S. 713.6 E. 3150.0	Aug. 1965	Double cased well by Layne. As above.	(4,18) Conc. punt foundation just above grade.		131	38-118	
Ui-7	S.7376.8 W.1453.0	Hay 1972	Double cased well by Layne. As above.	40.98 Conc. pump foundation	•	126	93-123*	*Screen interval of 93-123 was given to me, but I suspect 103-120 is more likely to be correct. (W.P.S.)
CC-8*	\$ 5.265.33 E. 0.00	01-11-46*	Double cased well by Layne. As above.	(G) 42.6		ארד – לנד	117-127	"Hell was plugged and abandoned about 1980. "Hell was completed in two acquifiers. Both above "Blue Clay" unit.
CC-31	§ 5.417.22 W.948.77	01-0569	Double cased well by Layne. As above.		44.46 Metal plate covering well.	139	74-69 == 119-129	"Well use plugged and abandoned about 1980 (7). ""Well use completed in two squifers, both above "blue clay" unit.
LH-10	S. 1290.4 H. 1386.3	Dec. 1973	Double cased well - mou- ified Layme type well. By Holland well.	43.91 S.W. holt on dis- charge head.			114-134	
.a-11	N. 474.6 W. 468.4	04-22-75	Duble cased well-modified tayne type well by Holland,	•	19.61 Top side of (N.W. corner) lower flame on discharge	T27	115-125	Mell was recompleted in 1978 - smaller dia, inner casing and screen also lower capacity (sum) purchased.

.indicates Never Elevation Data Based on Corrected (Adjusted) Base Elevation that is 0.2 ft Lower Than Older Elevation Data	JA 7.0 81 780		Establica chec 18 0.4 Pt Lover Than Older Elevation Data					Page 13 of 14	Ξ.
MELL. MELL. CC-17* Also marbered in-1 by corp in 1981	STAUFTER COCKOINNTES S. 69.7 W.1415.0	DATE COPPLETED 17-05-78*	WELL CONSTRUCTION 11 steel Comented in 30" hole to 71". 11 steel inner casing with 17" SS screen in 30" hole.	ELEWATION OF CONC. SIAB ON CANDE (G))). 89 Sole plate.	ELEWATION ELEWATION TOTAL DEPTH OF OF OF OF OF OF OF PARM OF SCHOOL CASHG OR OF INTERNAL CAMPE (G) HEAS. FT. HOLZ BELLOW CAMPE (G) 13.97 122 94-118 Sole (1 large on blate. there of discharge head.	OCFTM.	DEPTH OF SCHEDED LINE SCHEDED LINE SCHEDED CANOL DEPTH OF SCHEDED CA	**************************************	ž ž

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	Ground water Intercept well designed by Ground Water Associates, Inc. and Stauffer Geology, on line December 1960.	Ground water intercept well, (as above).	Gound water intercept well, (as dawe).
DEPTH OF SCHEDED INTERNAL.	(F-11)	72-122	74-134
2 0 G		~	ž (
DE TOP OF CASING OR	No. 41 Top of 1/2 PVC meas- uring tube (Nest Side)	17.06 Top of 1/2- FV measuring tube (East Side)	No of 1/2 FV: measur Inj tube (East Side)
OF CONC. SIJA OR	16.17 M.W. bolt on dis- charge head.		
ACL TIME OF) T L	Ochie cased 17.01 well as above, S.E. bolt on dis- charge heat.	Couble cased 15.97 well. As above S.E. bolt on dis- clarye leat.
DATE CASE		09-02-90	08-52-90
STAUPTER	5.7314.13 E.2017.23	S. 2403. 12 E. 1696. 88	5.7546.90 E.1166.33
TIEN	<u>.</u>	- N-2	

Schordinates not Established by James N. Carratt & Assoc., Inc. (consulting engineers & land surveyors). Indicates Never Elevation Data Based on Oprrected (Aljusted) Base
Elevation that is 0.2 Ft Lover Than Older Elevation Data

NFS 4/29/85 Page 16 of 17	OneDVIS Plugged and abandoned April 1981.	
continued to land merveyore)	DEPTH OF SCHEDNED INTERNAL BELLA GRADE 1302-1491 Injection ton (lef.	Bushing)
	ELEVATION ELEVATION YORK, OF CONC. OF YOP OF DEPTH SLAB OR CASING ON OF GRADE (G) NEXS. PT. HOLE (C) YO	1021
)justed) Base	ELEVATION OF CONC. SLAB OR GRADE (G) (C) YO	78.82
m Ocrected (A)	TYPE OF OF ONC. WELL ONETHICFION CANDE (G) Deep waste (G) 30 Injection well.	June 1974 Deep waste injection well. Jan. 1911 Deep waste Injection well.
Data Based o	ONTE CONVLITED Peb. 1969	June 1974 Jan. 1983
Elevation that is 0.2 ft lover Than Older Elevation Data	STAUFTER COODIMITIES \$ N. 70.38 E. 765.02	N. 960.4) E. 409.96 S. 075.13 E. 625.55
Elevation	#1-INI NJBAN TTJM	- INJ - J

\$Coordinates not Established by Garratt Engineers
.Indicates Never Elevation Data Based on Corrected (Adjusted) Base
Elevation that is 0.2 Ft Lover Than Older Elevation Data

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MELL MANBER CNA-1*	STAUFFER COORDINATES 5,2464.06 8,2140.97	DATE COMPLETED 1952	TYPE OF MELL CONST.	ELEVATION OF CONC. SLAB OR GNADE (G) 35.77	ELEVATION OF TOP OF CASING OR HEAS. PT.	HOLE	DEPTH OF SCHEDIED INTERVAL BELLM GRADE 86-118	**COPPENTS *Hell abandoned. Pump and motor have been removed. **Top of casing is covered with steel plate. **Double cased well by Layne 24" steel casing camented in 10" hole to 60 ft. 16" inner casing with 12" screen.
CNA-24	5.2840 ** E.1024	1952	888	45.64 Concrete pump foun. Top of foundation at grade.		ננו	57-127	*Well abandoned. Pump and motor have been removed. Top of casing is covered with steel plate. **Coordinates are approximate. ***Duble cased well by Layne 24° steel casing cemented in 30° hole to 80 ft. 16° inner casing with 12° acreen.
CNA-3	5.2050.44 E. 3293.76	1952		43.17 (Base Plate)	(Top of 1° pipe cap holding att	130 Vidalon :	95-125 92-122	*Double cased well by Layne 24" steel casing commented in 30" hole to 80 ft. 16" inner casing with 12" screen. *Duble cased well by tayne 24" steel casing commented in 30" hole to 80 ft. 16" inner casing with 12" screen.

APPENDIX XIII
TOPOGAAPHIC MAP

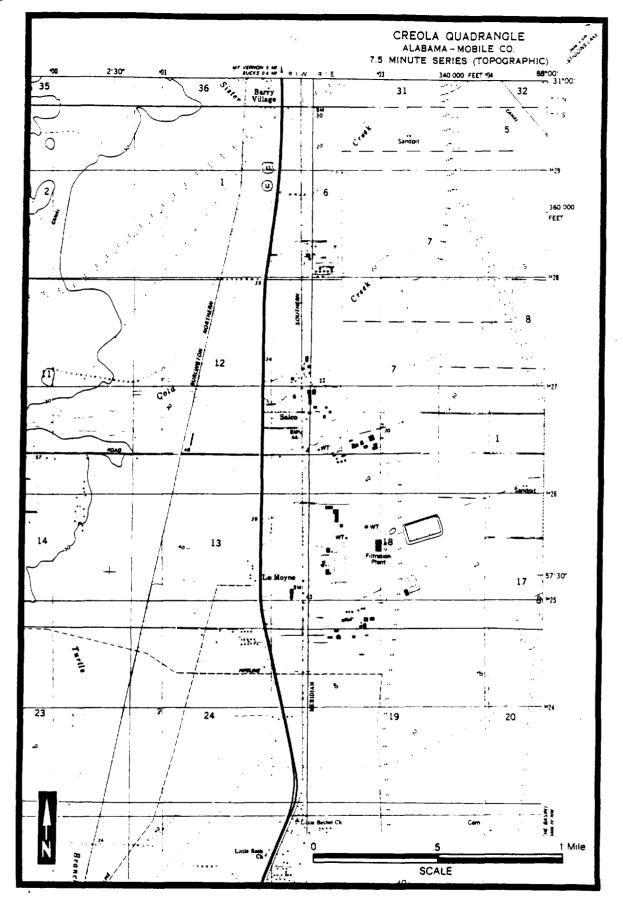


Figure 13-1 Topographic Map

APPENDIX XIV WATER RESOURCES OF THE MOBILE AREA, ALABAMA

UNITED STATES DEPARTMENT OF THE INTERIOR
Designs Lieksy, Secretary

GEOLOGICAL SURTEY Thomas E. Wolan, Director

CHOLOGICAL SURVEY CIRCULAR ST

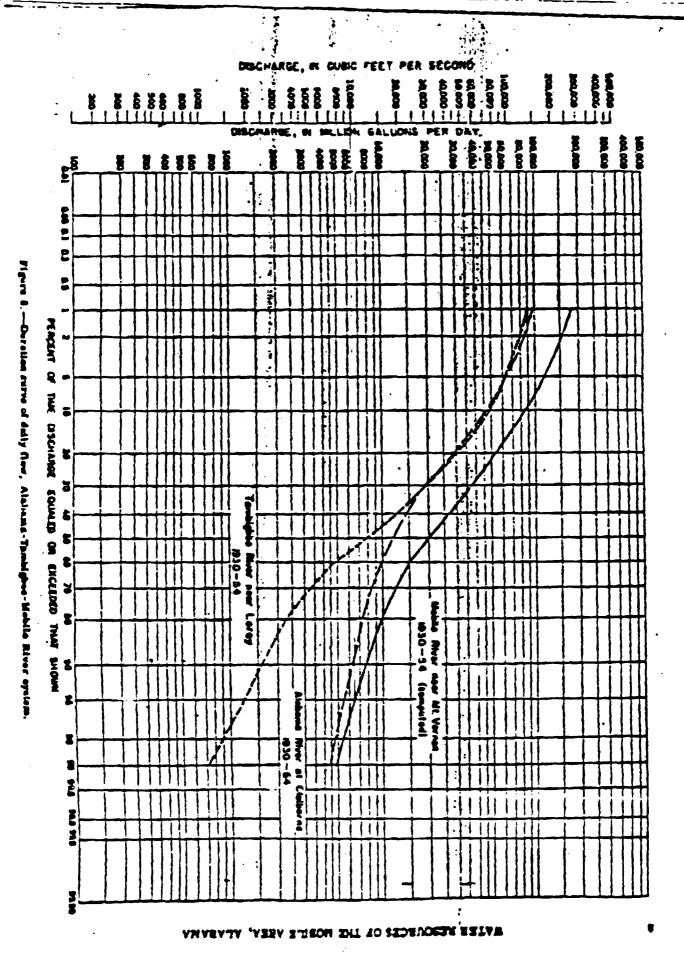
WATER RESOURCES OF THE MOBILE AREA, ALABAMA

Br W. H. Robinson, W. J. Powell and Engage Brews

Wife a section on malesty of the Mobile Moor by the Course of Engineers, E. S. Array, Mobile District

Wednesd R C. 188

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WATER RESOURCES OF THE HOBILE AREA, ALABAMA

Table 2. - Probable duration of low flow at supplemental gaging alter in the Midble area

Personal of	States Creek States Creek States Creek States Creek										
time desharps tasticated ups equaled	Basett tear Waj (Water year	P CTULE	Levis	Creek kinusa	Silbo near M (Water year	Creek cibtosh	Batto (Walet Wal	alwert.			
or exceeds	cis, sq mi	22461 84 127	c(s/ sq mt)	254/44 Br	C19/ 94 781	mgd/64 /mi	eful sq rei	™84\ 84 B			
0	0.47	6,30	8.40	0,36	0,40	0.28	0.37	0.24			
16	.27	.17	.18	.12	.16	.10	.14	.000			
9		.13	.11	.071	.10	.006	.913	.046			
0	.14	.000	.070	.043	.063	.041	.043	.028			
3	.11	.871	.047	.020	.542	.027	.626	.017			
M	.002	.010	.036	.623	.922	.021	.010	.012			
M	,006	,943	.023	.814	,520	.013	.010	.006.5			
K	.003	.934	.015	,0007	.613	,0004	.0064	.0041			
19	,034	,021	.0083	,0063	.0000	.0045	.0029	.0019			
Percent of			Discharge	-	alle of drain	age area		<u> </u>			
time discharge	Codar		Baye	n Sara	Zights	ile Creek		ile Creek			
time.	Coder near Mose (Water year	M Yarnes	Baye near S	 	Elghts mar E	alle Creek		richies			
time discharge indicated	Mar Mou	1239-54)	Baye sear S (Water yea	m Sore ioraland	Elghts mear E (Water ye	ile Creek	mar C (Water year	riebies re 1939-54			
time discharge indicated was equaled	mar Mose (Water year cla/sq mi	1239-54)	Baye sear S (Water yea	m Sara laraland ura 1929-54)	Elghts mear E (Water ye	kile Creek light Mile kra 1939-54)	mar C (Water yea	riebies re 1939-54			
time discharge indicated was equaled or exceeded	mar Mene (Water year cla/sq mi 0.53	2 Yernes 2 1239-54) med/sq mi	Baye mear 3 (Water year ofs/sq mi	ne Sara Inraland Lrs 1939-54) Lagd/sq mi	Bights near B (Water ye: efs/sq mi	nile Creek light Mile ars 1939-541 mgd/sq mi	Water year cfs/sq mi 1.6 1.6	richies re 1939-54 mgd/sq			
time discharge indicated was equaled or exceeded	rear Mone (Water year cla/sq ma 0.53 .38 .38	2 7ernes 2 1239-54) med/sq mi 0,34	Baye mear 2 (Water yea ofs/sq mi 1.1 .31 .88	s Sara iaraland ara 1920-54) angd/sq mi 6,71 .88 .44	Rights near E (Water ye) efs/sq mi 1.8 1.7 1.5	nile Creek light Mile ars 1939-541 mgd/sq mi	tear C (Water year cfs/sq mi 1.6 1.4 1.3	richies re 1939-84 mgd/eq 1.8 .50 .36			
time discharge indicated was equaled or exceeded	mar Mou (Water year cle/sq mi 0.53 .39 .31	2 Yarnes 2 1239-54) cued/sq mi 0.34 '.29	Baye near S (Water yea efs/sq mt 1.1 .31 .31 .36 .38	s Sars iaraland ure 1920-54) iaqui/aq mi 0,71 .38 .44 .36	Eights near E (Water yes efs/sq mi L.S 1.7 1.5 1.6	nile Creek light Mile ars 1939-54) mgd/sq mi 1.3 1.1 1.0	war C (Water year cfs/sq mi 1.6 1.6 1.3	richies re 1939-84 mgd/eq 1.8 .90 .86 .78			
time discharge indicated was equaled or exceeded	mar Mou (Water year cle/sa mi 0.53 .39 .39 .31 .31	# Yerses # 1739-54) mad/sq mi 0.34 '.29 '.21 '.17	Baye near S (Water yea efs/sq rat 1.1 .81 .88 .88 .48	## Sars araland araland aralists 1920-54 	Rights near E (Water ye) efs/sq mi 1.8 1.7 1.5	nile Creek light Mile ars 1939-541 mgd/sq mi 1.3 1.1 1.0 .97	war C (Water yea cis/aq mi 1.6 1.9 1.2 1.3	richiem re 1939-54 m.gd/eq 1.0 .90 .81 .78			
time discharge indicated was equaled or exceeded	mar Mou (Wover year cle/sa mi 0.53 .39 .39 .21 .21	2 Yermon 1739-54) mqs/sq mi 0,34 1,25 1,17 1,15 1,14	Baye near S (Water yea efs/sq mt 1.1 .31 .34 .36 .46 .48	## Sars araland araland aralises 1930-54 appl/sq mi 6,71 .85 .44 .35 .31 .37	Eights near E (Water yes efs/sq mi L.9 1.7 1.5 1.8 1.3 1.4	nile Creek light Mile ars 1939-54) mgd/sq mi 1.8 1.1 1.0 .97 .97	war C (Water year cis/aq mi 1.6 1.9 1.2 1.2 1.3	richien rs 1939-54 mgd/eq 1.8 .90 .96 .78 .79 .71			
time discharge indicated was equaled or exceeded	mar Mou (Water year cle/sa mi 0.53 .39 .31 .31 .31	# Yernes # 1739-54) mqd/sq mi 34 31 17 15 14	Bayes near S (Water yea efs/sq mt 1.1 .31 .36 .36 .48 .48	## Sars Sars Invalidation 1939-54 Inpal/set mid 6,71 .88 .44 .38 .31 .27	Eights near E (Water yes efs/sq mi L.9 1.7 1.5 1.8 1.9 1.9 1.9 1.9 1.9 1.9 1.9 1.9 1.9 1.9	nile Creek light Mile ars 1939-54) mgd/sq mi 1.3 1.1 1.0 .97 .97 .90	mar C (Water year cfs/aq mi 1.6 1.9 1.2 1.2 1.3 1.1	richien rs 1939-54 mgd/eq £.8 .96 .96 .78 .79 .71			
time discharge indicated was equaled or exceeded	mar Mou (Wover year cle/sa mi 0.53 .39 .39 .21 .21	2 Yermon 1739-54) mqs/sq mi 0,34 1,25 1,17 1,15 1,14	Baye near S (Water yea efs/sq mt 1.1 .31 .34 .36 .46 .48	## Sars araland araland aralises 1930-54 appl/sq mi 6,71 .85 .44 .35 .31 .27	Eights near E (Water yes efs/sq mi L.9 1.7 1.5 1.8 1.3 1.4	nile Creek light Mile ars 1939-54) mgd/sq mi 1.8 1.1 1.0 .97 .97	war C (Water year cfs/aq mi 1.6 1.9 1.2 1.2 1.3	richiem re 1939-54 m.gd/eq 1.0 .90 .96 .78 .78			

channels, beyone, and lakes fill on a rising tide and empty on a failing one. The tidel offset is known to be sufficient at times to cause the Mobile River at Mount Verson to flow upstream.

It has been generally recognized that the flow in the Mobile River to practically equal to the combined flow of the two headwater streams. Until recently more accurate definition of the pattern of flow did not appear to be warranted. Now, however, questions have arrain in connection with recent developments on the Mobile River that show the proof for more nearly accurate information.

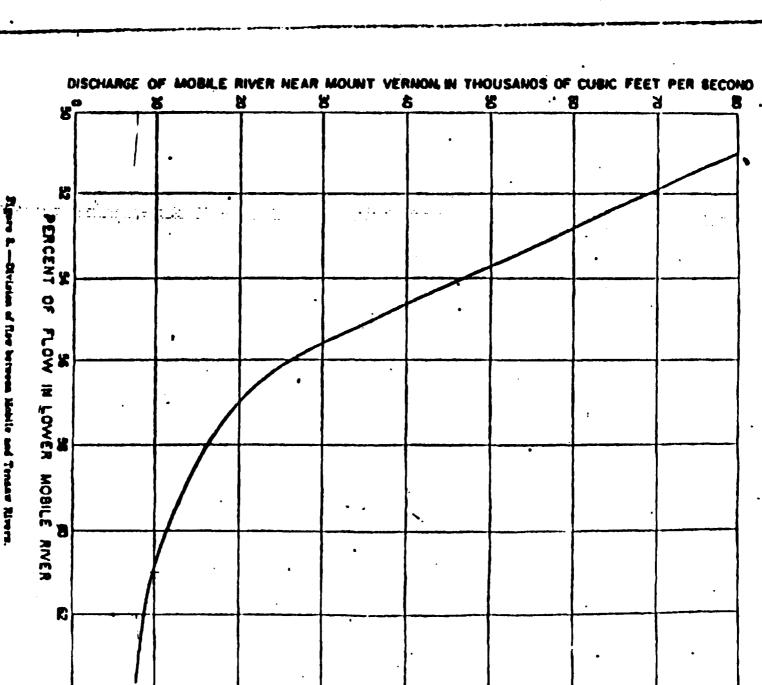
Discharge

In times of low flow the Alabama River in the major contributor to the Midule River. Its naturally good low-flow characteristics are improved by considerable storage capacity at power-plant receivairs on lin headwater circums.

The Tampigher River at the gaging station aran berry has a distinguishes compositive to be design age area of the Alphana River at Michanse. However, in periods of low sales, the yield of this aircum is much legs than that of the Alphana River. Some regtilation is exercised over the low flows in the stream by the savigational locks and dams. It is expected that additional control will be exercised by the new lock and dam at Demopolis.

The samual runoff of the Mobile River is the combined runoff of the Alabama River at Claiborne and the Temblighee River near Lerey, and a small runoff from the drainage area between the gaging stations and the confluence of the rivers. Revever, the flow of the Mobile River for any particular time can not be so easily related to the flow at the upstream gaging stations. Charmel storage and tide have a and startial effect on the rate of flow of the Mobile River.

A gaging station was established on the Mobile River at Monet Verney, and operated for I year to make possible open appraisal of the flow characteristics of the Mobile River and evaluation of the relation between the symbiated flows at the upstream gaging stations and the flow in the Mobile Miver. This station was located about 2 miles above the division take the Texase Miver and the lower Mobile Miver on the reach of river where the flow is garactally confined to one channel. Except desiring times of flood, the discharge at the station varied commercially during the daily tital excle. Negative, or upstream, flow accasionally occurred for a flow flows, or upstream, flow accasionally occurred for a flow flows. Typical stage and encharge typicagraphs for resocied tital cycles are alread in figure 6.



tion, but they will be publicated in the effect of water going into or causing out of starage on a generally rising or falling stage. Average discrets yes for the menta from Orinter 1953 through September 1954 has shown in figure 7. For comparison, the combined flow of the optificam distings to also shown in the control that a restablished amount of water is retained to always on a generally replace.

bing and to referred on a generally falling one. For this reason, months of high average flow are not always closely comparable to the companed flows of the against a stations. On the suber hand, the monthly averages for perturbe of low flow attent to be tout ecomparable to the monthly average of the complimed flow, expectably if the combined flow is connected for

1.

WATER RESOURCES OF THE MOBILE AREA, ALABAMA

Duration of flow data for the Mobile River at Mount Vernon are shown in figure 5. The maximum, mustman, and modian values of the average discharge for each month for the combined flows of the upstream stations for the period 1930 to 1854 are shown in figure 7. As occurrences in the past are indications of occurrences in the future, the diagram can be used to extimate the probable range of conditions that may be generally expected.

The Mobile River first divides into the Tansaw and lawer Mobile Rivers. The flow in these two streams is about equally divided at bankful stage, but at lower stages the more efficient channel of the lower Mobils River carries as increasingly larger proportion of the total flow (fig. 3). Although this relation is believed to be reliable for ment doily flows, it is not necessarily reliable if used for average flows for shorter puriods or for argumenting the division of flow for any particular instant, expectally when varying tidal effects occur within a tidal cycle. Once, for a short period, water was known to flow simultaneously up the Turnou River and down the lower Mobile River.

The Tensow River divides several times before it empties into Mobile Bay. No information is available on the distribution of flow at three divisions. The Mobile River does not divide further until just morth of the matropolities area, for do any streams of approximate size flow into this result.

Please

Not such information to available on floods in the Mobile River, Large flouds occurred to 1929, 1938, and 1848. Information is available on these floods at the spatream gaging stations, but because the flood flows were mostly confined to the uninshipted swampland little information is available about them on the Mobile River. Waver-surface profile data on a leaser floud in April 1931, were obtained from the Corpe of Engineers and are shown in figure 9. For comparison, a reported water-meriage elevation of the 1929 flood at the Moste Vernon gaging station and the approximate elevation of the 23-year flood are shown in the same figure. A "25-year flood" is a flood of the magnitude that would be equaled or exceeded on the average of each in 28 years.

Floods of another type occur on the lower reach of the Mehile River when strong winds from the Gulf of Mexico cause the water of Mehile Bay to pile up at the head of the bay. When these winds reach the valueities attained in the infrequent hurricanes, the piling up of water floods low-lying areas in the city. The most destructive hurricanes of this century coopered in Separaber 1906 and in July and Octaber 1916. The July 1916 seerm in reported to have produced a tide of 11.5 foot above mean are level at Mabile.

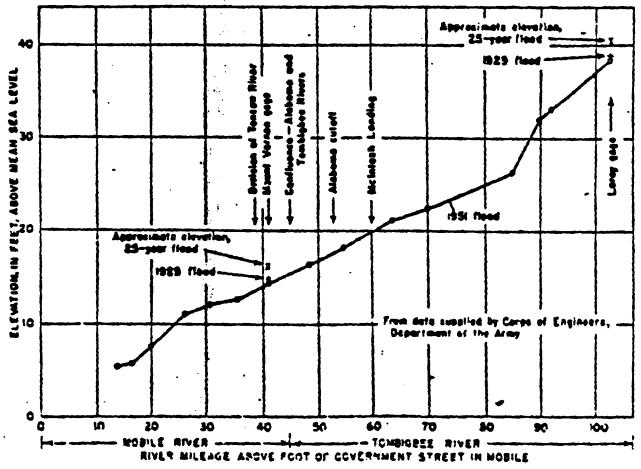


Figure & -- Profile of April 1851 Seed on Mobile River.

In command the command of the comman

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2-4795 MOBILE RIVER MEAR MOUNT VERNON ALA
LOCATION--LAT 3: 84 50 LONG 87 58 85 IN SE 1/4 SEC 4: TRU BIE AT 88AT PIER ON
UEST BK OF LAKE DAVID .5MI US FR LAKE OUTLET TO MOBILE: RV 2,5MI ME OF MT VERNON
• AT MILE 43.3 FR MOBILE /ORATNAGE AREA---43600 /RECORDS AVAILABLE---OCT 1953
TO SEPT 1954 / GAGE---STAGE RECORDER DATUM ABOUT 2FT BELOW MSL BY COMPARATIVE
OH MEADINGS AUR BAGE AT ALA 5T DOCKS

THOU 1967 MOUNT VERNON FLOW =1 CLAIBORNE FLOW • MY 43 FLOW) • 1.05
AFTER 1967 MOUNT VERNON FLOW =1 CLAIBORNE FLOW • COFFEEVILLE FLOW) • 1.07
MONTHLY AND VERRLY DISCHARGE IN CUBIC FEET PER SECOND

W.Y.	OC TOBER	MOVERBLA	OFCENBER	-MANAGY	FERRUARY	HARRI	APRIL		PANE NA		AUGUST 1	EP1/1616	THE WEAR
		•	•	•	•	•	•	• 1	•	, ,			1. 10 m
1920											24500.0		
1050								135300.0				2000.0	77720.0
1030	74000.0		103400.0		_			54304.0			16400.0	23440.0	65100.0
1931	10100.0	-	20445.	Bonne .	4.5344.4			34608.0	15448.4	1440.0	19500.0	10400.0	34500.0
1932	9190.0	9174.4	74944.4	124444	100200.0	100400-0	43200.0	49000.0	. 25900.0	A7444 - A	26700.6	35400.0	63710.0
1033	47100.6	72486.6	175104.4	211000.6	141000.4	147694.6	130500.0	46000.0	21500-0	37944-4	21200.0	19500.0	71100.0
1934	14300.0		21100.0					20204.0		24800-0	31700.0	19100.0	37000.0
1935	56394.0							75300.0		17000.0		16300.0	61200.0
•		•	•	•	•	•	•	•)	•	•	,
1936	12000.0							38404.0		24800.0	24500.0	15500.0	48944.0
1437	14200.0							153304.6		19500.0	16600.0	30500.0	66666.0
1038	20900.0	34100.0						43304.0	34900.0	41400.0	41400.0	17400.0	64200.0
1939	15500.0	14700.0						30106.0	49900.0		115700.0	27400.0	63600.0
1040	10500.0	15100.0	10000.0	*****	120700.0	102260.0	64386.0	40500.0	34388.0	141500.0	25444.9	15000.0	54700.0
1941	12500.0	19000.0		54200.0	• 54166. A		45884.4	* Z8400.0	1 1000.0	41400.0		15140.0	34700.0
1442	12400.0	14200.0						25700.0				19500.0	44500.0
1943	15000.0	15300.0	44000.0	12300.0	70100.0	153700.0	130200.0	39990.0	20405.0	23000.0	22800.0	16900.0	54300.0
1944	11600.0	17260.0							24200.0		27500.0	21000.0	49700.0
1945	13100.0	14400.0	26300.0	46400.0	115100.0	145200.0	101300.0	92496.0	23900.6	21200.0	10000.0	13200.0	55700.0
•		,	• ,	•	•	•		• •	•	• :	•	•	
1946	15300.0							71000.0			49308.0		07400.0
1947	16606.0	46306.0	40000.0	195100.0	150000.0	132900.0	157100.0	73000.0			17200.0	15000.0	74500.0
1948	15100.0	39000.0	60000.0	46500.0	167600.0	214300.0	135900.0	28300.0	50500.0	\$3860.0	29400.6	16600.0	60100.0
1949	15100.0	70000.0	235100.0	204300.0	223500.0	125600.6	132500.0	103404.4	44244.4	56000.0	29440.0	32300.0	105700.0
1720	17000.0	24400.0	39700.0	101300.0	115>00.0	134700.0	62313.5	52400.0	28400.0	35488.0	35300.0	69000.0	50400.0
	31300.0	22700 0	, ,,,,,,	48644 4	114344 6		********	45000.0	•	33344	15300.0	14100.0	40404.0
1951 1952	12600.0	22444	37200.0	91 740 - 4	54556.5	145744.4	77144	34900.0	24200.0	12000.0		12900.6	51400.0
1953	11200.0	11940.4	24444-8	100200.0	109000-0	150300.0	11100.0		22700.0			12600.0	39500.0
1954	15500.0	13500.0	60900.0	70400.0	73400.0	41000.0	74000.0	34700.0	14440.4	11000.0	10500.0	9430.0	36100.0
1955	7610.0	9420.0	12500.0	47200.0	89800.0	72400.0	160100.0	40500.0	24900.0		22240.0	10000.0	43400.0
•	•	•	•			,			•		•	•	
1956	10100.0		10300.0	13900.0	124100.0	137600.0	13760.0	36000.0	16400.0	24000.0	10400.0	11000.0	44440.0
1957	15400.0		48760.0	47788.8	127000.6	79000.0	159000.0	51000.0			12466.0	21700.0	5240 0.0 75100.0
145A 1459	34900.0		127704.0	6444	100400.0	142200.0	90200.0	105000.0	24000.0	58860.4	24880.0	17900.0	46606.0
1740	23400.0 29406.6	19000.0		30000,0	120444	0.000.0	87700.8		71100.0		13660.0	15000.0	58508.0
	* 7700.0	35500.0	- 4.70665	~ *******			-	30000.0	10600.0	13000.0			32344
1761	22100.0	21300.4	28500.0	37688.0	127900.0		179444.4	45500.0	44248.0	51800-0	21200-0	22000.0	77800.0
1962	13800.0	55000-0	217600.0	203000.0	151700.0	143100.0	186400.0	36560.0	25400.0	19200.0	12500.0	15400.0	87200.0
1963	17800.0	23400.0	24100.8	6860.0	75800.0	124100.0	38000.0	70700.0	41900.0	44300.0		13900.0	47100.0
1964	142 3	13100.6	37700.0	84400.0	141646.0	210306.6	277100. ■	L 33500.0	23700.0	30500.0	24160.0	15400.0	00500.0
•	•	•			•					•	•	•	

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2-4785 MOBILE RIVER MEAR MOUNT VERNON ALA
LOCATION--LAT 31 86 SO LOME 87 SO 95 IN SE 1/4 SEC 41 TRN RIE AT BOAT PIER OM
WEST OK OF LARE DAYID .5M1 US FR LARE OUTLET TO MOBILE RY 2.5M3 ME OF MT VERNON
• AT MILE 41.3 FR MOBILE /DRAIMAGE AREA---43000 /RECORDS AVAILABLE---OCT 1953
TO SEPT 1954 / GAGE---STAGE RECORDER DATUM ABOUT 2FT BELOW MSL BY COMPARATIVE
GH READINGS AUX GAGE AT ALA 67 DDC45
THRU 1967 MOUNT VERNON FLOW =1 CLAIBORNE FLOW • MY 43 FLOW) = 1.05
AFTER 1967 MOUNT VERNON FLOW =1 CLAIBORNE FLOW • COFFEEVILLE FLOW: = 1.07
MONTHLY AND VERNLY DISCHARGE IN CUBIC FEET PER SECOND
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ļ •.	W.Y.	october	MOVEMBER	DECEMBER	JANUARY	FEBRUARY	MAHCH	APRIL	MAY	JUNE	JULY	AUGUST S	RPONSTON	THE YEAR
1	•		•	•	•	•	•	•		•	•	•	•	•
í	1965	33694.0	24866.6	79366.6	44564.0	1 123500.0	130500.0	113000.0	21506.0	26406.0	29788.8	10705.0	17300.0	66266.0
'	•		•	•	•	•	•	•			•		•	
•	1966	32700.0		51500.0		151000.0			104000.0			19000.0		52500.0
i	1967	27888.0	37900.6	41200.0	50100.0	73400,8	50000.0	16466.6	48488.8	21700.0	52460.6	49600.0	44500.0	43600.0
4 -		25500.0	49605.0	366400.4	200400.0	47866.6	83646.6	95449.6	70000.0	23640.6	27500.0	23660.0	13700.0	49500.0
	1969							331404.0		20700.0	10100.0	17000.0	21400.0	53000.0
•	1970	21500.0						114700.0		43900.0	14400.0	23700.0		51400.0
							-						,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	24.0000
1		33444		•	-		347444	·	,		34300	33344 4	, 20200 0	70100
i	1971	23400.0	27300.0			133600.0						37300.0		70400.0
ſ	1972	17200.0	16500.0	99766.6	202200.(95366.6	114000.5	51300.0	43000.0	25888.0	26500.0	19200.0	18700.0	61100.0
1	1973	12700.0	24000.0	94400.0	147700.0	121900.0	172800.0	229400.0	141000.0	94940.0	44300.0	26060.6	19066.6	94204.8
j .	1974	19200.0	20000.0	78208.0	207300-4	104000.0	75200.0	143200.0	53000.0	57700.0	25340.0	27200.8	54200.0	30504.9
1	TOTAL													2874504.6
1	AVERAGE							121469.6						62489.1
1 .	MACHINE						1301000	1214010		36 200 2	3131400	6311465	£1/1500	-6
ı		MOUND OF F	AVERAGES	10.3 311	mir iradi	A COUNTY								

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	376	8stimated Discharge 355,000 152,000 2214,000 2312,000 134,000 234,000 236,000 236,000 237,000 237,000 341,000 352,000 353,000 353,000 353,000 353,000 353,000 353,000	
	376,000	152,000 154,000 179,00	

APPENDIX XV BACKGROUND LEVELS OF HEAVY METALS, ADEM'S LETTER 12/19/86

ALABAMA DEPARTMENT OF ENVIRONMENTAL MANAGEMENT



Leigh Pagues, Director

December 19, 1986

1761 Federal Drive Mantgomery, AL 36130 206, 271-7700

Mr. William M. Cawthra, P.E. Environmental Technical Manager Stauffer Chemical Company Dobbs Ferry, New York 10522

Field Offices:

Unit 806. Building 6 225 Oxmoor Circle Birminghem, Al. 35209 206/942-6166 RE: Request of November 18, 1986, for background levels of heavy metals in or around the vicinity of Stauffer Chemical's Cold Creek and Lemoyne plants.

Dear Mr. Cawthra:

P.O. Ben 963 Desetur, AL 36602 206/363-1713 We are in receipt of the above referenced request. After review of our records it has been determined that the Department does not have any data on file which could be deemed to be representative of background levels near the above referenced Stauffer plants.

2204 Perimeter Reed 1686. AL 18 479-2336 For your information I am including the following metals data as recorded from soil samples taken at two locations in Pike County by the Department in 1984.

TOTAL METALS (ug/g)

Near Pike County Lake

Arsenic	Calcium	Cadmium	Iron	Lead	Zinc
10.2	214.0	5.0	3,410.0	8.0	10.0
	Roadside	Sample - U.S. H	Wy 231 (Mile	79)	
Arsenic	Calcium	Cadmium	Iron	Lead	Zinc
29.0	472.0	2.0	6,997.0	220.0	44.0

After review of the above, should you have any questions, please do not hesitate to contact this office.

Sincerely,

E. John Williford, Chief Field Operations Division

FJW/RWC/mpt

cc: Joe Downey, ADEM

APPENDIX XVI MATRECON INC'S REPORT ON MEMBRANE TESTS AND INTERPRETATION

Interoffice Memorandum

TO: W. Cawthra - EEC

DATE: 12/5/86

FROM: L. E. Drake - EEC

CC:

SUBJECT: LeMoyne Cap Liner

FILE: 3955A012

P.F. 7.1-9

key word: LeMoyne

The Matrecon, Inc. test results on the samples taken from the duPont 3110 cap liner at LeMoyne indicate to me that there has been only a small change in the important properties of that material since installation in 1974. Its integrity should still be good and useful life expectancy high. A repeat of the tests in 5 years would be recommended.

L. E. Orake

LED/cm

5 IU UU3UZ

matrecon, inc.

RESEARCH . CONSULTING . TESTING . NON-METALLIC MATERIALS

P O Box 24075 Oakland, CA 94623 2811 Adeline Street Oakland, CA 94608

(415) 451-27

M87-343

RECRIVED
OCT 23 1986
ENVIRONMENTAL COMTROL

October 15, 1986

Mr. Danny Flack Stauffer Chemical Company Highway 43 Axis, AL 36505

Dear Mr. Flack:

Re: Report on the Testing of Exposed 3110 Cap Liners - P.O. #025-033112

Enclosed is our report, "Testing of Exposed 3110 Membrane Liner Samples Collected by Stauffer Chemical Company at their Mobile, Alabama Facility."

Also enclosed is the "Chain of Custody Record" signed by Gary Walvatne, which you asked to be returned.

If you have any questions, please contact me.

Sincecely,

Henry E. Haxo, Jr.

President

HEH/rmr

Enclosures: 1 copy of report

1 signed chain of custody record

<u>scc</u>

CC : R. HALSTEAD - oris.

T. SAYERS - WEPT

B. STILSON

B. JOYCE

00303

STAUFFER CHEMICAL COMPANY CHAIN OF CUSTODY RECORD CC/Lemoyne SF RI/FS

Sample Source LINERS Collected By: W.P.STILSON Date: 8 / 88 11 / 86 Time: YARIBUS Relinquished By: W.P.STILSON Date: 8 / 11 /86					_	_			Field Notebook: CERCLA No. 1 Log Notebook:												
No. Sample No. Cetegory & Sample No. Cetegor						•		•	Aste		20 A SI CE		NOTE: LINER MATERIAL IS DUPONT 3/10 20 MIL PUNCAGEDS AND METALSO MY 1979		3//0						
No.	Sample No.	Test Category	1	100	MAC	5	\$	Thick	2	8	ਬੰ	द्व	2	8	Pilocy	1	Shipment Container	Storage Condition —LAB		Commer	nta
1	LLF- IP															X	PLANTIC BAG		ASTM	D 638	TYPE 4
2	SLF-IP															X			••	4	,,
3	NLF-IP															X	44		•	4	",
- النساح				L					_												
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TESTING OF EXPOSED 3110 MEMBRANE LINER SAMPLES COLLECTED BY STAUFFER CHEMICAL COMPANY AT THEIR MOBILE, ALABAMA FACILITY

by

Henry E. Haxo, Jr. Benjamin B. White Gary Walvatne

Prepared for

Mr. Danny Flack Stauffer Chemical Company Highway 43 Axis, AL 36505

Stauffer Chemical Company P.O. #025-033112

October 15, 1986

Matrecon, Inc. 2811 Adeline Street Oakland, CA 94608 TESTING OF EXPOSED 3110 MEMBRANE LINER SAMPLES COLLECTED BY STAUFFER CHEMICAL COMPANY AT THEIR MOBILE. ALABAMA FACILITY

INTRODUCTION

Mr. Danny Flack of Stauffer Chemical Company requested, under P.O. #025-033112, that Matrecon, Inc. determine the tensile properties and volatiles content of three samples of exposed 20-mil 3110 elasticized polyolefin, taken from the cap liners of three separate landfill units. These samples were collected by Stauffer Chemical from their Mobile, Alahama facility. The cap liners had been installed in 1974 and were being checked for elasticity under a Remedial Investigation Feasibility Study (RIFS), as required by the U.S. Environmental Protection Agency. The purchase order called for tensile testing in accordance with ASTM D638-84, using a Die IV dumbbell.

No retained sample was available for comparative testing, nor were there any test data pertaining to the specific lots of 3110 that had been used. The tensile test data available from du Pont, the manufacturer of 3110, was based on ASTM D882-3, which they used in developing their specification for 3110. Consequently, Stauffer approved tensile testing of the exposed samples in accordance with ASTM D882-73; it was expected that specific test data on the unexposed sheeting, would become available from du Pont or the installer so that a comparison could be made with the original property values and, thus determine whether the properties had changed during exposure. However, no property values on the unexposed sheeting became available.

This report presents the results of tensile testing of the exposed 3110 sample, in accordance with both ASTM D638-84 and D882-73 tests, and results of the determination of volatiles content.

SUMMARY

The test values for the exposed cap liner sample, identified as LLF-1P, tested in accordance with ASTM D638-84 were equal to those in the Matrecon database for unexposed 3110 sample, also determined in accordance with the same test method. The values for the properties of the other two samples, SLF-1P and NLF-1P, were somewhat lower, indicating possible loss, but the differences are probably within testing error. The values exceeded the test values given in the du Pont specification (1). However, the du Pont values were obtained in accordance with ASTM Test Method D882-73, which requires strip specimens.

The results of the testing of the exposed samples, in accordance with ASTM D882-73, yield similar modulus values (stresses at 100% and 200% elongation), as obtained in the D638-84 tests. However, the tensile strength values were lower and the strip specimens slipped in the grips to yield false elongation values, when jaw separation was used to measure elongation. The samples were retested and the elongation was followed by benchmarks. Elongation at break values were somewhat lower than the values obtained with the dumbbell specimens, as were the tensile strength values. This is usual when results obtained with strips are compared with those obtained with dumbbell specimens, particularly when testing materials having high elongation.

⁽¹⁾Du Pont Specification E-90896.

Comparison of the D882-73 data with the du Pont specification values, indicate that there may have been some loss in values of tensile strength and elongation at break, and there was stiffening of the sheeting. However, comparison of test values obtained with the dumbbell specimens with those in the Matrecon database, also obtained with dumbbell specimens, indicate that little or no change in properties took place during the exposure since 1974.

We believe the latter data more accurately reflect the property values of the exposed 3110 membrane cap liner.

MATERIAL

On August 12, 1986, Matrecon received three field-exposed samples of 20-mil du Pont 3110 elasticized polyolefin liner. These liner samples identified as LLF-1P, SLF-1P, and NLF-1P by Stauffer, had been removed from the cap of each of three landfills by Stauffer Chemical Company at their Mobile, Alabama facility. These samples were given Matrecon identification numbers E475, E476, and E477, respectively. An inventory and dimensions of the samples are presented in Table 1. Sample LLF-1P (E475) was received with two small cuts (1/4-in. to 1/2-in. long), one 2-in.-long cut, and a hole at the apex of a dimple in the sheeting.

TEST METHODS

Tensile properties of the exposed 3110 samples were measured at the request of the client, in accordance with both ASTM D638-84 and D882-73 test methods.

Tensile testing was conducted in accordance with ASTM D882-73 using 1 in. by 6-in. strip specimens. Initial jaw separation was 2 inches and the rate of jaw separation was 20 inches per minute (ipm).

TABLE 1. INVENTORY OF EXPOSED DU PONT 3110 20-MIL LINER SAMPLES COLLECTED BY STAUFFER CHEMICAL COMPANY AT THEIR MOBILE, ALABAMA FACILITY

Matrecon identification	Client identification	Sample size, in.	Date received
E475	Sample No. LLF-1P, Le Moyne Landfill Cover 8/8/86 @ 10:55 AM, W.P.S.	19 x 19	8-12-86
E476	Sample No. SLF-1P, Cold Creek South Landfill Cover, 8/8/86 @ 2:30 PM, W.P.S.	19 x 19	8-12-86
E477	Sample No. NLF-1P, Cold Creek North Landfill Cover, 8/11/86 @ 8:45 AM, W.P.S.	19 x 19	8-12-86

Matrecon, Inc. M87-343 <1232> September 23, 1986 Tensile testing was also conducted in accordance with ASTM D638-84 using type IV dumbbell specimens and a rate of jaw separation of 20 ipm.

. The volatiles content was measured by weighing a specimen before and after heating it in an oven at 105°C for 2 hours.

RESULTS

Values for tensile strength and volatiles content are summarized in Table 2. Values for tensile strength, measured in accordance with ASTM D638-84, are compared with values obtained from Matrecon's data base for unexposed du Pont 3110 22-mil sheeting which was tested under the same conditions.

Tensile property values, measured in accordance with ASTM D882-73 with strip specimens, are compared with the du Pont specification values taken from the du Pont Brochure E906896. The specification values are generally conservative and may not be typical of the actual test data. During these tests, the strip specimens slipped in the tester grips. This resulted in false elongation-at-break values, that were substantially in error on the high side, as the elongations were measured by jaw separation. Also, the tensile strength values were significantly lower than the values in the specification.

The tests of tensile properties, in accordance with ASTM D882-73 using strip specimens with benchmarks, were repeated. The results presented in Table 3 show comparable moduli (S-100% and S-200%) to those obtained with dumbbell specimens. The ends of the specimens still slipped in the grips and 10 of the 12 specimens broke outside the benchmarks, resulting in low tensile values.

Henry P. Haxo, Jr.

President

Rendmin R. White

Physical Testing Laboratory

rvallyathe

inde Materials Laboratory

HEH/BBW/GW: rmr

Attachments: 3 Tables

4

TABLE 1. INVENTORY OF EXPOSED DU PONT 3110 20-MIL LINER SAMPLES COLLECTED BY STAUFFER CHEMICAL COMPANY AT THEIR MOBILE, ALABAMA FACILITY

Matrecon identification	Client identification	Sample size, in.	Date received
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E477	Sample No. NLF-1P, Cold Creek North Landfill Cover, 8/11/86 @ 8:45 AM, W.P.S.	19 x 19	8-12-86

Matrecon, Inc. M87-343 <1232> September 23, 1986

TABLE 2. TENSILE PROPERTIES AND VOLA LINER SAMPLES COLLECTED BY STAUFFER CHEI

OMPANY AT THEIR MOBILE, ALABAMA FACILITY

		sample	Stauffer identific	ation	Du Pont	Du Pont 3110 22-mil liner (36) from Matrecon database
Property	Test direction	LLF-1P (E475)ª	SLF-1P (E476)*	NLF-1P (E477) a	specification value ^b	
Volatiles, %		0.49	0.32	0.43	•••	0.15
Tensile properties measured in accordance with ASTM D882-73C						
Tensile at break, psi	Machine Transverse	2555 2230	2360 1990	2325 2360	2400 2400	•••
Elongation at break, %	Machine Transverse	d d	d d	d d	600 600	•••
Stress at 100% elongation, psi	Machine Transverse	1015 885	995 930	975 970	750 750	•••
Stress at 200% elongation, psi	Machine Transverse	1075 980	1075 1015	1030 1020	800 800	•••
Tensile properties measured in accordance with ASTM D638-84e						
Tensile at break, psi	Machine Transverse	2810 2495	2570 2215	2445 2610	•••	2715 2525
Elongation at break, %	Machine Transverse	715 635	615 550	600 615	• • •	675 655
Stress at 100% elongation, psi	Machine Transverse	955 940	995 925	985 960	•••	940 905
Stress at 200% elongation, psi	Machine Transverse	1035 1005	1085 1020	1055 1045	•••	1035 1000

^aMatrecon identification numbers.

bfrom du Pont Brochure E90896.

CTested specimen: strip 1 in. x 6 in.

dSlipped in grips.

elest specimen: Type IV dumbbell.

TABLE 3. RETENTION OF TENSILE PROPERTIES OF EXPOSED SAMPLES IN ACCORDANCE WITH ASTM D88-73 USING STRIP SPECIMENS WITH BENCHMARKS

•		sample	Du Pont		
Property	Test direction	LLF-1P (E475)b	SLF-IP (E476)b	NLF-1P (E477)b	Specification value ^C
Tensile at break, psi	Machine	2305	2085	2115	2400
	Transverse	2075	1815	2205	2400
Elongation at break, %	Machine	635	575	600	600
	Transverse	610	505	635	600
Stress at 100% elongation, psi	Machine 'Transverse	970 905	935 880	980 950	750 750
Stress at 200% elongation, psi	Machine	1020	1015	1020	800
	Transverse	965	975	990	800

^{*}Each sample was tested in duplicate in each direction; values are averages. . Out of the 12 specimens tested in the retest, 10 of the specimens broke outside the benchmarks, i.e. between the benchmarks and the grips.

bMatrecon identification numbers.

^CFrom du Pont Brochure E90896.

Matrecon, Inc. M87-343 <1232> October 14, 1986



SPECIFICATION DU PONT 3110° ELASTICIZED POLYOLEFIN SHEETING

P	10	perty

Tensile Strength Modulus @ 100% 200%

Elongation at Break Water Absorption

Cold Bend Test
Brittleness Temperature

Ozone Resistance

After Heat Aging
Tensile Strength
Elongation at Break
Specific Gravity
Graves Tear
Dimensional Stability (linear)

Volatiles or Plasticizer Loss

Test Method

ASTM D-882-73 ASTM D-882-73

ASTM D-882-73 ASTM D-471 (14 days @ 212°F.) ASTM D-2136 ASTM D-746 Procedure "B" }

ASTM D-1149
3 ppm @ 30% Strain
@ 104°F. for 70 hrs.
Procedure "A"
ASTM D-412
(14 days @ 212°F.)

ASTM D-792 (25/4°C.)
ASTM D-1004
ASTM D-1204
After 3 hours @ 212°F.
ASTM D-1203
Method "A"
Thickness 0.021"

Requirement

Min. 2400 pei.
Min. 750 pei.
Min. 800 pei.
Min. 600%
Max. Wt. Change
3.0%
No Cracks at
-60°C. (-76°F.)**

No Cracks
6x Magnification

Min. 2300 pai. Min. 500% 0.90 ±0.05 Min. 300 lb./in.

Max. 4% Max. Wt. Loss 0.5%

"The brittleness temperature is lower than -60°C. (-76°F.), but this was the limit of the test equipment.

Port Trademark for electicized polyelefin shooting and welder.

E. I. DU PONT DE NEMOURS & CO. (INC.) ELASTOMER CHEMICALS DEPT. SUITE 724, BANK OF DELAWARE BUILDING WILMINGTON, DELAWARE 19898 Phone for: Order Service: (302) 774-5929

Technical Assistance: (302) 774-8445 (302) 774-4758



APPENDIX XVII MERCURY IN SWAMP MAP, DWG. 1.3

OVERSIZED DOCUMENT MAP

APPENDIX XVIII TRACE CHEMICAL ELEMENT CONTENT OF NATURAL SOILS

Appendix XVIII

TRACE CHEMICAL ELEMENT CONTENT OF NATURAL SOILS

Element	Common Range	Average (ppm)
Aluminum	10,000-300,000	71,000
Antimony	2-10	-
Arsenic	1-50	5
Barium	100-3,000	430
Beryllium	0.1-40	6
Boron	2-100	10
Bromine	1-10	5
Cadmium	0.01-0.7	.06
Cesium	0.3-25	6
Chlorine	20-900	100
Chromium	1~1,000	100
Cobalt	1-40	8
Copper	2-100	30
Fluorine	10-4,000	200
Gallium	0.4-300	30
Gold		1
Iodine	0.1-40	5
Lanthanum	1-5,000	30
Lead	2-200	10

Appendix XVIII

TRACE CHEMICAL ELEMENT CONTENT OF NATURAL SOILS - CONTO.

Element	Common Range (ppm)	Average (pom)
Lithium	5-200	20
Magnesium	600-6,000	5,000
Manganese	20-3,000	600
Mercury	0.01-0.3	.03
Molybdenum	0.2-5	2
Nickel	5-500	40
Radium	8 x 10-5	
Rubidium	50-500	10
Selenium	0.1-2	.3
Silver	0.01-5	.05
Strontium	50-1,000	200
Tin	2-200	10
Tungsten		1
Uranium	0.9-9	1
Vanadium	20-500	100
Yttrium	25-250	50
Zinc	10-300	5 0
Zirconium	60-2,000	30 0

REF: USEPA Office of Solid Waste and Emergency Response, HAZARDOUS WASTE LAND TREATMENT, SW-874 (April, 1983) Page 273, Table 6.46.

APPENDIX XIX COMPREHENSIVE TOXICITY REVIEW

Mercury

Mercurv exists as the metallic, inorganic and organic forms, which can be changed in the environment. For example, inorganic mercury can be converted to methyl and dimethyl mercury by microorganisms both aerobically and anaerobically. Most of the mercury in drinking water is in the inorganic form.

The ACCIH has established the following TLVs for mercury compounds:

mercury vapor 0.05 mg/m³ TWA aryl and inorganic mercury 0.10 mg/m³ TWA alkyl mercury 0.01 mg/m³ TWA 0.03 mg/m³ STEL

The WHO guideline (1984) for total mercury in drinking water is 1 ug/1. The EPA MCL for mercury is 2 ug/1.

Mercury can be taken up by fish, bioaccumulate, and produce toxic effects in humans ingesting contaminated fish. Mercury concentrations of $0.20~\rm ppm$ or less are normal for freshwater fish. The FDA has established a mercury concentration of $0.5~\rm ppm$ in the edible portion of fish tissue as a criteria for recommending legal action to the Division of Regulatory Guidance.

Carbon Tetrachloride

The ACGIH has established 5 ppm TWA as the TLV for carbon tetrachloride. The STEL of 20 ppm has been deleted until more information is available.

The risk estimates for carbon tetrachloride have been reported by the <u>Safe</u> Drinking Water Committee (1977) based on the chronic animal studies and using a dose per surface area conversion factor. The estimates of <u>lifetime risk</u> for man ingesting 1 us/l are $4.5 - 5.4 \times 10^{-8}$. The 95% upper confidence limits are $1.0 - 1.1 \times 10^{-7}$.

Carbon tetrachloride is considered to be slightly toxic to aquatic species. The EC50 in daphnia at 48 hr is 35.2 mg/l. The LC50 in bluegill at 96 hr. has been reported to be 125 mg/l, 27 mg/l and 38 mg/l. The LC50 in tidewater silverside at 96 hr. is 150 mg/l. The LC50 in limanda at 96 hr. is 50 mg/l. The LC50 in limanda at 16 hr. is 50 mg/l. Carbon tetrachloride has a low potential to bioaccumulate in fish (RTECS, HSDB, 1986).

Rainbow trout were given diets containing 3200 and 12800 ppm of carbon tetrachloride for 20 months. Hepatomas were found in 4/44 animals at the low dose, 3/34 animals at the high dose and none in the controls (RTECS, HSDB, 1986).

Nickel

The average daily oral intake of nickel has been reported to be 400 to 500 ug/day (FR Vol. 50, No. 219, 46977). Most of the nickel that is ingested is not absorbed and is excreted in the feces. Fecal excretion of nickel is 100 times greater than urinary excretion. There appears to be a mechanism that limits the intestinal absorption of nickel in mammals, despite the relatively large amounts of nickel in their food (Drinking Water and Health, 1977).

3 10 00321

The ACGIH has established the following TLVs for nickel compounds:

```
metal 1.0 mg/m<sup>3</sup>
insoluble compounds 1.0 mg/m<sup>3</sup>
soluble inorganic compounds 0.1 mg/m<sup>3</sup>
```

The average concentration of nickel in tap water is approximately 4.8 ug 1 (FR Vol 50, No. 219, 46977). Because of the low toxicity of nickel in food and drinking water, the low concentrations present in drinking water, and the small daily intake of nickel in drinking water (compared with food), there has been no need to establish safe levels for nickel in drinking water. The USEPA National Interim Primary Drinking Water Standards and the WHO European standards for drinking water do not include standards for nickel. However, a guidance level of 0.15 mg/l has been determined based on a two year feeding study in the rat (NOAFL of 5 mg/kg/day) and the estimated intake of nickel from food and air (FR Vol. 50, No. 219, 46978).

The LC50 for Channa punctatis at 96 hr. is 307 mg/l for nickel. The LC50 for Daphnia magna at 3 wk. is 0.13 mg/l. The LC50 for Acroneuria lycoria at 96 hr. is 4 mg/l. The LC50 for Artemia salina at 48 hr. is 163 mg/l. (RTECS. HSDB, 1986).

Nickel affected the hatching process, delayed hatching and increased mortality of embryos when eggs, alevins, swim-up fry and 8 month old Atlantic salmon were exposed to nickel. The maximum acceptable toxicant concentration was approximatel 50 ng/l (RTECS, DSDB, 1986).

Although aquatic organisms may accumulate nickel from their surroundings, there is little evidence for significant biomagnification of nickel along food chains (RTECS, HSDB, 1986).

The EPA has recently issued the following ambient water quality criteria for nickel: The four day average fresh water concentration is $e^{(0.8460[\ln(hardness)]+1.1645)}$ ug/l. The one hour average fresh water concentration is $e^{(0.8460[\ln(hardness)]+3.3612)}$.

Chromium

The ACCIH has established the following TLVs for chromium compounds:

metal	0.5 ma/m3
Cr ⁺²	0.5mg/m^3
Cr ⁺³	0.5mg/m^3
Cr ⁺⁶ (water soluble compounds)	0.05mgg/m^3
Cr ⁺⁶ (certain water insoluble compounds) Cr (chromate ore processing)	0.05 me/m ³

The interim drinking water standard for chromium is 0.05 mg/l. The ambient water quality criterion to protect human health from Cr^{+6} is also 0.05 mg/l (RTECS, HSDB, 1986).

The ambient water quality criterion for total recoverable, Cr^{+6} to protect freshwater aquatic life is 0.9 ug/l as a 24 hr. average, and a ceiling of 21 ug/l. The concentration of total chromium should not exceed 0.05 mg/l for the protection of freshwater organisms (RTECS, HSDB, 1986).

Cyanide

Cyanide is relatively uncommon in most U.S. water supplies. The average cyanide concentration was 0.09 us 1 and the highest evanide concentration was 8 ug/1 in 2595 water samples of 969 U.S. public water supply systems in 1977 (FR Vol. 50, No. 219, 46978).

The ACCIH has established the following TLVs for cyanide compounds

sodium cyanide 5 mg/m³ potassium cyanide 5 mg/m³

The WHO considers a guideline value of 0.1 mg/l to be reasonable for cvanide. The EPA's ambient water quality criteria for cyanide is 3.77 mg/l. Since cyanide levels in drinking water are so low, the EPA has decided not to propose an RNCL for cyanide (FR Vol. 50, No. 219, 46978).

Zinc

Zinc is an essential trace element in human and animal nutrition. The recommended daily allowances for zinc recommended by the NAS are as follows: adults, 15 mg/day; growing children over one year old, 10 mg/day; and additional supplements during pregnancy and lactation. As far as human health in the general population is concerned, the major concern with zinc is not with toxicity, but rather with marginal or deficient intake.

Levels of zinc in drinking water are typically less than 0.2 mg/l, the maximum is 1.5 mg/l (Drinking Water and Health, 1980). Food is the major source of zinc, approximately 12 mg/day. The contribution of drinking water to the daily nutritional requirement for zinc is negligible under most circumstances.

The ACCIH has established the following TLVs for zinc compounds:

zinc chloride 1 mg/m³ TWA
2 mg/m³ STEL
zinc chromates 0.05 mg/m³
zinc oxide 5 mg/m³ TWA fume
10 mg/m³ TWA total dust
10 mg/m³ STEL fume
zinc stearate 10 mg/m³ TWA total dust

The NAS Safe Drinking Water Committee stated that "zinc is an essential nutrient for humans. There is evidence of borderline deficiencies of the element in children in the United States as well as in other parts of the world.... The possibility of detrimental health effects arising from zinc consumed in food and drinking water is extremely remote" (FR Vol. 50, No. 219, 46981). The secondary drinking water standard and recommended primary interim drinking water standard for zinc are 5 mg/l (Drinking Water and Health, 1977).

3 10 Q0323

Pesticides

Drinking water concentration guidelines have been developed by Stauffer Chemical Company for EPTC, butvlate, vernolate, pebulate, molinate and cycloate. The guidelines are derived from the acceptable daily intakes established by the EPA or from the no observable effect levels from various Stauffer toxicology studies. In the calculation of these guidelines, it was assumed that the body weight was 10 kg and that 1 l of water was ingested per day. The proposed safe drinking water levels are listed below:

Compound	Proposed Safe Drinking	Water	Level	(mg/1)
EPIC	0.5			
Butylate	2.0			
Vernolate	0.1			
Pebulate	4.0			
Molinate	0.2			
Cycloate	0.05			

The proposed safe drinking water levels for EPTC and butylate are based on the acceptable daily intake established by the EPA as listed in the Registration Standards. The proposed safe drinking water level for vermolate is based on the no observable effect level for the 2-generation reproduction study in the rat (1 mg/kg/day) and a safety factor of 100. The proposed safe drinking water level for pebulate is based on the no observable effect level in the chronic feeding study in the mouse (40 mg/kg/day) and a safety factor of 100. The proposed safe drinking water level for molinate is based on the no observable effect level in the 3-generation reproduction study in the rat (0.2 mg/kg/day) and a safety factor of 10 $_{\odot}$. The proposed safe drinking water level for cycloate is based on the no observable effect level in the chronic feeding study in the rat (0.5 mg/kg/day) and a safety factor of 100.

Molinate

Subchronic Feeding in the Rat

Molinate was fed to rats at dietary levels to provide 0, 35, 70 and 140 mg/kg/day. Signs of toxicity include increased thyroid and adrenal weights, kidney damage, changes in the ovaries and slight changes in the adrenals. A no effect level was not defined in this study.

In a second study, molinate was fed to rats at dietary levels to provide 0, 8, 16 and 32 mg/kg/day. Signs of toxicity include increased thyroid weights and slight changes in the adrenals. The no deleterious effect level was 8 mg/kg/day.

Subchronic Feeding Study in the Dog

Slightly increased thyroid weight was observed in dogs fed molinate at 41 mg/kg day. No effects were observed at 20 mg/kg/day.

Subchronic Inhalation Study in the Rat

Two subchronic inhalation studies in the rat were performed at Stauffer Chemical In the first study, animals were exposed to molinate at concentrations of 0, 0.1, 0.6, 1.8, or 4.0 mg/m 3 for 6 hr/day, 5 days/wk for 13 weeks. Signs of toxicity included nasal epithelial and sinus alterations, necrotizing rhinitis, slight degeneration of the testicular germinal epithelium and decreased fertility.

In the second study, male animals were exposed to molinate at concentrations of 0, 0.07, 0.16, 0.30, 0.64 or 1.6 mg/m³ for 6 hr/day, 5 days/wk for 4 weeks. Decreased fertility was observed for animals at the 0.64 and 1.6 mg/m³ levels. No effects on fertility were observed at the 0.07, 0.16 or 0.30 mg/m³ levels.

Fertility Study in the Male Mouse

Animals were treated at dosages of 0, 2, 20, 100 or 200 mg/kg/dav by gavage for seven weeks. Decreased fertility was observed in males at the 100 and 200 mg/kg/day levels. The decreased fertility was reversible after a 4 week recovery period. No effects on fertility were observed at the 2 or 20 mg/kg/day levels and no histopathological changes were observed at any of the dosage levels.

Fertility Study in the Male Rat

In the preliminary phase of this study, animals were treated at dosages of 0, 12, or 60 mg/kg/day for 5 days. Subsequently the animals were mated with 1 female/wk for 10 weeks. The results of this study indicate late spermatid development is affected by molinate. No effects were observed at 12 mg/kg/day.

In the final phase of this study, animals were treated at dosages of 0, 0.2, 4, 12 or 30 mg/kg/day for 5-10 wk. Subsequently the animals were mated with 2 females/wk for 1-2 wk. Decreased fertility, changes in sperm viability morphology, motility and concentration, and an increase in degenerating spermatids were observed at the 4, 12 and 30 mg/kg/day levels. No effects were observed at the 0.2 mg/kg/day level for 5 weeks.

Fertility Study in the Male Rabbit

Animals were treated at dosages of 0, 2, 20 or 200 mg/kg/day in capsules for 6 weeks. No effects on fertility or histopathology were observed at any of the dosage levels.

3 10 00325

Sperm Production_Study in the Male Monkey

Animals were treated orally at dosages of 0, 0.2, 10 or 50 mg/kg/day five days; wk for 12 weeks. No effects were observed in sperm motility, sperm morphology, sperm concentration, seminal fluid volume, serum luteinizing hormone, serum follicle stimulating hormone or serum testosterone levels at any of the dosage levels.

Reproduction Study in the Rat

A three generation reproduction study in the rat was performed with molinate at dosage levels of 0, 0.063, 0.2 and 0.63 mg/kg/day in the diet. A reduction in the number of litters produced per number mated and a decrease in pup survival were observed at the highest dosage level. No effects were observed at the low or mid dosage levels.

Rats appear to be the most sensitive to the fertility effect of molinate. Further studies in rats of timed matings and electron microscopy of sperm demonstrated that the fertility effects correspond to the time the sperm are extruded from the Sertoli cells. Flectron micrograph studies showed that the membrane of the midpiece of sperm appeared to rupture or disappear as an early event, even when mitochondria appear normal. Subsequently, head and midpiece separate and total disorganization of the midpiece occurs. This suggest the site of action may be directly on the sperm and that it occurs either just before or just after extrusion from the Sertoli cells. Gross morphology of rat sperm differs considerably from rabbit and monkey sperm, especially with respect to the relative length of the midpiece and the manner of its attachment to the head. In this aspect, human sperm more closely resemble rabbit and monkey sperm. The male monkeys reproductive system is most similar to humans with respect to physiology, hormonal system and sperm morphology.

Teratology Study in the Mouse

Molinate was fed to mated female mice at dietary levels providing dosages of 0, 8 and 24 mg/kg/day on gestation days 6-15. No signs of maternal toxicity were observed and no teratogenic effects were observed in the fetuses.

Teratology Study in the Rabbit

Molinate was administered by oral gavage to mated female New Zealand White rabbits on gestation days 7 through 19 at dosages of 0, 2, 20 and 200 mg/kg/day. Embryofetotoxicity was observed at the 200 mg/kg/day level, but was considered to be secondary to maternal toxicity. The no effect level for maternal and embryofetotoxicity was 20 mg/kg/day. No teratogenic effects were observed at any dosage level.

Chronic Study in the Mouse

Molinate was fed to CAF hybrid mice at dietary levels providing dosages of 0, 3.6, 7.2 and 14.4 mg/kg/day for 99-101 weeks. Additional groups were exposed in utero and during gestation followed by similar dietary feeding for a total exposure up to 76 weeks. Survival at the high dosage level in mice exposed in utero may have been slightly reduced. The no effect level was 7.2 mg/kg/day.

Chronic Study in the Rat

Molinate was fed to Fisher rats at dietary levels providing dosages of 0, 8, 16 and 32-mg/kg/day for the first 18 weeks, and 0, 0.63, 2.0 and 6.32 mg/kg/day for the remainder of the experiment (at least 104 wk). Signs of toxicity included decreased body weight gain, increased testicular weight and possibly decreased kidney weight. No effects were observed at 0.63 mg/kg day.

Genetic Toxicity Studies

Molinate was negative in a Rec-assay with <u>Bacillus subtilis</u> H-17 and M-45 strains. Molinate was negative in tests with <u>Eschericha coli WP2 hor strain</u> and five strains of <u>Salmonella typhimurium</u> both with and without a metabolic activation system. Molinate was negative in a host mediated assay in mice using <u>Salmonella typhimurium</u> G-46. Molinate was negative in a micronucleus test in the mouse. Molinate was negative in an <u>in vitro</u> chromosomal aberration and sister chromatid exchange test with mouse lymphoma cells.

Neurotoxicity Study in the Hen

Molinate was administered to hens orally at 0, 0.02, 0.063, 0.20, 0.63 and 2.0 g/kg two times with a 3 week interval. Mortality was associated at the 0.63 and 2.0 g/kg levels. Degeneration of nerve axons in brain, spinal cord and peripheral nerves was found in all hens at the 0.63 and 2.0 g/kg dosage levels. However, none of the molinate-treated hens exhibited clinical signs of delayed neurotoxicity similar to those seen after administration of the positive control.

Toxicity Study in Mallard Ducks

Molinate was fed to mallard ducks at dietary concentrations of 0.1, 0.18, 0.32, 0.56, 1.0, 1.8 and 3.2% for five days. The LC50 was 1.3% and the no effect level was 0.1%.

Subchronic Toxicity Study in Coturnix Quail

Molinate was fed to quail at dietary concentrations of 10, 100 and 1000 ppm for nine weeks. Signs of toxicity included decreases in feed intake, body weight and hatchability. No effects were observed at the 10 ppm level.

Toxicity Studies to Aquatic Species

The LC50 of molinate in rainbow trout is 0.54 mg/l at 24 hr, 0.29 mg/l at 48 hr and 0.20 mg/l at 96 hr. The LC50 of molinate in Leiostomus xanthurus, an estuarine fish, is greater than 1 mg/l at 24 and 48 hr. The EC50 of molinate in in oysters is greater than 1 mg/l at 96 hr. The EC50 for molinate in Penaeus azterus is greater than 1 mg/l at 24 and 48 hr.

3 10 00327

The toxicity of these materials to aquatic species is listed below:

Compound	Specie	Time Point (hr.)	Endocint	<u>Value</u>
EPIC	bluegill sumfish	96 96	1C50 1C50	14 mg 1 10-35 mg 1
	daphnia	48	LC5 0	14 mg/l
Butylate	bluegill sunfish	96	1020	6.4 mg/l
	trout	96	1020	7.0 mg/l
	daphnia	48	1020	11.9 mg/l
Vernolate	bluegill sunfish	96	1250	8.4 mg/l
	trout	96	1250	9.6 mg/l
	daphnia	48	1250	7.6 mg/l
Pebulate	bluegill sunfish	96	LC50	7.4 mg/l
	trout	96	LC50	7.4 mg/l
	daphnia	48	LC50	2.1 mg/l
Molinate	bluegill sumfish	96	TL50	18.8 mg/l
	trout	96	TL50	6.97 mg/l
	daphnia	48	LC50	19.4 mg/l
Cycloate	bluegill sunfish daphnia	96 48	LC50	4.6-6.8 mg 1 24 mg 1

In addition, a 25 day bioconcentration study has been conducted with radiolabelled cycloate in bluegill sunfish. By day 14 of the depuration phase, 94-98% of the cycloate was eliminated. This indicates that bioconcentration of cycloate is not remarkable and that residues are rapidly cleared from fish tissues once exposure is terminated.

EPIC

Subchronic Rat Feeding Study

EPTC was fed in the diet to groups of rats to provide dosage levels of 0, 8, 16 and 32 mg/kg/dav for 13 wks. Signs of toxicity at the 32 mg/kg day level include decreased body weight and food intake and slight irregularity in hepatic cell size with some glycogen depletion. No adverse effects were observed at the 8 and 16 mg/kg/dav levels.

Subchronic Dog Feeding Study

EPTC was fed in the diet to groups of dogs at dosage levels of 0, 450, 900 and 1800 ppm. Signs of toxicity included decreased brain cholinesterase levels, hair loss and changes in gastric mucosa.

Subchronic Rat Inhalation Study

Animals were exposed to 0, 8.3, 58 or 290 mg/m 3 for 6 hours/day, 5 days/week for three months. Signs of toxicity included decreased food consumption (mid and high dosage levels), increased SGOT (high dosage level), mild transient histopathological changes in nasal epithelium and simuses (mid and high dosage levels), decreased brain cholinesterase levels (high dosage level) and myocardial degeneration (high dosage level). The latter observation was considered by the EPA to be virally induced. No treatment-related effects were observed at the low dosage level.

Rat Teratology Study

EPTC was administered to groups of mated female Charles River CD⁸ rats by gavage at dosage levels of 0, 30, 100 and 300 mg/kg/day on gestation days 6 through 15. Maternal toxicity was observed at the highest dosage level. Slight embryotoxicity may have been observed at the mid-dosage level. No effects were observed at the low-dosage level. No teratogenic effects were observed at any dosage level.

Reproduction Study in the Rat

EPTC was fed to groups of rats at dietary levels of 0, 40, 200 and 1000 ppm for two generations with two matings per generation. No adverse reproductive effects were observed.

Chronic Mouse Feeding Study

EPTC was fed in the diet to groups of mice to provide dosage levels of 0, 5, 20 and 80 mg/kg/day for two years. Signs of toxicity included decreased body weight and food consumption.

Chronic Rat Feeding Study

The rats received EPTC in the feed to provide dosage levels of 0, 5, 25 or 125 mg/kg day for two years. Signs of toxicity included mortality (males at the high dosage level) decreased food consumption and weight gain (both sexes at the mid and high dosage levels), coagulopathy (males at the high dosage level), cataracts (females at the high dosage level), altered carriage of the testes (at the high dosage level), and hindlimb weakness and muscular atrophy, hindlimb peripheral nerve and spinal cord axonal degeneration (both sexes at the mid and high dosage levels). A spontaneous level of mild heart muscle degeneration was observed at the control and low dosage levels. EPTC increased the severity of heart muscle degeneration at the mid and high dosage levels. No tumors were considered to be induced by EPTC administration and no treatment related effects were observed at the low dosage level.

Butylate

Subchronic Feeding Study in the Rat

Butylate was fed to groups of rats at dosage levels of 0, 8, 16 and 32 mg/kg/day for 13 weeks. No effects were observed at any dosage level.

Teratology Study in the Mouse

Butylate was administered to groups of mated female mice at dosages of 0, 4, 8 and 24 mg/kg/day on day 6 through termination of gestation. No signs of maternal toxicity and no teratogenic effects were observed in this study.

Teratology Study in the Rat

Butylate was administered to groups of mated female rats by gavage at dosages of 0, 40, 400 and 1000 mg/kg/day during gestation days 6 - 20. The no effect level for general toxicity was 40 mg/kg/day. No teratogenic effects were observed in this study.

Reproduction Study in the Rat

Butylate was fed to groups of rats at concentrations of 0, 200, 1000 and 4000 ppm for two generations. The no effect level for general toxic and reproductive effects was 200 ppm.

Chronic Feeding Study in the Rat

Butylate was fed to groups of rats at dosage levels of 0, 50, 100, 200 and 400 mg/kg/day for 2 years. Signs of toxicity included decreased weight gain, increased food consumption, increase liver weights and increased hepatocellular hypertrophy. No effects were observed at 50 mg/kg/day.

Pebulate

Subchronic Inhalation Study in the Rat

Groups of rats were exposed to mean pebulate concentrations of 0, 3.4, 16.1 and 79.3 mg/m³ for 6 hr/day, 5 days/wk for 14 weeks. Signs of toxicity include increased in vitro coagulation times, inhibition of brain and erythrocyte cholinesterase levels, kidney damage and changes in masal epithelium. No effects were observed at 3.4 mg/m³.

Subchronic Feeding Study in the Rat

Pebulate was administered to groups of rats at dosages of 0, 8, 16 and 32 mg/kg/day for 14-15 weeks. The only sign of toxicity was irritability at the 32 mg/kg/day level. No effects were observed at the 16 mg/kg/day level.

Teratology Study in the Mouse

Pebulate was administered to groups of mated female mice at dosages of 0, 8 and 24 mg/kg/day during gestation days 6 through termination. No signs of maternal toxicity and no teratogenic effects were observed in this study.

Neurotoxicity Study in the Hen

Pebulate was administered to groups of hems at two dosages of 8.9 g/kg with a three week interval between each dosage. No signs of acute delayed neurotoxocity were observed.

Chronic Feeding Study in the Mouse

Pebulate was administered to groups of CD®-1 mice at dosage levels of 0, 10, 40 and 160 mg/kg/day for two years. Signs of toxicity include decreased erythrocyte count, decreased hemoglobin, decreased hematocrit and increased liver weights.

Vernolate

Subchronic Feeding Study in the Rat

Vernolate was administered to groups of rats at dosages of 0, 8, 16 and 32 mg/kg/day for 14 weeks. No signs of toxicity were attributed with certainty to vernolate.

Subchronic Feeding Study in the Dog

Vernolate was administered to groups of dogs at dietary levels of 0, 400, 900 and 1800 ppm. No signs of toxicity were attributed with certainty to vernolate.

Teratology Study in the Mouse

Vermolate was administered to groups of mated female mice at dosages of 0, 8 and 24 mg/kg on gestation days 6 through termination of gestation. No signs of maternal toxicity or teratogenicity were observed at any dosage level.

Teratology Study in the Rabbit

Vernolate was administered to groups of mated female rabbits at dosages of 0, 2, 20 and 200 mg/kg/day for gestation days 6 through 21. No signs of maternal toxicity or teratogenicity were observed at any dosage level.

Reproduction Study in the Rat

Vernolate was administered to groups of rats at dietary levels of 0, 20, 100 and 500 ppm for two generations with two matings per generation. Signs of toxicity included decreased body weight and food consumption. No adverse reproductive effects were observed at any level.

Neurotoxicity Study in the Hen

Vermolate was administered to hens in two dosages of 10 ml/kg with a 21 day interval between dosages. No evidence of delayed paralysis was observed.

Cycloate

Subchronic Inhalation Study in the Rat

Animals were exposed to 0, 1.2, 13 and 119 mg/m^3 for 6 hours/day, 5 days/week for 68 exposure days (three months). An increase in the incidence and/or severity in Wallerian degeneration of the peripheral nerves was observed at the high exposure level. No histopathological effects were observed in peripheral nerves from animals exposed to low or mid dosages.

Teratology Study in the Rat

Cycloate was administered to groups of mated female rats at dosages of 0, 10, 75, 175 and 400 mg/kg/day for gestation days 6 through 15. Signs of maternal toxicity were observed at dosage levels of 175 and 400 mg/kg/day. No signs of embryofetotoxicity or teratogenic effects were observed.

Neurotoxicity Study in the Hen

A neurotoxicity study in the hen was performed at the very high dosage levels of 10,170 mg/kg (2 doses) and 3,051 mg/kg (10 dosages). The LD50 of RO-NEET in the hen is greater than 10,170 mg/kg. No behavioral or histopathological evidence of muscle or nerve degeneration was observed indicating cycloate does not produce acute delayed neurotoxicity.

Chronic Feeding Study in the Mouse

A two year chronic mouse feeding study was performed at dosages of 0, 20, 60 and 180 mg/kg/day. Histopathological examination included brain, spinal cord, nerve and muscle tissues. No effects on nerve or muscle tissue were detected.

Chronic Feeding Studies in the Rat

The first chronic rat study was performed at Hazleton Laboratories with cycloate administered in the feed to provide dosage levels of 0, 8, 24 and 72 mg/kg/day. The second chronic rat study was conducted at Stauffer Chemical Company's Environmental Health Center (EHC) at dosages of 0, 0.1, 0.5, 3 and 16 mg/kg/day. Neuronal and muscular effects occurred principally in older animals and were identical to age-related changes seen in controls. The changes were distinguishable in treated and control animals only by differences in frequency and severity. An increase in the incidence of posterior neuro-muscular weakness was also observed in the first study; however, no evidence of neuromuscular impairment or neurotoxicity was observed clinically in the second study. A no observable effect level (NOEL) was not defined in the first study. The Environmental Protection Agency (EPA) concurs that the NOEL in the second study was determined to be 0.5 mg/kg/day.

Chronic Mouse Feeding Study

EPTC was fed in the diet to groups of mice to provide dosage levels of 0, 5, 20 and 80 mg/kg/day for two years. Signs of toxicity included decreased body weight and food consumption.

Carbon Disulfide

Carbon Disulfide (CS₂) is a clear, colorless, flammable liquid which is only slightly soluble in Reter (2,200 mg/l # 22° C). The pure chemical boils at 46.3°C. Very little information could be found regarding its toxicity in liquid or solution form. (The oral toxicity to humans, LDL₀ is 14 mg/kg.) NIOSH, in its criteria document for CS₂ vapor (1977), recommended a TWA of 1 ppm, with a 10 ppm ceiling for any 15 minute period.

Thiocyanate

Thiocyanates are not normally dissociated into cyanide. They have a low acute toxicity. Sodium Thiocyanate (NaCNS) has an Oral LD $_{50}$ for rats of 764 mg/kg.

Prolonged absorption may produce various skin eruptions, running nose, and occasionally dizziness, cramps, nausea, vomiting and mild or severe disturbances of the nervous system.

V. 34 NT 1 JUN 0 D 137 REC 31 1 2

ORDRAH ®

(Common Name : Molinate)

Selective Herbicide

TECHNICAL DATA REVIEW

Written by :

G.J. Nohynek, PhD Manager Environmental Services Stauffer Chemical SA

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1. INTRODUCTION

Ordram \odot , containing the active ingredient molinate, is a selective thiolcarbamate herbicide for the control of grass weeds such as Echinochloa spp. in rice. Ordram is used as a emulsifiable concentrate, Ordram 6 E, containing 720 g active ingredient per litre, or as granular formulations containing \gt , 7. \gt or 10% active ingredient. Ordram is applied prior to planting followed by soil incorporation or post-flood, postemergence. Typical application rates are 3.0-4.0 of active ingredient per hectar.

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2. IDENTITY

CHEMICAL NAME AND STRUCTURE 2.1

IUPAC Name :

S-Ethyl hexahydro-1H-azepine-1-carbothioate

CAS - Number :

2212 - 67 - 1

• Structure:

Molecular Weight:

157.30

• Empirical Formula : $C_9H_{17}NOS$

2.2 Common Name

Molinate

2.3. Composition of Technical Product

Purity:

minimum 95% (w/w)

Commercial Formulations 2.4

Ordram 6E:

emulsifiable concentrate containing

720 g molinate/litre

Ordram 5,7.5,10 G:

granular formulations containing

5,7.5 and 10% (w/w) molinate

3. PHYSICAL AND CHEMICAL PROPERTIES - Ordram technical

3.1 Physical Properties

3.1.1 Appearance: Clear, amber liquid

3.1.2 Melting Point : below- 30°C

3.1.3 Specific Gravity: 1.063 at 20°/20°

3.1.4 Boiling point : 202°C (extrapolated)

3.1.5 \apor Pressure: 5.6×10^{-3} mmHg at 25°C

3.2 Chemical Properties

3.2.1 pH : 9.0 (4:4:2 Ordram/acetone/water)

3.2.2 Flash Point: 110°C (TOC) 93°C (TCC)

3.2.3 Solubility at 20°C:

Kerosene: Miscible

Xylene : Miscible

Acetone : Miscible

Ethanol: Miscible

Water : 970-27mg/l

. 4 -

3.2.4 Partition Coefficient - Octanol/Water :

760 at 25°C

3.2.5 Hydrolysis and Solubility Study :

The solubility of molinate in deionized water at 25°C was determined to be 970 ± 27 mg/l. The results of the hydrolysis study indicated that under the experimental conditions employed, i.e. at p $^{\circ}$ 5.0, 7.0 and 9.0 and at 25 and 40°C, no significant hydrolytic break-down of molinate occurred within a 30-day test period. (Stauffer, RRC 64-51, September 1954).

4. ANALYTICAL METHODS

4.1. Formulation Analysis

Active ingredient content of formulations is determined by means of a gas-liquid chromatograph equipped with a 6ft \times 1/4 inch Pyrex glass column packed with 3% OV 17 on Gas Chrom Q 60-80 mesh, a linear temperature programmer and a flame ionization detector. Ro-Neet (cycloate) is used as internal standard. (Stauffer, WRC 75-1, 1975).

4.2. <u>Determination of Crop Residues</u>

Potential residues are extracted from plant samples by steam distillation and are determined by gas chromatography using a 10% 0\ 17 on 80-100 mesh Gas Chrom Q column and a flame photometric detector equipped with a 394 nm filter for specific response to sulfur compounds. The sensitivity of the described method is 0.01 mg/kg Ordram. (Stauffer, WRC 72-43, 1972).

4.3 Water Analysis

Ordram is extracted from the water sample with isooctane and determined by gas chromatography using an OV-17 column and a flame photometric detector in the sulfur mode. Sensitivity of this method is 10 µg/l. Use of an Amberlite XAD-2 macroreticular resin absorption column, subsequent elution with toluene and gas chromatography determination results in lower detection limit of 1 µg/l. The sensitivity of both methods may be further improved by use of gas chromatography / mass spectrometry employing mass fragmento-

graphic detection resulting in a lower detection limit of 50 ng, litre. (Stauffer, RRC 75-13, 1975 and RRC 76-21, 1976).

5. BEHAVIOR OF ORDRAM IN SOIL, WATER AND PLANTS

5.1 Soil Persistence/Metabolism

In a laboratory study the degradation of ¹⁴C-molinate in soil under flooded and nonflooded conditions was investigated. Under moist soil conditions 50% of the applied Ordram was lost within 3 weeks. Under flooded-conditions, dissipation was reduced to 50% loss at 10 weeks. Under aerobic flooded conditions volatilization of was the primary mode of dissipation and little degradation occurred. In moist soils, degradation pathways induce hydroxylation of the azepine ring further oxidation to the respective ketone, oxidation to the sulfoxide and cleavage to yield the imine followed by further degradation reactions, resulting finally in evolution of ¹CO₂:

(V.M. Thomas and C.L. Holt. 1980)

In a study on soil degradation of ¹⁴C-molinate carried out by a Japanese group essentially similar results were found: rapid degradation under unflooded conditions (half-lives: 5-2) days, depending on soil type) and slower soil degradation under flooded conditions (half-lives: 40-160 days). The degradation pathway identified in this study included several routes, all of them resulting in total breakdown to ¹⁰⁰CO₂:

Possible degredation pathways of molinate in soil.

(Imai and Kuwatsuka, 1982)

5.2 AQUATIC DEGRADATION

The fate of [14C] Ordram was studied in natural creek water, with and without a sediment layer of soil. From creek water alone (aerobic system), 50% of the applied C dissipated in 9 days, while from creek water above sediment (anaerobic system) 50% of the radioactivity dissipated in 2 days. Eight weeks after treatment, most of the radioactivity disappeared from the water in both cases, but the sediment still contained 32.0% of the initial radioactivity, the majority of which (25.3%) was in a "bound" state. About 50% of the extractable radioactivity in the sediment at 8 weeks was shown to be undegraded Ordram.

- 7 -

In water without sediment, Ordram rapidly disappeared (not detected after 5 weeks). Many organo-soluble breakdown products were observed with the major products being identified as 4-hydroxy Ordram (42.1%) and 3-hydroxy Ordram (20.7%). In water with sediment, Ordram losses were more rapid than in the aerobic system. The major breakdown products were identified as 4-hydroxy Ordram (72.6%), 4-keto Ordram (14.1%) and 3-hydroxy Ordram (13.3%). The majority of the extractable radioactivity found in the sediment after 5 weeks (3.7%) was identified as parent compound.

After 8 weeks, only a few polar metabolites remained in the flood water held over sediment, with the major product (41%) identified as S-carboxymethyl hexahydro-1H-azepine-1-carbothioate (carboxy Ordram).

After 6 weeks in the non-sedimented system, many polar breakdown products were detected of which hexamethyleneimine (28.9%) and carboxy-Ordram (16.2%) were major. Caprolactam was only detected in both systems the differences in polar metabolite concentrations were attributed to different adsorptive properties of the sediments. The following degradation pathway was proposed:

Proposed degradation pathway of Ordram in water.

5.3. DISSIPATION OF ORDRAM IN RICE FIELDS

follows :

Several reports have been published on this subject :

In a study conducted in Californian rice fields commercially treated with molinate it was shown that the principal route of dissipation was volatilization, followed, by soil absorption photolysis, metabolism and plant uptake. The average half-life in the water was 3 days. The principal routes of molinate dissipation were estimated as

Routes of Molinate loss from an Ordram-Treated Field:

Process	Estimated loss, %	
Soil adsorption and metabolism	10	
Plant uptake and metabolism	5	
Aqueous microbial metabolism	1	
Hydrolysis	1	
Photolysis	5-1 0	
Volatilization to atmosphere	75-85	

(C.J. Soderquist et al., 1973)

• In an investigation conducted by Stauffer Chemical Company in small simulated rice paddies, the half-life of Ordram was 3.5 - 5.5 days and 97% of the applied material had disappeared 21 days after application. The initial water concentration was 2.63 - 2.93 mg/l decreasing to 0.006 - 0.08 mg/l after 21 days. (Gray et al., Stauffer Mt. View Research Center)

- In a study conducted 1975 in Texas, persistence and half-life were evaluated with respect to intermittent and continuous flow irrigation schemes at normal and excessive application rates of 3.4 and 11.2 kg ai/ha of granular Ordram, respectively. Persistence at statistically significant levels ranged from 96 to 354 hours following the application. Half-life values averaged 96 hours in the intermittent flow plots and 54 hours in the continuous flow plots. Application rates had little effect on half-life. From laboratory studies these authors suggested that biological degradation was the principal mode of dissipation. (L.E. Devel et al., 1978).
- In a field study carried out in France (Camargue) a field half-life of 3 days was measured. From initial values of 1.8 - 5.0 mg/l the water concentration stabilized in the vicinity of 10 µg/litre, a level it reached 25 days after application. (Podlejski, 1976).

5.4. OCCURENCE OF MOLINATE IN THE AQUATIC ENVIRONMENT

- In a USSR study, molinate concentrations of discharge water from rice fields were monitored: Peak concentrations of 7-10 µg/l occurred during June to drop to 0.5 - 1 µg/litre during August (Chubenko and Strekosov, 1976).
- Investigations conducted 1976 and 1978 by the Californian Central Valley Water Quality Control Board in cooperation with the State Department of Fish and Game monitored occurence of rice pesticides in drainage systems from rice growing areas. Peak concentrations of molinate were 0.30 - 1.8 mg/litre molinate which occurred 26 - 45 hours after application. Water Residues levelled to 0.2 - 0.55 mg/l 123 - 144 hours after application. (Stauffer, Unpublished Draft Report, 1978).
- In an investigation in France measuring concentrations of several pesticides in effluent waters from rice fields molinate concentrations ranged from maximum 1.5 mg/l on day 2 after application to 0.14 mg/l on day 13 after application of the herbicide. (Podlejski, 1978)

A 3-year study conducted by a group of U.C. Davis, California, in experimental rice paddies monitored molinate concentrations in spill-waters from a flow-through water management system: water residues ranged from 1.5 mg/litre on day 0 to 0.1 mg/litre 3 days after application.
(Tanti et al., 1974)

5.5. LEACHING BEHAVIOR OF MOLINATE

In laboratory column leaching experiments using 5 different soil types.i.e. loamy sand, sandy loam, loam, clay and peat soils and irrigation with 200 mm water, molinate was shown to leach to a depth of 30-40 cm in the two light soils. In heavier soil leaching depth did not exceed 7.5 cm. (Gray and Weierich, 1965). In field studies, no molinate was found in soil cores taken from 2.5 - 50 cm and 17.5 - 20.0 cm soil depth. (Denel et al., 1975). A soil adsorption/desorption study conducted by Stauffer using loamy sand, silty clay and loam soils resulted in identification of adsorption coefficients from 0.72 to 3.03. Molinate soil adsorption is correlated with increasing cation exchange capacity (Thomas and Holt, 1975)

The results of the leaching and adsorption studies indicate that molinate shows little or no tendency to leach in heavy soils and soils with high cation exchange capacity. Limited leaching may be expected in light, sandy soils with low cation exchange capacity.

5.6. METABOLISH AND DEGRADATION IN PLANTS

After application of both chain-labelled Ordram-C and ring-labelled Ordram-C to the roots of rice plants at a rate of 6.7 kg/ha, radioactive carbon dioxide was rapidly evolved from the plants. In six days after application, 4.0% of the applied chain-labelled Ordram was evolved as radioactive carbon dioxide. In thirteen days after application of the ring-labelled Ordram to the roots, 11.4% of the applied radioactivity was evolved as carbon dioxide. Apparently the cyclic imine ring of Ordram was broken open, degraded into small fragments and evolved as CO₂.

The alkyl chain was also broken down and converted into CO₂. The cell sap from the roots and shoots of the treated rice plants was analyzed by one-dimensional and two-dimensional paper chromatography followed by scanning the chromatographs for radioactive components and preparing autoradiographs of them. These analyses showed the presence of seven radioactive amino acids as metabolites of chain-labelled Ordram-C in rice plants. These radioactive amino acids were identified as asparagine, glycine, threonine, alanine, tryptophane, phenylalanine and isoleucine. Several organic plant acids including lactic acid, glycolic acid and three other unidentified organic acids were also found as radioactive metabolites of chain-labelled Ordram-C in rice plants.

Cellulose and other water insoluble materials from the roots and leaves of rice plants became highly radioactive in just three days after treatment with either the chain-labelled or the ring-labelled Ordram-C⁻⁻. The proteins in the cell sap also became labelled with carbon-14. Ring-labelled Ordram was also broken down in rice plants and the radioactivity was found in several amino acids and organic acids. These findings indicated that both the chain-labelled and ring-labelled Ordram-C⁻⁻ were rapidly metabolized in rice plants. degraded into small fragments, converted into radioactive carbon dioxide. amino acids, organic plant acids related to the Krebs cycle, proteins, and cellulose. Thus, most of the radioactivity of the metabolites of Ordram-C⁻⁻ was accounted for in these various natural plan constitutents. (Gray, 1969)

6. METABOLISH OF ORDRAM IN FISH AND MAMMALS

6.1 Metabolism in Japanese Carp

The fate of the herbicide, molinate, in Japanese carp, (Cyprinus carpio) var. Yamato koi, was investigated. [Ring- C] molinate was applied to the water at 0.2 ppm. The overall 'C residues in tissues of fish were low with an average bioaccumulation value of 1.35% during 1- to 14-days exposure. Molinate disappeared rapidly from water containing the carp. It accounted for only 3.8% of the extracted radiocarbon present in the water 14 days after treatment. Molinate was readily converted into various organosoluble and water-soluble degradation products shortly after addition to the water. Molinate sulfoxide, ring-hydroxylated molinate, isomers of keto molinate, keto hexamethyleneimine, hexamethyleneimine, and other metabolites were detected both in the water and/or fish bile. Polar metabolites, but no unchanged molinate, were found in the bile. In vitro, carp liver microsomal mixed-function oxidase systems also produced organosoluble metabolites identical to those found in the water containing living fish. The data suggest that major metabolic pathways for molinate in carp involve sulfoxidation, oxidation to both hydroxy and keto derivatives, and possible conjugation involving carbamoylation of endogenous SH-groups and O-conjugation of HMI.

Proposed metabolism of [ring-14C]-molinate by carp in vitro and in vivo.

6.2 <u>Metabolism in the Rat</u>

6.2.1 Balance and Tissue Residue Study

Excretion of $\frac{14}{4}$ C after administration of [ring- 14 C] Ordram, S-ethyl hexahydro [2-10] azepine-1-carbothioate, to rats was rapid, with approximately 97% of an oral dose (72 mg/kg) being excreted within 45 h after dosing. The major routes of excretion were via the urine (55%, urine - cade wash) and feces (11%). Less than 1% of the dose was detected in the sodium hydroxide air traps. No significant differences were observed between the rates and routes of excretion from male and female rats. Similarly, no significant differences in residual tissue 'C attributable to sex effects were observed. With the exception of blood, residues associated with most tissues substantially decreased over the 7-day period after dosing. Wholebody residues decreased from an average of 13.8% of the administered C at 1 day to 3.7% after 7 days. Seven-day blood residues were dose dependent, decreasing markedly from 40.1 ppm Ordram equivalents at a relatively high dose of 60 mg/kg to less than 1 ppm at a dose of 5 mg/kg. (De Baun, 1975 A).

6.2.2 PATHWAY OF METABOLIC DEGRADATION

Ordram (S-ethyl hexahydroazepine-1-carbothioate) is readily degraded by the rat to more polar products which are excreted primarily in the urine. Unchanged Ordram accounts for only 0.1% of the urinary C after an oral dose (72 mg/kg) of [ring- C] Ordram. The major metabolic pathway involves sulfoxidation and conjugation with glutathione, giving rise ultimately to a mercapturic acid derivative which accounts for 35.4% of the urinary C. Ring hydroxylation to give the 3-and 4-hydroxy-Ordram derivatives (0.8% free, 26.1% as 0-glucuronides) represents another major metabolic route. Hydroxylation in the 2 position of the ring and subsequent ring cleavage represent a minor pathway. Hexamethyleneimine (14.6%) and 3- and 4-hydroxyhexamethyleneimine (10.3%) are major metabolites presumably formed by hydrolysis of sulfoxidized Ordram and its hydroxy derivatives. Although there are small quantitative differences, the metabolism of [ring- C] Ordram in female and male rats is qualitatively the same.

The following metabolic pathway has been proposed:

Proposed metabolism of [ring- 14 C] Ordram in the rat. Percentage are average values for female and male 0-48 h urine after oral dosing with 72 mg/kg. Expressed as percent urinary 12 C.

(De Baun, 1978 B)

7. TOXICOLOGICAL INVESTIGATIONS

7.1. Acute Toxicity

7.1.1 Ordram, technical

Ordram technical has been evaluated for acute toxicity properties in numerous studies with the following results:

Study	Results
Acute oral LD _{>0} , male rat	- : 794 - 926 mg/kg
Acute oral LD _{>0} , female rat	: 501 - 1260
Acute oral LD _{>0} , male mouse	: 795 mg/kg
Acute dermal LD ₅₀ , rabbit	: > 4640 mg/kg
Eye irritation, rabbit	: Slight to Moderate. Reversible Irritant
Skin Irritation, rabbit	: Mild to Moderate Irritant
Acute Inhalation Toxicity:	
4-hour LC ₅₀ , male rat	: 2.9 mg/l
4-hour LC,0, female rat	: 2.4 mg/l
4-hour LC ₅₀ male mouse	: 2.1 mg/l

(Stauffer Chemical Company Toxicology Dept.: T-1402A. T-1402B, T-1646, T-1719, T-1870, T-4401, IRDC / T-6242, T-6>98A)

7.1.2 Signs and Symptoms of Acute Oral Intoxication

Within several minutes following dosage of 1000 and 2150 mg/kg Odram tech., rats appeared depressed and ataxic and exhibited salivation, excessive masticatory movements and lacrimation. At levels of 1000 mg/kg the animals appeared extremely weak and depressed, exhibiting ataxia, intermittent tremors, bloody

exudate around the eyes, signs of excessive urination, labored respiration and hypothermia. These symptoms reached maximum intensity at 24 hours after administration and persisted up until death or disappeared within 72 hours.

Gross autopsies revealed congested and hemorrhagic lungs.

7.1.3 Acute Toxicity of Ordram Formulations

Formulation Form T	ulation * ype	Study	Results
Ordram 5 E (960	g/1) EC	Oral LD ₅₀ , male rat _	794 mg /kg
Ordram S E	•	Oral LD ₅₀ , fem. rat	926 m g/kg
Ordram 5 E		Dermal LD ₅₀ , rabbit	3563 mg/kg
Ordram S E		Dermal Irritation. rabbit	Hoderate Irritant
		Eye Irritation, rabbit	Moderate Irritant
Ordram 6 E (720	g/1) EC	Oral LD ₅₀ , male rat	794 m g/kg
Ordram 6 E Ordram 6 E		Oral LD ₅₀ , fem. rat Oral LD ₅₀ , male mouse	651 mg/kg 1260 mg/kg
Ordram 6 E		Dermal LD ₅₀ , rabbit	10000 mg/kg**
Ordram 6 E		Eye Irritation, rabbit	
Ordram 6 E		Dermal Irritation,	Moderate
Ordram 6 E		rabbit Acute 1-hour inhalatio LC ₅₀	Irritant on → 2.1 mg/l
Ordram 10 G	C	Acute oral LD ₅₀ ,rats	> 5000 mg/kg**
		Acute dermal LD ₅₀ , rabbit	> 5000mg/kg**
		Eye Irritation, rabbit	Mild Irritant
Ordram 7.5 G	G	Oral LD _{5O} , male rat	> 4640mg/kg**
		Oral LD ₅₀ , fem. rat	> 4640mg/kg**
		Dermal LD ₅₀ , rabbit	> 4640mg/kg**

- EC = Emulsifiable Concentrate: G = Granular
- •• Highest Dose Tested

(Stauffer Toxicology Reports T-4956, T-4971, T-64-10, T-65-6, N-3548).

7.2. Dermal Sensitization

The dermal sensitization potential of Ordram Technical was tested using the Open Epicutaneous Test (OET). The test animals, Hartley guinea pigs, were induced over a four week period and challenged on day 29 and day 43. Concentrations of Ordram Technical used for the daily induction application ranged from 3 to 100%. Each animal was challenged with 100, 30, and 10% Ordram Technical, as well as vehicle. There were no apparent adverse effects of Ordram Technical to the guinea pigs, as measured by weight gain, general appearance of behavior.

Repeated application of Ordram Technical did not cause any notable skin reaction during the induction phase. Challenge and rechallenge with Ordram Technical did not produce positive skin reactions in guinea pigs induced with this material. In contrast, the positive control material, DNCB, produced sensitization at both challenge and rechallenge.

In conclusion, Ordram Technical did not produce dermal sensitization in the OET and has therefore none or very low sensitization potential.

(T.R. Castles, Stauffer Chemical Co., T-12007, August 1955).

7.3 Subacute/Subchronic Toxicity Evaluations

7.3.1 Ordram 6 E: 21-Day Subacute Dermal Toxicity in Rabbits

A total of 15 daily dermal doses of undiluted Ordram 6-E appeared well tolerated by male and female albino rabbits at a level of 0.1 ml/kg/day. Moderately irritating to both intact and abraded skin, these signs rapidly subsided within a two week post-treatment observation period.

A level of 1.0 ml/kg/day proved to be toxic to rabbits, resulting in 90% mortality (18 of 20 animals) during the course of the study. Typical of the response were severe skin irritation with secondary infection, anorexia, diarrhea and weight loss. Apparent liver pathology and an abnormal blood picture accompanied several animals.

(G. Meyding, Stauffer Chemical Co., Rep. 64-3, 1964)

7.3.2 Ordram: Safety Evaluation by Dietary Administration to Dogs for 43 weeks.

Groups of 4 purebred Beagle dogs received daily diet containing 450. 200 and 1500 ppm Technical Ordram corresponding to an approximate daily intake of 15,30 and 60 mg/kg body weight over a period of 13 weeks.

No measurable changes were observed in the dogs of the 15 and 30 mg/kg/day groups. In the group of dogs exposed to 60 mg/kg/da/Ordram the following observations were recorded:

- -Slight increase in blood area nitrogen content in 1/4 days.
- -Slightly increased thyroid weights in 3/4 dogs.

450 ppm. corresponding to 15 mg/kg/day was considered as the NOEL under conditions of this study.
(Woodard Research Corporation, Nov. 1964).

7.3.3 Ordram - Safety Evaluation by Repeated Oral Administration to Rats for 13 weeks.

Groups of 15 male and 15 female Charles River rats were administered for 13 weeks diets containing Ordram levels adjusted to daily intakes of 35.70 and 140 mg/kg/day. In the 35 mg/kg/day groups slightly reduced food intake body weight gain was recorded. No other changes when compared to controls were observed at this dose level.

At doses of 70 and 140 mg/kg Ordram produced marked toxicity, expressed as:

- Marked body weight reduction and food intake
- Decrease in hemoglobin and hematocrit values at the high-dose males.
- Increased relative liver, kidney, adrenal, thyroid, testes and ovary weights at high-dose and sometimes at mid-dose level.
- Histopathological findings included at high-dose hepatic cell glycogen depletion, hepatic cell size irregularity, kidney nephron tubular degeneration in the high-dose males, concurrent with testes degenerative seminiferous tubule cell changes and aspermatogenesis.

Under conditions of this study, a precise NOEL could not be defined. (Woodard Research Corporation, October 1964)

7.3.4 Ordram - Safety Evaluation by 13-Week Feeding Study in the Rat

Groups of 15 male and 15 female Charles River Rats were administered a diet containing Ordram at levels adjusted to correspond to a daily intake of 6.16 and 32 mg/kg/day.

At the high and mid-dose levels the following observations were made :

- Reduced body weight and food intake at high dose.
- Increases in relative adrenal, thyroid, testes weights in the males and adrenal/thyroid weights in the females at high dose.
- Ovarian stromal cell vacuolation in mid-and high-dose females and adrenal cortical cell vacuolation in mid-and high-dose males.

No changes, when compared to controls were observed at 8 mg/kg day and this level was considered the NOEL in this study. (Woodard Reseranch Corporation, April 1967).

7.4. Chronic Toxicity/Oncogenicity

7.4.1 Ordram - Repeated Oral Administration to Mice for Lifetime

Groups of 20 male and female CAF1 hybrid mice were administered for 99 to 100 weeks a diet containing Ordram corresponding to a daily intake of 3.6, 7.2 and 14.6 mg/kg/day. Additional groups consisting of 36 male and 28/25/31 female CAF, mice were exposed in utero and through weaning via parental dietary administration and were continued for 76 to 78 weeks on respective dietary levels of 3.6, 7.2 and 14.4 mg/kg/day.

The conclusions of the study were as follows:

Ordram produced no neoplastic changes as revealed by the histological examination of lung, liver, kidney, testis, ovary, and uterus as well as all other tissues appearing abnormal at necropsy. No compound-induced non-neoplastic changes in these same tissues were observed by similar examination. Groups could also not be distinguished on the basis of growth, behavior, body weights, food consumption, organ weights and gross necropsy evaluations. Survival in the high dose mice exposed in utero may have been slightly reduced. The other groups were indistinguishable with respect to survival. Based on the data presented in this study, the no-effect level of Ordram is 7.2 mg/kg/day. (Woodard Research Corporation, June 1977).

7.4.2 Ordram - Safety Evaluation by Repeated Oral Administration to Rats for 104 Weeks.

Groups of 60 male and 60 female Fisher Rats were administered for 2 years dietary concentrations of Ordram corresponding to daily intakes of 8, 16 and 32 mg/kg for the first 18 weeks, followed by dose adjustion to 0.63, 2.0 and 6.32 mg/kg for the remainder of the study. Reduction of dosage was justified by appearance of weight less jeopardizing survival over the study period.

At termination of the study the following observations were recorded in the high-dose (6.32 mg/kg/group):

Slight initial and intermittent reduction in <u>food</u> consumption and weight gain, increased absolute and relative testicular and decreased relative kidney weights. No differences were found concerning behavior, survival, clinical chemistry values and urine analysis, hematology, gross necropsy and histopathological evaluations. No changes attributable to compound administration were recorded at the mid-dose and low-dose level with the exception of silghtly increased testicular weights.

No indications for compound-related induction of neoplastic lesions or non-neoplastic lesions were observed during the histopathological evaluations of the tissues. It was concluded that Ordram has no oncogenic activity and that the NOEL at 2-year dietary administration to Fisher rats is 0.63 $\rm mg/kg/day$. (Woodard Research Corporation, June 1977).

7.5. REPRODUCTIVE TOXICOLOGY

7.5.1 TERATOLOGY STUDY IN MICE

Groups of 20 pregnant mice were exposed to dietary administration of technical Ordram corresponding to a daily intake of 8 and 24 mg/kg from day 6 of gestation to termination (groups II and III) and during days 6-15 of gestation (groups IV and V). On day 18 of gestation Cesarian Section was performed on 10 females/group, the remainder being permitted to deliver naturally.

No signs of maternal toxicity was observed in respect of behavior, body weights, food consumption and gross necropsy observations. No changes were monitored in total number of fetuses, fetal development, resorption sites and fetal weights. Evaluation of cesarean-derived fetuses and naturally delivered pups resulted in no indications of soft-tissue and sceletal malformations. It was concluded that Ordram was nonfetotoxic and nonteratogenic under conditions of this study with a NOEL of 24 mg/kg/day when administered over gestation.

(Woodard Research Corporation, April 1967).

7.5.2 Teratology Study in Rabbits

This study was conducted to assess the effects of Ordram on the pregnant rabbit and unborn conceptuses when administered to dams from day 7 through day 19 of gestation. Ordram was administered by oral gavage at levels of 0, 2, 20, or 200 mg/kg/day in 0.5 ml/kg in corn oil. Data collected included clinical observations, body weights and feed intakes. Necropsy findings included maternal organ weights, counts of corpora lutea, implants resorptions, fetuses and anomalies detected in fetuses on external, soft tissue, or skeletal examinations.

Maternal toxicity occurred in the 200 mg/kg/day dose group as indicated by significant body weight losses during gestational days 14-21 and increased absolute and relative liver weights. The relatively large body weight losses in these dams were associated with an increased occurrence of abortions and a resorbed litter. Embryofetotoxicity at the 200 mg/kg/day dose level was also indicated by a slight delay in sternebral ossification and a reduction in extra ribs. Embryofetotoxicity observed in the 200 mg/kg/day dose group was secondary to the maternal toxicity.

Therefore, treatment of pregnant dams with 200 mg/kg/day of Ordram during gestation caused maternal and embryofetotoxicity. The noeffect level for maternal and embryofetotoxicity was 20 mg/kg/day. No teratogenic effects were observed at any dose level. (S.L. Wilczynski, Stauffer Chemical Co., T-11566, June 1955)

7.5.3 3-Generation Reproduction Study in Rats

Groups of 25 male and 25 female Charles River CD rats were administered dietary concentrations of Ordram over a 3-generation, 2-mating/generation study using doses corresponding to average daily intakes of 0.063, 0,2 and 0,63 mg/kg.

No changes were noted in all groups concerning food consumption, organ weights, reproductive behavior, gross necropsy and histopathological evaluations. The body weight of the high-dose males was slightly reduced. In the high-dose group (0.63 mg/kg/day) there was a slight reduction of litter numbers but not of litter sizes concurrent with slight reduction of pup survival. No pup weight reduction and no indications for pup abnormalities were observed. The NOEL level of this study was 0.2 mg/kg/day. (Woodard Research Corporation, 1975).

7.5.4 Male/Female Fertility Study in Rats

Groups of 8 male and female rats were administered a diet over 12 weeks containing Ordram levels corresponding to a daily intake of 8, 16 and 32 mg/kg. After 12 weeks each animal was mated to an untreated control animal that was kept over the same period under identical housing conditions. The results of this study indicated no reproductive effects following mating of untreated males with treated females. Mating of treated males with untreated females resulted in reduced reproductive performance. (Woodard Research Corporation, May 1975).

7.5.5 Fertility Study in Male Rats

Groups of 20 male rats received dietary daily doses of 0.2, 1.0 and 5.0 mg/kg for nine days and 5,0 mg/kg per gavage for 14 days. 10 and 14 days after initiation the males were mated with untreated females; the 5.0 mg/kg group males at 2 and 4 weeks after end of treatment period. Slightly reduced reproductive performance was observed two weeks after cessation of Ordram at 5.0 mg/kg, with recovery 4 weeks after treatment. No effects were observed at lower doses.(Litton Bionetics Inc., Project No 2621, 1976)

7.5.6 Fertility Study in Rabbits

Thirty-seven male. Dutch Belted rabbits of proven fertility were assigned to four dose groups. There were 10 males in the vehicle control group and 9 males each in the 2, 20, and 200, mg/kg dose groups. The test material was administered daily for 6 weeks in gelatin capsules. Male fertility was determined by mating each male with 2 untreated females during the sixth week of the dosing period and again in the fifth week of the recovery period. Five control males and four treated males from each of the treatment groups were sacrificed at the completion of dosing. The remaining males were sacrificed at the completion of the postrecovery fertility test. The testes plus epididymides, pituitaries, thyroids plus parathyroids, and adrenals were weighed, collected, and submitted for histopathology.

No evidence of impaired fertility was found. There were no treatment-related effects on mating behavior, number of pregnancies, mean number of pups/litter, mean weight/pup, length of gestation, or the mean number of viable pups/litter.

Necropsy results from the interim and final sacrifices revealed no treatment-related macroscopic lesions. There were no treatment-related changes in the mean weights of the testes plus epididymides, pituitary, thyroids plus parathyroids, or adrenals. There were no histological changes with an incidence or severity pattern related to the administration of Ordram.

It was concluded that administration of Ordram at doses up to 200 mg/kg/day over 6 weeks to male rabbits induces no negative effect upon reproductive performance and fertility.
(J. Killinger, Stauffer Chemical Co., T-10176, Nov. 1980).

7.5.7 Fertility Study in Nonhuman Primates

The effect of Ordram on nonhuman primate sperm production and reproductive hormone levels was investigated. Adult male cynomolgus monkeys(Macaca fascicularis) were treated orally with 0, 0.2, 10.0, or >0.0 mg/kg/day of Ordram five days a week for twelve weeks. Sperm samples were collected and analyzed weekly. Blood samples were collected and analyzed every four weeks.

There were no treatment-related changes observed in sperm sample motility, morphology, volume, concentration, or total count. Also, there were no treatment-related changes in serum luteinizing hormone, follicle stimulating hormone, or testosterone. There was a significant depression in plasma cholinesterase activity in the >0.0 mg/kg dose group indicating that Ordram was well absorbed. Thus, Ordram did not cause any measurable effects on nonhuman primate sperm production or reproductive hormone levels.

(J. Killinger, Stauffer Chemical Co., T-10714, Dec. 1951).

7.5.8 Fertility Study in Male Mice

One hundred male, CD-1 mice of proven fertility were randomized into five dose groups; vehicle control, 2, 20, 100, and 200 mg/kg/day. The males were dosed daily by oral gavage for 7 weeks. Fertility was determined by mating each treated and control male with 2 untreated females after 2, 4, and 6 weeks of treatment and again after a four-week recovery period. An interim sacrifice was performed on 5 males/group at the completion of dosing and a final sacrifice was performed on the remaining males after the end of the recovery period. Testes plus epididymides were weighed. The thyroids, pituitaries, testes and epididymides were collected and subjected to microscopic examination.

Treatment-related antifertility effects were observed. After 2 weeks of dosing, there were significant decreases in the number of pregnancies and in the number of implants per pregnancy in the 100 mg/kg group. After 4 and 6 weeks of dosing, there were significant decreases in the number of pregnancies and in the number of implants per pregnancy in the 100 and 200 mg/kg groups. The study clearly shows a no-effect level of 20 mg/kg for the observed antifertility effects. No antifertility effects were observed in the fertility test after the four-week recovery period, demonstrating complete reversability. Ordram did not cause macroscopic changes in the testes or other organs. Necropsy results from both the interim and final male sacrifices showed no dose-related lesions. There were no dose-related changes observed in the mean testes plus epididymides weights from either the interim or the final sacrifice. There were no histological changes indicative of a compound-related effect on the testes, epididymides, thyroids or pituitaries in any treatment group. While treatment-related antifertility activity was observed at relatively high doses (100 and 200 mg/kg/day), no tissue changes occurred as a result of Ordram treatment, determined by microscopic examination.

(J. Killinger, Stauffer Chemical Co., T-10121, Dec. 1980).

7.5.9 Fertility Study in Male Rats-Mechanism and Site of Action

The site and mechanism of Ordram's antifertility effect in male rats, particularly the stage of sperm development affected, was investigated. The study was composed of 4 parts which contained 3 experimental designs linked together by a common dose level.

In part I, male rats received either 0, 12, or 60 mg/kg of Ordram for 5 days, after which they were mated with 1 female per week for 10 weeks. The results of this study phase identified late spermatid development as the stage of spermatogenesis sensitive to Ordram exposure. This is concluded from the substantial reduction in male fertility during the third post-treatment week.

In part II, III, and IV, male rats were treated with either 0, 0.2, 4, 12, or 30 mg/kg of Ordram for either 5 or 10 weeks, after which they were mated with 2 females per week for either 1 or 2 weeks. At terminal sacrifice, blood, sperm samples, and reproductive tissues were taken for evaluation. A correlation was made between dose, changes in plasma hormone levels (testosterone, FSH, LH, TSH, T_3 and T_4), sperm morphology, motility, and viability, and morphologic changes in the testes and epididymides.

There were no measurable treatment-related changes in serum hormone concentrations which correlated with the reduction in male fertility. The dose-response relationship observed for male fertility could be correlated with changes in sperm viability, morphology, motility, and concentration. Histological examination of the testes and epididymides revealed a slight increase in degenerating spermatids. Generally, only a few tubules were involved and often only a small portion of the tubule was affected.

In part I, the no-effect level was shown to be 12 mg/kg/day of Ordram for 5 days. In part III and IV, a clear dose-response relationship could be shown at 4, 12, and 30 mg/kg/day, whereas treatment with Ordram at 0.2 mg/kg/day did not result in a measurable toxicological response. Therefore, 0.2 mg/kg/day of Ordram for 5 weeks is the no-effect dose in this study.(J.L. Minor, Stauffer Chemical Co., T-10421, May 1981)

7.6. Special Toxicological Investigations

7.6.1 Study on Coagulation Parameters, Hematology and Blood Chemistry in Male and Female Rats Exposed to Ordram

The purpose of this study was the determine dose and time/response relationships in coagulation parameters in groups male and female rats administered daily doses of \flat , 20, 80 and 160 mg/kg for 12 days. Blood parameters were evaluated after 4 (\flat), 11 (12) and 20 (21) days of study.

A mild and reversible coagulopathy was produced in male and female rats at a dose level of 160 mg/kg as evidenced by slight increases in APTT and by slight decreases in Factor X at day 4. These values had returned to normal by day 12 in the females and day 21 in the males. There were no mortalities and no abnormalities noted in the hematological and blood biochemical data.

The only adverse gross observations noted were (1) a mild to severe hind-limb weakness which was dose related and reversible in spite of continued treatment and (2) a mild to severe inhibition of body weight gain in both male and female rats. The latter was also dose related and varied from 6% at 20 mg/kg/day to about 40% at the 160 mg dose level. At autopsy slight to moderate hemorrhage was noted in 4/5 male rats at day 4 and in only 1/5 male rats at day 12. At termination at day 21 all male rats appeared normal at autopsy.

In summary, Ordram can produce a mild coagulopathy at very high doses, 160 mg/kg/day which reversed despite continued dosing. The no effect level in the experiment was 80 mg/kg/day for both male and female rats.

(Stauffer Chemical Co., RRC, T-6216, June 1977).

7.6.2 Toxicity to Cattle Sheep and Chicken

In a study conducted by the U.S. Department of Agriculture 3 cattle (age 9 to 16 months) were administered 10 daily doses of 100, 100 and 250 mg/kg. 10 sheep received 10 daily doses of 25, 25, 50, 50, 100, 100, 175, 250, 375 and 500 mg/kg. Groups of 5 chicken were dosed with 100, 250 and 500 mg/kg for 10 days.

Signs of poisoning in cattle and sheep were salivation, anorexia, diarrhea, and ataxia. One sheep had tympanites and another had lethargy and muscular spasms.

At necropsy, the mucosa of the abomasum and intestines was hemorrhagic and lymph nodes were congested. The liver was swollen and often light brown. The spleen was distended and the kidneys were engorged with blood. The respiratory mucosa was often reddened and the thyroid was congested. Chickens had swollen, congested kidneys, light-brown liver, distended gall bladder, and reddened intestinal mucosa.

The NOEL in this study for cattle, chicken and sheep was 100. 100 and 100 mg/kg, respectively. (J.S. Palmer, 1972).

7.7. Mutagenicity Evaluations

7.7.1 Ames Test

Ordram was evaluated for its potential to induce reverse mutations in the Salmonella typhimurium strains TA-1535, 1537, 1535 and TA-100 in presence and absence of a rat liver S 9 metabolic activation system. No indication for mutagenic potential was observed in either strain both in presence and absence of metabolic activation. (Woodard Research Corporation, May 1975)

7.7.2 Ames Test/ Saccharomyces Cerevisiae D 4 Test

Ordram Technical was tested for mutagenic activity in <u>S. thypnimurium</u> strains TA-1535, TA-1537, TA-1536, TA-95, TA-100 and in Saccharomyces Cerevisiae D 4 both in presence and absence of a rat liver S 9 metabolic activation system.

No genetic activity was observed for any of the microbial indicator strains in direct or activation plate tests. (Litton Bionetics Inc, Project No 2547, July 1975)

7.7.3 <u>Mutagenicity Battery - Institute of Environmental Toxicity / Japan</u>

The genotoxic potential of molinate was investigated in the following test systems:

- Rec Assay using Bacillus subtilis strains H 17 and M 45

- Ames test using <u>S. typhimurium TA-1535</u>, TA-1537, TA-1535, TA-95. TA-100 in presence and absence of metabolic activation.
- Escherichia coli WP 2 hor in presence and absence of metabolic activation.
- Host-Mediated Assay in male mice using 2 doses of 30 and 100 mg/kg and S. typhimurium G 46 as indicator strains.

No indication of mutagenic / genotoxic potential was observed in any of these test, while the respective positive control substances induced reproducible responses of the test systems. It was concluded that molinate has no or very low mutagenic potential. (Y. Shirasu et al. Institute of Environmental Toxicology, Sept. 1977).

7.7.4 Mouse Lymphoma L 5178 TK +/- Forward Mutation Assay

Ordram Technical was incubated with Mouse Lymphoma L 5178 cells directly in a dose range of $0.0125-0.28\,\mu l/ml$ and in presence of a rat liver S 9 metabolic activation system.

No mutagenic activity was observed in absence of metabolic activation. In presence of the S 9 system, mutant incidence was marginally increased (2x-5x) toxic at dose levels resulting at < 10% cell survival. However, following current interpretation of the Mouse Lymphoma Test System, only mutant incidence of more than 3x is considered as a mutagenic response if occurring at toxic doses inducing less than 10% cell survival (Clive et al. 1979). It may therefore be concluded that Ordram has low or marginal mutagenic activity in presence of S 9 under conditions of this assay. (J.B. Majeska, Stauffer Chemical Company, T-11840, Sept. 1954).

7.7.> Combined In-Vitro Cytogenetic and SCE Assay in Mouse Lymphoma L 5178 Y cells

Ordram Technical was evaluated for its ability to induce chromosomal aberrations or an increase in sister chromatid exchanges in L5175) mouse lymphoma cells.

In the direct assay, Ordram Technical was tested in the dose range of 0.0125 to 0.2000 ul/ml. With the inclusion of an exogenous metabolic activation system from Aroclor 1254 induced rat livers, Ordram Technical was more toxic, and the dose range was decreased to 0.0025 to 0.0400 ul/ml.

No increase in aberrations or SCE was seen in the direct assay.

In the presence of an activation system, there was a slight increase in aberrations in one out of 4 tests. This response does not meet the criteria for a positive response, and there was no increase in aberrations in the other trials.

In the SCE portion of the assay, no reproducible increase was seen in the presence of an activation system. In Trial 1, the apparent increase at the low dose was influenced by the large variation in solvent control values. A minor increase in SCE at 0.02 and 0.04 ul/ml in Trial 2 was not reproduced in either Trial 1 or Trial 3.

It may therefore be concluded that Ordram Technical is neither clastogenic nor an inducer of SCE in L>179Y mouse lymphoma cells when tested directly or in the presence of an Aroclor 1254 induced rat liver activation system.
(J.B. Majeska, Stauffer Chemical Co. T-11856, Dec. 1953)

7.7.6 Bone Marrow Micronucleus Test in Mice

Ordram was evaluated for its ability to induce micronuclei when administered by oral gavage to mice.

The results of an initial rangefinding study indicated no survival at doses of equal to or greater than 800 mg/kg administered in a single dose. Two consecutive doses of 400 mg/kg administered approximately 24 hours apart did not appear to affect the animals. With a single dosing of 600 mg/kg, there was a reduction in surviving animals; the female animals were apparently more sensitive to Ordram than the male animals. A reduction in PCE frequency (PCE/1000 erythrocytes) was seen at 400 mg/kg (2 dosings) and 600 mg/ml (1 dosing) in both sexes. From these results, a single administration of 200, 400, and 600 mg/kg was chosen for the males and a single administration of 100, 200, and 400 mg/kg was chosen for the females in the micronucleus assay.

No significant increase in the numbers of micronuclei relative to the concurrent solvent controls was seen in the male animals (200, 400, and 600 mg/kg) or female animals (100, 200, and 400 mg/kg) at sacrifice times of 24, 45, or 72 hours after dosing.

It was therefore concluded that Ordram Technical was non-clastogenic under conditions of this test.
(J.B. Majeska, Stauffer Chemical Co., T-11820, Nov.1953)

8. ECOTOXICOLOGICAL STUDIES

8.1. Toxicity to Avian Wildlife

8.1.1 Subacute Toxicity in Mallard Ducks

Groups of 10 Mallard ducklings were administered for 5 days a diet containing 0.1, 0.18, 0.32, 0.56, 1.0, 1.8 and 3.2% Ordram and then returned to untreated diet for 3 days. The 5-day dietary LC 50 in this study was 13000 ppm (= 1.3%). The no-effect level in this study was 1000 ppm (0.1%) Ordram. (Woodard Research Corporation, August 1965).

8.1.2 2-Months Subacute Feeding of Ordram in Coturnix Quail

Technical Ordram was fed in the diet to groups of 10 Coturnix quails at levels of 10, 100 and 1000 ppm for nine weeks. Behavior, body weight, fertility, egg hatchability, relative and absolute organ weights at the 10 ppm group was comparable to controls. Higher dietary levels induced reduced hatchability, feed intake and body weights. 10 ppm was the NOEL in this study.

8.2 Review of the Acute and Subacute Toxicity and Bioaccumulation of ORDRAM in Fish and Aquatic Invertebrates

Surrary

Acute LC53 values obtained from static aquatic toxicity tests of up to 96 hr duration indicated that Ordram is slightly toxic to several species of warm water fish (13-43 ppm), and moderately toxic to rainbow throut (0.2-7.0 ppm). The 48hr LC53 values for Ordram in several species of warm-and cold-water invertebrates varied from 0.2 to 385 ppm. Common carp were exposed to Ordram, Machete or Saturn (Bolero) for 21 days at a range of exposure concentrations. Ordram at 2 ppm produced no effects in carp and was less toxic than either Saturn or Machete. Channel catfish exposed to Ordram 6E formulation for 11 days showed no effects at levels as high as 3 ppm (a.i.). In carp, Ordram is rapidly metabolized to polar metabolites. Bioaccumulation studies with catfish and bluegill indicated that Ordram is concentrated to only a small extent and that these low tissue residues rapidly declined when the exposure was withdrawn. These data are consistent with the conclusion that Ordram presents minimal hazard to the aquatic environment. Acute Toxicity of Ordram to Freshwater Fish

The acute toxicity of Ordram to several species of freshwater fish was determined using the static test procedure. Briefly, Ordram neat, or diluted in organic solvent, was mixed into aerated water and the fish were placed in the test media (water plus test chemical). No additional test material was added nor was the test media aerated during the 4 day (96 hr) test. The acute LC50 values were calculated from the mortality data after 24, 48 and 96 hrs using one of several different statistical methods. To facilitate comparison, LC50 values were recalculated by this reviewer using a standard method (21). However, in many studies, the mortality observed after 24 or 48 hrs was insufficient to permit calculation of LC50 values.

The data for the acute aquatic toxicity of Ordram to several species of freshwater fish are summarized in Table 1. The 96 hr LC50 values for warm-water fish including carp, goldfish, bluegill sunfish, mosquito fish, fathead minnow, and catfish ranged from 5 to 43 ppm. Where more than one test was performed on a particular species, the results were usually consistent and LC50 values varied only by a factor of 2 or 3. In acute studies conducted by Brown et al (6), catfish were treated with Ordram in tap water and in two samples of water from rice paddies. The 96 hr LC50 values for Ordram in tap water and in two samples of paddy water were quite similar (33 and 29 ppm, respectively, see Table 1). However, a much lower LC50 (5 ppm) was obtained for Ordram in a third sample of paddy water. Since only 75% of the controls survived in the latter study the authors suggest that an unknown toxicant was probably present in the paddy water.

For rainbow trout, a cold-water species, the 96 hr LC50 values varied from 0.2 to 7 ppm and were lower than the LC50 values obtained for any of the warm-water species. The 0.2 ppm value obtained by a USDI Laboratory (8) was several fold lower than values obtained in the other two tests in this species (1.3 ppm and 7 ppm). Since the description of the test system used in the USDI study was incomplete, no comparison of test methods can be made to resolve the apparent inconsistency.

In acute tests with fish, signs of toxicity were similar for all the species tested. After 1 or 2 days of exposure to Ordram at the very high toxic doses, the skin coloration became darker and the fish tended to inhabit the upper layers of water and surface often. The fish exhibited hyperactivity, erratic swimming behavior and had labored respiration (exaggerated gill movements). Progressively, the fish lost their ability to swim upright and sank to the bottom and died. Most deaths occurred between day 2 and 3.

Acute Toxicity to Freshwater Invertebrates

The available information on acute toxicity of Ordram to several species of freshwater invertebrates is summarized in Table 2. Most of the data are derived from the scientific literature. Studies were performed under static conditions using an exposure system similar to that described for acute fish tests. Unfortunately, many important details of the test conditions were not provided in the literature reports and for many tests, 96 hr LC50 values were not reported.

For aquatic invertebrates, the 48 hr LC50 values ranged widely from 0.6 to 325 ppm. As shown in Table 2, many apparent inconsistencies in LC50 values are evident within a particular species. For example, the reported 48 hr LC50 values varied several fold for water flea (32 fold), grass shrimp (20 fold), and one species of crayfish (4 fold). Also, marked differences in 48 hr LC50 values are evident between genus and species of Scud (a freshwater crustacean) and crayfish. As with the acute fish studies, the description of the test system was often incomplete, which precluded determination of the reason for the unexpected variability observed in these studies. However, it is clear from the information available that the toxicity of Ordram to aquatic invertebrates is not remarkable.

Similar toxic signs were described for the 8 species of aquatic invertebrates (excluding clam). In general, test organisms exhibited hyperactivity and/or hyperexcitability immediately after addition of the test material which persisted for a few to several hours. Thereafter, markedly reduced activity was noted throughout the test period. As observed in the fish tests, death was usually delayed 1 to 3 days after exporsure.

Subchronic Toxicity of Ordram to Freshwater Fish

Common carp (232 mm, 210 g) were exposed for 21 days in a static test to a range of concentrations (nominal) of Ordram Tech. (0.01, 0.05, 0.1, 0.5, 2.0, 5.0,

10.0 and 15 ppm), the Ordram formulation Mamet SM (0.75, 1.5, 3.0, 7.5, 15.0, 30.0 and 60 ppm), Byram Tech. (0.05, 0.5 and 2 ppm), Saturn Tech. (0.5, 2.0, 5.0, 10.0, and 15.0 ppm), Saturn S (15, 30, and 60 ppm), Saturn SM (15, 30 and 60 ppm) and Machete (0.05, 0.5 and 2.0 ppm). Carp (thirty per dose group) were maintained in 500 liter tanks with sufficient aeration to maintain adequate 0_2 levels (>5.5 mg/1). Biological loading was 30 g/1 and carp were fed once per day during the test. On days 0, 3, 7, 10, 14 and 21, hematocrit (HCT), hemoglobin (HG), erythrocytes (RBC) and white blood cells (WBC) were measured in carp and ammonia was measured in the water.

Ordram Tech. at concentrations of 2 ppm produced no mortality or toxic signs.
Carp exposed to 5 ppm of Ordram or more exhibited a dose-related incidence of toxic signs and mortality. At 5 ppm of Ordram Tech., carp exhibited no mortality but here observed to be dark in color and lethargic throughout the test. Carp exposed to 10 or 15 ppm of Ordram Tech. exhibited hyperactivity and after several days were observed to have areas of apparent hemorrhage on their ventral surface. At the highest exposure levels (10 and 15 ppm), mortality was observed beginning on day 4 and all carp died by day ll. Except for an increase in WBC count at the highest dose, no significant changes in ammonia production, RBC count, HCT or HG were noted. Ordram granular formulation (Mamet SM), at the highest concentration tested (60 ppm of product, 4.8 ppm a.i.), produced decreased HCT and RBC values but no mortality or toxic effects were noted at any of the lower concentrations tested.

Exposure to Byram Tech. at 2, 0.5 and 0.05 ppm for 21 days produced no significant adverse effects on behavior or survival of carp.

Carp exposed to Saturn Tech. at \leq 2 ppm did not exhibit any mortality or toxic signs. However, carp exposed to 10 and 15 ppm of Saturn Tech became dark almost immediately upon initiation of exposure and all carp in both of these treatments died within 24 hours. Carp exposed to the Saturn granular formulations, Saturn S

and Saturn SM, at concentrations \geq 15 ppm of product (1.05 ppm and 1.5 ppm a.i., respectively), expanienced significant mortality and dose-related toxic signs. The toxic signs included darkening of the body, exopthalmos, lethargy, hemorrhagic patches, and decreased RBC and/or WBC counts.

Machete Tech. at \geq 0.5 ppm produced a dose-related incidence of mortality and toxic signs (darkening of the skin, lethargy and apparent loss of equilibrium). At 0.5 ppm, mortality was observed beginning on day 11 of exposure and all carp had died by day 14. At 2 ppm, carp became dark in color and lethargic shortly after exposure, a loss of equilibrium was evident by day 1, and all carp died by day 4. The carp that died after exposure to Machete had evidence of massive internal hemorrhage at necropsy.

Bioaccumulation Studies

Bluegill sunfish were continuously exposed in a dynamic test system to a nominal concentration of 1.0 or 0.01 mg/l 14 C-Ordram for 35 days, after which the remaining fish were transferred to flowing uncontaminated water for 14 days (15). Radiometric analyses of water and fish tissues were performed in intervals during both exposure and withdrawal. Mean measured concentration of 14C-residues in water during the 35-day exposure period was 1.04 ppm for the high level and 0.012 ppm for the low level. No mortality was observed in either dose group. Throughout the test, fish in both treatment levels fed readily and behaved normally. Equilibrium concentrations of accumulated 14C-residue in the edible portion were obtained in both treatment groups during the 35 day exposure period and were approximately 12 times (x)and 16.4% the concentration of Ordram in water, respectively. Fish exposed to 1.04 ppm 14C-Ordram for 35 days had 4.5% more 14C-residue in the non-edible portion than in the edible partion. The corresponding value for fish exposed to 0.012 ppm 14Cordram was 10.2%. Those fish placed in uncontaminated flowing water eliminated 57% and 54% of those 14C-residues present after 35 day exposure for the high and low level, respectively, within 24 hours.

Juvenile channel catfish (Ictalurus punctatus) were continuously exposed in a dynamic test system to nominal concentrations of Ordram 6E equivalent to 0.12, 0.60 and 3.0 ppm of pure Ordram for 11 days, after which the remaining fish were transferred to flowing, uncontaminated water for 7 days (16). Gas-liquid chromatographic analyses of water and fish tissues were performed at intervals during both exposure and withdrawal. Mean concentrations of Ordram measured in the water during the 11-day exposure period were 0.13, 0.62 and 3.17 ppm for the low, intermediate and high levels, respectively. No mortality was observed at any concentration. Based on maximum Ordram concentrations detected in the carcass after 11 days of exposure, concentration factors for the 0.13, 0.62 and 3.17 ppm concentrations were 9.3X, 6.5X and 10.8X, respectively. Those fish placed in clean flowing water metabolized and/or eliminated more than 90% of the Ordram residue present in the carcass within 24 hours. Throughout the study period, the fish appeared normal, fed readily and appeared to be in excellent physical condition.

Field Studies

A cooperative study was conducted by the California Central Valley Regional Water Quality Control Board (24) and the California Department of Fish and Game (25) to assess the toxic effects of commonly used rice pesticides under actual field conditions. Juvenile channel catfish (Ictalurus punctatus) were housed in wire mesh cages placed in drainage water from rice fields. Ordram residues were measured in water samples taken from fields and drains. Mortality, packed cell volume, total plasma protein and brain acetylcholinesterase were measured in the fish. The fields were treated with Ordram 10G at recommended application rates (4-5 lbs a.i. per acre). In addition, methyl parathion and MCPA were also applied to some of the fields during the study. In fields where flood water was held for 6-7 days after application, as is the recommended practice, Ordram residues had declined to approximately 0.2 ppm when the water was released into drains. Where

the irrigation water was not held, Ordram concentrations in drain water reached 0.6 ppm immediately after application (24). Ordram residues in fish placed in drains where water was not held were approximately three fold higher (2.1-4.9 ppm) than those in fish from water that was held before release (0.6-1.4 ppm). A significant drop in blood plasma proteins occurred in the fish from the former treatment, but not from the latter. No mortality was seen in any of the fish tested, nor were any effects on acetylcholinesterase or packed cell volume noted. Metabolism of Ordram in Caro

The metabolic fate of Ordram was investigated in Japanese Carp (Cyprinus carpio) variety Yamato Kai (17). [Ring-14C] Ordram was applied to water at 0.2 ppm in a static system. The overall 14C residue in tissues of fish were low. A maximum concentration factor in the edible portion (muscle tissue) of 1.15 was attained during the 14 day study. Ordram disappeared rapidly from the water containing the carp. It accounted for only 3.8% of the extracted radiocarbon present in the water 14 days after treatment. Ordram was converted into various organo-soluble and water-soluble degradation products shortly after addition to the water. Polar metabolites, but no unchanged Ordram, were found in the bile. These data show that major metabolic pathways for Ordram in carp involve sulfoxidation and oxidation to both hydroxy and keto derivatives. A later study showed that Ordram mercapturate was a major metabolite recovered from the bile of Ordram treated carp (18).

Discussion

In acute toxicity tests under static conditions, Ordram had rather low toxicity to seven (7) species of warm-water fish and moderate toxicity to rainbow trout, a cold-water species. Where more than one test was performed on a species, the results were usually consistent. In similar acute static tests with six (6) species of aquatic invertebrates, LC50 values were variable, ranging from 0.2-385 ppm.

Considerable intra-and inter-species variability was evident in the toxicity of Ordram to aquatic invertebrates. After exposure to toxic doses of Ordram, both fish and invertebrates exhibited hyperexcitibility and/or hyperactivity followed by reduced activity. Death was usually delayed 1-3 days after exposure.

Subchronic exposure of common carp to Ordram technical or granular formulations under static conditions produced no dose-related mortality or incidence of toxic signs at concentrations of ≤ 2 ppm (a.i.). Effects on blood parameters (elevated white blood cell count and/or decreased hematocrit and red blood cell counts) and red discolored areas on the skin (suggestive of hemorrhage) were observed at Ordram concentrations of ≥ 5 ppm (a.i.). Under the same test conditions, the competitive products Saturn (technical and granular formulations) and Machete technical were much more toxic to carp than Ordram as evidenced by higher mortality, and the faster onset and greater severity of toxic signs. Exposure of channel catfish to concentrations of Ordram-6E formulation as high as 3 ppm (a.i.) for 11 days resulted in no adverse effects during the exposure period or the following 7 day observation period.

In a 35 day dynamic test with bluegill, equilibrium bioconcentration (BC) values of 10 to 16 were obtained after 35 days of exposure. More than 4.5 times as much Ordram was detected in the nonedible portion (viscera) as in the edible portion (muscle). In an 11 day dynamic study with catfish, the BC values measured (6.5-10.8) were only slightly lower than those obtained in the longer term study. In both studies, most of the Ordram residue in the fish (54-90%) was eliminated within 24 hours after the fish were placed in flowing uncontaminated water. In carp, Ordram is rapidly and extensively metabolized to more polar products and rapidly excreted (17, 18).

Recent reports in the scientific literature support the conclusion that

Ordram is eliminated rapidly from tissues of fish once exposure is withdrawn. In

comparing the overall bioaccumulation potential of several major categories of pollutants. Barrows et al (22) showed that when bioconcentration factors were low (\leq 1000) the equilibrium time and the biological half-lives were short (\leq 7 days). Barrows et al indicate that these compounds would not be concentrated or retained to any great degree by aquatic organisms. Macek et al (23) studied the bioaccumulation of over 50 pesticides and concluded that, in general, the rate of bioaccumulation of those materials into bluegill tissues is inversely correlated with water solubility. Thus, tissue residues for pesticides such as Ordram with relatively high solubility (800 ppm at 20°C), would be expected to be small and short-lived, provided the source of contamination is not continuous or the duration of exposure prolonged.

Environmental monitoring and field toxicity studies conducted by the State of California support the results of Stauffer sponsored fish subchronic toxicity tests. The environmental monitoring studies indicate that when Ordram is used as recommended, the Ordram residues in drain water are very low (24). The field toxicity studies showed that the small Ordram residues in fish did not produce mortality under actual exposure conditions (25). Bioaccumulation and metabolism studies reported herein clearly show that under both static (19) and dynamic conditions (15, 16). Ordram is concentrated to only a small extent in tissues of fish. Furthermore, these low tissue residues are eliminated rapidly when the exposure is withdrawn (17, 18, 22, 23).

The available aquatic toxicology information is consistent with the conclusion that Ordram presents a minimal hazard to the aquatic environment.

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ACUTE TOXICITY OF ORDINAY TO FRESHMATER FISH

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	Reinbow Trout Salmo geindneri	Stickleback Gasterosteus aculeatus	2002	Posculto Fish	Fathead minnow Proposales profess				Catfish, comon		760 7607	Elvegill Sunfish	Coldfish Carressius auratus	Carp. Common Carping Carpings	Test Organis- (Genus Species)
<u>ه</u>			30.7		742	.	34 (28.4. 45.5)	(33-35)				717		\$1 (42-51)	;;;;;
1.8 (1.9-3.5)	ž		21.4			35.5 (24.4 - 59.2)	15.7 (13.3 - 18.1)	(32-35)			37 (23-60)		>32	t	Certisence Limits
1.3 (0.9-1.9)	7.0 (5.2-9.3)	194	16.4	26	25 (20.5 -	29.45 (19.7 - 45.4)	÷.	(32-35)	(91-11)	t .1	(20-40	(17-21)	(16-53)	\$	70 77
9) ie ch 3). 8;	Tech 992	Ordram 6E Formulation	Ordram 6E Formulation	Ordram 6E	Tech 952						Tech 97.82	Analytical	Tech 97.82	Te c 7	Purity of Ordran
1.29 4.4cm	1.5g 4.9cm) · • •	1.5 - 4.5cm	0.89 4.2cm	0.3	0.39	0. J _y	1.8 - 5.19	0.9g 4.1g	1.5g	1.59	1.5g 3.5cm	4.8g 7cm	FISH Size
•	2				7.1	6.4	6.4	8.5	7			7.1		7.3	0
	13 - 0.5			3.22-02	18 ± 0.5	23 - 2	2 - 52	27 5 5	16 - 17		16 - 19	10 + 0.5	16 - 19	22 ± 0.5	() U
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⁴ LCSO was calculated by the method of Litchfield and Milcomin (1948) unless otherwise indicated.

b Toxicity determined in tap weter.

c Toxicity determined in rice paddy water 1, see reference for details.

I logically determined in rice paddy water II, see reference for details.

LCSO calculated according to Gaddum et al. Br. J. Pharmacol. and Chemotherapy 11:2, 156-60, 1956.

-	ACI	UTE TOXIC!	OF ORDRA	1 10 A 1C	INVERTEBRATES			
Test Organism (Genus Species)	LC50,(TL50), appm 95% Confidence Limits 24 hr 48 hr 96 hr			Purity of Test Organism Ordram Size		Temp. (°C)	рН	Reference
Water flea Daphnia Magna		0.6 19.4		Tech. Tech	EI EIp	21 17	7.4 7.6	10 20
Seed Shrimp Cypridopsis Vidna				Tech.	EI	21	7.4	10
Scud Cammarus fasciatus	• `	0.39 :		Tech.	EI .	15.5	7.4	10
	0.6 (0.3-0.9)	0.4 (0.2-0.6)	0.3 (0.1-0.7)	Tech.			,	
Gammarus lacustris	9.8 (6.9-1.5)	7.6 (6.1-9.5)	(3.5-5.8)	Tech	2 mo. old	15.5	7.4	10
Grass Shrimp Palaemonetes		1.0	`	Tech	EI	21	7.4	10
Kadiakensu	22	20	15.9	8E	25-31 mm			8
Crayfish Ocohectes Nails		5.6		Tech	El Z ,	15.5	7.4	10
Procambarus clarkii		14.0			25-35 mm /	3	8.4	12
	·	58 (42-80)						13
Procambarus simulans	34.7	33.2	21.8	8E	60-70 mm			8
Nactrid Clam Rungia cyneata	750	385	197	8 E		Ý		8

The criterion employed for death was failure to move when appendages and/or antennea were stimulated. The LC50 — were calculated by the method of Litchfield — Wilcoxin (1948).

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APPENDIX XX DATA SUPPORTING SWAMP AS GROUND-WATER RECHARGE AREA

The attached data is offered as evidence that the Cold Creek swamp does recharge the groundwater at the Stauffer LeMoyne/Cold Creek Complex and that the groundwater flow is generally to the south. The attached figures are described as follows:

- Pigure 1 Shows Cl⁻ Concentration Trends in Wells #2, #5, #6, & #7 during 1972.
- Figure 2 Shows Cl⁻ concentration Trends in Wells #5 & #6 from 1974 until 1978.
- Figure 3 Static Water level elevations with contours for Four Ouarters of 1984.
- Figure 4 Hypothetical Representation of Area Groundwater flow after three barrier wells are installed on Stauffer south property line. Contours by Ground Water Associates Inc. in 1978. Covers from Virginia Chemicals to Shell Chemical. Does not include M&T Chemical. Stauffer wells #2, #5, #6, #7 and #10 locations marked.
- Figure 5 Ground Water Associates Water Table Surface Map (1978) showing predicted results of installing three intercept wells on Stauffer south property line. Location of Stauffer wells #2, #5, and #6 are marked.

From Figure 1, it can be determined that Wells #5 and #6 both show strong increasing trends in chloride concentration. Referring to Figure 5, we see that these two wells are located south of the Cold Creek swamp and are down gradient of the swamp.

From Figure 2, we can see a significant decrease in chlorides concentration beginning in early 1976 for both wells #5 and #6. The major reason for this rapid decline in groundwater chloride concentration is that the effluent pipeline to the river was installed at the end of 1975 and plant effluent (high in chlorides) was no longer allowed to flow thru the Cold Creek swamp.

Figure 3 shows the contours of the water table surface through the plant area in 1984 (Quarterly RCRA Data) and is a further proof of groundwater flow to the south.

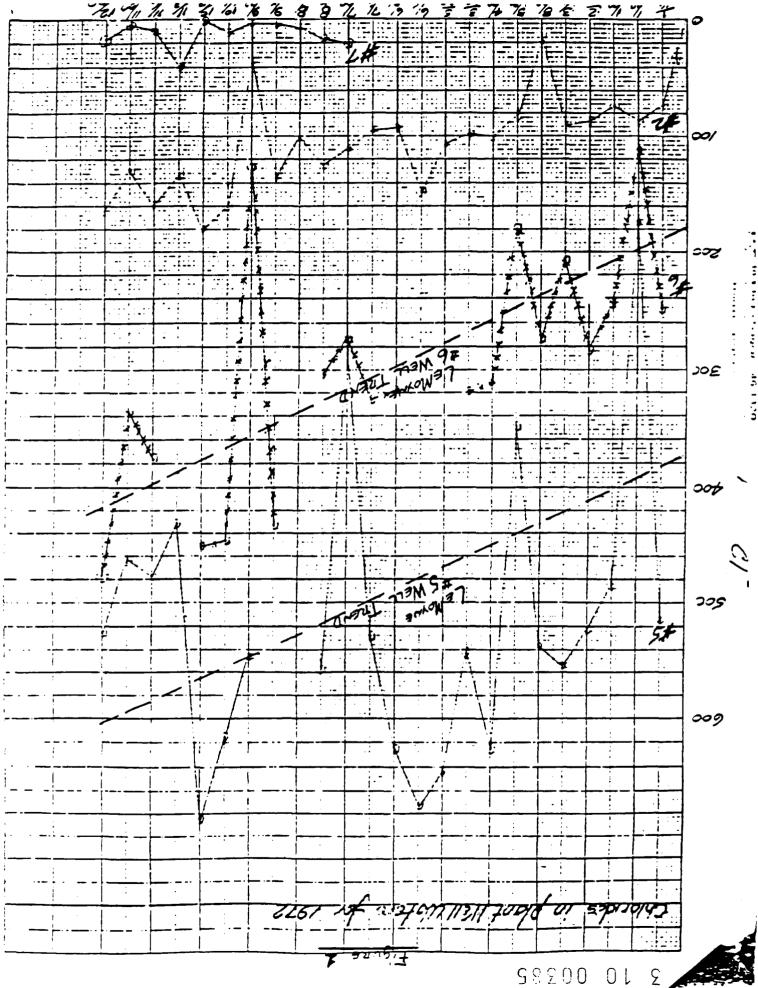
Figure 4 shows a hypothetical contour of the water table surface (1978) after installation of the three intercept wells. It is a further indication of groundwater flows towards the south. Note that the groundwater south of Courtaulds flows to the north and that Courtaulds has, by far, the largest influence on groundwater flows in the area.

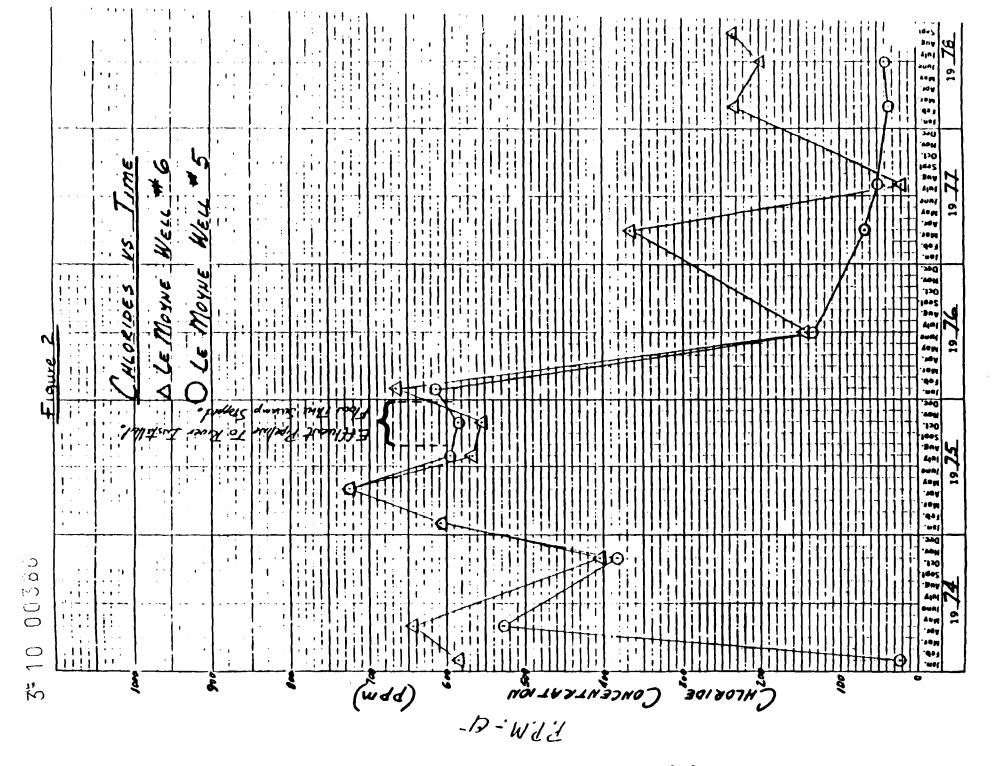
Page 2

Based on this information, we can conclude that the groundwater flow at the Cold Creek/LeMoyne complex is generally to the south and that the Cold Creek swamp recharges the aquifer. We may further conclude that the swamp has been washed constantly into the aquifer and by surface flow to the river since the end of 1975 and that all water soluble contaminates have now been removed.

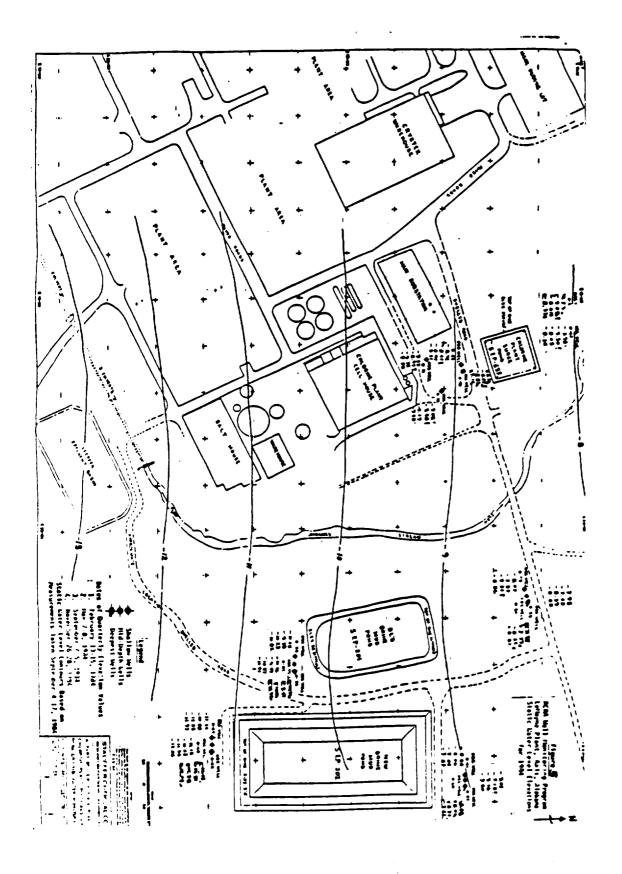
Rule Haltead, P.E.

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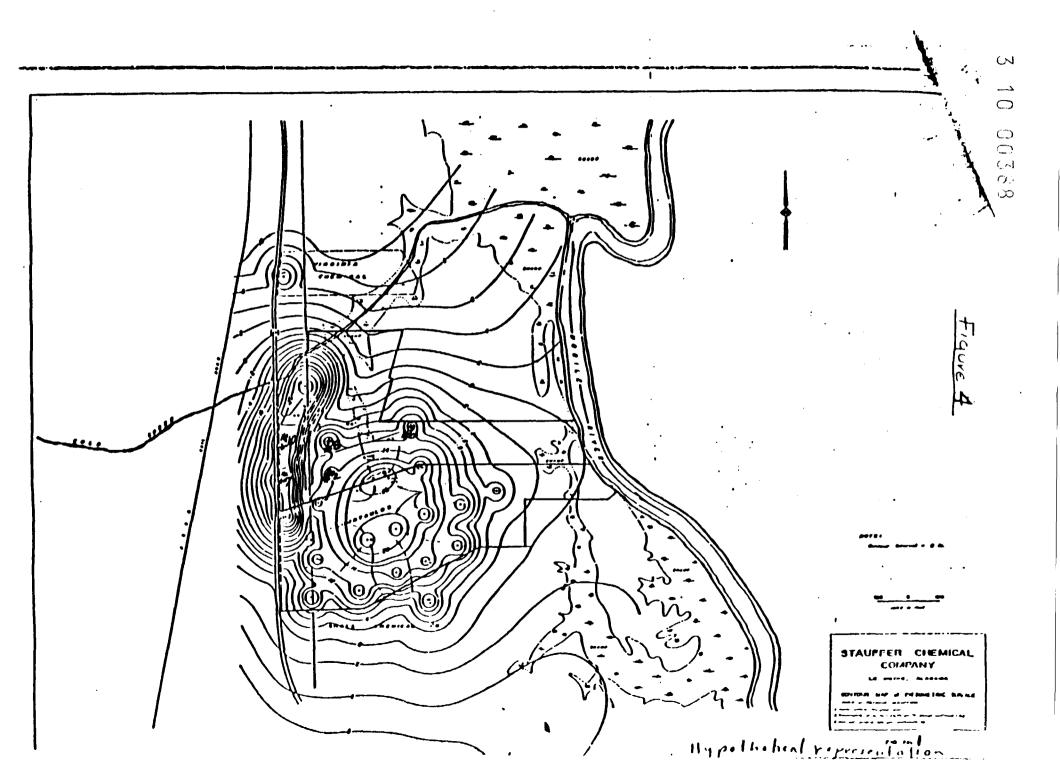


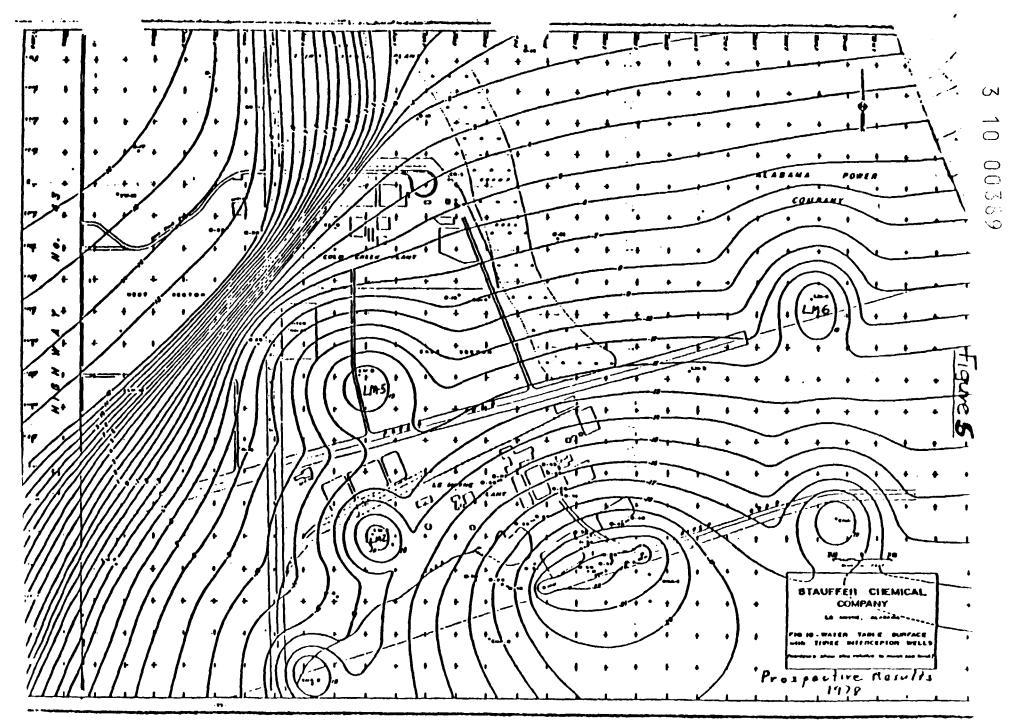


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Figure

APPENDIX XXI IDENTIFICATION OF SAMPLED FISH

3 10 00391

OF THE No. 10

SUBTECT: BIOLOGICAL (FISH) SAMPLING

REPORT FOR STATE OF ALA.

DEPT. OF CONSERVATION AND NATURAL RECOU

DIVISION OF GAME AND FISH

- D DATE OF COLLECTION: THURSDAY AUGUST 7, 1986
 TIMES OF COLLECTION: YARIOUS BETWEEN 10:30 AM AND 4:30 PM
- PLACE OF COLLECTION: STAUFFER CHEMICAL CO. PROPERTY AID ADJACENT ALABAMA POWER COMPANY LANDS, NEAR AXIS, ALABAM (MOBILE COUNTY). SWAMP ALONG COLD CREEK AND DID UN NAMED. TRIBUTARY TO COLD CREEK.
- 3 SAMPLING PARTY: GEORGE YOUNG STAUFFER CHEMICAL CO.

 WILLIAM STILSON STAUFFER CHEMICAL CO.

 MIKL NAME THOMPSON EMINGERING & TESTING CO. (WIRKING FOR STAUFFER & MC KEE (CDM) FUR FED. EPA

 MALCOME PIERCON ALABAMA POWER CO.

 STEVE KROTZER ALABAMA POWER CO.
- (4) SAMPLING METHOD: BACIL PACK ELETRIC SHOCKER UNIT OWNED & OPERATED BY

 ALABAMA POWER COMPANY. PERSONNEL

 FISH SEINE USED AT ONE LOCATION.

 MINNOW TRAPS TRIED AT TWO LOCATION WITH VERY LITTLE SUC
- 5 SAMPLE SIZE: STATION BA-1 ZO SMALL FISH, (110 Grouns) 12 FISH (160 gm) = 210 gm.
 41.7

STATION BA-Z GFISH (ZSOGRAMS) PLUS ONE AMERICANO EEL 12'UNG ZSO (NOT INCLUDAD IN SAMPLE)

14.3

37.5

STATION BA-3

7 FISH (100GRAMS), 4 FISH 150CRMS = ZSOGRAM

STANFFER SPLIT (1/2 OF SAMPLE) (ALA. POWER CO. TOOK EQUAL AMOUNT)

STATION BA-4 ZI VERYSHALLFISH (50 gms), (0 FISH (150 gms)) = ZOU STAUFFER SPLIT (1/2 OF SAMPLE) (ALA. POWER CO. TOOK EQUAL AMOUNT)

STATION BA-5 10 FISH (260gmo) + 1 FISH (300 pmo) = 260 NOT INCLUDED INSUPUR STAUFFIR SPLIT (1/2005 SAMPLE) (ALA. POWER CO. TOOK EQUAL AMOUNT) DOEKFERRY

(DISPOSITION OF SAMPLE: STAUFFER SAMPLES WERE FRUZEN AND SHIPPED TO DUR EASTERN RESEARCH L WHERE THE WERE GRUND UP TO POWDER AND KEPT FROZEN.

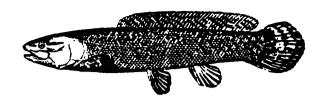
State or Country: Alabama	Field No
County: Mobile Map:	
Locality: Swamp tributary to th	e Mobile River 3.0 miles
County: Mobile Map: Locality: Swamp tributary to the SSE of Bucks	TISRIE Seci
Water: Turbid with alsas bloom Vegetation: Eckerel Wed Panicum	
Vegetation: <u>Fickerel Weel Fanicum</u>	n. Hairgrass, etc.
Dans dark fatil wise	actually silt and organic material
Cover: locs; limbs, stumps etc.	Tomairy Site and organic material
Shore:	_Temp:
Dist. offshore:	Streem width:
Depth of capture: To 2 fect	_Stream width:
Depth of capture: To 2 fect Collected by M. Pierson and S.	Knteer
Tide:	Date: / August 1986
Mathod of conture: 34/KOCK FI	PCT mmfm C B i i A a
Orig. preserv.: All Fishes forcen	for Time: 1030 - 1230
mercury analysis	# of specimens approximate
, , , , , , , , , , , , , , , , , , ,	
1. Amia calva	2 Bowlin
2. Anguilla rostrata (n.t.	retained) American Ecl
3. Notemigonus crysoleucas	18 Goden shirer
4. Erimyzon sucetta	1 Lake chubsucker
5. Ictalurus natalis	Yellow bullreat
b. Fundalus chrysotus	18 Golden topminnows
7. Aphedoderus sayamus	5 Pirate Paren
8. Lepomis gulosus	6 Warmouth
9 Lepomis macrochirus	8 Bluezill
la Lepomis marginatus	5 Dollar sunfish
11. Lepomis megalotis	6 Longar sunfish
2. Lepomis punctatus	6 Spotted sunfish
3. Micropterus salmoides	6 Largemouth bass
	,

Town hip on Swith, Roge one East, as one

State or Country: Alabama	= AA OUI _ C
State or Country: Habama	_Field No
County: Mobile Man: Locality: Cold Creek 3. miles 55	W of Bucks downstream of
	S RIW Sec 11
	N., LongW.
Water: Clear W/ Tannin stain	
Vegetation: Typha, Panicum	
Bottom: silt, sand	
Cover: heavy vegetation Te	emp:
Shore:Cu	urrent: <u>moderate - slow</u>
	ream width: <u>20 - 30 </u>
	epth of water:
•	rotzer
Tide:Da	ite: 7 August 1986
Method of capture: Backpack Elect	tratishing
	Time: 1400 - 1500
mercury analysis	# of specimens
1 National and Completion	12 Golden shiner
1. Notemiganus crysoleucas	12 Golden Shirter
2. Fundulus olivaceus	2 Blackspotted topminnou
3. Lepomis gulosus	3 Warmouth
	10
4. Lepomis macrochirus	10 Bluegill
	
5. Lepomis marginatur	4 Dollar sunfish
I leavis manalatis	6 Lonaear Sunfish
6. Lepomis megalotis	6 Longear sunfish
7. Lepomis microlophus	1 Redear sunfish
7.12.010	1000
& Lepomis punctatus	3 Spotted sunfish
	, <u>, , , , , , , , , , , , , , , , , , </u>
9. Micropterus salmoides	1 Largemouth bass
	42
(No. of speci	mens estimated)

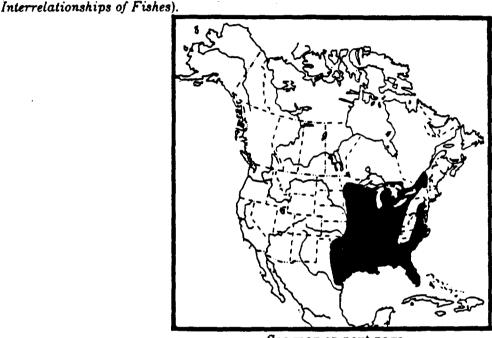
Amia calva Linnaeus Bowfin

Order Amiiformes Family Amiidae



Female (Jordan and Evermann 1900).

TYPE LOCALITY: Charleston, SC (Linnaeus 1766. Systema_naturae, Laurentii Salvii, Holmiae. 12 ed., 1:1-532). SYSTEMATICS: Sole living representative of Amiiformes, which first appeared in Triassic and was well developed in middle Mesozoic (Bailey 1971. McGraw-Hill Encycl. Sci. Technol., 3 ed: 339-40). Genus Amia extends back into Upper Cretaceous, and occurs in early Tertiary in North America (Patterson in Greenwood et al. (eds.) 1973.



See map on next page

DISTRIBUTION AND HABITAT: Known from St. Lawrence and Ottawa rivers and Lake Champlain west throughout Great Lakes, including Georgian Bay and lakes Nipissing and Simcoe, ON; south in Mississippi basin from Lake Winnibigoshish, MN, to LA: in lower TX drainages west to Colorado River; and along Coastal Plain from AL to eastern PA. Changes in environmental quality have probably all but eliminated it from Missouri system, where it was historically known as far north as eastern SD (Churchill and Over 1933. Fishes of South Dakota). Pflieger (1975. Fishes of Missouri) stated that all Missouri River records may be attributable to introduction, a view not subscribed to here. Introduced in a number of localities in IA, IL, NC, and CT. Inhabitant of sluggish, clear, often vegetated, lowland waters.

ADULT SIZE: 457-610 mm TL. 870 mm TL maximum.

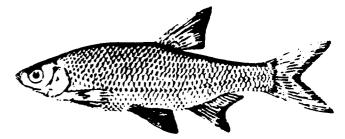
BIOLOGY: Commonly used as laboratory test animal because of status as a "living fossil", ease of maintenance, and interesting behavioral and physiological attributes. Capable of gulping air at the surface and can withstand high temperatures, and is even known to aestivate (Neill 1950. Copeia: 240). Males build circular nests in shallow, vegetated areas in the spring, and protect schooling young after hatching. Mansueti and Hardy (1967. Development of Fishes of the Chesapeake Bay Region) described early development. Voracious predator that favors fish but will consume virtually any type of animal. Hoffman (1967. Parasites of North American Freshwater Fishes) listed parasites. May live at least 30 years (Carlander 1969. Handbook of Freshwater Fishery Biology Vol. 1). Biology summarized in Scott and Crossman (1973. Freshwater Fishes of Canada) and Pflieger (1975).

Compilers: G. H. Burgess and C. R. Gilbert. November 1978.

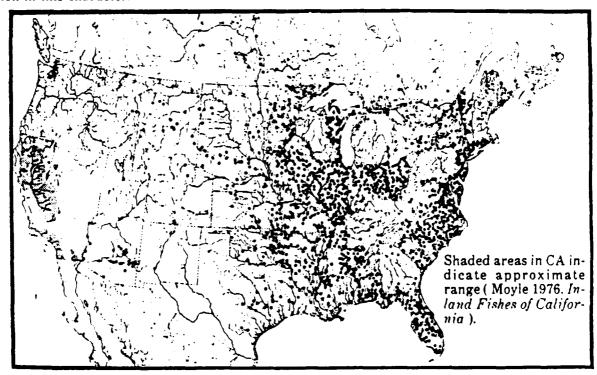
Notemigonus crysoleucas (Mitchill) Golden shiner

TYPE LOCALITY: New York (Mitchill 1814. Rept. on Fishes of New York: 1-30). SYSTEMATICS: Possibly more closely related to certain Eurasian cyprinids than to any North American group (Gosline 1974. Jap. J. Ichthyol. 21: 9-15). Three subspecies have been recognized — N. c. crysoleucas in northeast, and N. c. auralus and N. c. bosci in south — but recent authors have not considered these valid. Variation in anal fin ray count appears to be influenced by water temperature during development (Hubbs 1921. Trans. III. State Acad. Sci. 11: 147-51; Schultz 1927. Pap. Mich. Acad. Sci. Arts Letts. [1926] 7: 417-32). Scott and Crossman (1973. Freshwater Fishes of Canada) discussed and provided additional data on geographic variation in this character.

Order Cypriniformes Family Cyprinidae



MD: Anne Arundel Co., Lake Waterford, 101 mm SL (NCSM).



DISTRIBUTION AND HABITAT: On Atlantic slope from Maritime Provinces south to FL, west to TX, and north to SA. Widely used as bait and ornamental thus transplanted into many areas including parks, in United States. Miller (1952. Calif. Fish Game 33:7-42) provided details on stocking in southwest. Prefers clean, quiet, vegetated water with access to extensive shallows. Common to abundant in ponds and lakes. Often in streams and rivers where, in sluggish sections, it may be abundant.

ADULT SIZE: 53-234 mm SL.

BIOLOGY: Because of value as forage species, considerable information is available concerning propagation and biology (see Carlander 1969. Freshwater Fishery Biology Vol. 1 for summary). Keast and Webb (1966. J. Fish. Res. Board Can. 23: 1845-67) studied foods and concluded it was surface and midwater feeder. Scott and Crossman (1973) provided summary of general biology.

Compiler: D. S. Lee. March 1978.

3 10 00396

Erimyzon sucetta (Lacepede) Lake chubsucker

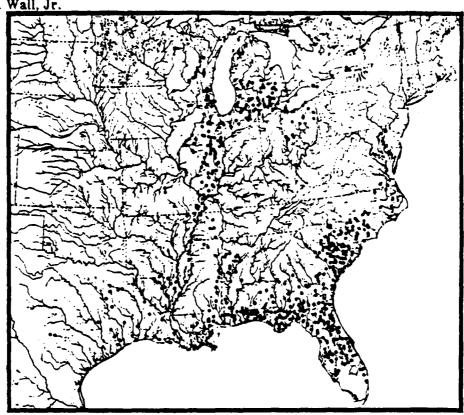
TYPE LOCALITY: South Carolina (Lacepede 1803. Histoire Naturelle des Poissons 5:1-803).

SYSTEMATICS: Hubbs (1930. Misc. Publ. Mus. Zool. Univ. Mich. 20:1-47) recognized subspecies E. s. sucetta and E. s. kennerlii (sic), but Bailey et al. (1954. Proc. Acad. Nat. Sci. Phila. 106:109-64) and most subsequent authors have regarded differentiation invalid. Variation in genus currently being studied by B. R. Wall, Jr.

Order Cypriniformes Family Catostomidae



SC: Barnwell Co., Savannah River system. 173 mm SL (Smith-Vaniz 1968).



DISTRIBUTION AND HABITAT: Atlantic slope north to VA and south to tributaries of Lake Okeechobee, FL; Gulf slope south to Charlotte Harbor drainage, FL, and west to Guadulupe River system. TX. Mississippi Valley in LA. AR, southeast MO. MS, west TN, west KY. IL, IN, and OH. In southern tributaries of Great Lakes drainage to lakes Michigan, Huron, Erie, and Ontario. Found on Canadian side of lakes Erie and St. Clair. Common on lower Coastal Plain; less abundant in inland portions of range. Occupies ponds, oxbows, sloughs, impoundments, and similar waters of little or no flow that are clear and have bottoms of sand or silt mixed with organic debris. Aquatic vegetation usually present. Abundance apparently declining in areas subject to siltation.

ADULT SIZE: 130-386 SL mm.

BIOLOGY: Odum and Caldwell (1955. Copeia:104-06) investigated respiration in anaerobic FL spring. Hildebrand (1967. Trans. Am. Fish. Soc. 96:414-16) studied effects of herbicides on fertilized eggs and fry. Reproductive biology summarized by Scott and Crossman (1973. Freshwater Fishes of Canada). Overall summaries by Trautman (1957. The Fishes of Ohio). Pflieger (1975. The Fishes of Missouri), and Smith (1979. The Fishes of Illinois).

Compilers: B. R. Wall, Jr. and C. R. Gilbert. January 1980.

3 10 00337

Ictalurus natalis (Lesueur) Yellow bullhead

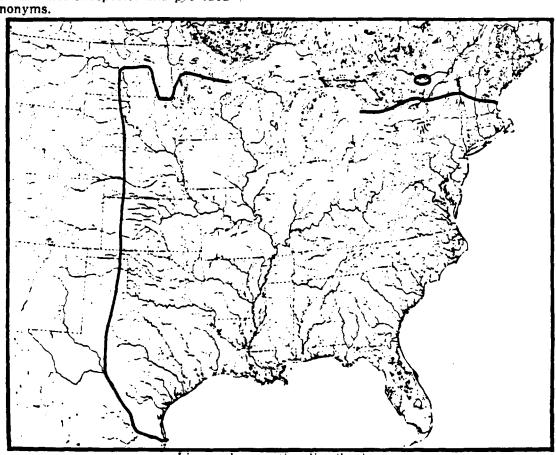
TYPE LOCALITY: "North America" (no locality mentioned) (Lesueur 1819. Mem. Mus. d'Hist. Nat. Paris 5:148-61).

SYSTEMATICS: No definitive systematic study of species. Two subspecies recognized at one time but synomized by Bailey et al. (1954. Proc. Acad. Nat. Sci. Phila. 106: 109-64). Scott and Crossman (1973. Freshwater Fishes of Canada) commented that I. n. erebennus and I. n. natalis still often considered distinct subspecies and provided list of synonyms.

Order Siluriformes Family Ictaluridae



FL: Alachua Co., Gumroot Creek, 64 mm SL (NCSM).



Line encloses native distribution

DISTRIBUTION AND HABITAT: Originally throughout eastern and central United States; also introduced outside native range. Common in areas of heavy vegetation in shallow, clear bays of lakes, ponds, and slow moving streams (Scott and Crossman 1973). Tends to inhabit smaller, weedier bodies of water than *I. nebulosus* in southern part of range. Common.

ADULT SIZE: ca. 380 mm TL.

3IOLOGY: Breder and Rosen (1966. Modes of Reproduction in Fishes) discussed reproduction. Carlander (1969. Handbook of Freshwater Fishery Biology Vol. 1) sum-

marized fishery information, diet, and age class-length relationships. Scott and Crossman (1973) discussed parasites and predators. Yerger (1953, Copeia:115) reported a snake predator. Taste system structure and function described by Atema (1971, Brain Behav. Evol. 4:273-94). Orientation and taste discussed by Bardach et al. (1967, Science 155:1276-78). Chemical recognition, agnostic behavior, and establishment of social hierarchies discussed by Todd (1971, Sci. Am. 224[5]:98-108). Effects of detergents on taste buds reported by Bardach et al. (1965, Science 148:1605-07).

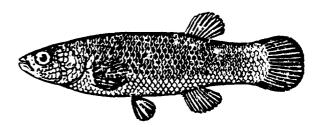
Compiler: G. S. Glodek, August 1979.

Fundulus chrysotus (Günther)
Golden topminnow

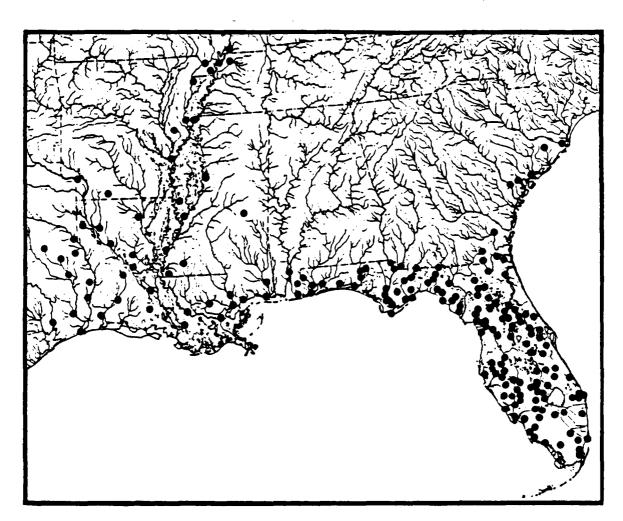
TYPE LOCALITY: Charleston, Charleston Co., SC (Günther 1866. Catalogue of the Fishes in the British Museum 6:1-368).

SYSTEMATICS: Subgenus Zygonectes. In past frequently confused with similar F. cingulatus. Brown (1956. Copeia:251-55) diagnosed and compared the two species and defined ranges.

Order Atheriniformes Family Cyprinodontidae



FL: ca. 56 mm SL (Jordan and Evermann 1900).



DISTRIBUTION AND HABITAT: Lowland areas (below Fall Line) from SC, GA, and FL west along Gulf to eastern TX and north to extreme southeastern MO and western KY. Inhabits backwaters and pools of ditches and slow-moving streams, usually associated with heavy submergent aquatic vegetation. Occasionally in brackish water along coast. Relatively common in preferred habitat throughout most of range.

DULT SIZE: 30-50 mm SL, 57 mm SL aximum.

BIOLOGY: Feeds mainly on insects and other aquatic invertebrates, near or at the surface (Hunt 1953. Trans. Am. Fish. Soc. [1952] 82:13-33). Leitholf (1917. Aquatic Life 2:141-42) described reproductive activity in aquaria and noted that eggs were deposited on submerged plants, stones, and side of aquarium. Eggs laid a few at a time over period of a week or more. (Pflieger 1975. The Fishes of Missouri).

Compiler: J. R. Shute. September 1978.

22 pairs.

Fundulus olivaceus (Storer) Blackspotted topminnow

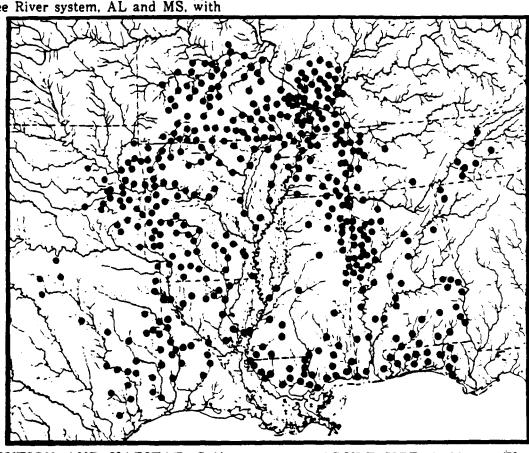
Order Atheriniformes Family Cyprinodontidae

TYPE LOCALITY: Florence, Lauderdale Co., AL (Storer 1845. Proc. Boston Soc. Nat. Hist. 2:47-49).

SYSTEMATICS: Subgenus Zygonectes. Regarded by Hubbs and Burnside (1972. Copeia: 862-65) as distinct genus. Most closely related to F. notatus (Braasch and Smith 1965. Copeia: 46-53; Thomerson 1966. Tulane Stud. Zool. Bot. 13:29-47) and an apparently undescribed species from Tangipahoa River, MS and LA. Setzer (1970. Trans. Am. Fish. Soc. 99:139-46) reported basic difference in diploid chromosome number between F. notatus (40) and F. olivaceus (48) throughout respective ranges, but Black and Howell (1978. Copeia: 280-88) found race of F. notatus in upper Tombigbee River system, AL and MS, with



LA: Lincoln Parrish, Ouachita River drainage, 45 mm SL (Smith-Vaniz 1968).



DISTRIBUTION AND HABITAT: Gulf slope, from Galveston Bay drainage, TX, east to Choctawhatchee River system, FL, and middle Chattahoochee River drainage, GA; north in Mississippi Valley to central MO and southern IL, and east to eastern TN. Prefers small to large fast-flowing, relatively clear, sand-gravel bottom streams where often occurs along margins near thick stands of emergent vegetation. Common but rarely collected in large numbers.

ADULT SIZE: 60-90 mm TL, 97 mm TL maximum.

BIOLOGY: Food habits discussed by Rice (1942. J. Tenn. Acad. Sci. 17:4-13), and further studies on biology by Thomerson (1966).

Compiler: J. R. Shute. March 1978.

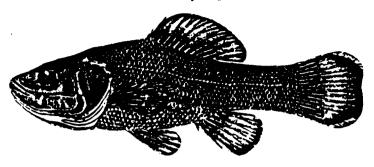
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Aphredoderus sayanus (Gilliams) Pirate perch

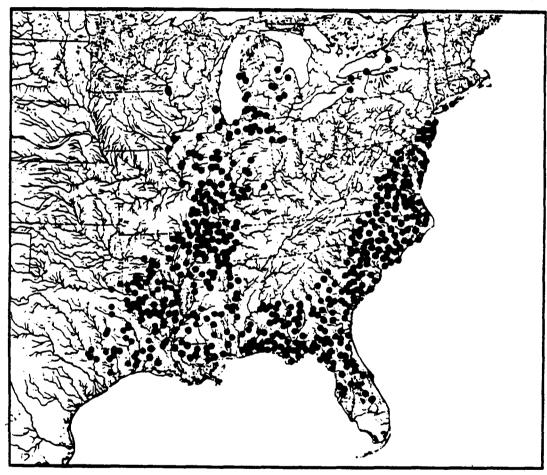
TYPE LOCALITY: Fishponds, Harrowgate, "near Philadelphia" (Gilliams 1824. J. Acad. Nat. Sci. Phila. 4:80-82).

SYSTEMATICS: Only living member of family. Two recognized subspecies A. s. sayanus on Atlantic slope and A. s. gibbosus in west. Precise range limits of subspecies and zone of intergradation not recorded in literature, but presumed to be similar to that shown by Crossman (1966. Copeia: 1-20) for Esox americanus.

Order Percopsiformes Family Aphredoderidae



DE: Sussex Co., Raccoon Pond. 69 mm SL. (NCSM)



DISTRIBUTION AND HABITAT: Widespread throughout lowlands of Atlantic and Gulf slope. Mississippi Valley. Disjunct populations in Lake Erie and Lake Ontario drainages, western NY. In lakes, ponds, quiet pools, and backwaters of low gradient streams with an abundance of aquatic plants, organic debris and other cover. Rare and localized toward periphery of range (particularly in north), but elsewhere may be very common in preferred habitat.

ADULT SIZE: 64-144 mm TL.

BIOLOGY: Little is known. Recent summary by Pflieger (1975. The Fishes of Missouri) provided observations on age and growth. Early growth and development described by Mansueti (1963. Copeia:546-57). Huish and Shepherd (1975. J. Elisha Mitchell Sci. Soc. 91:76) investigated age, movement, density, period of fecundity, and stomach contents of a NC population. Insects are the major food (Becker 1923. Occas. Pap. Mus. Zool. Univ. Mich. 138:1-4; Flemer and Woolcott 1966. Chesapeake Sci. 7:75-89).

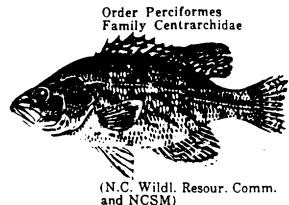
Compiler: D. S. Lee. April 1978.

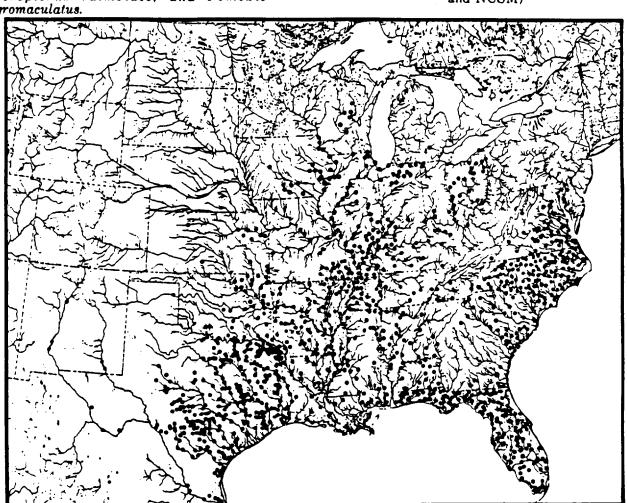
Lepomis gulosus (Cuvier) Warmouth

TYPE LOCALITY: Lake Pontchartrain, New Orleans, LA (Cuvier in Cuvier and Valenciennes 1829. <u>Histoire Naturelle des</u> Poissons 3:1-500).

Poissons 3:1-500).

SYSTEMATICS: Long placed in monotypic genus Chaenobryttus. Placed in Lepomis by Bailey et al. (1970. Am. Fish. Soc. Spec. Publ. 6:1-150). Known to hybridize with at least four other Lepomis spp., Micropterus salmoides, and Pomoxis nigromaculatus.





DISTRIBUTION AND HABITAT: Common in ponds, lakes, and occasionally streams, from KS and IA to southern WI, MI, and west PA, south to Rio Grande and FL. Presumed native on Atlantic slope north into VA, perhaps to MD. Transplanted west of Rockies and to portions of Atlantic slope. Occasionally reported in brackish water up to 4.1 ppt. Abundant where introduced in saline water of lowermost Colorado River. AZ (Minckley 1973. Fishes of Arizona).

ADULT SIZE: 75-260 mm TL; 284 mm TL maximum.

BIOLOGY: Larimore (1957. Ill. Nat. Hist. Surv. Bull. 27:1-83) studied life history in IL and found nesting from mid-May to August. Nests usually constructed near cover and guarded by male. Feeds mainly on insects, crayfish, and fish. Carlander (1977. Handbook of Freshwater Fishery Biology Vol. 2) summarized published information and tabulated data on age, growth, length, and weight.

Compiler: D. S. Lee. December 1978.

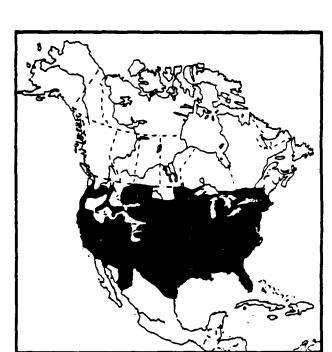
Lepomis macrochirus Rafinesque Bluegill

TYPE LOCALITY: "Ohio River" (Rafinesque 1819. J. Physique 88:417-29).

systematics: Three subspecies are recognized. Lepomis m. macrochirus occurs in the Great Lakes and north Mississippi basin, L.m. speciosus in TX and Mexico and L.m. purpurescens on the Atlantic slope from coastal VA to FL (Hubbs and Lagler 1964. Fishes of the Great Lakes Region). Widespread introductions have resulted in extensive mixing of these gene pools. Avise and Smith (1974. Evolution 28:42-56) studied geographic variation and subspecific intergradation, and Avise and Smith (1977. Syst. Zool. 26:319-35) studied relationships to other centrarchid species using electrophoretic data. Commonly hybridizes with several other species of Lepomis, particularly in areas of ecological disturbance. Considered to be most closely related to L. humilis (Branson and Moore 1962. Copeia:1-108).



Former Distribution



Order Perciformes

Family Centrarchidae

(N.C. Wildl. Resour. Comm.

and NCSM)

Present Distribution

See map on next page

DISTRIBUTION AND HABITAT: Originally restricted to western and central North America where it ranged from coastal VA to FL, west to TX and northern Mexico, and north from western MN to western NY. Widely transplanted elsewhere in North America and introduced into Europe and South Africa. Inhabits shallow warm lakes, ponds, and slow-flowing rivers and creeks often with abundant aquatic vegetation.

ADULT SIZE: 178-203 mm TL, 405 mm TL maximum.

BIOLOGY: Spawning has been summarized by Breder and Rosen (1966. Modes of Reproduction in Fishes). This is a generalized wide spectrum feeder. Carlander (1977. Handbook of Freshwater Fishery Biology Vol. 2) provided a lengthy summary and discussion of the general biology and of growth data obtained from numerous management studies.

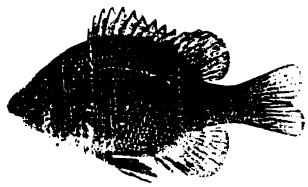
Compiler: D.S. Lee. February 1978.

Lepomis marginatus (Holbrook) Dollar sunfish

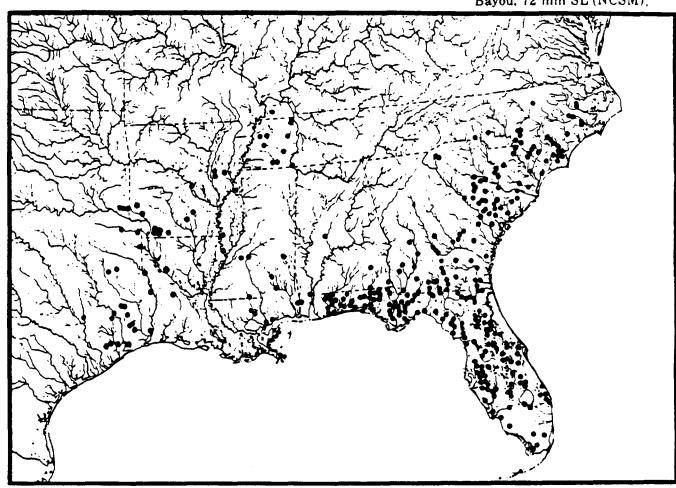
TYPE LOCALITY: St. Johns River, FL (Holbrook 1855. J. Acad. Nat. Sci. Phila. [Ser. 2] 3:47-58).

SYSTEMATICS: Closest relative L. megalotis, the two comprising subgenus Icthelis. Considered monotypic by previous investigators, but recent studies by compiler indicate possible polytypy. Reeves and Moore (1949. Proc. Okla. Acad. Sci. 30:41-42) gave diagnostic characters for separation from L. megalotis.

Order Perciformes Family Centrarchidae



AR: Calhoun Co., Locust Bayou, 72 mm SL (NCSM).



DISTRIBUTION AND HABITAT: Southern coastal drainages from NC to TX and north through central Mississippi basin to KY and AR. Usually common to abundant. Species of swamps and sluggish streams.

ADULT SIZE: 36-100 mm SL.

BIOLOGY: Very little information available. McLane (1955 Ph.D. diss., Univ. Florida) classified it as insectivorous and reported breeding season as extending from April to September in St. John's River, FL.

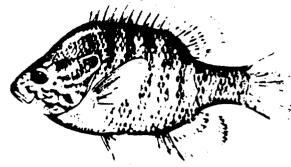
Compiler: B. H. Bauer. December 1978.

Lepomis megalotis (Rafinesque) Longear sunfish

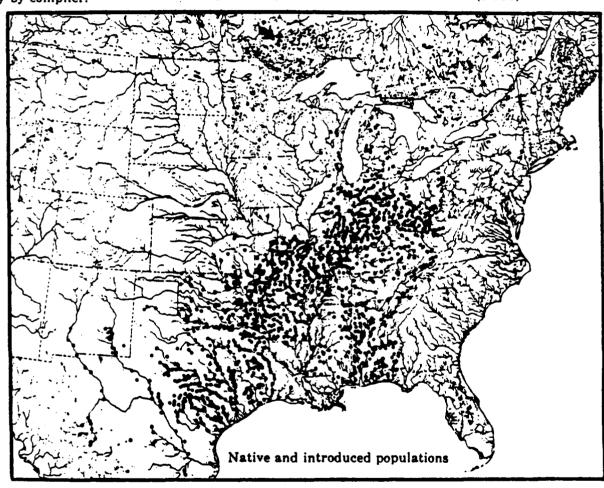
TYPE LOCALITY: Kentucky, Licking, and Sandy rivers. KY (Rafinesque 1820. Ichthyologia Ohiensis).

SYSTEMATICS: Closest relative L. marginatus, these two species comprising subgenus Icthelis. Hybridizes extensively with other Lepomis. Most polytypic member of family Centrarchidae, consisting of from four to six subspecies. Presently under study by compiler.





(NMC)



DISTRIBUTION AND HABITAT: Restricted to fresh waters of east-central North America. West of Appalachians, occurs from southern QU south to Gulf of Mexico in AL and western FL. Extends west through TX and Rio Grande tributaries in northeast Mexico, north through eastern parts of the states from OK to southern ON. Now thrives in reservoirs, but typically inhabits small streams and upland parts of rivers, generally absent from downstream lowland sections.

ADULT SIZE: 41 - 200 mm SL.

BIOLOGY: Various aspects of life history well studied. Hubbs and Cooper (1935. Pap. Mich. Acad. Sci. Arts Lett. 20:669-96) and Bacon and Kilambi (1968. Proc. Ark. Acad. Sci. 22:44-57) reported on age and growth. Applegate et al. (1966. Proc. 20th Ann. Conf. Southeast. Assoc. Game Fish Comm: 469-82) studied food habits. Numerous authors have studied breeding and mate selection behavior.

Compiler: B. H. Bauer. August 1978.

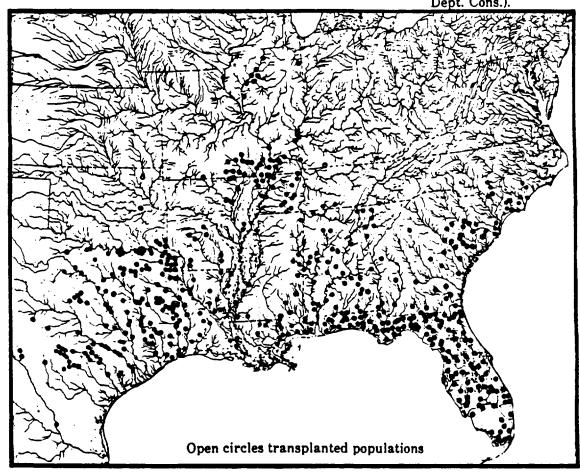
Lepomis punctatus (Valenciennes)
Spotted sunfish

TYPE LOCALITY: Charleston, SC (Valenciennes in Cuvier and Valenciennes 1831. Histoire Naturelle des -Poissons 7: 1-531).

SYSTEMATICS: Two subspecies recognized: L. p. punctatus ranging from NC to FL, and L. p. miniatus in Mississippi and most Gulf slope drainages; zone of intergradation apparently in extreme west FL (Carr and Goin 1959. The Reptiles, Amphibians, and Freshwater Fishes of Florida) and probably AL.



MO: New Madrid Co., Sikeston, 111 mm SL (Mo. Dept. Cons.).



DISTRIBUTION AND HABITAT: Southeastern United States from eastern TX east to and including peninsular FL, north along Atlantic slope to southeastern NC. In Mississippi basin north to IL. Common in quiet or moderately flowing waters with heavy vegetation or other cover.

ADULT SIZE: 55-140 mm SL, ca. 200 mm TL maximum.

BIOLOGY: Forbes and Richardson (1920. The Fishes of Illinois) reported spawning in May in IL. Pflieger (1975. The Fishes of Missouri) noted nesting in July in MO and commented on various aspects of life history. Carr (1946. Q. J. Fla. Acad. Sci. 9:101-06) studied courtship, spawning, and nest defense habits in FL, and recorded spawning from early spring to November. Courting males make grunting sounds (Gerald 1971. Evolution 25:25-87). Carlander (1977. Handbook of Freshwater Fishery Biology Vol. 2) summarized available weight, age, and growth information.

Compiler: D. S. Lee. November 1978.

Micropterus salmoides (Lacepede) Largemouth bass

TYPE LOCALITY: "les rivieras de le carolina": Charleston, SC, regarded as probable type locality (Lacepede 1802. Histoire Naturelle des Poissons 4:1-728).

SYSTEMATICS: Subfamily Lepominae, tribe Micropterini. Formerly placed in monotypic genus Huro (Hubbs 1926. Misc. Publ. Mus. Zool. Univ. Mich. 15:1-77; Hubbs and Bailey 1940. Misc. Publ. Mus. Zool. Univ. Mich. 48:1-51). Hubbs and Bailey (1940) reviewed systematics, and Bailey and Hubbs (1949. Occas. Pap. Mus. Zool. Univ. Mich. 516:1-40) defined and mapped distinctive subspecies. M. s. floridanus, endemic to peninsular FL.



Former Distribution

DISTRIBUTION AND HABITAT: Original range from northeastern Mexico to FL, much of Mississippi River, north to southern QU and ON and on Atlantic slope north only to southern or central SC. Precise original distribution masked by numerous undocumented transplants. Robbins and MacCrimmon (1974. Biomanag. Res. Cent.:1-196) extensively surveyed nearly worldwide introductions. Prefers clear, quiet waters vith aquatic vegetation. Common to abunant throughout most of range.

ADULT SIZE: ca. 120 mm - 700 mm TL.

Order Perciformes Family Centrarchidae



(N.C. Wildl. Resour. Comm. and NCSM)



Present Distribution

BIOLOGY: One of most important North American gamefish and much information available. Summaries of life history and propagation information available in Scott and Crossman (1973. Freshwater Fishes of Canada) and Carlander (1977. Handbook of Freshwater Fishery Biology Vol. 2). Heidinger (1974. An Indexed Bibliography of the Largemouth Bass. Micropterus salmoides) provided extensive bibliography.

Compiler: D. S. Lee. February 1978.

APPENDIX XXII LOCATION OF CARBON TETRACHLORIDE STORAGE AND SPILL AREA AT THE LEMOYNE PLANT

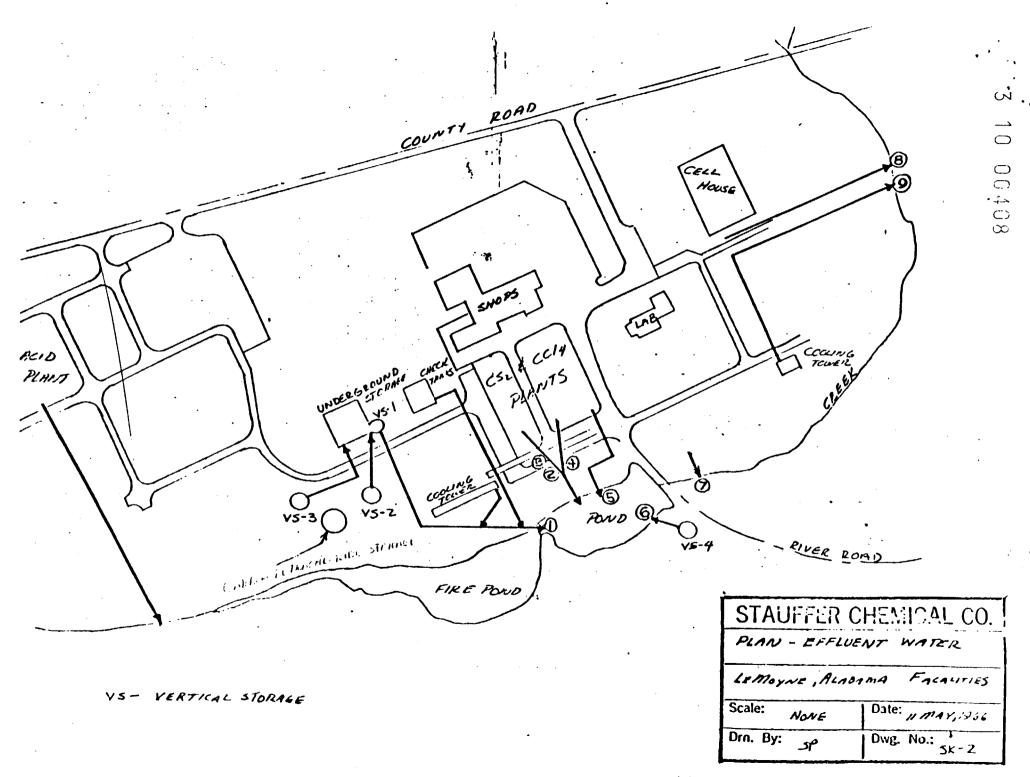


Figure 22-1 Location of Carbon Tetrachloride

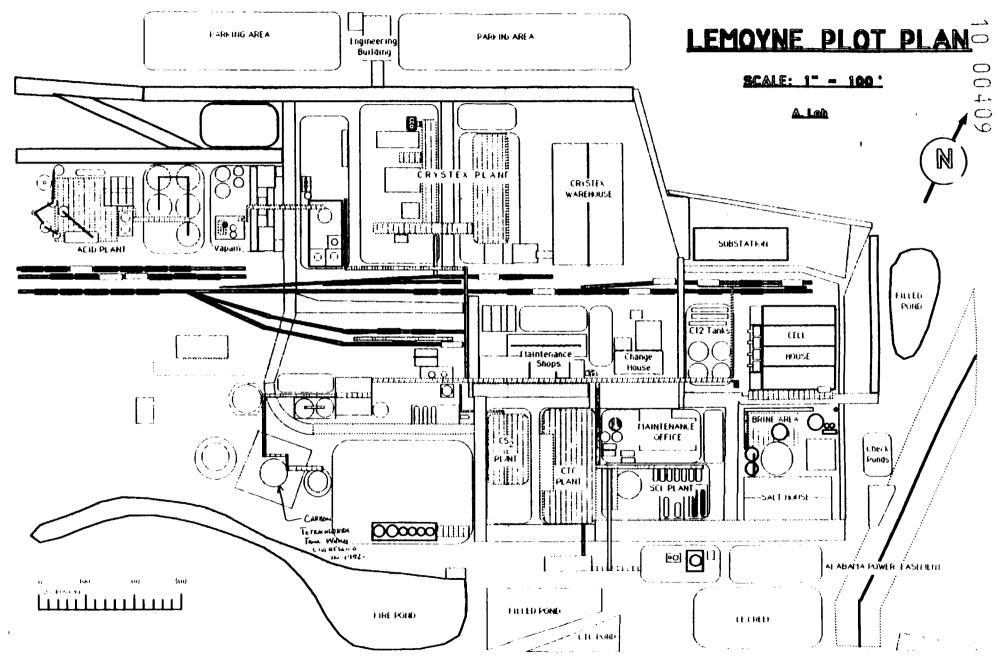


Figure 22-2 Location Carbon Tetrachloride Storage 1 .k and Spill Area

APPENDIX XXIII
GROUND-WATER INTERCEPT
AND TREATMENT PROGRAM

1. GROUND-WATER INTERCEPT SYSTEM

The interceptor well system was originally designed to depress the ground-water table to the extent necessary to act as an effective hydraulic barrier to the southward movement of contaminants. Normally, without the intercept system, approximately 0.5 million gallons of ground water per day pass the southern property boundary of the site. At maximum design pumping rates of 1500 gallons per minute (gpm), the intercept wells remove about 2 million gallons per day. Since 1982, pumping rates have varied between 1,000 and 1,300 gpm. These pumping rates are sufficient to remove the contaminated ground water as evidenced by the very low contaminant levels south of the site boundary (see Section 5.0).

The only major problem associated with operation of the intercept wells has been the measurement of flow in the wells, not flow itself. A minor problem associated with the wells has been the precipitation of aluminum oxide in wells IW-2 and IW-3. The precipitation of aluminum oxide has resulted from the dissolution of clay minerals in the strata surrounding the wells by the slightly acidic ground water that the wells intercept. The aluminum precipitate has been cleaned from these wells on an as-needed basis in order to maintain adequate flow rates. Thus, wells IW-2 and IW-3 have been out of service periodically for cleaning and/or repair since 1980. A submersible pump was used while the wells were being serviced in order to maintain adequate flow rates.

2. GROUND-WATER TREATMENT PROGRAM

In December of 1979 and January of 1980, Stauffer Chemical Company conducted treatability tests on contaminated ground water taken from the aquifer in the proposed interceptor well system area. The attached "Treatability Report, January, 1980, Air Stripping" describes the results of those tests. The effects of air stripping the two volatile chemicals, carbon tetrachloride (CTC) and carbon disulfide (CS₂), by spraying through nozzles and by floating mechanical aerators were measured. The test results indicated that ground water containing relatively high concentrations of carbon tetrachloride (up to 80 ppm) and carbon disulfide can be successfully aerated to produce an effluent with volatile concentrations of 50 ppb or less.

Since 1980, when the ground-water treatment system was designed, the concentrations of carbon tetrachloride and carbon disulfide in the ground water have decreased from up to 80 ppm to less that 10 ppm. Because the spray nozzles in the ground-water treatment system perform the bulk of the removal of the contaminants (greater than 90% removal), and because NPDES permit levels for discharge of CS₂ and CCl₄ in treated water were not being exceeded with use of the spray nozzles alone, a decision was made to discontinue use of the mechanical aerators.

Attached to the treatability report are notes on potential remedial action technology for CMA's Solidwaste Work Shop (8/23/83), and a schematic of the aeration system.

ATTACHMENT 1

TREATABILITY REPORT

JANUARY, 1980

AIR STRIPPING

VOLATILE GAS STRIPPING SYSTEM DESIGN RECOMMENDATIONS BASED on TEST CAMPAIGN RESULTS

LAMOYNE GROUNDWATER IMPROVEMENT PROGRAM

APPROPRIATION REQUEST # 1625A660

January 1980 T. Helfgott Ph.D., P.E.

CUTLINE

_		Pages
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II.	SUMMARY	2
III.	TEST DESCRIPTION AND RESULTS	3
IV.	INTERPOLATION ANTICIPATED PERFORMANCE	8
	FIGURES	
	TABLES	
	APPENDIX	

Nomenclature Terms Used

ø	•	diameter
MG	•	Million Gallons
mg	•	milligrams
l ppb	•	1 pg/1
1 mg /1	•	1 ppm = 1000 ppb = 1000 ug/1
Gpm	•	Gallons per minute
hp	•	horsepower

I. PURPOSE

3 10 00416-

- treatment system that takes 1500 gpm water from a carbon tetrachloride and carbon bisulfide laden aquifer by means of intercept wells, serates the water with spray nozzles and mechanical serators, raises the pH of the water to meet river discharge standard (pH 6-9) and discharges the water essentially free of the two volatile gas (each less than 50 ppb) (50 pg/l) into the Mobile River. The discharge standard of less than 30 mg/l total suspended solids must also be met.*
- B. The intent of this report is to offer the basis for design recommendations for the volatile gas stripping system in the LaMoyne Groundwater Improvement Program. The one month** long test campaign at the LaMoyne facility had three major objectives:
 - to size mechanical aerators in terms of horsepower rating and select number of aerators necessary to achieve required residual volatile gas levels.
 - to affirm that the 50 pg/l standard could be met; and
 - 3. to collect nozzle stripping performance data.

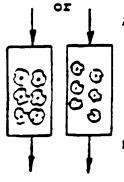
^{*} Since the groundwater contains 4-30 mg/l Fe $^{++}$ it is anticipated that at least part of the time greater than 30 mg/l TSS will be generated largely as hydrated iron floc - [Fe(OH)₃ • 6 H₂O] .

^{** 17} December 1979 thru 15 January 1980. As of this time, however, all requested tests have not been done, and all data is not yet available.

3 10 00417

II. SUMMARY

Based on the test campaign results at the LeMoyne facility "Carbon Tet" pond, the following volatile gas stripping system is recommended:



- A. Six (6) mechanical aerators each of 20 horsepower rating in a pond arranged in either of the conformations shown in the margin. See picture of the floating aerator in the Appendix.
- B. Pond size can be selected on the basis of marginally overlapping zones of turbulence between aerators in order to realize a relatively long narrow pond and smallest area.
- C. Detention time under aeration is not critical controlling parameter for these rapid surface renewal machines arranged as suggested.* For the eight foot depth of pond the pumping function of the mechanical aerators should easily satisfy the need for rapid turnover of the water under treatment.
- D. Twelve full cone spray nozzles rated at 125 gmp and 30 psig non-clogging (1 3/4" #) orifice are recommended**.

Detention time would be a problem for unmixed ponds of very short retention periods or for very long detention times for low flowthrough.

^{**} See in appendix vendors letter of 7 January 1980.

II. SUMMARY - (cont'd)

Compared to the preliminary design, the recommended gas stripping equipment, based on the test campaign, can save 40% of the capital cost for the mechanical aerators and over 50% of the operating power cost since 6 units at 20 horsepower are suggested rather than 10 units at 25 horsepower.

III. TEST DESCRIPTION AND RESULTS

Figure 1 is a schematic flow sheet of the "Carbon Tet" pond at LeMoyne that was used to treat the groundwater. Only 2 (out of 4) of the wells in the well field of the aquifer could be used at the time of the test. Each pump was rated at 75 gpm. Due to production needs, 30 gpm of low carbon tetrachloride (roughly 10 to 1000 ppb) containing effluent at pH 7 also went into the pond. Flat spray nozzles (1 inch opening) at an operating pressure and flow (though much below rated capacity) were used. As shown on the flow sheet the pond was 1.5 MG size with a baffle at a 2/3 point to attempt to segregate the aeration zone from the settling zone. At the (Western) $\widehat{\sigma}^{A3}$ end of the pond a simple overflow collection though is used. Typical operating parameters are flagged on the flow sheet and the 5 major sampling points are indicated. Table I lists the data summary for the 5 tests (A through E) for which data are available at this point. Data accumulated are attached to the appendix of this report.

Wes influxed with the well water containing about 86 ppm CCl₄.

The concentration in the pond rose from 0.001 ppm (1 mg/1) to just below 7 ppm where the concentration started to plateau after six days. Figure 2 shows the concentration charges of CCl₄ with time that would imply a detention time of the pond (without mixing) of roughly 6 days, or 4 days for up to the baffle (2/3 of the 1.5 MG volume of the pond). This corresponds to the time in this pond for the flow rate anticipated.

The outlet concentration of CS₂ was 0.37 ppm (370 pg/1) while the well water had an average concentration of about 3.2 ppm (3,200 pg/1). Since there was no mechanical aerator used in this test most of the removal is to be due to the nozzles (88%).

B. In the next test, two aerators were turned on which drew 31-32 amperes per unit. In less than a day or so, the final effluent went to 0.3 ppm CCl₄ (300 pg/l) and remained about this value. This is taken as the lowest value that can be achieved by one set of two aerators following an initial reduction of better than 90% by the nozzle.

Since only 171-18 hp (horsepower) were drawn by each 25 hp aerator these could be specified as 20 hp units. The 25 hp units would be more appropriate if this were a thicker wastewater rather than a relatively clean groundwater.

Although CS₂ data is sparse values as low as 12 ppb (12 pg/1) were noted for the effluent in this test. The average well water CS₂ concentration was about 3.2 mg/1 (3,200 pg/1) so an adequate overall reduction of better than 99% is achievable.

- C. In the third test, one serator was left on and the effluent quality went to about 0.7 ppm (700 pg/l) in less than the anticipated full detention time of the pond indicating a highly mixed system. In general, better than 90% removal is still noted for the nozzles which are doing the bulk of the removal in terms of lbs CCl₄ stripped. The mechanical aerators are needed to polish the groundwater to the discharge standard.
- D. The fourth test used two aerators with the nozzles at 2 psig.

 The average removal across the pond assignable to the 2

 mechanical aerators alone was 89% (0.450 to about 0.05 ppm for

 7 sets of average data).
- E. It was necessary to be sure that a 50 ppb (50 µg/l) or better of CCl₄ could be reached. In this last test, the wells were shut off and the production effluent concentrations were noted. For average beginning of pond values of about 100 ppb (100 µg/l), end of pond values below 10 ppb CCl₄ were achieved. For the few data points for CS₂ the values were below 50 ppb (41 & 13 ppb).

In this test samples were also taken from the middle of pond and compared to the outlet concentration: for 5 match sample sets there was no significant difference in concentration. From this it is concluded that except for the inlet side of the pond (East), the pond is uniformly mixed from at least the middle area to the effluent area. A visual flow pattern could be seen in the pond due to the black-brown iron sludge content: see reactors shown in flow sheet. The iron was dispersed uniformly after the first mechanical aerators.

F. Quantitative Results

Quantitative results are summarized on Table II. Because the pH was not controlled in the test, the pH ranged from 3.0 to 5.5 at the influent point; it was 4.6 as an average. Typically the well water has a pH of 5.3 but because some iron precipitated there was acid generation. The pH of the production water was 7.0. Suspended solids, roughly correlated to turbidity, did not change much. The settling sone beyond the baffle was not very effective. Turbulent eddies could be seen by-passing the plastic baffle curtain. The alkalinity of the well water was generally low, typically about 17 mg/l CaCO₃ eq., and the acidity is typically about 70 mg/l CaCO₃ eq. Figure 3 is a titration curve from earlier tests on the groundwater showing other water quality information and alkaline requirements

At this time, sampling in depth in the pond has not been done yet but for the high mixing reigm of the aerators uniform distribution from bottom to top of pond is expected.

III. TEST DESCRIPTION AND RESULTS - (cont'd)

in terms of caustic, lime, or calcium carbonate to achieve a pH of about 6. A major advantage of CaCO₃ is that it would tend to buffer the water system at about 6.3 while not adding sodium and improving the overall water quality in terms of hardness and alkalinity.

IV. INTERPOLATION AND ANTICIPATED PERFORMANCE

With the limited data on hand, the task now is to interpolate the information for a full design. At this time, we have the following information: (1) We know what the nozzle can do — better than 90% in most cases. (2) We know that a set of 2 aerators with a starting influent concentration of about 3-4 ppm can reduce this to about 0.3 ppm (300 pg/1). (3) We know there is little influence (less than 10% CCl₄ removed) accountable for the settling zone; that is, between the baffle and the pond outlet there is no significant change in CCl₄ concentrations. (4) We know that if we start into the pond at lower concentrations using a set of 2 mechanical aerators the following CCl₄ levels can be accomplished:

Test D: 0.450 ---- 0.049 (89%)

Test E: 0.103 ---- 0.010 (90%)

The data for CS_2 is very sparse but the very limited information indicates about 90% plus removal by the nozzle is possible and that effluent concentrations below 50 ppb (50 pg/l) can be reached. The initial concentration of CS_2 is less than 10% of that for CCl_4 and it is anticipated that CS_2 is more readily stripped from the water phase into the air.

From this data we can assemble a CCl₄ removal diagram, Figure 4, so that each bar represents a set of 2 aerators following the initial removal by the nozzle. From the graph we can see by extrapolation that three sets of 2 aerators would more than be minimal to go below 50 ppb and the 2 sets of 2 aerators following the nozzles would be marginal at best.

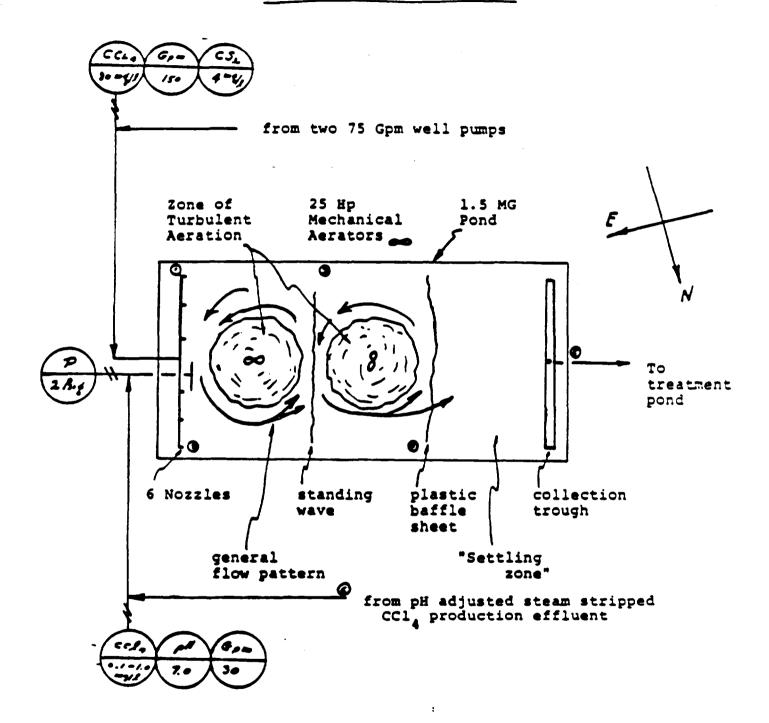
IV. INTERPOLATION AND ANTICIPATED PERFORMANCE - (cont'd)

Table V shows the anticipated performance of the system assuming three levels of well water CCl₄ -- 30 ppm CCl₄ the original design figure; and the 60-80 ppm CCl₄ that was seen during the recent test campaign. Therefore, a conservative design would be for 6 aerators and 4 or 5 aerator units would be the margin selection. The aerator rating can be 20 hp based on the actual power drawn and noted during the tests.

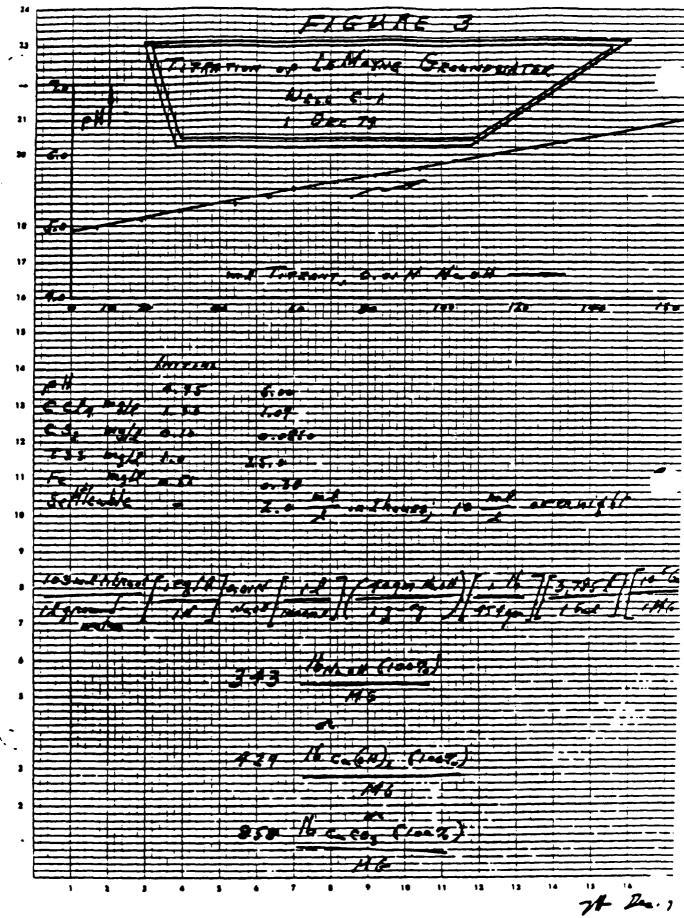
Since the preliminary design called for 250 hp (10 \times 25 hp units) the test campaign has realized a better than a 50% power saving as well as a lesser number of mechanical aeration units.

Figure 1

"Carbon Tet" Pond at LeMoyne



Size of zone of turbulent aeration: approximately 100 ft \emptyset Five major sampling points indicated . Approximately 1.0 MG water under mixing and rapid aeration



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DATA SUMMARY -- GROUNDWATER DEPROVEMENT CAMPAIGN
LeMoyne, Alabama

December	1979	-	January	1980
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Test	No. of Aerators in Use	Flow From	Major Information Gained
A	0	Wells and Production	Detention Time: 4-6 days Overall Reduction 86 7 ppm CCl ₄
3	2	Wells and Production	Power Drain: 18 hp each aerator Nozzle Removal: 86-99% CC14 for an average of 96% 40 0.3 ppm CC14
c	1	Wells and Production	Nozzle Removal: 93% CCl ₄ Overall Removal: 98.8% 60 0.7 ppm CCl ₄
D	2	Wells and Production	Aerator Set Removal: 0.4 — 0.08 ppm CCl4
E	2	Production Only	Aerator Set Removal 0.1 — 0.01 ppm CCl ₄

Removal % CCl4 from Influent

Summary of all Tests

Mechanism	Average 1	Range 1
Mossle	92	77-99
1 Aerator	82	-
2 Aerators	91.	78-99

TABLE II

Quantitative Information

Turbidity in	12 NTU
Turbidity out	12 NTU
Total Suspended Solids in	11.9 mg/l
Total Suspended Solids out	11.4 mg/1
Iron in	3.7 ppm Fe
Iron out	3.4 ppm Fe
pH in Avg. (Range)	4.6 (6.5-3.0)
pH out Avg. (Range)	3.9 (5.0-2.9)
Well Water pH	5.3 (typically)
Water Temperature	66° F (19° C)
Well #29	162 ppm ccl ₄
Well #31	9.5 ppm CCl4
Ambient Air	1-2x10 ⁻⁴ ppm CCl ₄
Air Temperature	38-70° F (4-21° C)
Acidity	16 ppm CaCO ₃ eq.
Alkalinity	~ 70 ppm caco₃ eq.

TABLE III
Conservative Design

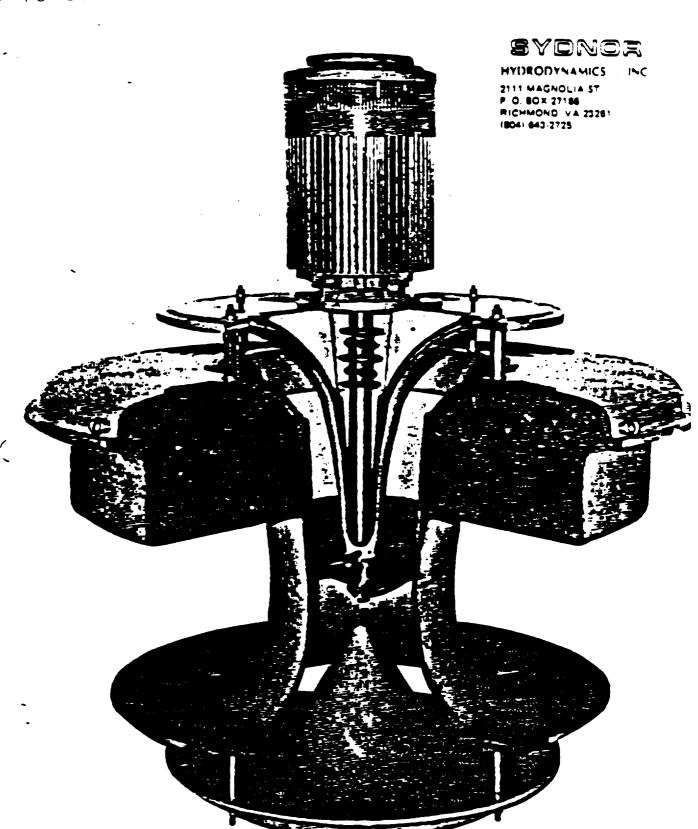
90%+ Stripping of CCl₄ by Nozzles
90% Stripping of CCl₄ by each set of 2 Aerators

	Concent	rations, pr	E CC1
Inlet Concentration	30	60	80
Nozzle Outlet	· 3	6	8
End of 1st Set	0.3	0.6	0.8
End of 2nd Set	0.03	0.06	0.08
End of 3rd Set	0.003	0.006	0.008

Discharge Required 50 ppb - 0.050 ppm - 50 µg/l

APPENDIX

- 1. Picture of System Floating Aeration Used.
- Letter from vendor, Wm. Steiner Mfg. Co., recommending nozzle specification for the groundwater improvement program.
- 3. Data Collection



CUT AWAY OF THE SYDCO OXYGENATOR SHOWING THE PROPRIETARY DESIGN. QUALITY WORKMANSHIP AND MATERIALS, AND SUPERIOR CONSTRUCTION

STEINEN <

Wm. Sieinen Mfg. Co., 29 East Haisey Rd., Parsippany, N.J. 07054 + 201-887-6400 + Capie., STEINEN + TWX 710-986-8212/STEINEN PAPY

January 4, 1980

Dr. T. Helfgott Stauffer Chemical Co. Dobbs Ferry, NY 10522

Dear Dr. Helfgott:

This will confirm our discussion yesterday regarding the change of pressure specifications for the spray nozzles in the gas stripping application.

We suggest consideration be given to the following:

Twelve units of the Steinen Hi-Flo Tan-Jet nozzle \$THM or THF 14516. This nozzle is rated \$ 126 g.p.m. \$ 30 p.s.i. with a spray angle of 96°. This will provide a total system flow of 1,512 g.p.m. \$ 30 p.s.i.

Due to the solid content of 2,000 p.p.m. as well as a pH of approximately 5 of the solution, we suggest consideration of 316 stainless steel as a material of construction. Since this is a casting, we would require a minimum of six weeks for shipment after receipt of order. The approximate orifice diameter of this nozzle is $1-3/4^{\circ}$.

If you develop any further questions, please let me know.

Sales Manager

Industrial Division

JJP: js



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ATTACHMENT 2

NOTES ON POTENTIAL REMEDIAL ACTIONS

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Notes on Potential Remedial Action Technology For CMA's Solidwaste Work Shop

I. Groundwater Interceptor System (South Eastern U.S.)

Because of leaks and spills over the years, the groundwater beneath this plant had become contaminated with fairly high concentrations of volatile organics. After an extensive groundwater study and pilot testing, a groundwater interceptor and treatment system was designed. Two rather significant groundwater parameters came to light during the study which had a bearing on the overall design. These were low pH (3.6 to 6.2) and high dissolved iron (10 to 15 ppm). It became apparent that the most cost effective method of organic removal was some sort of aeration. Any aeration would however oxidize the dissolved iron to insoluble ferric hydroxide and further reduce the pH. This eliminated the possibility of using a packed bed air stripper and groundwater reinjection.

From the hydrogeologic investigation, it was determined that three (3) interceptor wells located in a line 300 ft. apart and pumping at 500 GPM each would form a depression to intercept the contaminated groundwater. It was also determined that by sparging the water through spray nozzles, about 80 to 90% of the volatile organics would be driven off. To meet surface water discharge limits, organic removals of 99.8% were necessary and the pN of the effluent had to be between 8 and 9. Total suspended solids were limited to 30 ppm.

A rectangular shaped pond with a capacity of 3,200,000 gal. was installed. Because of space limitations the pond was located some 3,600 ft from the interceptor wells. The well pump headers were connected to a 12 in. PVC pipe which terminated at the pond by a spray header with (12) - 2 in. spray nozzles.

In order to achieve better organic removals, 8 floating mechanical aerators were installed and a partial beffle used to limit back-mixing. A quiescent zone near the pond's effluent end allowed for settling of the ferric hydroxide.

The pH was controlled most uniquely with a dual head metering pump feeding 50% caustic directly into the 12 inch pipe line near the well pump end and just upstream of a static mixer. A small pump was used to continuously take samples from the pond and pump it through a pH measuring assembly which sent a set point signal to a pH controller back near the well pumps. This controller compared the pH after caustic addition with the set point and adjusted the metering pump accordingly. A rangeability of 1500 to 1 was deemed necessary so the metering pump was designed with automatic stroke adjust on each head and SCR variable speed control on the pump motor. With a 4 to 20 MA DC input control signal, the pump output

flow ranges from 0.0003 to 0.45 GPM.

The treated effluent was collected by a flume connected to a sump outside the pond. A continuous sampler was placed in the line between the sump and the river. Flow was by gravity.

The entire system has been in operation for almost 3 years. The total installed cost was about \$2,000,000 and the operating cost is about \$60,000 per year. Thus far, the concentration of organic contaminants in the groundwater have been reduced by more than 50%. Concentrations are starting to level off and there is no telling how long it will take to adequately clean the aquifier.

ATTACHMENT 3

SCHEMATIC OF AERATION SYSTEM

APPENDIX XXIV SUMMARY TABLE OF MONTHLY THIOCARBAMATE DISCHARGE ANALYSIS, TABLE 24-1

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TABLE 24-1 SUMMARY TABLE OF MONTHLY THIOCARBAMATE DISCHARGE ANALYSIS (THIOCARBAMATE CONCENTRATIONS IN POUNDS/DAY)

		1984			1985			1986			1987	C.7
<u> Month</u>	Lemoyne	Cold Creek	Total	Lemoyne	Cold Creek	Total	Lemoyne	Cold Creek	Total	Lenoyne	Cold Creek	Total 🔘
												\bigcirc
January	N/A	N/A	N/A	0.15	0.07	0.22	0.06	2.54	2.60	0.14	2.51	2.65
February	N/A	N/A	N/A	0.17	0.50	0.67	0.33	0.43	0.76	0.10	0.54	0.64
March	W/A	n/a	H/A	0.14	0.99	1.13	0.15	0.30	0.45	0.16	1 0.49	0.65
April	0.46	<0.10	<0.56	0.10	0.02	0.12	0.16	0.25	0.41	0.00	1.10	1.10
May	0.40	0.10	0.50	N/A	0.05	0.05+	0.27	<0.01	0.27	0.07	0.61	0.68
June	0.29	0.14	0.43	0.11	0.23	0.34	0.01	0.16	0.17	0.26	1.58	1.84
July	0.68	0.64	1.32	0.10	4.00	4.10	0.07	0.21	0.28	0.20	1.60	1.80
August	0.30	0.31	0.61	0.12	1.44	1.56	0.01	0.38	0.39	0.09	0.12	0.21
September	0.29	0.02	0.31	0.09	0.71	0.80	0.16	0.08	0.24	0.10	1.05	1.15
October	0.39	0.10	0.49	0.10	0.27	0.37	0.09	0.11	0.20	0.20	0.16	0.36
November	0.11	0.06	0.17	0.16	0.70	0.86	0.89	0.16	1.05	0.28	0.03	0,31
December	n/a	0.40	0.40+	0.16	0.02	0.18	0.07	0.62	0.69	N/A	N/A	N/A

APPENDIX XXV EXPLANATION OF WHY THE MERCURY LOCATED IN THE COLD CREEK SWAMP IS IN THE INORGANIC MERCURY SULFIDE FORM

APPENDIX XXV

EXPLANATION OF WHY THE MERCURY LOCATED IN THE COLD CREEK
SWAMP IS IN THE INORGANIC MERCURY SULFIDE FORM

Stauffer Chemical Company started up a plant to produce carbon tetrachloride at the LeMoyne site in 1964. This plant is still in operation today. The following year, a mercury-cell caustic-chlorine plant was started up at the LeMoyne site. Wastewater from both plants was discharged into a natural stream which flows across the property from the southwest to the northeast and then discharges into the Cold Creek Swamp. Waters from the natural stream combine with waters from Cold Creek, Sisters Creek and other natural drainage to recharge the swamp, which not only recharges the groundwater, but also discharges to the Mobile River.

The discharge from the carbon tetrachloride plant contained sulfides formed in the final stages of sulfur purification. The chemical reaction equations for the process are as follows:

1)
$$3Cl_2 + CS_2 \longrightarrow CCl_4 + S_2Cl_2$$

2)
$$2S_2Cl_2 + CS_2 \longrightarrow CCl_4 + 3S_2$$

The recovered sulfur is contaminated with a small amount of S_2Cl_2 , which is then removed in a hot water wash process producing the sulfide contaminated wastewater.

This wastewater stream was discharged into the natural drainage creek upstream of the chlorine plant wastewater discharge. When the mercury-laden chlorine plant wastewater entered the stream, the mercury immediately reacted with the sulfides, producing the insoluble mercury sulfide, which was then carried into the Cold Creek Swamp prior to precipitation onto the swamp bottom. This practice was changed somewhat in 1970 when the chlorine plant wastewater treatment system was installed to react the mercury with sodium hydrosulfide and settle/filter the mercury sulfide prior to discharge of the wastewater into the natural creek.

3 10 00448

In 1974, all discharges to the natural creek were stopped, and a plant-wide system, including effluent from the Cold Creek plant, was installed to transport the effluent to the river through a pipeline under provisions of an NPDES permit.

The mercury sulfide that settled on the swamp bottom is essentially immobile due to the extremely low solubility of mercury sulfide in water and the numerous dams between the plant and the river (natural growth and beaver dams), which tend to prevent washing of the swamp bottom as the result of heavy rain runoff.

The ground-water wells in the area were becoming increasingly contaminated with chlorides from swamp infiltration prior to 1974. Since that time, the chloride concentrations have been dropping due to the plant effluent (high in chlorides) being transported through a pipeline to the river instead of being discharged into the swamp. If the mercury had been in a soluble form, it too would have shown up in the plant water and monitoring wells. This has not been the case.

APPENDIX XXVI
PLANT FACILITIES
INFORMATION AND DRAWINGS

APPENDIX XXVI

PLANT FACILITIES INFORMATION AND DRAWINGS

Detailed graphic depictions of eight plant facilities (in pockets), the new carbon tetrachloride wastewater treatment pond, and the LeMoyne acid plant WWT pond are included in this appendix. The information contained in these drawings includes a plan view of the facility, one or more cross sections, facility dimensions, liner locations, and all closure aspects such as depth and type of fill, landfill vent localities, and liner coverage. Locations of adjacent monitoring wells are shown in Figure 1-2 in Section 1.2 and Figures 4-2 and 4-3 in Section 4.4 of this report.

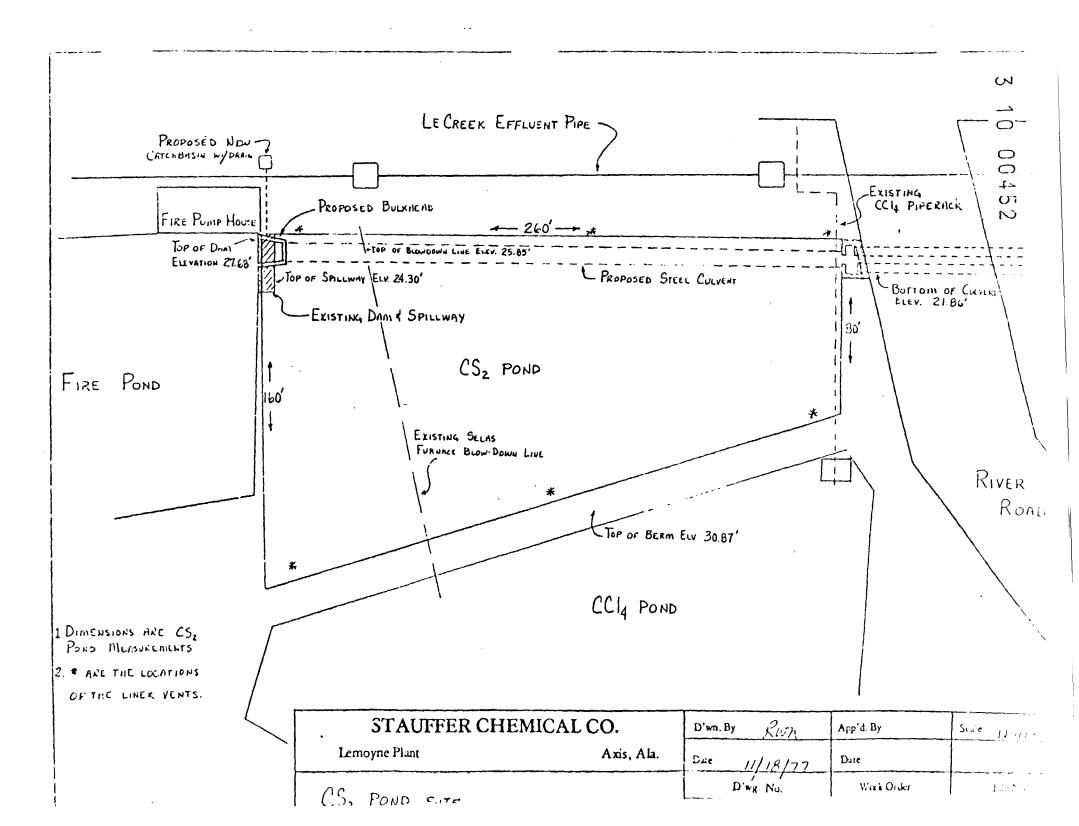
As discussed in Section 4.4.2, the stratigraphy across the site is fairly uniform. Therefore, the stratigraphy in the vicinity of the plant facilities, which is of interest to a depth of approximately 20 feet, the maximum depth of any of the facilities, can be summarized as follows. As shown in Table 4-1 in Section 4.4.2, a red or yellow to brown stiff clay unit with a basal sandy clay section that pinches out locally, is present from ground surface to approximately 8 to 22 feet below ground surface. Sand and clay interbeds that grade laterally into sand and the upper unit are present at depths of from approximately 10 to 15 feet to approximately 11 to 74 feet.

The estimated quantities and kinds of wastes placed in these facilities are summarized as follows:

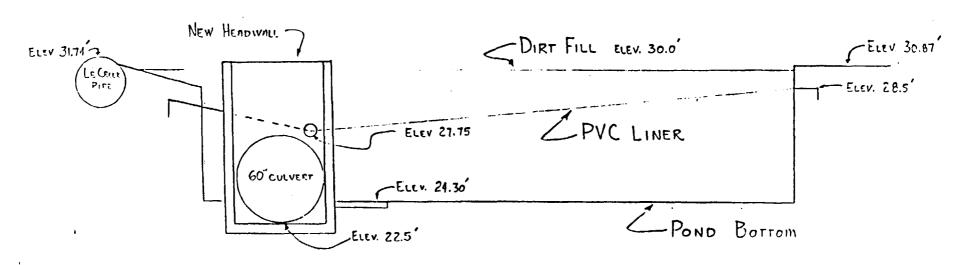
- 1) Cold Creek North and South Landfills The exact quantity of wastes placed in these landfills is unknown. The kinds of waste stored in the two Cold Creek landfills include water treatment plant sludge, used sandblast sand, generator coke, incinerator ash, and filter aid waste.
- 2) LeMoyne Landfill It is estimated that approximately 11,000 to 12,000 tons of brine mud, plant refuse, and absorption oil were disposed of in this landfill.

3 10 00451

- 3) Old Chlorine Plant WWT Pond Approximately 30,000 pounds of mercury sulfide and miscellaneous fill were placed in this pond.
- 4) Old Brine Mud Pond Between 4600 and 5500 tons of brine mud were stored in the Old Brine Mud Pond.
- 5) Old Carbon Tetrachloride Plant WWT Pond It is * estimated that approximately 1900 cubic yards of sulfur sludge were disposed of in this pond.
- 6) Old Carbon Disulfide/Carbon Tetrachloride Plant WWT Pond An estimated 2000 to 3000 cubic yards of sulfur sludge from operation of the carbon disulfide and carbon tetrachloride processes were stored here.
- 7) New Carbon Tetrachloride WWT Pond Approximately 300 cubic yards of sulfur sludge from operation of the carbon tetrachloride process are stored in this pond.
- 8) LeMoyne Acid Plant WWT Pond Approximately 100 tons of iron sludge from the acid plant are stored in this pond.



VIEW FROM RIVER ROAD THROUGH THE 60" CULVERT

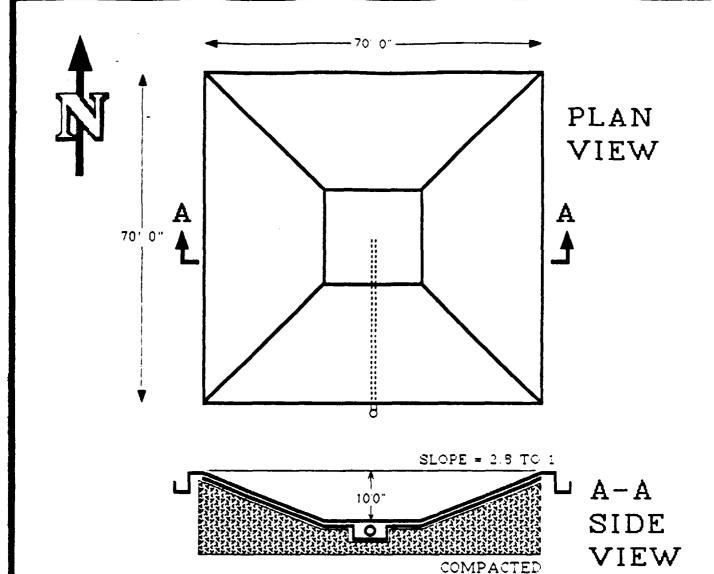


VIEW FROM FIRE POND THROUGH THE 60" CULVERT

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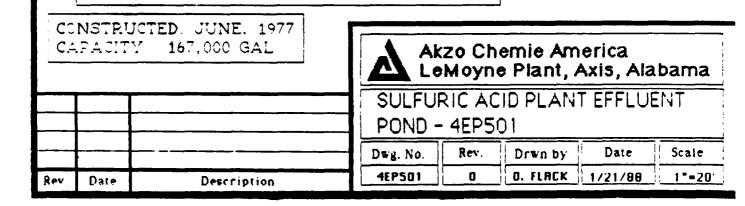
CLAY

HYPALON

LEAK DETECTION POLYETHYLENE SHEETS

6" PVC PIPE

NOTE: DRAWING ASSEMBLED FROM PROJECT FILE AND FIELD MEASUREMENTS



APPENDIX XXVII GRAPHIC AND DESCRIPTIVE WELL LOGS

GRAPHIC WELL LOGS

COLD CREEK PLANT, BUCKS, ALABAMA EXPLORATORY BORING 0-5

4/1: -/11/4 CD CHOS <u>}</u> , i 701 105 Tr 2837208E F2.1531 ! 3 Air Lina FLAST ATTEMPT 17107 CEMENTES INSTACCES 4/11/73 STARTES 75.75 CASSAO (3) 4 (A) લ 9 (v) 0.5 PROTECTIVE NIPPLE BALL VALVE ASSEMBLY (3' LING) 9-5 LENGTHS 4" PVC WELL SCREEN 4" T. 3" REDUCING BUSHING @94.5 4" PVC BLANK PIPE 2.5 LENGTH IN 3 PVC BLANK PIPE 20' LENGTH OF 4" PYC BLANK PIPE SLOTTES 32،5 مماك "1000 PIPE. (0.012" SLOT SIZE) 10 I 2 - 10 LENGTHS OF 3" CONCRETE PAD DEPTH OF ABOUT 15 0 1-20 LENGTH (4 - 1 - 13) 2.5'- 3" PVC 4, ×4. × 6. Bottom CASING ·677 TOP OF CASING DEPTH + 2.5 HHITE 7245 3018 してたく 747 DIAS DIAS မ္တ 22 () [1] 9 9 o a ۶ 8 0// /20 0 5

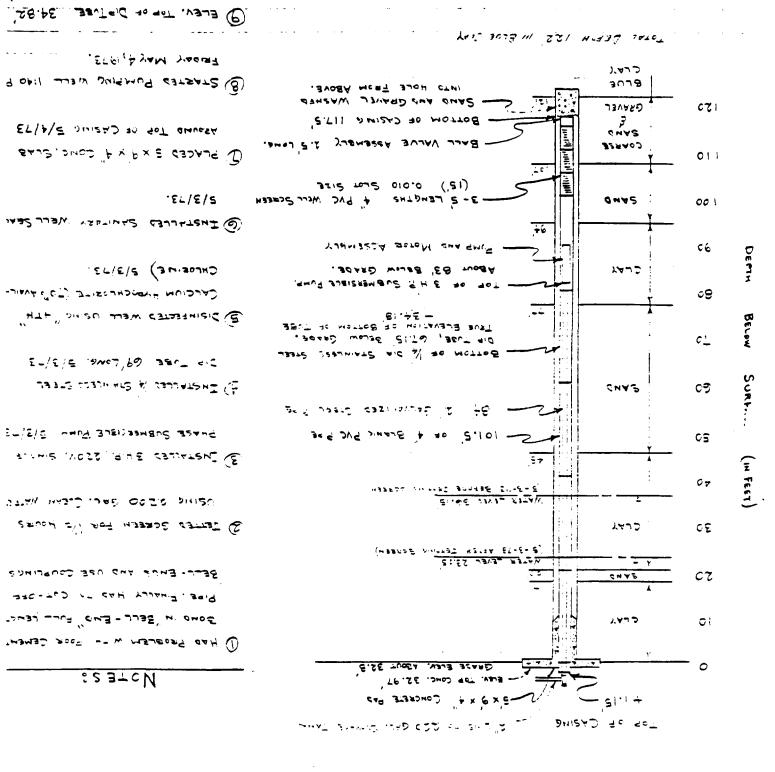
W.P.S. 4-18-73

ALABAMA 000 Axis PLANT EXPLORATORY E MOYNE

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STALLER CHEMICAL COMPANY GRAPHIC LOS DINGARM

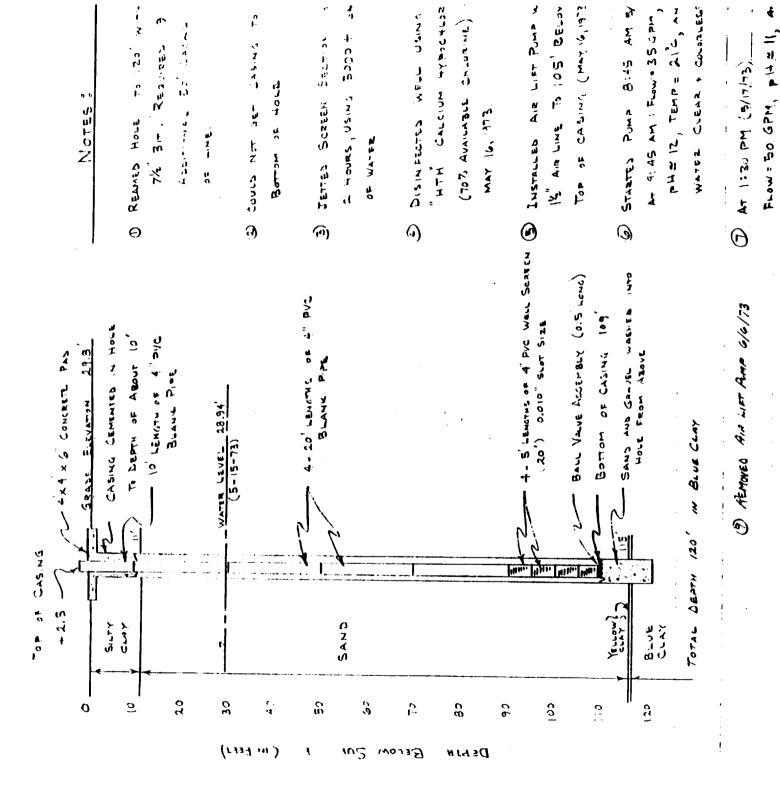
COLD CREEK PLANT, BUCKS, ALABAMA EXPLORATORY BORING 0-8



W.P.S. 6-11-73

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COLD GREEK PLANT, BUCKS, ALABAMA EXPLORATORY TO BORING 0-14



W.P.S. 8-30-73

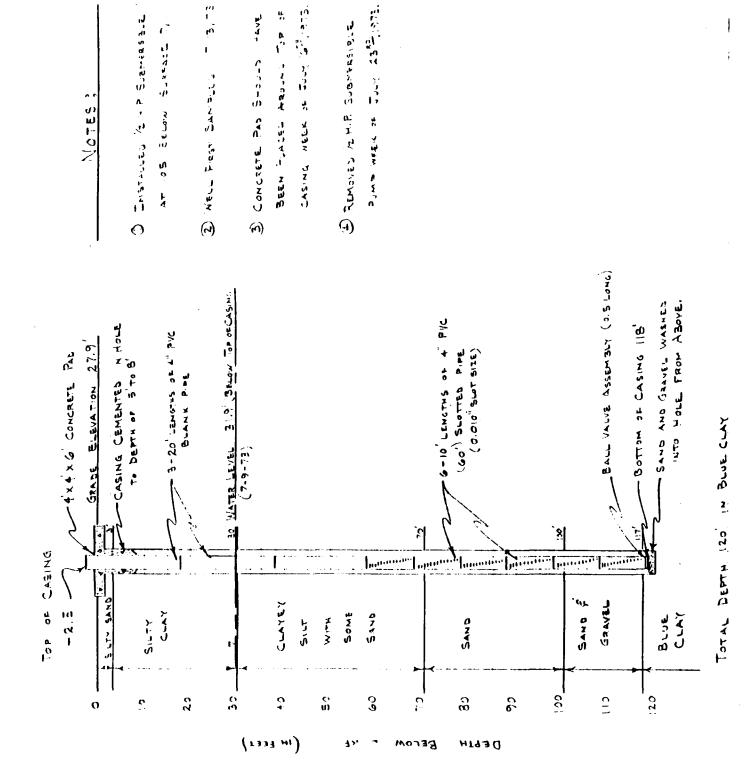
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COLD CREEK PLANT, BUCKS, ALABAMA Explorator/CLBORING 0-21



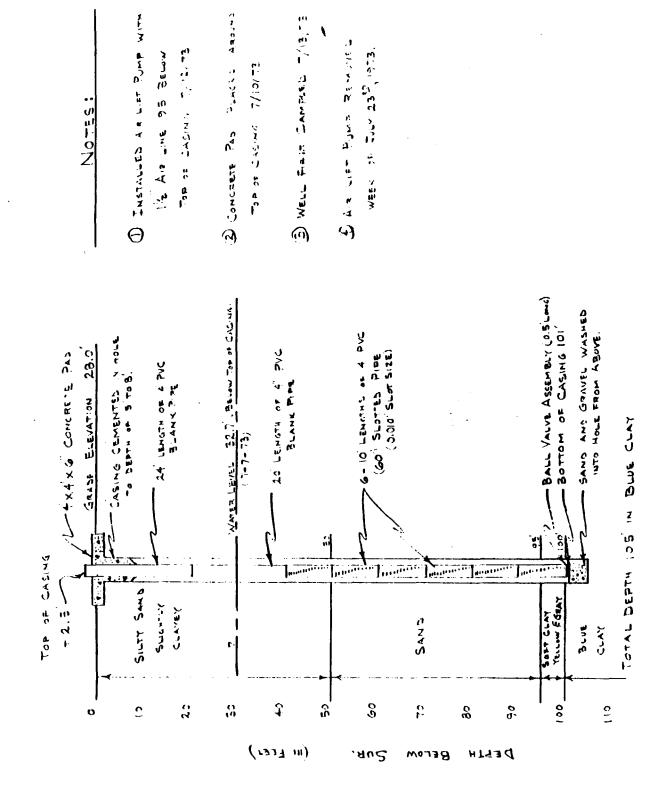
STAUFFER CHEN CAL CONDENC GRAPHIC LOG AND CASHS DIRECTOR

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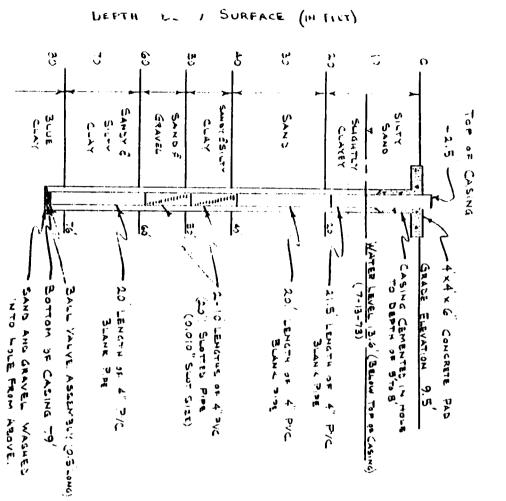
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LE MOYNE PLANT, AXIS, ALAZANA EXPLORATORY BORING 0-22



LE MOYNE EKPLORATORY BORING PIANT Axis アレアロハスト 0-23



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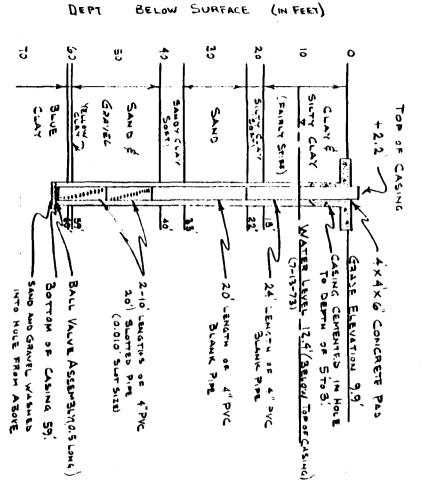
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LE MOYNE EXPLORATORY BORING PLANT V Axis, ALPURTA 0-24



● MELL DRILLED AS PISSIBLE POTABLE WATER WELL

VOTES 3

D INSTACCED I HIP ITE FLUE CAS-49 7/12/73. THE THE STEEL STEE

(3) CONCRETE PAS PLACED MESUMA : l 35 CASING 7/17/73

A WELL FIRST SAMPLES 7.31/73.

" IN BLUE CLAY

More - E M EKKIK -ELEV. FRURE 'n

CORRECT CIRADE = ECE' 13 13.9 FEET

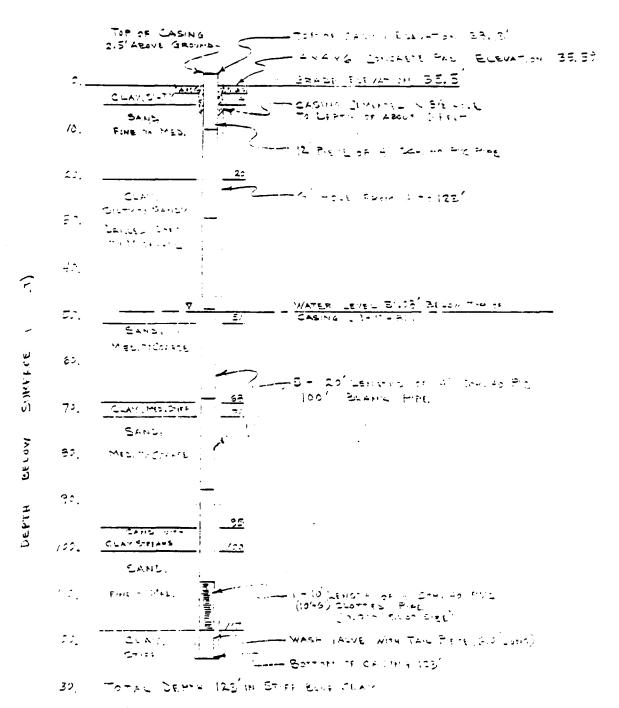
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W.P.S.

9-5-73

STAUFFER CHEMICAL COMPANY GRAPHIC LOG AND CASING DIRECTAR

LE MOYNE PLANT : AXIS : ALAMANIA MONITOR WELL : O-59 UPGRADIEM NEW BRITE MUD PIND (DRILLED 3-9-31)



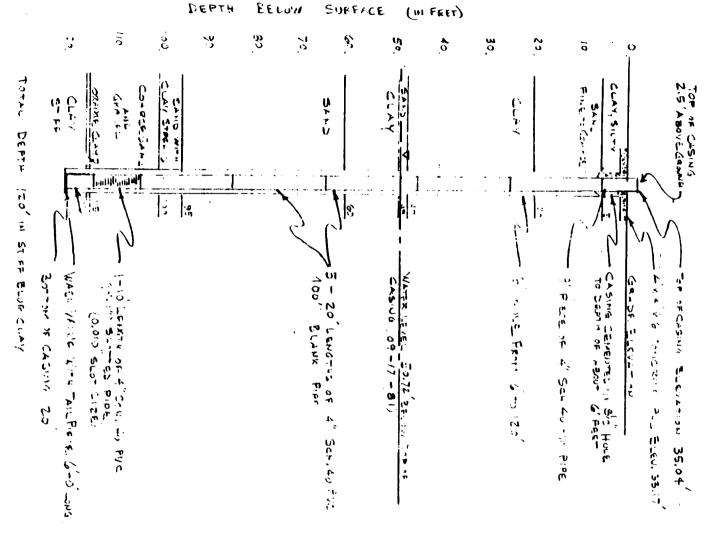
GRAPHIC LOG AND CHOING DIAGRAY

LE MOYNE PLANT, AXIS, ALAZAY.

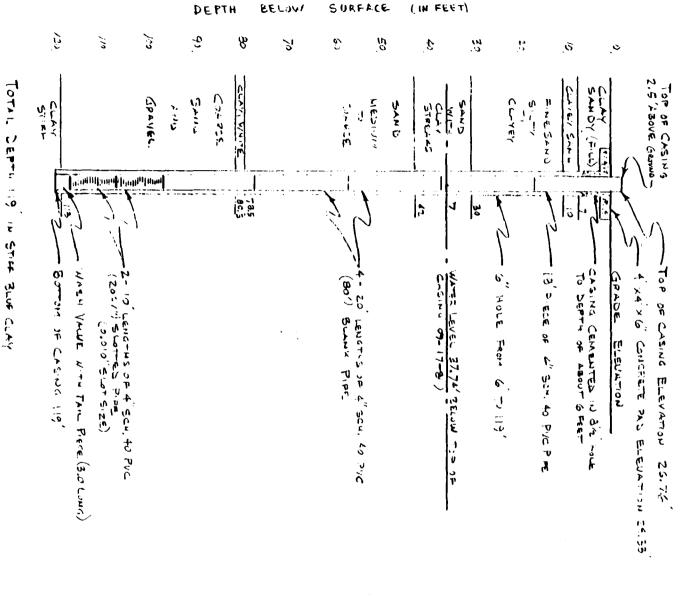
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DOWNGRADIENT NEW BRUSE MUC POND

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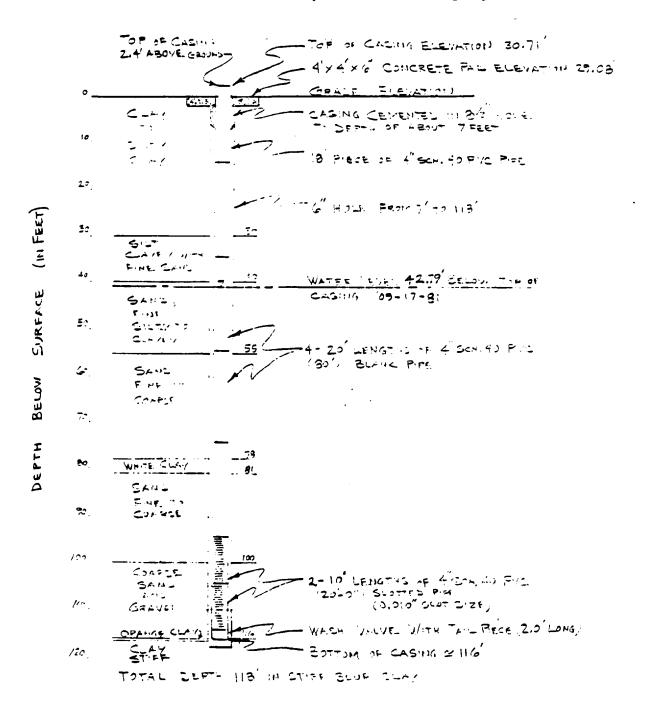


UPGRADIENT CHORINE PLANT m MOYNE PLANT MONITOR WELL DRICLES 9-19-5) AXIS, ALABERT 60 (Acmor



STAUFFER CHEMICAL COMPANY
GRAPH & LOS AND CASING DIAMERAN

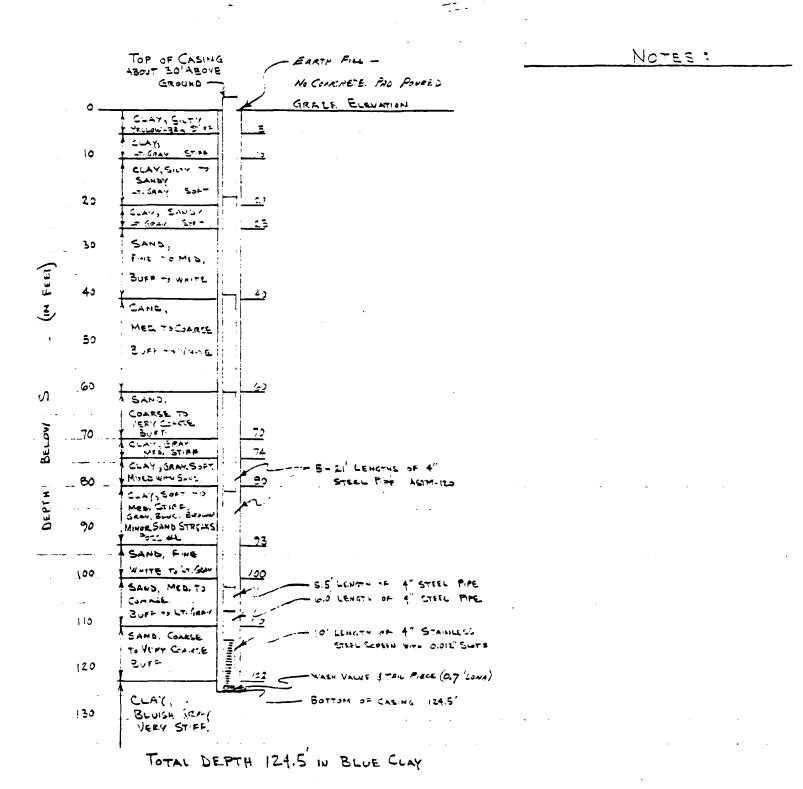
LE MOYNE PLANTS AXIS, ALLEMA
MONTON MELL 0-70
DOWNGOLD TON CHEST ME PLANT SURVEY FORL
LEG 3-23-31



3 10 00469

STAUFFER CHEMICAL COMPANY GRAPHIC LOG AND CASING DIAGRAM

COLD CREEK PLANT, BUCKS, ALABAMA TEST WELL FOR PROPOSED WATER WELL NO. 12



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	MARCH 20 THROUGH 25, 1975	2) SEE NOTES & JEMSANS IN LOG	WATER UNDER HATESIAN FRESH	BLW MEXCHES AT ABOUT 11	A COMP FOREX TO		(A) OFUT IN PRECSURE NATIONS	About 15 PEIS ON MARCH 28 17	STORAGE YOLDPE, OF CASING	170日の日 ス・シンプラ へつじょ キュ アの日		Ø WELL FIRST SAMPIED 3/23/75	,	Constitution for Man, Constitution	10 10 10 10 10 10 10 10 10 10 10 10 10 1					i			WELL 0-28 SHEET LOE.		
T GRADE ELEYATION 31.3			HALLE REGISES TO 11/3"	122.5 FEET.	, , , , , , , , , , , , , , , , , , , ,	123,2 OR 6 SCW. 40 STEEL	Par - Terrator	***	TAT TRESEDUE CEMENT TOR	123 T SURFACE	(4.0 th 0.3) Jun 44.		76		<i></i>	N. N. N		1	<u>1</u> 7	- 5/2 HOLE DRILLED TO 610'	FA2 OF 41, 00 x0227 Val	ASTA A-120 THRAME	1/22 P. PE (SCH. 40 Pine 10.79 / Pr		071
\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	CLAX, Street Clay, Property Property Property Street Construction Received Construction	Meb.Ster to Set	SAUS	11.11.11.11.11.11.11.11.11.11.11.11.11.	**************************************	COARTE	1 /		777	MOCERATION	54. EF	3	777	777	- ARAS	MED TO COARGE AND CARD	77: 3,473	4 4 160				1.372		STIFF SILTY TO SAUDY	Ī
0	ə	ન		\$		0 u)	;	B	i.		<u>လ</u>	i	(), 6	: 5	og:	011	120		6.	6.	. A.		C%)	ij	ø.

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	190	.,			, ,		• •							
_	120	CLAY	~~											
	200	-	200											
-	2.50	CLAY	200											
•	414	CLAY, SILTY TO SANDY, SOFT	- 210						•					
	210		210											
	225	SAMD, SILTY TO ITEY FME, MIXED WITH CLAY, SOFT												
	230	SOFT '												
			236					•						
	240	CLAY, STIFF,	;											
	250	SUGATCY SAURY	}: :-											
<i>^</i>	.		!											
(III FEET)	220	:	• •											
=	4	SAND	266											
こ	270	SAND	270											
u J		CLAY CLAY STEEKS												
Ο,	280_	:	280											
	290	CLAYS												
3			296											
BELOW	2 00	SAND(?)	300											
بِك		CLAY,							•					
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DEPTH		1		-										
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	350		•											
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	37)	·	370										٠	
		CLAY,	•				-							
	£80	VERY STIFF										WELL 0-	18 SHE	et Zo
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420		420
430	CLAY, SILTY, TO	
1 40	SWG-TLY SANDY	
450		450
	CLAY,	
460	NITH STREAKS	
	DE WERIN	
470	SAND	
754		
490		4.2
	CLAY. Styf	
500		<u>300</u>
310	CLAY, ST PR, NICK SUIGHT	
	EMOVET SE	
52)	FUE SANS	
	_	
530		
540		
		582' 0= +1/2" 0.D. x 0.237" WALL,
550	CLAY,	550 III /FT., ASTM A-120 THREADED
	MED. STIFF TO	PIPE (Sch. 40 Pipe 10.77 16/67.)
5పు	SAND,	<u>569</u>
570	FINE TO MED. WITH STREAMS	570
	SAND,	The state of the s
530	MESIUM TO	STAINLESS STEEL WELL SCREEN
	COMBELL	사용 : 사용사
5 %0		WASH VALVE WITH TAIL PIECE (0.5 LONG)
		BOTTOM OF CASING 591
630	··	100 WELL 0-18 SHEET 30F
	SAND, MED. TO COARSE	TOTAL DEPTH GIO'IN SAND W.P.S. 1-12-8
610	PERSON IN CORP.	(BUT ESPECT TO BE ALMOST AT BOTTIM)

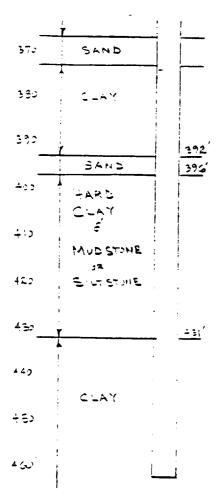
COMPANY GPAPHIC LOG AND CACING DIAGRAM STAUPPER CHEMICAL

COLD CREEK PLANT, BUCKS, ALABAMA Exploratory Borns 0-9

SEASE ELEVATION 33. 129 - 6 SCH. 40 STEE POPE - THERDED ENDS HALLISORIAN PRESSUES CEMENT JOB, 129 TO 40!. 52 HOLE DRILLED TO 40!. 52 HOLE DRILLED TO 40!. (TOTAL LENGT AND 200.)	U SEE REMARKS ON LOG	2) INSTALLED A.R. LIET (2) A. L. A. A. L. M. L. A. C. D. C. L. M. L. A. C.	() (C				
	ELFVATION		Dature and Bease.	Pares E. 45		HOLE DRILLES TO	LEMATUS JE Sch. 40 Pipe. Stal Lematura

						,	DEPTH	Bec	ow C	3 0.4	()	(r)							
<u> </u>	ф в в		(a)	ш С.,	(.)	233	м ф О		6°	250	2-2)) 	ю 10 10	270	 .	190	E	176	3 10
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4												COT SE OF CASENO	BALL PALVE	100 P.M 0.010					

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TOTAL CEATH 461 IN CLAY

DESCRIPTIVE WELL LOGS

EEGUN 4/6/73 FIRESHED 4/6/73 LOGGED BY WPStilson DRILLED BY Holland Drilling DRILL HOLE 0-5

ELEV. 29.3 (Grade) TOTAL DEPTH 120' ICCATION 662' East and 28' North of NW Corner of "Old" Property Boundary (28' onto newly acquired parcel)

Bit: 5½" 3-Blade Drag Bit Mud: None - Plain Water, but had to use lime. THECE FOUTAGE TESS J THOLOGY REMARKS ESYT'C FRCN Approx log from mud 3 Sand, fine to med., some humus material, roots 0 3 and plant material pit excavation. Soft, easy digging. 0 3 As above 3 17 20 Sand, fine to med, clean, white to buff Added 4-50# sacks 3 of lime. Added 4-50# sacks 20 70 50 Sand, fine to med, clean, white of lime. 70 74 Sand, med. to coarse, mostly white to buff, contains about 5% colored pebbles (orthoclase and chert) 76 2 Clay, med to dark gray, "Blue Clay" Drilling a little 74 harder. 76 80 4 Sand, fine to med., white to gray 80 95 15 Clay, med to dark gray, grades to olive-brown then to lt. bluish gray, contains a few wood fragments 95 100 5 Sand, coarse to med Added 4-50# sacks 4 of lime. \bigcirc 100 105 5 Sand, fine \bigcirc 118 105 13 Sand, med to coarse Clay, blue to bluish gray 120 2 118 Took sample from bit 120 T.D. Clay contains gravel from above it up to 3/4"dia. , wellSTRUPPER CHERTCAL CONTANT DRIBLE EOG Page 1 of 2

LeMoyne Plant, Ax Λlabama

Exploratory Boring I CGGED BY WPStilson DRILLED BY Holland Drilling DRILL HOLE 0-6 PEGUN 4/9/73 FINISHED 4/9/73

Approx
10CATION 300' East of RR Track (Main Line); 175' South of South ELEV. 40.9 (Grade) TOTAL DEPTH 136' Fence Line

	FUCTA	G ខ	THICK	RECUV-	1 LTHOLOGY	REMARKS
	FRCM	TC	HESS	15-1.7	1 1 1HOJ.OG ((Arasana)
	0	1.0	1		Silty clay, with some fine sand, olive to lt.	Approx log from mud
					tan to gray contains roots and some humus	pit excavation,
					material	moderate digging
ł	1.0	3.5	2.5		Silty clay, with some fine sand, yellow to	
					yellow brown	
	0	3.5	3.5		As above	
	3.5	10	6.5		Silty clay, yellow to yellow to yellow brown	
١	10	15	5	ļ	Clay, light gray to cream	
	15	20	5		Clay, silty with small amt. fine sand, mostly	Drilling a little
		ļ		\\\\\	gray but contains a few red streaks	harder
	20	52	32		Clay, with 10 to 20% silt and fine sand, 1t. gray	Drilling easier
1					with a few small red streaks through interval	
	52	60	8	1	Sand, fine to med, clean, white, contains a few	Added 2-50# sacks o
	-		ļ	ļ	small shell fragments	lime
	60 .	80	20	 	Sand, fine to med, clean, white, one streak of	
,	!			<u> </u>	yellow clay less than 0.5' thick at about 68'.	
-	,	02	1.0	·	Becomes med to coarse below 75'.	
)	80	93	13	·	Sand, med to fine, mostly white	Added 4-50# sacks o
)		100				lime.
) -	93	100	7	·	Sand, med to coarse with 5 to 10% 1/8" to 1/4"	
					size. Increase in number of colored grains (pink,	·
)			ļ 		orangish, and yellow-chert and orthoclase). Con-	
					tains a few wood fragments. White grades to buff	
	ļ	I	1	1	to tan. Becomes re coarse with depth.	1

Exploratory Boring

A/9/73 FIRISPED 4/9/73 10000 BY WPStilson DRILLED BY Holland Drilling DRILL BOLE 0-6 Approx
ELEV. 40.9 (Grade) TOTAL DEPTH 136' | ICCATION 300' East of RR Track (Main Line); 175' South of South
Fence Line

	Bit:	5½" 3-Bla	ade Drag	Bit	Fence Line Mud: None - Plain Water, but had to use lime								
	PC CTA	GE	THICK - PESS	REGUVA EBY	1 (THCLGGY	REMARKS							
	FRCH	J.C.	1. (% 10)	1911 J	T COROJA/OC	Win Alley							
	100	130	30		Sand, fine to med, clean, white to lt. gray	Added 3-50# sacks							
						of lime							
	130	132	2		Sand, coarse to pea gravel 1/8" to 3/8"								
- 1	132	134	2		Clay, yellow to yellow brown								
	134	136	2		Clay, light gray to light bluish gray								
ļ	T.D.	136											
						and the supplementation of the supplementatio							
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Cold Creek Plant, ks, Alabama

Exploratory Boring 5/2/73 FIGURE 5/2/73 ICGGED NYW.P.Stilson DRILLED NYHolland Drilling DRILL NOIS 0-8

ELEV. 32.8 (Grade) TOTAL DEPTH 122' 10CATIONNorth of New Plant Const. (N. 902.7', W. 014.0' PlantGrid)

Bit: 4½" 3-Blade Drag Bit	Mud: None - Plain Water, but had to add lime.	
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FC TA	GE	THICK - LESS	RMGC VIII. BRY	1. 10012-1-2229	REMARKS
FRCTi	'(C.	1,1700	15.1 (1 TTHOLOGY	WENNING
0	12	12		Clay, some silty, red, gray, and brown, mostly	Fairly hard drilling
· 			· · · · · · · · · · · · · · · · · · ·	gray 6' to 12'	
12	14	2_		Clay, to silty clay, yellow brown streak	
14	17	3		Clay, silty, mostly gray, silt content increases with depth	
17	20	3		Sand, fine, white to buff, contains some small white shell fragments	
20	35	15		Clay, silty to sandy, buff to light gray	"Soft" - drills easily
35	45	10		Clay, light to med. gray with a few red streaks	1
45	55	10		Sand, med. to coarse, white to buff, shells as above	
55	60	5		Sand, coarse (approx. 1/8" size) white to buff,	
			1	5 to 10% colored grains	
60	74	14		Sand, med. to coarse (just slightly finer than above), white to buff	
74	77'	3		Sand, coarse, white to buff	"Active Sand,"-caved in on bit. Started
					to lose water, adde
77	87	10		Clay, color varies - dark gray to olive gray with chocolate brown streaks	
87	90	3		Clay, light gray to light bluish gray	
90	94	4		Clay, light gray to light bluish gray with a few chocolate 1 n streaks	

Exploratory Boring

FEGUR 5/2/73 FINISPED 5/2/73 ICGGED BYW.P.Stilson DELLED BY Holland Drilling DELL HOLE 0-8

ELEV. 32.8 (Grade) TOTAL DEPTH 122' ICCATION North of New Plant Const. (N. 902.7', W. 014.0'Plant Grid)
Bit: 4½" 3-Blade Drag Bit Mud: None - Plain Water, but had to add lime. Bit: 4½" 3-Blade Drag Bit

FC CTA	AGE	SHICK -	RECO	1.70007.17.027	REMARKS
Pacif	ηc.	1,683	MAA	1 CTROLOUZ	1019.2(07.3)
94	100	6		Sand, fine, white to buff	
100	107	7		Sand, med. to coarse, buff	
107	121	14		Sand, very coarse to pea gravel	"14' of chatter cuts extra good"
121	122	1		Clay, bluish gray "Blue Clay"	
T.D.	122_			In Blue Clay	
					
					
	 				
					
			1		
			<u> </u>		
· · · · · · · · · · · · · · · · · · ·					
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EEGIE 5/14/73 FIRISHED 5/14/73 | LOGGED BYW.P.Stilson DRILLED PYHolland Drilling DRILL HOLE 0-14

ELEV. 29.8' (Grade) CTAL DEPTH 120' ICCATION About 180' North of East End of Wooden Bridge (N. 922.70) E.1232.04', Plant Grid)

FC CTA	GE	THICK	RECOVA-	L LTHOLOGY	REEARKS
encti	<u>'['</u> ['].			1 1 1111/13/2019	
0	1			Sand, clayey, organic material (humus-forest	
 -				debris, roots)	
1	5	4		Clay, slightly silty, red to reddish brown	
5	11	6		Clay, gray and yellow, with a few small streaks	
				of red	
11	20	9		Sand, fine, white to buff sand with minor amounts	
				of buff to yellow and tan silt and clay,	
				contains a few small shell fragments	
20	30	10		Sand, fine, buff to tan, shells as above	والمستقد والمراقب والمستقدم والمستقداء والمراقب والمستقداء والمراقب والمستقداء والمراقب والمستقد والمستقد والمستقداء والمراقب والمستقد والمستقداء والمراقب وال
30	35	5		Sand, medium to fine, buff to tan	
35	40	5	·	Sand, medium to coarse, buff to tan	Added 1-50# sack
					of lime
40	50	10		Sand, coarse with some 1/8" plus material, buff	"Lively Sand"
				to tan, about 5% colored grains	
50	60	10		Sand, same as above	
60	70	10		Sand, fine to coarse (mostly medium), buff to	Added 1-50# sack
<u>:</u>				lt. gray	of lime
70	80	10		Sand, fine to coarse (mostly coarse), buff to	
				tan, small amount of gravel about 76' to 79'	
·				zone. Hit thin streak (less than 1/2 foot)	
				medium gray clay at about 79'.	
80	90	10		Sand, fine to coarse (mostly medium), buff to tan,	
				possible thin streak of reddish yellow clay	
		1			

Cold Creek Plant, Bucks, Alabama

Exploratory Boring

FEGUR 5/14/73 FINISHED 5/14/73 ILEGED BYW.P.Stilson DRILLED BY Holland Drilling DRILL HOLE 0-14

ELEV. 29.8' (Grade) TOTAL DEPTH 120' LOCATION About 180' North of East End of Wooden Bridge (N. 922.70',

FC TA	GE			· ·	REFARKS
Pach	10	1505	WTY.	1 (THCL) GY	CANASAN
90	100	10		Sand, fine to coarse (mostly coarse), buff to	Added 1-50# sack
				tan, increase in colored grain content to about 5%	of lime
100	110	10	- · 	Sand, mostly coarse, same as above	"Chatter" - "cuts
					well"
110	115	5		Sand, coarse to very coarse, same as above	"Chatter"
115	116	1		Clay, yellow (maybe only ½' yellow clay then blue-gray)	
116	120	4		Clay, 1t. bluish gray to 1t. gray. "Blue Clay"	
T.D.	120			In Clay	
-					
<u>. </u>					
		·			
- /					
 					

Cold Creek Pla Bucks, Alabama BEGUN //73 FINISHED 7/3/73 LOGGED BY WPStil 1 DRILLED BY Holland Well Co. DRILL NOLE 0-21

Explc ory Boring

(Grade)

ELEV. 27.9' TOTAL DEPTH 120 LOCATION East side of swamp - about 600' east of treatment pond

Mud: Johnson's Revert Bit: 6" 3-blade drag bit

FOC	TAGE	THICK-	RECOV-		
FROM	то	NESS	ERY	LITHOLOGY	REMARKS
0	11	1		Silty sand - slightly clayey, dark brown to yellow	Added 1-25# sack
~_				brown roots & forest debris top ½ foot	Johnson's Revert
1	3	2	,	Silty sand (hard clayey silt mixed with fine sand)	
				red to reddish brown	
3	10	7		Silty clay-clayey silt, fairly soft, red brown and	
				yellow brown grades to gray with only streaks of	
·				red with depth	
10	20	10		Silty clay, fairly soft, light gray	1
20	30	10	·	Silty clay, about same as above but increase in silt	
				and decrease in clay content with depth - a couple	
				of small streaks of yellow-brown, silty clay	
30	40	10		Clayey silt, somewhat sandy (fine sand) mostly lt.gray	
				with only a few minor streaks of yellow-brown	·
40	50	10		Clayey silt, about same as above except for increase	
t	_			in silt and amount of fine sand. A few streaks of	
)) 	_			med. gray, clayey sand	
⊃ <u>50</u>	60	10		Clayey silt, generally about same as above except	
				starting to pick up med. to coarse sand and a small	
,				amount of gravel. more streaks of med. gray	1

Cold Creek Plant, Bucks, Alabama

BEGUN 7/3/73 FINISHED 7/3/73 LOGGED BY WPStilson DRILLED BY Holland Well Co. DRILL HOLE 0-21

Exploratory Boring

(Grade)
ELEV. 27.9' TOTAL DEPTH 120 LOCATION East side of swamp - about 600' east of treatment pond
Bit: 6" 3-blade drag bit Mud: Johnson's Revert

FOO	TAGE	тиіск-	RECOV-		
i'ROM	ТО	NESS	ERY	LITHOLOGY	REMARKS
60	70	10		Clayey - silty sand, similar to above but grading	
				into a sand, few streaks of lt. gray clay, and a	
			,	few wood fragments	
70	80	10		Sand, mostly coarse with some pea gravel, a few streak	S
				of gray clay which has traces of red, also streaks of	
		[med. gray clayey sand. Streaks of sand grains with a	
				reddish-brown coating (iron oxide-hydroxide).	
80	100	20		Sand, med. to very coarse with some pea gravel, buff	1
	ļ			to light gray with 10% colored grains	
100	117	17		Sand and gravel, sand is buff to tan to lt. brown,	"Chatter 105-110"
:0 				some streaks of sand with (iron oxide-hydroxide?)	
				coatings and streaks of sand grains cemented with	
<u> </u>				this material. Gravel - about 40 to 50% 1/8" to 3/8"	
	1			with about 20% of this portion colored grains, 5 to	
·×				10% of grains larger than 3/8"	
117	120	3		Clay, stiff, light gray to lt. bluish gray-"blue clay"	
T.D.	120			In blue clay.	
		-			
1		1			

LeMoyne Plr Axis, Alabama Explc ory Boring

BEGUN 3/73 FINISHED 7/7/73 LOGGED BY WPStil n DRILLED BY Holland Well Co. DRILL ... OLE 0-22 (Grade)

ELEV. 28.0' TOTAL DEPTH _ 105' LOCATION About 25' south of River Road opposite west edge of industrial

Bit: 6" 3-blade drag bit Mud: Johnson's Revert

FOC	TAGE	THICK-	RECOV-		
FROM	TO	NESS	ERY	LITHOLOGY	REMARKS
0	15	15		Silty sand, just slightly clayey, red to reddish brown	Added 1-25# sack
				with streaks of med. brown; roots & forest debris	Johnson's Revert
				top 1/2 foot	
15	20	5		Silty sand, slightly clayey, reddish brown with	
				streaks of cream and lt. gray clay containing traces	
				of red ·	
20	40	20		Silty sand, slightly clayey, mostly reddish brown,	
·				contains streaks of med. gray silty sand. Hit	ì
	_	_	-	streak of red-orange-to-yellow clayey silt at 38'	
		_		to 39'. (More clay streaks in 30 to 40' than in 20'	
				to 30')	
40	50	10		Silty sand, slightly clayey, about same as 30 to 40'.	
		_	_	Hit one or more additional streaks of the orange-yello	w.
50	60	10.		Sand, mostly med. with 2 to 3% pea gravel. Sand is	-
		_		lt. brown to reddish brown	
60	70	10		Sand, fine to coarse, with 2 to 3% pea gravel. Sand is	
	_			tan to reddish brown.	
70	80	10	_	Sand, med. to coarse, tan with about 5% orange and red-	Some "chatter"
				orange grains (iron ide-hydroxide?) coatings & stain	ing

LeMoyne Plant, Axis, Alabama

BEGUN 7/6/73 FINISHED 7/7/73 LOGGED BY WPStilson DRILLED BY Holland Well Co.

Exploratory Boring DRILL HOLE 0-22

(Grade)

ELEV. 28.0' TOTAL DEPTH 105' LOCATION About 25' south of River Road opposite west edge of industrial waste heap

Bit: 6" 3-blade drag bit

FOC	TAGE	THICK-	RECOV-		
FROM	ТО	NESS	ERY	LITHOLOGY	REMARKS
80	95	5		Sand, fine to med., lt. yellow-brown to tan	
95	100	5		Clay, fairly soft, streaks of yellow-orange, cream,	Went from sand to
		_		lt. and med. gray	clay without hit-
100	105	5		Clay, stiffer, lt. to med. gray and bluish gray	ting gravel
	_	-	-	"Blue Clay"	
T.D.	105			In Blue Clay	
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LeMoyne Plan Axis, Alabama

Explc ory Boring REGUN .0/73 FINISHED 7/10/73 LOGGED BY WPStil a DRILLED BY Holland Well Co. DRILL .OLE 0-23 (Grade)

ELEV. 9.5' TOTAL DEPTH 80' LOCATION Adjacent to east side of industrial waste heap about 250' north of River Road

Bit: 6" 3-blade drag bit

FOO	TAGE	THICK-	RECOV-		
ГROM	OT	NESS	ERY	LITHOLOGY	REMARKS
0	10	10	\	Silty sand, slightly clayey, lt. tan, reddish brown,	Added 1-25# sack
				and 1t. to med. gray, dark brown; roots and forest	of Johnson's Reve
				debris top 1/2 to 1 foot	
10	20	10		Silty sand, slightly clayey, reddish-brown to yellow-	
				brown	
20	35	15		Sand, fine to coarse, tan to 1t. brown	Some "chatter"
35	40	5		Sand and clay, gradational contact	
40	50	10	\ <u></u>	Clay, sandy and silty; soft, lt. grayish brown, some	
	_			lt. gray clay with traces of red, and some yellow-	
				brown color	
50	60	10		Sand and gravel (med to coarse sand with gravel),	
	_			strong orange-brown coloration from both coatings	
·				on grains and intermixed orange silty clay	
$\infty \frac{60}{}$	76	16		Sandy clay and silty clay, fairly soft, contains	
χ Τ΄				streaks of sand, some of which contain thin seams of	
) 				cemented grains. Cementing agent is red to yellow-	
<u> </u>				brown (iron oxide-hydroxide). Clay gets stiffer with	
76	<u> </u>			depth	
76 T.D	80 80	4		Clay, stiff, bluish ' "Blue Clay" In Blue Clay	

LeMoyne Plant, Axis, Alabama

Exploratory Boring BEGUN 7/11/73 FINISHED 7/11/73 LOGGED BY WPStilson DRILLED BY Holland Well Co. DRILL HOLE 0-24

(Grade)
ELEV. 9.9' TOTAL DEPTH 63' LOCATION About 100' south and 150' west of southwest corner of barge

slip at River

Bit: 6" 3-blade drag bit

FOO	TAGE	THICK-	RECOV-		
FROM	то	NESS	ERY	T LAIOTOCA	REMARKS
0	8	8		Clay, silty and sandy clay mostly gray with red	Added 1-25# sack
				streaks, dark brown to tan sandy silty with roots	of Johnson's Rever
			,	and forest debris top 1/2 to 1 foot	
8	13	5		Silty clay with small amt. fine sand, mostly gray	
				with some yellow and yellow brown	
13	18	5		Silty clay, fairly stiff, slight amount of fine sand,	
				lt. gray	
18	22	4		Silty clay, soft, sand content increases with depth,	1
			,	lt. gray	
22	35	13		Sand, fine to very coarse (mostly med. to coarse),	
	-			buff to lt. brown with yellow brown silt and grain	
				coatings	
35	40 '	5		Clay? 5' interval of soft clay intermixed with fine	
<i></i> 	-			sand, lt. brown to med. gray	
× 40	50	10		Sand, med. to coarse with about 5% pea gravel, buff	Looks like a fairly
5 ——	-			to white with minor streaks of gray - about 5%	clean sand
- 	_			colored grains	
	<u> </u>				
n					

LeMoyne Plan Axis, Alabama

BEGUN 1/73 FINISHED 7/11/73 LOGGED BY WPStill DRILLED BY Holland Well Co.

Explo ory Boring

(Grade)

DRILL HOLE 0-24

ELEV. 9.9' TOTAL DEPTH 63' LOCATION About 100' south and 150' west of southwest corner of barge

slip at River

Bit: 6" 3-blade drag bit

FO	OTAGE	THICK-	RECOV-		
FROM	ТО	NESS	ERY	LITHOLOGY	REMARKS
50	59	9		Sand, fine to coarse (mostly med. to coarse), gets	"Locked-up" bit at
	_	_	·	more coarse with depth, buff to lt. gray, about	55'
			3	5% colored grains	
59	60	1		Clay, silty, soft, yellow-brown and lt. and med. gray	
60	63	3		Clay, stiff, lt. gray to lt. bluish gray, "Blue Clay"	
T.D.	63			In "Blue Clay"	
				·	
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0 C					
0					
<u></u>					
2					

	3/9/81	_ FINISH	ED 3/10/81 LOGGED BY W.P. StilsonDRILLED BY Holland Well Company	DRILL HOLE 0-59
GRADE DLEV.	35.3	TOTAL	DEPTH 123 LOCATION Stauffer Grid Coordinates S. 1261.5 E.	2563.5
Bits:	,			
FOO	TAGE	THICK-		
FROM	то	NESS	LITHOLOGY	REMARKS
0	4	4	Clay, silty to sandy, fairly stiff, red-brown, and some yellow-brown	
4	10	6	Sand, fine, buff to tan, mixed with minor silty clay	
10	20	10	Sand, fine to medium (mostly fine), white to buff	
20	40	20	Clay, silty to sandy, buff to tan with streaks of soft lt. gray clay	"Drilled soft"
40	51	11	Clay, silty to sandy, soft to moderately stiff, white to lt. gray	
			intermixed with streaks of lt. gray clay	
51	60	9	Sand, medium to coarse, traces of gravel, white to lt. gray	"Light chatter 55-60"
60	67	7	Sand, medium to coarse, buff to white	Taking fluid
67	68	1	Sand, medium to coarse, minor gravel, color change - red-brown to orange-	
			brown, staining on sand and gravel	Taking fluid
63	71	3	Clay, moderately stiff, lt. gray, traces of orange staining & wood fragments	
71	80	9	Sand, medium to coarse, lt. gray to tan	"Lively sand" Taking fluid
30	90	10	Sand, medium to coarse, (mostly coarse) buff to tan, lt. yellow-brown	
<u>y —</u>	'		staining on # 5% of grains	Taking fluid
90	100	10	Sand, medium to coarse, with minor fine, buff to tan, lt. gray and cream,	
0			clay streaks, probable 95-100', traces of wood fragments, staining on	
0			≃ 5% of grains	Taking fluid
№ 100	'7	17	Sand, fine to medium, mostly fi buff to tan, hit hard streak just	H1 *1 / CH

			LeMoyne Plan' Axis, Alabama	ET 2 OF 2					
BEGUN	_ <u>'81</u>	_ FINISH	ED 3/10/81 LOGGED BY W.P.St. onDRILLED BY Holland Well Company	DRILL LE 0-59					
GRADE ELEV.	35.3	TOTAL	DEPTH 123 LOCATION Stauffer Grid Coordinates S. 1261.5 E.	2563.5					
	Bits: 31/2" Drag Bit 0 to 6'								
	6"	Drag Bit	t 6' to T.D. Mud: Johnsons Revert						
FOO	TAGE	THICK-							
FROM	то	NESS	LITHOLOGY	REMARKS					
117	120	3	Clay, stiff, lt, gray to lt, bluish gray						
120	123	3	Clay, stiff, bluish gray						
T.D.	123		in stiff bluish clay						
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o —		<u> </u>							
23									
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LOGGED BYW.P.StilsonDRILLED BY Holland Well Company DRILL HOLE 0-62

BEGUN GRADE ELEV.	33								
Bits:	8 1/2" 6"	Drag Bi	t 0 to 6' t ^{6'} to T.D. Mud: Johnsons Revert						
FOO	TAGE	THICK-							
FROM	ТО	NESS	LITHOLOGY	REMARKS					
0	5	5	Clay, silty to sandy, hard, (top 2 to 3' was compacted fill-similar material)						
5	10	5	Sand, silty to clayey (fine to medium sand) lt. yellow-brown						
10	15	5	Sand, fine to medium, white to buff						
15	20	5	Sand, coarse and some gravel to + 1/4"						
20	20 40 20 Clay, silty, moderate to soft, lt. gray and tan, some medium gray with								
	ļ <u>.</u>	-	inter-bedded streaks of fine sand						
40	47	7	Clay, moderately stiff, lt gray to almost white	······					
47	49	2	Sand, medium to coarse, buff to white						
49	60		Clay, moderate to soft, lt. gray to buff, thin sandy streaks probable						
60	80	20	Sand, fine to coarse, (mostly medium) buff to white						
80	95	15	Sand, fine to med. (some coarse), buff to white, possible clay streaks	····					
95	100	5	Sand, fine, white to buff, with 1' to 2' thick clay streak (1t. gray)						
100	114	14	Sand, med. to very coarse, minor gravel, moderate to heavy staining on						
93-	'		more than 5% of grains (rust to brown)						
<u> 114</u>	115	1	Sand, coarse, cemented (?) "sandrock" - drilled very hard, heavy staining + 5%						
115	116	1	Silt to silty clay, orange, (less than 1' thick)						
	120	4	Clay, stiff, lt. bluish gray						
∾ T.D.	0		In stiff bluish clay						

LeMoyne Plan+ Axis, Alabama 9 1 of 2

[8] FINISHED 8/25/8] LOGGED BY W.P.Sti on DRILLED BY Holland Well Company DRILL E 0-68

GRADE TOTAL DEPTH 119' LOCATION Stauffer Grid Coordinates S. 1028.18 E. 1546.41 ELEV. 24' Bits: Drag Bit 0 to 6' 8 1/2" Drag Bit 6' to T.D. Mud: Johnsons Revert FOOTAGE THICK-NESS LITHOLOGY REMARKS FROM TO 0 6 Clay, to sandy clay, firm, fill material contains old paving (clam) shells and debris Silt to clayey silt, black organic material, roots, etc. (top of depressed 6 surface of pre-filled swamp in this area) Sand, silty to clavey, fine peaty (organic rich), fairly soft, tan to 10 lt. vellow-brown and gray with streaks of black decayed swamp vegetation Sand, fine, silty to slightly clayey, lt. to medium gray "Drilled fairly firm" 20 10 10 Sand, as above with sand content increasing, lt. to medium gray 20 24 4 "Fairly soft" 24 30 6 Sand, fine silty with trace of clay, yellow-brown "Drilled somewhat soft" Sand, fine to medium (mostly fine) slightly silty, lt. yellow-brown to tan, .__30 42 12 thin streaks of lt. gray clay "Drilled slightly Sand, medium to coarse, with 10% gravel + 1/8" to + 1/4", tan, light orangesofter than above, 42 50 8 but had a light chatter" brown (iron) staining on some of gravel, minor clay streak possible 4 ON 50 Taking fluid Sand, medium to coarse, with minor gravel, tan to gray, traces of staining 60 10 \bigcirc (iron) on grains, minor streaks of buff, tan, and lt. gray clays, traces of dark gray to black silt or clayey silt \bigcirc Takino fluid Sand, medium to coarse, minor amount fine, with trace of gravel, tan to 70 10 60 "Light c' +ter" 2 1t. gray, thin clay streaks ble, traces of staining

BEGUN

BEGUN	8/25/8	<u>l</u> FINISH	ED 8/25/81 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company	DRILL HOLE 0-63					
GRADE DLEV.									
Bits:									
FOO	TAGE	THICK-							
FROM	то	NESS	LITHOLOGY	REMARKS					
70	79	9	Sand, medium to very coarse, 3 to 5% gravel +1/4", minor amt. +1/2", lt. to	Taking fluid					
			medium gray, thin streaks of lt. gray clay possible	"Light chatter"					
79	80	11	Clay, fairly firm, lt. gray - almost white (about 1-1/2' to 2' thick)						
80	90	10	Sand, fine to coarse (mostly coarse) trace of gravel, tan to lt, gravish tan	Taking fluid "Chatter"					
90	100	10	Sand, fine to coarse (mostly coarse), minor gravel, lt. gray	Taking fluid "Chatter"					
100	117	17	Sand, very coarse to gravel, 50% +1/8", about 30% +1/4"	Taking fluid "Heavy chatter"					
117	118		Sand, medium to coarse, partially cemented, thin "sandrock" gray to lt. brown	Mud pit almost empty					
			red-brown (iron) staining slight color change in drilling fluid						
113	119		Clay. stiff. lt. bluish grav						
T.D	119		In "blue clay"						
<i>(</i>)	'								
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<u> </u>									
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LeMoyne Plar Axis, Alabama

ge 1 of 2 LE 0-70 28/81 FINISHED 8/28/31 LOGGED BYW.P.St. onDRILLED BY Holland Well Company DRILL BEGUN — GRADE CLEV. 28.7 TOTAL DEPTH 118 LOCATION Stauffer Grid Coordinates S. 1259.32 E 1662.19

Bits: 81/2" Drag Bit 0 to 7'

	6"	Drag Bi	t 7' to T.D. Mud: Johnsons Revert	
FO	DTAGE	THICK-		
FROM	то	NESS	LITHOLOGY	REMARKS
0	0.5	0.5	Topsoil	
0.5	20	19.5	Clay, to silty clay, trace of fine sand, firm, lt. gray with minor yellow-brn	
20	30	10	Clay, to silty clay, minor fine sand, moderately firm, medium to dark gray	
30	40	10	Silt, clayey, with fine sand, moderate to soft, lt. to medium gray possible	
			sand streak at 38'	
40	50	10	Sandy silt - silty sand, fine, clayey, soft, lt. and medium gray, traces of	
			yellow-brown, minor lt. brown sand	
50	55	5	Sand, silty to clayey sand grades to sand, lt. gray	
55	60	5	Sand, fine to coarse, traces of gravel, lt. gray to buff	"Light chatter" Taking fluid
60	78	18	Sand, fine to coarse, traces of gravel, buff to tan, clay streaks at 62'	
			possible at 68', other thin streaks possible (lt. and medium gray)	
78	81	3	Clay, moderate to soft, lt. gray to white	
81	90	9	Sand, medium to coarse, some fine,tan to lt. gray, staining (iron) on	
, - -			3 to 5% of grains	
- 	100	10	Sand, fine to coarse with minor gravel ~ 5% +1/4",tan to lt. gray, inter-	
) 			bedded streaks of soft lt. and medium gray clay probable.	
⊃ <u>100</u>	110	10	Sand, very coarse and gravel, about 20% +1/4" and 3% +1/2", buff to gray,	
γ <u> </u>	- <u></u>		staining (iron) on 1 to 3% of ins	

0-70								
28.7 TOTAL DEPTH 118 LOCATION Stauffer Grid Coordinates S. 1259.32 E. 1662.19 8 1/2" Drag Bit 0 to 7' 6" Drag Bit 7' to T.D. Mud: Johnsons Revert								
KS .								

Since !

BIGUN 3 1/2 FINISHED 3/21/1/ LOGGED BY WILLIAM DRILLED BY WILL CAST OF DRILL HOLE TW-12

PRICE TO THE PROPERTY OF THE PROPERTY

ELEV. TOTAL DEPTH INT. LOCATION COMAN WAS A PROTECTION OF THE STATE OF

Tradition of the control of the cont

FOO	PACE	THICK-			
PROM	TO	NESS		LITHOLOGY	REMARKS
2), ; .	5	- 4	CLAY BULLY DOWN COLD FOR DAY CLAY Survey Cold Survey CLAY Survey Down Survey CLAY Survey To Chin Survey CLAY AN AND GRADE SURVEY CLAY AN AND COLD SURVEY CLAY AN AND COLD SURVEY CLAY AND AND COLD SURVEY CLAY AND COLD SURVEY CLAY AN AND COLD SURVEY CLAY AND COLD SUR	.
<u> </u>	!0	5	•	Company Low Source Low Group	
10	2.5	<u> </u>	,	CLAY SULLY TO STORE SON SON SON STORE VINE VINE	
20	25	5	,	CIAL ANTERESTICATION OF THE STATE OF THE STA	
25	30	!-		SAMO, FOR 12 110 1110 1110 1110 11 11 11 11 11 11 11	
2,	4)	;)		JAM FOR LOWE - WOLL TOPLY	Forms, allow to from
4)	50	10		LAME, MELLINGER WITH TIME A LOW BOX & COM STORE	
				TRACE OF THE STORES OF THE STREET STREET,	
50	197	(1)		CAND, AEDT JOING FORENT HO CONTINUES	English France
(50)	75	[0		SAME, CONFIRM TO VERY EXPENSE (CART OF) BUTT WITH 10-70% CALORED GRAME.	TRAIGHT FASICE
73	74	1		CLAY MEDICHER, LT. Tolker, May - 1000 July 100 Style 11. March May	
74	80	10		CLAY, CORD & MORELYOUR MET DOWN IN SOME 20 1 SOLD SHOW SHEET	
Bo :	93	13		CLAY, SOFT TO MEL. DITE, MARLY DIES - MEL. GRAY WAR MINES PATT.	
<u>ي</u>	, 			DE La . De mode de la g. Alto Mare de la Bernan-alte par y Clay	
-1 -1				Crim States, Portions	
○ 73	100	7		Seal Nery trabaction, Miner of the may	
€ 100	110	10		SAME Met, to your as a liter on you the bone is to down sold on the	TARMA COME
M 110	122	12		SAIL COARLE TO VERY COLD & (TENTE OF ME) BOFF, AROUS BOOKS SOME SOME	LAMUIN TOUL " LIGHT CHANGE
177	1 11	-, -			

DRILL HOLE TW-12

BEGUN 3/25/27 FINISHED 3/25/77 LOGGED BY WELLOW DRILLED BY Logged BY Logged BY ELEV. 3. 1000 TOTAL DEPTH 1.45 LOCATION 10.67. 10.61 3 (10.000)

FOOTAGE		THICK-	Control of the Contro	
POM	ТО	NESS	LITHOLOGY	REMARKS
T.D.	124.5		Now the 10 sections of some of the section of the s	
			CONTRACT TO THE STATE OF SHIP SORTER.	to
			111 Mayreon 1813, firm a real Processing	BEST AVAILABLE COPY
			1 Same 140 100 100 100 100 100 100 100 10 10 20"	SILAB
 			Girphone, Ann All Co Day, Joseph Mar 1 400	EL WY
			TO When you then Man They would at 25 "	ØE.
			Hore to Table por Milett & Comme (1) to 22.	
			20'DE 12" Dignered , 0.025 That Signes Creek Brown	
			Section Was to an ONL of M. STITLE HOLE CALIDE.	
			Wale the 11 " to be are you Mer was on	
			Secretal They Stong Thy or Letter Decrease of 118.	
			Cross 2 1 1 1 1 1 13 - 113	
		,	COMMENT: IF THE MODEL OF THE SCHEEN WAS	
			PLACEL ACOUT GO FIRE TO NOTH. THEY MISSED THE	
			ROTTON GOT THE CORE WALL THE THE THE	······································
			OF THE COEFFEE WAS OFFORING FINE CHARLE	
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			DIA. CTENTICE DOLLER AND 10 or 8 July Les Was Less (2)	
			The second secon	

COLD CREEK PLAN BUCKS, ALABAMA PAGE 1 of 6

Explor y Boring DRILL OLE 0-28

TREGUE 3, 75 PHILEHED 3/25/75 LOCGED BY W.P.SLIP DELLED BY Molland Well Co.

Grade CTAL DEPTH 610' LOCATION N.899.67', E.315.0' Plant Grid

,	Bit	6" 3-B	lade Drag	Bit	Mud: Carloss Hi-Gel Bentonite	
i	on jak	. С	THICK-	PECOV		
		10	NUSS	EPY	1.FT = 0 LOGY	REMARKS
		10	10		Clay, silty, fairly stiff, yellow with streaks of light gray, and minor streaks of red.	
10)	16	6		Clay, fairly stiff, light gray, with minor streaks of red.	
16	j	_17	11		"Sand rock," 6" to 8" thick streak of limonitic(?) cemented sand.	"drilled hard"
17		28	11		Clay, slightly soft, medium to dark gray.	
28	3	33	5		Sand, fine, light gray to white.	
3.	3	34	11		Sand, fine to medium, reddish brown.	
34		40	6		Sand, medium, buff to light tan, contains iron stained grains.	Added 3-50# sacks Hi-Gel
!						Bentonite.
40)	50	10		Sand, medium to coarse, buff to white.	Formation taking fluid,
	_					Added 6-50# sacks Hi-Gel
	· .	 - 				Bentonite.
$\stackrel{\smile}{\hookrightarrow} \underline{50}$)	57	7		Sand, coarse.	
$\stackrel{\square}{\supset} \underline{5}$	7	64	7		Sand, coarse, and small gravel.	
O 6/	·	70	6		Clay, slightly stiff, light gray.	
						
2		-				
l	1		1		1	}

				OVER DIRECT FRANT, DUCKS, MEMDARIA TAME & UT O	Evaloratory Roring
$10^{\circ} GUM$	3/20/15	FIRESHED	3/25/75	LOCGED BY W.P.Stilson DRILLED BY Holland Well Co.	Exploratory Boring DRILL HOLE 0-28

ELEV.Approx 35' TOTAL DEPTH 610' LCCATION N.899.67', E.315.0' Plant Grid Grade

Bit 6" 3-Blade Drag Bit Mud: Carloss Hi-Gel Bentonite

0 (1 6 3-B	lade Drag	וטונ	Mud: Carloss Hi-Gel Bentonite	
$i = F \circ O$	Pagt;	THICK-	RUCOV-	•	
10167	TO	MESS	ЕРҮ	Terreords	REMARKS
	77	7		Clay, sandy, slightly stiff, light gray.	
	80	3		Clay, moderately stiff, medium to dark gray, contains wood	
,				fragments.	
30	94	14		Clay, moderately stiff, light to medium gray to slight bluish gray.	
94	100	6		Sand, medium to coarse, and gravel.	
100	113	13		Sand, coarse, and gravel.	
113	122	9		Clay, very stiff, bluish gray.	
				Note: Drilling of 6" hole was stopped. Redrilled hole with	
1				11-7/8" bit to depth of 123 feet. Ran in 123'-2½" of 6"	
				steel pipe and pressure cemented in hole. Started drilling	
				5½" hole inside 6" pipe.	
118	_123	5		Drilled out cement.	
123	130	77		Clay, slightly silty, stiff, bluish green.	
130	160	30		Clay	
160 <u>160</u>	170	10		Clay, silty to sandy, moderately stiff, possible sand streaks.	
O ₁₇₀	190	20		Clay, sandy, soft, (interbedded sand streaks).	
0					
Μ					
	1	1		1	1

DECUME 3 75 FIRESHUD 3/25/75 LOCGED BY W.P.Stil DELLED BY Holland Well Co.

Explora DRILL "y Boring DE 0-28

Grade

Bit 6" 3-Blade Drag Bit Mud: Carloss Hi-Gel Bentonite

()	3-1	stage brag	טונ	mud: Carloss Hi-Gel Bentonite	
i'en:	rizetti.	THICK-	! LEECOV= :		
• • • •	To	miss	ЕБА	1/PSOLOGY	REMARKS
100	200	10		6.1	
190	_200	10	<u> </u>	Clay.	
200	210	10	<u> </u>	Clay, silty to sandy, soft.	
210	220	10		Sand, mostly fine sand, with minor clay streaks, (clay could be	
				from up-hole).	
220	236	16		Sand, mixed with clay	"Drilled soft." "Slight
				,	chatter just before
					236 feet."
236	240	4		Clay, stiff, medium gray and greenish gray.	
240	250	10		Clay, slightly sandy, stiff, bluish green.	
250	266	16		Clay, slightly sandy, stiff, bluish gray, (grades to medium to	
				dark gray, soft, sandy clay).	
266	270	4		Sand, sand streaks in a stiff clay.	"Some chatter."
270	280	' 10		Sand streaks interbedded in a light bluish gray stiff clay, (minor	
,				gravel noted that could be from the 266-270' interval - circulated	
)	.			10 minutes).	
280	290	10	}	Clay, stiff.	
290	296	6		Clay, stiff, bluish green.	
1	}	1	}	\	

COLD CREEK PLANT, BUCKS, ALABAMA PAGE 4 of 6 LUCH URLEK PLANT, BUCKS, ALABAMA PAGE 4 of 6 Exploratory Boring 145000 3/20/75 FIRESHED 3/25/75 LOCGED BY W.P.Stilson DULLIED BY Holland Well Co. DRILL HOLE 0-28

ELEV-Approx 35' TOTAL DELTH 610' EXTATION N.899.67', E.315.0' Plant Grid Grade

Rit 6" 3-Rlade Drag Rit

Mud. Carloss Hi-Gol Routonita

B	it 6" <u>3-</u> [Blade Drag	Bit	Mud: Carloss Hi-Gel Bentonite	
	17815 	THTCK -	EBCOV-		
1.039	'('')	HESS	EPY	LTP*sot.cx;Y	REMARKS
296	300	4		Sand(?) or streaks of fine sand in soft clay.	"Drilled soft."
300	310	10		Clay, stiff, bluish green, (hit log at 305' - some wood fragments	
				and also some brown-black silty clay).	
310	320	10		Clay, stiff, blue-green and medium gray.	"Hard drilling."
320	330	10		Clay, stiff, as above but with some olive green.	"Some chatter at
					320-323 feet." (?)
330	350			Clay, soft.	
350	360	10		Clay, soft, bluish green, light to medium gray sandy clay, and	
				brown-black silty clay, also minor wood fragments.	
360	370	10		Clay, soft, bluish green, more green and less of the brown-black	
				than above.	
370	380	10		Clay, very stiff, mostly green and gray green, with minor amount	"Hard drilling."
		•		soft, brown, silty to sandy clay.	
380 ~	390	10		Clay, stiff, light to medium gray, and green tones (almost no	
(sand at all).	
390		30		Clay, stiff, mostly light gray to greenish gray (almost_no sand	
) 			at all).	
K) <u></u>				
	•	1	1	1	1

PAGE 5 of 6 COLD CREEK PLANT BUCKS, ALABAMA

PEGUY 3, 75 FINISHED 3/25/75 LOCGED BY W.P.Still DELLIED BY Holland Well Co.

Explor

Explor y Boring DRILL .JLE 0-28

ETTY - Approx 35' TOTAL DEPTH 610' JONATION N.899.67', E.315.0' Plant Grid Grade

Rit 6" 3-Rlade Drag Rit Mud. Carloss Hi-Gol Rontonito

	3-l	Blade Drag	Bit	Mud: Carloss Hi-Gel Bentonite	
For.	HWIII	LIHECK-	PECOV-		
1 - 1 (3)	TO	BESS	EPY	LUTSOLOGY	REMARKS
420	430	10		Clay, stiff, but just slightly softer than last 30 feet; also,	
1				slightly more silty than last 40 feet.	
430	440	10		Clay, slightly silty with traces of sand, sand content increasing	
			ļ	with depth, stiff, mostly light gray to greenish gray.	
440	450	10		Clay, about same as above, but sand content increasing.	
450	460	10		Clay, with streaks of medium sand.	"Chatter 450-460 feet."
460	470	10		Clay, with streaks of medium to fine sand.	"Chatter 460-466 feet,
\				·	soft 466-470 feet."
470	490	20		Clay, medium gray sandy clay, light gray stiff clay, softer cream-	"Light chatter." (?)
	-			colored clay with fine sand, green clay with medium sand, and	Washed sample does show
				olive green clay.	some sand.
490	500	10		Clay, about same as above (470-490), but starting to get more	
:				stiff blue-green clay again.	
500	± 510	10		Clay, about same as 490-500, but with slight increase in amount	
L.				of fine sand.	
510	520_	10		Clay, about same increase in amount of fine sand.	"Some chatter."
	-				
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7		1	I		

PAGE 6 OF 6

COLD CREEK PLANT, BUCKS, ALABAMA

THEOUR 3/20/75 FIRSTERD 3/25/75 LOCGED BY W.P.Stilson DILLED BY Holland Well Co.

Exploratory Boring DRILL HOLE 0-28

DESV-Approx 35' TOTAL DEPTH 610' 1000 N.899.67', E.315.0' Plant Grid Grade

Bit 6" 3-Blade Drag Bit Mud. Carloss Hi-Gel Rentonite

B	it 6" 3-	Blade Drag	Bit	Mud: Carloss Hi-Gel Bentonite	
	(_ ?	BECOV-		
	1.1	1.1.05	017	LITTO LOGY	REMARKS
520	550	30		Clay, medium gray to a bluish gray stiff clay, some olive green	
		_		clay, minor amount of soft cream clay containing fine sand.	
550	560	10	,	Clay, not as stiff as above, medium to dark gray, also dark brown	
- · · · - ·				to black soft clay. Definite increase in fine sand content.	
560	570	10		Sand, fine to medium, mixed with soft clay streaks.	"Chatter 566-570 feet.
570	580	10		Sand, medium to coarse.	"Good chatter."
580	600	20		Sand, fine to coarse, mostly medium to coarse.	
600	610	10	<u> </u>	Sand, medium to coarse.	"Gotsomewhat soft, did
		_			not cut as good as
T.D.	610			In sand, but probably near the bottom of the sand.	above."
			· · · · · · · · · · · · · · · · · · ·	Notes: (1) Installed 10' section of 4" diameter, 0.010" slot	
				well screen. Screen is set from 580 to 590 feet below ground surface (approx.+1.0'). Came out of	
1				hole with 582 feet of $4\frac{1}{2}$ " O.D. x 0.237" wall, 11 lb/ft, ASTM A-120 threaded pipe.	
0.5	,			(2) Checked well on 3/28/75. Well will flow at rate	
(L)				as per Water Well Handbook). Shut-in pressure at	
00			<u> </u>	surface-is-about-15-psi	
——————————————————————————————————————			-		
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with the restriction of the rest

Cold Creek Plant, 2ks, Alabama

Explore ry Boring

5/3/73 FIRISHED 5/7/73 LOGGED BYW.P.Stilson DRILLED BY Holland Drilling DRILL HOLE 0-9 Approx.
E) 33' Grade 'TC'AL DEPTH 461' 10CATION 10' North of Paint Shop (S. 002.71', E. 577.72'PlantGric) Bi : 4'." 3-Blade Drag Bit Mud: Lime and Bentonite Used When Reaming First 128' of Hole to 11 7/8"Dia

₩(%,	ggro Taras	ACS A LITHOLOGY	REKARKS
TACE TO			
. 0 3	3	Sand, clayey, red, (fill)	
3 21	18	Clay, slightly silty, yellow	
21 30	9	Sand, med. to fine, buff to white, contains	<u>a</u>
30 40	10	few small white shell fragments Sand, fine, same as above	Added 1-50# sack of lime
40 55	15	Sand, med., same as above	
55 60	5	Sand, coarse, grades to gravel, buff to whi	te
60 70	10	Sand, coarse to med., buff to white, about pea gravel	10%
70 80	10	Sand, coarse to med., buff to white, about pea gravel, increase in colored grain cont	
		to about 10%. Hit one or more thin streaks	of
		gray clay	
80 100	20	Sand, fine to coarse, buff to white. Looks	like Red coloration come
		a "red sand" but grains are 80% white, buf	
		lt. gray, and colorless with 20% colored g	rains silt or red silty
·		· · · · · · · · · · · · · · · · · · ·	clay
100 112	12	Sand, med. to coarse, red coloration decrea	ses
		with depth, 5 to 10% pea gravel, contains	a few
		small shell fragments	
112 118	6	Sand, very coarse, grades to pea gravel	Added 2-50# sacks o
			lime "Chatter"
118 121.5	3.5	Gravel, 3/8" to 1'~"	"Chatter" nok

FIRISIED 5/7/73 TOGGED BYW.P.Stilson DELLED by Holland Drilling DRILL HOLE 0-9 BEGUE: 5/3/73 Approx.

TOTAL DEPTH 461' LOCATION 10' North of Paint Shop (S. 002.71', E. 577.72'PlantGrid, ELEV. 33' Grade Bit: 41;" 3 Blade Drag Bit Mud: Lime and Bentonite Used When Reaming First 128' of Hole to 11 7/8"Dia

	"%C	iš		RECO	1 THG1.6GY	REKARKS
11.0	E.,[.	<u>TC</u>				
121.	5	128	6.5		Clay, gray to light bluish gray "Blue Clay"	T.D. for day May 4, 1972. Reamed
				,		11-7/8" hole to 128'. Installed 129' 6"
						casing. Pressure
			'			cemented casing in hole.
128		140	12		Clay, bluish gray, "Blue Clay"	Resumed drilling
						5/7/73. Only slight amount of
						cement at bottom of hole.
140		142	2		Sand, fine	QuestionableDid_ not see cutting
142		159	9		Clay, gray and bluish-green	
159		163	4		Sand, fine to very fine, buff to light gray	
163		171	8		Clay, color varies, 70% blue to bluish green,	
	2				30% olive	
171	50	181	10		Clay, about same as above except some gray or	
181	00	191	10		<pre>lt. bluish gray coming in Clay, about same as above except less olive.</pre>	
191	10	201	10		Clay, about same as above	Cutting somewhat
	8					slower; cuttings are finer and "less
			١			gritty"

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Cold Creek Plant, ks, Alabama

Explora y Boring

PEGUE 5/3/73 FIRISHED 5/7/73 LOGGED BYW.P.Stilson DRILLED BY Holland Drilling DRILL HOLE 0-9

Approx.

ELEV. 33' Grade TOTAL DEPTH 461' LOCATION 10' North of Paint Shop (S. 002.71', E. 577.72'PlantGrid)
Bit: 4'2" 3-Blade Drag Bit Mud: Lime and Bentonite Used When Reaming First 128' of Hole to 11 7/8"Dia

_ F∈€"A	FCCTAGE				RECOV-	1 TBUICLOOV	REKARKS
FBGi	TC	ness	ЕВУ	1 17HCLOGY	MERANG		
201	207	6		Clay, about same as above			
207	211	4		Sand, med. to fine	"Lively Sand" -		
					losing water		
211	221	10		Sand, med. to coarse but mostly coarse last 5',	Added 2-50# sacks		
	<u> </u>			grains clear, white and lt. gray, about 5%	of lime		
				colored grains with a few black grains			
221	228	7		Sand, fine to coarse (mostly coarse), grains white	Still losing water.		
				to light gray with about 3% colored grains	Added 2-50# sacks		
					of lime		
228	242	14		Clay, light bluish gray, gray, and blue	Brought up wood		
	-				fragment		
242	251	9		Clay, about same as above. Brought up a slight	Drilling harder.		
	ļ			amount of black material in the 245 to 248'	Brought up a few		
				range. Material is soft, silt sized, and has	wood fragments.		
		_		no odor. (Soft lignite?)			
.251	271	20		Clay, mostly blue but some bluish gray and bluish	Drilling easier aga		
	-			green with a slight amount of gray			
271	. 286	15	ļ	Sand, fine to medium	Very little recover		
					of sand in cutting Returns mostly cla		
286	291	55	<u> </u>	Clay, bluish gray to bluish green with increasing	Kerding mostry C19		
				amounts of medium to dark gray			
291	295	4		Clay, same as above	Drilling hard		
295	301	6		Clay, same as above, gray tones increasing			
301	311	10		Clay, same as al `	Drilling 1		
31.	341	30	1	Clav, more blui, and less grav	1		

with their name on without out to tage 4 of 4

Cold Creek Plant, Bucks, Alabama

Exploratory Boring

FERRISED 5/7/73 FURTHER 5/7/73 FURTHER BYW.P.Stilson DRILLED BY Holland Drilling DRILL HOLE 0-9

Approx.
ELEV. 33' Grade 'CCTAL DEPTH 461' | LCCATION 10' North of Paint Shop (S. 002.71', E. 577.72'PlantGrid)

Bit: 4½" 3-Blade Drag Bit Mud: Lime and Bentonite Used When Reaming First 128' of Hole to 11 7/8"Dia

FC TA	GE.	SHICK -	RECES	1.60016.1.4057	REEARKS
Pacif	10	EESS	E 17	1 (THC1 CGV	UM:MIMO
341	351	10		Clay, same as above	Drilling hard
351	366	15		Clay, same as above	Drilling easier
366	372	6		Sand	Very little recovery
					of sand in cuttings
					returns are mostly
					clay
372	381	9		Clay, same as clay above sand	The state of the section of the sect
381	392	11		Clay, about same as above	Drilling hard
392	396	4		Sand	About 4' of sand -
					very little recover
396	401	5		Clay, probably thin interbedded seams of mudstones	Drilling very hard
				or siltstones	399' to 401'
401	431	30		Clay	Drilling very hard;
		ļ			cuttings are mostly
					clay.
431	441	10	ļ	Clay	Drilling easier
	1				below 433'
441	451	10		Clay, increasing in blue-green tones, decreasing	
]C	D			in gray tones	
451	O 461	10	- ·	Clay, about same as above	
T.D. C	⊃ 461			In Clay	No more drill steel.
C	D				May 8, 1972. Reamed
				pipe in 21'sand at 207 to 228'. Came out of	5½" hole from 128'
	ი			hole with 4" galv. pipe.)	to 228'. Tostalled
					20' of 4' ted PV

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BEGUN 34 FINISHED 4/6/84 LOGGED BY W.P.Sti DRILLED BY Holland Well Company DRILL & CNAM-34 GRADE
ELEV. 42.2' TOTAL DEPTH 128.5' LOCATION Stauffer Grid Coordinates 5.2480+5' E.1890+5'

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

5-3/4" Dray Bit 10' to T.D. Mud: Johnsons Revert

	5-3/4"	Drag Bi	t 10' to T.D. Mud: Johnsons Revert	
\- <u></u>	l' N GE	THICK-		
FROM	ТО	NESS	LITHOLOGY	REMARKS
0	8	8	Clay, stiff, red brown to light yellow brown.	
8	10	2	Clay, fairly stiff, silty to sandy, light yellow brown to red brown.	
10	16	6	Clay, slightly stiff, silty to sandy (grades into sand), tan, light reddish	
			brown and yellow brown.	
16	30	14	Sand, fine to coarse, with minor gravel, buff to light gray.	"Good chatter"
30	44	14	Sand, fine to coarse, buff to light gray, contains thin streaks of light	
			gray clayey silt.	
44	50	6	Clay, soft, Silty to sandy, light gray, light reddish brown, cream, and	
			trace of reddish brown.	
50	60	10	Clay, fairly soft, silty to sandy, trace of gravel, light yellow-brown,	
			cream and light gray and red mottled	
60	75	15	Clay, fairly soft, silty to sandy, light gray, minor medium gray, traces	
:	'		of black silty sand(?), also yellow brown and reddish brown clays.	
75	80	5	Sand, fine to coarse, with minor gravel, also with clay streaks, light	
10	ļ		gray and light yellow brown.	
0 80	95	15	Sand, fine to coarse (mostly medium) with 2 to 4% gravel + 1/4", buff to	
O			light gray, thin clay streaks.	
95	101	6	Clay, moderately stiff to soft, light gray.	
Ñ		1		

benegic rant, AXIS, Alabama

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BEGUN 4/6/84 FINISHED 4/6/84 LOGGED BY W.P. Stilson DRILLED BY Holland Well Company DRILL HOLE CNAM-34

GRADE

ELEV. 42.2' TOTAL DEPTH 128.5' LOCATION Stauffer Grid Coordinates 5. 2480+5'

E. 1890±5'

9-3/4" Drag Bit 0 to 10'; Bits:

Drilled on Courtaulds North America Property

5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

F00'	3-3/4 LVGE	<u> </u>	t 10 to 1.0. Mud; Johnsons Revert	
FROM	ТО	THICK- NESS	L I THOLOGY	REMARKS
101	110	9	Sand, fine to coarse (mostly medium), tan to light gray.	
110	120	10	Sand, fine to medium with 5% coarse + 1/8", tan to light gray, thin	
			clay streaks possible.	
120	127	7	Sand, fine to coarse with 3 to 5% gravel + 1/4", tan to light gray,	
			thin clay streaks possible.	
127	128.5	1.5	Clay, stiff, light bluish gray.	
T.D.	128.5		In light bluish gray clay.	
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LeMoyne Plant Axis, Alabama

BEGUN / 34 FINISHED 4/6/84 LOGGED BYW.P.Sti DRILLED BY Holland Well Company DRILL 1 ; CNAM-33 GRADE

ELEV. 42.2' TOTAL DEPTH 96' LOCATION Stauffer Grid Coordinates 5.2470+5' E. 1885±5' Bits: 9-3/4" Drag Bit 0 to 10';

Drilled on Courtaulds North America Property

5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

	3-3/4	17.49	to to t.p. Mid. Johnsons Rever	····
F00	TAGE	титск-		
FROM	ТО	NESS	1'1.HOTOCA	REMARKS
0	5	5	Clay. stiff. reddish brown to light yellow brown.	
5	8	3	Clay, stiff, tan to light yellow brown and light gray and red mottled.	
8	10	2	Clay, fairly stiff, silty to sandy, light yellow brown to reddish brown.	
10	17	7	Clay, slightly stiff, silty to sandy, (grades into sand), tan, light	
			brown and light reddish brown.	
17	30	13	Sand, fine to coarse, with minor gravel, buff to light gray	"Good chatter"
30	44	14	Sand, fine to coarse, buff to light gray, with thin streaks or light gray	
			clayey silt, also has streaks of cream clay.	1
44	50	6	Clay, soft, silty to sandy, light gray, light reddish brown, cream, and	
			trace of reddish brown.	
50	60	10	Clay, fairly soft, silty to sandy (with trace of gravel), light yellow	
			brown, cream, light gray with red mottled.	
60	75	15	Clay, fairly soft, silty to sandy, light gray, grace of medium gray, trace	
CI			of black silty sand(?), also yellow brown and reddish brown clays.	
10 75 O —	80	5	Sand, fine to coarse, with minor gravel, also with clay streaks, light	
0			gray and light yellow brown.	
<u>80</u>	90	5	Sand, fine to coarse, with 2 to 3% gravel + 1/4", buff to light gray,	"Fair to good chatter"
M			contains thin clay streaks.	
		I		

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BEGUN 4/6/84 FINISHED 4/6/84 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-33

GRADE

FINISHED 4/6/84 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-33

GRADE

FINISHED 4/6/84 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-33

Bits: 9-3/4" Drag Bit 0 to 10;

Stauffer Grid Coordinates 5.247015' E. 188545'

Drilled on Courtaulds North America Property

5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

	5 - 3/ 4"	Drag Bit	t 10° to T.D. Mud: Johnsons Revert	
	FOOTAGE			
FROM	TO	NESS	LITHOLOGY	REMARKS
90	94.5	4.5	Sand, fine to coarse, with 3 to 5% gravel + 1/4", buff to light gray.	"Heavy chatter"
94.5	96	1.5	Clay, fairly stiff, light gray.	
T.D.	96		In light gray clay.	
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LeMoyne Plant Axis, Alabama E CNAM-32 84 FINISHED 4/5/84 LOGGED BY W.P.Stil IDRILLED BY Holland Well Company DRILL I BEGUN ... GRADE ELEV. 35.4' TOTAL DEPTH 121' LOCATION Stauffer Grid Coordinates S.2550 E. 1560 Drilled on Courtaulds North America Property 9-3/4" Drag Bit 0 to 10'; Bits:

5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

	5-3/4	Ding Di	10 to r.b. Mud: Johnsons Revert	
	r a ge '	THICK-	•	
FROM	TO	NESS	LITHOLOGY	REMARKS
0	6	6	Clay, fairly stiff, reddish brown.	
6	8	2	Sand, clayey, light reddish brown to buff.	
8	12	4	Sand, fine to medium buff to tan.	
12	19	7	Sand, fine to coarse, with minor gravel, buff to tan, yellow-brown	"Formation taking fluid"
			staining on 1% of grains	
19	20	1	Clay, slighlty stiff, light gray and light reddish brown, trace of yellow	
			brown.	
20	24	4	Clay, fairly soft, light gray and tan, with yellow-brown clayey silt.	1
24	26	4	Sand, fine to coarse, light yellow brown.	"Drills smooth"
26	30	4	Sand, fine to coarse, light yellow brown.	"Formation taking fluid
30	40	10	Sand, fine, buff to tan, with thin clay streaks light gray, buff, and	
			reddich tan clay m minor yellow-brown staining.	
40	55	15	Sand, fine to coarse, with 3 to 5% gravel + 1/4", buff to tan, mixed with	"Drilled smooth" "Like a soft clay or
4	,		clay streaks, buff and light yellow-brown, (clay increases 50-55')	fine sand"
in 55	60	5	Clay, soft, silty to sandy, buff to yellowish tan, with gravelly streaks.	"A lot of foam on mud"
0 60	73	13	Clay, soft, buff to tan with minor light gray, thin sand streaks probable.	
<u></u>	80	10	Sand, fine to coarse, buff to tan.	
M <u>80</u>	, , , , , , , , , , , , , , , , , , ,	10	Sand, fine to coarse, buff to tan, we'very light yellowish clay(?) streaks.	
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nerwyne i rame, AXIS, Alabama

BEGUN 4/5/84 FINISHED 4/5/84 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-32

GRADE
ELEV. 35.4' TOTAL DEPTH 121' LOCATION Stauffer Grid Coordinates S. 2550. E. 2560.

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

	5-3/4"	Drag Bil	t 10' to T.D. Mud: Johnsons Revert	
FOOTAGE		THICK-		
FROM	то	NESS	LITHOLOGY	REMARKS
90	100	10	Sand, fine to coarse, with 3 to 5% gravel + 1/4", yellowish as above, possible clay streak @ 94', also traces of light gray and cream clay.	Fluid gone; stopped. Mixed 1 sack Revert.
100	120	20	Sand medium to very coarse, with 5% pea gravel + 1/4", yellow-brown	"Heavy chatter"
	ļ		staining on 1 to 2% of grains.	
120	121		Clay, stiff, yellowish silty clay on top of bluish clay.	
T.D.	121		In stiff bluish clay.	
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FINISHED 4/5/84 LOGGED BY W.P.Stil DRILLED BY Holland Well Company DRILL H

CNAM-31

GRADE TOTAL DEPTH 87.5' LOCATION Stauffer Grid Coordinates 5.2565. E. 1565. ELEV. 35.2'

Drilled on Courtaulds North America Property Bits: 9-3/4" Drag Bit 0 to 10';

5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

BEGUN 4/

	3-3/4	Dray br	t 10 to 1.D. mud: Johnsons Revert	
FOO	TAGE	THICK-		
FROM	то	NESS	LITHOLOGY	REMARKS
0	6	6	Clay, stiff, light reddish brown to yellow-brown with minor gray.	
6	8	2	Sand, clayey, light reddish brown to buff.	
8	12	4	Sand, fine to medium, buff to tan.	
12	18	6	Sand, fine to coarse, buff to tan with yellow-brown staining on about 1%	
			of grains.	
18	24	6	Clay, fairly soft, light gray with red (mottled), tan and buff.	
24	30	6	Sand, fine to coarse, tan to light yellow-brown.	
30	40	10	Sand, fine to medium, with minor coarse, buff to tan, with minor amount	
			of yellow-brown staining, thin clay streaks.	
40	50	10	Sand, fine to coarse, with 2 to 3% gravel + 1/4" thin clay streaks possible.	
50	60	10	Clay, fairly soft, tan and light gray, traces of medium gray clayey silt,	
			also some yellow brown clay, yellow brown staining ("seams") in tan	
		•	and gray clay.	
\D 60	70	10	Clay, silty, fairly soft, tan and light gray with small amount staining as	"Drilled soft and smooth
ю О ===			above, thin sand streaks possible.	
O 70	76	6	Clay, as above, grades into sand.	
	80	4	Sand, fine to coarse, with 2 to 3% gravel + 1/4", buff to tan	"Fair chatter"
№ 80	R7.5	7.5	Sand, fine to coarse, with minor gray contains thin clay streaks.	
+ 5				

nemovine ream, AXIS, Alabama

GRADE

3/4/84 FINISHED 3/4/84 LOGGED BY W.P. Stilson PRILLED BY Holland Well Company DRILL HOLE CNAM-30

38.7' TOTAL DEPTH 132' LOCATION Stauffer Grid Coordinates S. 2730'

ELEV.

Drilled on Courtaulds North America Property

E. 1155'

Bits:

9-3/4" Drag Bit 0 to 10'; 5-3/4" Drag Bit 10' to T.D.

Mud:

Jo	hnsons	Reve	rt

FOC	TAGE	THICK-		
FROM	TO	NESS	LITHOLOGY	REMARKS
0	14	14	Clay, silty, traces of sand, light reddish brown to tan, and gray-red	
		<u> </u>	mottled clay.	
14	20	6	Clay, silty to sandy (grading into sand), softer than above, otherwise	
			the same.	
20	30	10	Sand, fine to medium (mostly fine) tan to light reddish brown, traces	
			of yellow brown staining.	
30	39	9	Sand, fine to coarse, buff to tan, thin clay streaks possible.	
39	50	11	Clay, slightly stiff, light gray with traces of red, also some cream to	:
			very light yellowish clay and minor tan to light brown clay (must be	
	İ		intermixed with sand streams), buff to white.	
50	60	10	Sand, fine to coarse, with 5 to 10% gravel + 1/8" (must be intermixed	"Drilled like a clay with sand streaks"
			with light gray and cream clay streaks).	
60	80	29	Sand, fine to medium with minor coarse, buff to light gray (clay streaks	"Light chatter"
<u> </u> 	,		possible).	
∑ <u>80</u>	90	10	Sand, fine to coarse, with 2 to 4% gravel + 1/4", tan to light gray,	
) 			brown staining on 1 to 2% of grains.	
⊃ 90	100	10	Sand, medium to coarse, with 3 to 5% gravel + 1/4", buff to tan, brown	"Chatter"
γ			and yellow brown staining on 2 to 3% of grains, traces of clay,	
:		1		ļ

LeMoyne Plant Ris, Alabama

BEGUN 3, 34 FINISHED 3/4/84 LOGGED BYW.P.Still PRILLED BY Holland Well Company DRILL HO.

RADE

ELEV. 38.7' TOTAL DEPTH 132' LOCATION Stauffer Grid Coordinates S. 2730' E. 1155'

Drilled on Courtaulds North America Property

Hits: 9-3/4" Drag Bit 0 to 10';

31081			t 10' to T.D. Mud: Johnsons Revert	
	TAGE	THICK-		
FROM	ТО	NESS	T11HOT'X!A	REMARKS
100	110	10	Sand, fine to coarse with trace of gravel, tan to light brown (traces of clay)	"Heavy chatter"
110	120	10	Sand, coarse to very coarse (minor fine), buff to tan, brown staining on	
			less than 1% of grains.	
120	130	10	Sand, coarse to very coarse (minor fine to medium but more than above), buff.	
130	132	2	Clay, stiff bluish gray, with trace of orange silty clay on top.	
T.D.	132		In stiff bluish gray clay.	
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151721111	1/2/01	FINISHED 4/3/84		YW.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-29
BEGUN	47 37 64	F INISHED 4/3/04	regional D	TW.1. Stillsoid Ringer Bi notified well company Dieter Hole
GRADE				
ELEV.	33.4'	TOTAL DEPTH 35'	LOCATION	Stauffer Grid Coordinates S. 2740. E. 1170.
Bits:	9-3/4" 1	Drag Bit 0 to 10':		Drilled on Courtaulds North America Property

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Cou. 5-3/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

130.05	3-3/4	Trag bit	t 10 to 1.D. Mud: Johnsons Revert	1
FROM	TAGE TO	THICK- NESS	LITHOLOGY	REMARKS
0	8	8	Clay, silty, moderately stiff, tan to light brown, minor gray.	
3	12	4	Clay, silty to sandy, soft, light yellow brown.	
12	15	3	Sand, fine, white to light yellowish tan.	
15	20	5	Clay, sandy, fairly soft, light reddish brown to light yellow brown.	
20	30	10	Sand, fine, buff to tan, with small amount of yellow brown silt to clayey	
			silt.	
30	38	8	Sand, fine to medium (mostly fine), tan, with traces of yellow brown	"Light chatter"
			staining.	
38	50	12	Clay, slightly stiff, light gray with traces of red, also minor buff to	
			tan silty to sandy clay.	
50	60	10	Sand, fine to coarse, with traces of gravel, buff to light gray.	
60	76	16	Sand, fine to coarse, buff to light gray, possible clay streaks.	
7.6	Ω D 77	1	Clay(?), light gray.	
77	⊃ , 80	3	Sand, fine to medium, buff to light gray.	
80 4	○ - 84	4	Sand, fine to coarse, buff to light gray.	
84 1	∽ 85	1	Clay, light to medium gray.	
T.D.	85		In gray clay.	

Lemoyne Plant Axis, Alabama

BEGUN 2 '84 FINISHED LOGGED BYW.P.Stin OPRILLED BY Holland Well Company DRILL 1 2 CNAM-28 GRADE

ELEV. TCTAL DEPTH LOCATION Stauffer Grid Coordinates S, 2609.8±2 E, 2103.±3

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

·		mag bit	. 10 to T.D. Mud: Johnsons Revert	······································
ļ	PAGE	THICK-		DEMARKS
FROM	то	NESS	LITHOLOGY	REMARKS
0	11	11	Silt, sandy, brown and yellow brown (top soil).	
11	5	4	Clay, silty, with trace of sand, stiff, light yellow brown to red brown.	
5	10	5	Clay, silty, with minor sand, softer than above, gray and red mottled.	
10	20	10	Sand, medium to coarse, buff to light yellow brown (most be a layer of high	
			color intensity yellow-orange silty clay above 15') 3 to 5% yellow-orange	
			staining on grains.	
20	30	10	Sand, medium to coarse, tan to light yellow brown, 2 to 3% staining.	Formation taking fluid.
30	38	8	Sand, fine to medium with minor coarse, buff.	Formation taking fluid.
38	44	6	Clay, slightly stiff, light gray and red, also yellow-brown.	
44	50	6	Sand, fine to medium, with traces of coarse, white to buff.	"Drilled soft"
50	60	10	Sand, fine to medium (mostly fine), light gray to white, thin streaks of	
C	.7		light gray clay likely.	
60 ⊂	⊃ <u>64</u>	4	Sand, fine to coarse, buff.	
64	69	5	Sand, medium to coarse, with 1 to 2% gravel + 1/4", buff to tan.	
69	69.5	0.5	"Sand rock" (less than 6"), medium to coarse sand and gravel, weakly	
~)		cemented, yellow brown to rust.	
69.5	70	0.5	Clay, soft, cream, with yellow brown staining.	
70	71	1	Clay, slightly firm, medium gray.	
	•			

BEGUN 2/6/84 FINISHED 2/6/84 LOGGED BY W.P.Stilson PRILLED BY Holland Well Company DRILL HOLE CNAM-27
GRADE

ELEV. 44' TOTAL DEPTH 132' LOCATION Stauffer Grid Coordinates S. 2690. E. 1780.

Bits: 9-3/4" Drag Bit 0 to 10; Drilled on Courtaulds North America Property

FOOT	rage	THICK-		
FROM	то	NESS	LITHOLOGY	REMARKS
0	0.5	0.5	Silt, clayey, brown (topsoil).	
0.5	5	4.5	Clay, silty, fairly firm, light yellow brown with some reddish brown.	
5	10	5	Clay, silty to sandy, softer than above, light reddish brown.	
10	17	7	Clay, sandy, grades into sand, light reddish brown to light yellow brown.	
17	37	20	Sand, fine to coarse, mostly medium with traces of gravel, tan to light	
			_ yellow brown.	
37	39	2	Clay, light gray.	
39	45	6	Sand, fine to medium, tan.	
45	61	16	Clay, fairly firm, light gray, cream, and light yellow brown.	
61	71	10	Clay, fiarly firm, lgiht gray, tan, minor sand streaks possible.	
71	80	9	Clay, fairly firm, traces of gravel, light gray, tan with minor light	
			reddish brown.	
80	90	10	Sand, fine to medium, with traces of gravel, also with some clay streaks,	
	1 		(80% + sand - clay less than 20%).	
90 ℃	. 100	10	Sand, fine to coarse, with gravel, mixed with clay streaks (50% sand -	
			clay about 50%).	
100 ⊆	2 110	10	Sand, fine to coarse, with gravel, mixed with medium gray to light brownish	"Good chatter"
- ~	·····		gray clay streaks (90%+ sand - clay less than 10%)	

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BEGUN 7 34 FINISHED 2/6/84 LOGGED BY W.P.St. INDRILLED BY Holland Well Company DRILL E CNAM-27

GRADE
ELEV. 55' TOTAL DEPTH 132' LOCATION Stauffer Grid Coordinates S. 2690 E. 1780

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

	6-1/4"	Drag Bit	10' to T.D. Mud: Johnsons Revert	· · · · · · · · · · · · · · · · · · ·
FOO FROM	TAGE TO	THICK- NESS	LITHOLOGY	REMARKS
110	120	10	Sand, fine to coarse, with about 5% gravel + 1/4", light gray	"Good chatter"
				dood chacter
120	131	11	Sand, fine to coarse, with minor gravel, tan to light brown, yellow brown	
	<u> </u>		staining on 5 to 10% of grains.	
131	132	1	Clay, thin layer of yellow-brown clayey silt on top of light bluish gray	
	ļ		clay.	
T.D.	132		In stiff bluish clay.	
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BEGUN 2/8/84 FINISHED 2/8/84 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-26

GRADE
ELEV. 36.5' TOTAL DEPTH 126' LOCATION Stauffer Grid Coordinates 5.2500. E. 1725.

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

	0-1/4	Drug Dr	10 to 1.b. mud: Johnsons Revert	· · · · · · · · · · · · · · · · · · ·
FOO	TAGE	титск-		
FROM	ТО	NESS	LITHOLOGY	REMARKS
0	11	1	Silt, sandy, brown, (top soil).	
11	5	4	Clay, stiff, reddish brown, minor gray.	
5	8	3	Clay, grades to sand, light reddish brown and light yellow brown, (fine to	
		 	medium sand).	
8	20	12	Sand, fine to coarse, tan to light yellow brown, minor streaks of light	
			_ gray and cream clay.	
20	35	15	Sand, fine to coarse, buff to tan, thin clay streaks, traces of yellow-	
		·	brown staining.	,
35	39	4	Clay, fairly firm, light gray with traces of red.	
39	50	11	Sand, fine, silty, with thin clay streaks, tan to light yellow brown	"Drilled soft"
50	57	7	Sand, fine, silty, with thin clay streaks.	
57	60	3	Clay, fairly firm, light gray with traces of red, minor light yellow brown.	
60	70	10	Sand, fine to medium, buff to yellow tan.	Formation taking fluid "Chatter"
70	78	8	Sand, fine to coarse (mostly medium), buff to tan	Formation taking fluid "Chatter"
78	50 72 <u>84</u> 150	6	Sand, medium to coarse, buff to light gray	Formation taking fluid Added 1 sack Revert.
84	□ 90 □	6	Clay, fairly stiff, light gray, cream, minor light yellow brown, trace	
	O		of red.	
1	- 100	10	Sand, fine to medium with minor coarse, buff to light yellow tan, staining	Taking fluid. "Chatter"
1 ,			on 10 of ansing	1

behovine riant Axis, Alabama

184 FINISHED 2/8/84 LOGGED BY W.P.Sti IDRILLED BY Holland Well Company DRILL

3 CNAM-26

BEGUN GRADE

ELEV. 36.5' TOTAL DEPTH 126/ LOCATION Stauffer Grid Coordinates S. 2500. E. 1725.

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

FOO	TAGE	THICK-		
FROM	то	NESS	LITHOLOGY	REMARKS
100	105	5	Sand, as above, but with yellow-brown staining 2 to 3% of grains.	
105	107	2	Clay, fairly firm, light gray.	
107	124	17	Sand, medium to coarse, with 5% gravel + 1/4", buff to light gray, tan	
			clay streaks possible (120-124').	
124	126	2	Clay, 1/2' of yellow-brown clayey silt on top of stiff bluish gray clay.	
T.D.	126		_ In stiff bluish clay.	
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LeMoyne Plant, Axis, Alabama

BEGUN 2/2/84 FINISHED 2/2/84 LOGGED BY W.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CHAM-25
GRADE
ELEV. 41.0' TOTAL DEPTH 131' LOCATION Stauffer Grid Coordinates S. 2795 E.1450.
Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

	0-1/4	Diag bit	10 to T.D. Mud: Johnsons Revert	
F00	TAGE	титск-		
FROM	то	NESS	LITHOLOGY	REMARKS
0	0.5	0.5	Silt, sandy, brown, (topsoil).	
0.5	11_	0.5	Clay, silty, light yellow born.	
1	10	9	Clay, firm, yellow brown and reddish brown to 4', gray and red mo-tled	
	<u> </u>		4' to 10'.	
10	21	11	Clay, firm to 16', moderate 16' to 21', light gray to white, traces of fine	
			white sand.	
21	30	9	Sand, fine to coarse, with traces of gravel, buff to tan, some yellow	·
			brown staining on grains, (trace of clay).	
30	41	9	Clay, fairly soft, light gray, light yellow brown, and light red-brown, with	
			streaks of fine to coarse sand.	
41	55	14	Clay, fairly stiff, light gray to white, with traces of red mottling, minor	
			sand streaks, trace of gravel.	
55 :	64	9	Clay, soft, light gray, with minor sand and gravel streaks, gravel is buff	
I	ر ر		to white with traces of yellow brown staining.	
64 L	Ω 81	17	Sand, fine to coarse, mostly medium, with minor gravel, light gray to buff.	Formation taking fluid. "Light chatter"
		20	Sand, fine to coarse (mostly medium coarse) with 5% gravel + 1/4", buff	
	<u> </u>		to tan, traces of yellow brown staining on 5% of grains, clay streak	
	۰		likely at 93 _ 101, other clay promible.	
I	1	1	•	

Lemoyne Plant. Axis, Alabama

Tayo & UT .

BEGUN 2 34 FINISHED 2/2/84 LOGGED BY W.P.Sti. DRILLED BY Holland Well Company DRILL B CNAM-25
GRADE
ELEV. 41.0' TOTAL DEPTH 131! LOCATION Stauffer Grid Coordinates S. 2795. E. 1450.

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

	6-1/4"	Drag Bi	t 10' to T.D. Mud: Johnsons Revert	
FOO	TAGE	THICK-		
FROM	ТО	NESS	LITHOLOGY	REMARKS
101	130	29	Sand, medium to coarse, with 5 to 10% gravel + 1/4", light gray to buff,	
			yellow brown staining on about 5% of grains, thin (1' or less) clay	
			streak at 114'.	
130	131	1	Clay, stiff, bluish gray (thin layer of yellow brown silt and sand rock	
			on top of blue).	
T.D.	131		_ In stiff bluish clay.	
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BEGUN 2/7/84 FINISHED 2/7/84 LOGGED BYW.P.StilsonDRILLED BY Holland Well Company DRILL HOLE CNAM-24

 \mathbf{GRADE}

ELEV. 36.5' TOTAL DEPTH 127' LOCATION Stauffer Grid Coordinates S, 2610.

Е. 1395.

Bits: 9-3/4" Drag Bit 0 to 10'; Drilled on Courtaulds North America Property

FOOTAGE		THICK-	r 10 to 1.D. Hud. Johnsons Revert	
FROM	то	NESS	LITHOLOGY	REMARKS
0	1	11	Silt, slightly sandy, brown, (top soil).	
1	6	5	Clay, fairly stiff, light reddish brown and light yellowish brown.	
6	10	4	Sand, fine to medium, buff to light yellow brown.	
10	20	1	Clay, silty to sandy, slightly soft, light gray and tan, trace of light	
			yellow-brown.	
20	36	16	Sand, fine to coarse, (mostly medium) buff to tan, with light yellow brown	
			staining on 2 to 5% of grains, (minor light gray clay).	
36	39	3	Clay, moderatly stiff, light gray and light brownish red.	
39	50	11	Sand, fine to medium, with trace of coarse, buff, 40 to 50%/mised with	
			50 to 60% soft clay, light gray and light brownish red.	
50	60	10	Sand, fine to medium with minor coarse, buff, with light yellow brown	
			staining on 1 to 2% of grains, contains, streaks of soft gray clay	
			(80% san 20% clay).	
60 ∴	- 70	10	Sand, fine to coarse, buff to tan, yellow brown staining on 1% of grains.	
70	80	10	Sand, medium to coarse, buff to light gray, traces of clay(?) may be from	
ō			uphole.	
<u>80</u>	100	20	Sand, medium to coarse, (minor fine) buff to light gray, traces of clay,	"Heavy chatter"
Μ			yellow brown staining on 1% grai	

LeMoyne Plant - Axis, Alabama Taye L U L

34 FINISHED 2/7/84 LOGGED BYW.P.St DIDRILLED BY Molland Well Company DRILL BEGUN . E CNAM-24

GRADE

TOTAL DEPTH 127' LOCATION Stauffer Grid Coordinates S, 2610 ELEV. <u>36.51</u>

9-3/4" Drag Bit 0 to 10';

Drilled on Courtaulds North America Property

		t 10' to T.D. Mud: Johnsons Revert	
AGE	титск-		
ТО	NESS	LITHOLOGY	REMARKS
110	10	Sand, fine to coarse (mostly coarse), 2 to 5% gravel + 1/4", buff to	
		light gray, rest as above.	"Heavy chatter"
125	15	Sand, fine to coarse (mostly coarse), with 5 to 10% gravel + 1/4", buff	"Heavy chatter"
		to light gray, yellow brown staining on about 3% of grains.	
127	2	Clay, stiff, 1' of yellow brown clayey silt on top of light bluish gray.	
127		In stiff bluish clay.	
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	110 125 127	TO NESS 110 10 125 15 127 2	TO NESS LITHOLOGY 110 10 Sand, fine to coarse (mostly coarse), 2 to 5% grayel + 1/4", buff to light gray, rest as above. 125 15 Sand, fine to coarse (mostly coarse), with 5 to 10% gravel + 1/4", buff to light gray, yellow brown staining on about 3% of grains. 127 2 Clay, stiff, 1' of yellow brown clayey silt on top of light bluish gray. 127 In stiff bluish clay.

Derwyne Liant, Mis, Midband

BEGUN	2/9/84	FINISHED 2/9/84	LOGGED BY	W.P.StilsonDRILLED BY Molland Well Co	mpany DRILL	HOLE	CNAM-23
GRADE							
ELEV.	33.9'	TOTAL DEPTH 119!	POCYLION	Stauffer Grid Coordinates 5.2331 2	E. 2366.9		
Bics:	9-3/4" [Oraq Bit 0 to 10';		Drilled on Courtaulds North America	Property		-

	0-1/4	1	to to r.b. Mud: Johnsons Revert	
FOO	TAGE	титск-		
FROM	ТО	NESS	LITHOLOGY	REMARKS
		· · · · · · · · · · · · · · · · · · ·		
0	5	5	Clay, fairly stiff, reddish brown and light yellowish brown.	
5	7	2	Sand, fine to coarse (mostly coarse), tan to light yellow-brown, yellow-	
			brown staining on 5% of grains.	
7	10	3	Clay, silty to sandy, fairly soft, light yellow brown, minor red-brown	
			and gray with red mottled.	
10	20	10	Sand, fine to coarse, tan to light brown, minor clay streaks, soft,	
			light red and yellow-brown.	
20	30	10	Clay, soft, tan to light reddish brown (60%) with (40) sand, fine to medium.	Drilled soft.
30	37	7	Clay, fairly firm, light gray and red mottled.	
37	40	3	Sand, clayey, soft, light yellow brown to reddish brown.	
40	50	10	Clay, silty to sandy, drilled soft.	
50	62	12	Clay (est. 80% clay/20% fine sand), silty to sandy, fairly firm, cream	
			and light gray clay, minor red brown and yellow brown.	
62	67 70	8	Sand, fine to medium, with minor coarse, buff to light gray, minor light	
	0.5		reddith brown, thin clay streaks.	
70	76	6	Sand, fine to coarse (mostly medium), buff to light gray, clays are cream	
			and light gray, thin clay streaks.	
76	ر س	4	Sand, coarse to gravel - 1/4".fair" heavy yellow-brown staining on gravel.	
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LeMoyne Plant Axis, Alabama

BEGUN 34 FINISHED 2/9/84 LOGGED BYW.P.Stin DRILLED BY Holland Well Company DRILL 1. CNAM-23 GRADE

ELEV. 33.9' TOTAL DEPTH 119' LOCATION Stauffer Grid Coordinates S. 2331.2 E. 2366.9

Bits: 9-3/4" Dray Bit 0 to 10'; Drilled on Courtaulds North America Property

FOO	TAGE	титск-	TO CO 1.D. Mad. Johnsons Revert	
FROM	ТО	NESS	1.1THOLO3Y	REMARKS
90_	95	5	Sand, fine to coarse, with about 5% gravel + 1/4" buff to light gray.	
95	100	5	Clay, slightly firm, light gray and cream.	
100	110	10	Sand, fine to medium, with minor coarse, tan to light brown.	
110	117	7	Sand, fine to coarse, with 5% gravel + 1/4" (mostly coarse sand and	
	ļ	.	gravel) light gray.	
117	119	2	Clay, fairly stiff, 1' of yellow brown to orange clayey silt on stiff	
			bluish gray clay.	
T.D.	119	-	In stiff bluish clay.	
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BEGUN 1/31/84 FINI	SHED <u>1/31/84</u> LOGGED B	YW.P.StilsonDRILLED BY Hollan	d Well Company	DRILL HOLE 0-75
GRADE				
ELEV. 34.3 TOTA	M. DEPTH <u>125'</u> LOCATION	Stauffer Grid Coordinates	S, 2537.8 E	. 1270.7

Bits: 9-3/4" Drag Bit 0 to 10'; 6-1/4" Drag Bit 10' to T.D. Mud: Johnsons Revert

FOO	TAGE	THICK-	That yourself	
FROM	TO	NESS	LITHOLOGY	REMARKS
00	3	3	Clay, silty to sandy, reddish brown (fill),	
3	10	7		
			light red-brown.	
10	17	7	Clay, sandy to silty, fot to moderately stiff, reddish brown.	
17	20	3	Sand, fine, firm, buff to tan,	
20	33			Formation taking fluid Added 1 sack Revert
33	40	7	Sand, medium to fine, with minor coarse sand and gravel, also minor	Drilled soft
			streams of light gray, and light yellow-brown clay.	
40	50	10	Sand, medium to coarse, buff to tan, with streaks of light gray, dark	
			gray, and light yellow-brown clay.	
50	60	10	Sand, fine to coarse with minor gravel, mixed with medium to dark gray	
	<u> </u>		clay, minor tan, and traces of yellow-brown clay.	
60	80	20	Sand, fine to coarse, with minor gravel, buff to tan.	Formation taking fluid "Light chatter"
80	90	10	Sand, same as above, with thin streaks of cream, light gray, and medium	Drilled firm.
L	i) 		gray clay	
90 C	⊃ <u>100</u>	10	Sand, fine to coarse, with 5 to 10% gravel + 1/4", clay streaks as above	Drilled firm.
100	⊃ <u>124</u>	24	Sand, fine to coarse, mostly coarse sand and 10% gravel + 1/4", tan to buff	"lleavy chatter"
124 N	O 195		Clay, stiff, light bluish gray.	
	-		The state of the s	

rage 1 of Z

LeMoyne Plant Axis, Alabama

4 FINISHED 2/1/84 LOGGED BYW.P.St. DIDRILLED BY Holland Well Company DRILL

E 0-76

GRADE

ELEV. 31.7' TOTAL DEPTH 114' LOCATION Stauffer Grid Coordinates S. 2309.5

E. 2097.

Bits: 9-3/4" Drag Bit 0 to 10';

6-1/4" Drag Bit 10' to T.D.

Mud: Johnsons Revert

FOC)T A GE	THICK-		
FROM	TO	NESS	LITHOLOGY	REMARKS
0	12	12	Clay, silty to sandy, firm, reddish brown to light yellowish brown.	
12	18	6	Sand, fine to coarse, with minor gravel, tan to buff.	
18	29	11	Clay, silty to sandy, firm, light gray to buff, intermixed sand streaks	
			possible.	
29	39	10	Clay, silty to slightly sandy, firm, buff intermixed sand streaks possible.	
39	49	10	_Clay, silty to sandy, white to cream and light grays with intermixed fine	
			to coarse sand.	
49	53	4	Clay, sandy, gets more sandy with depth, buff to light gray, with fine	Grades into sand.
			white sand	
53	59	6	Sand, fine to coarse, with minor gravel, light gray to buff.	"Light chatter"
59	71	12	Sand, fine to medium with minor coarse sand and gravel, white to buff,	
			with light gray clay streaks. At 71' hit a hard zone, probably thin	
	,	<u>'</u>	"sand rock" yellow-brown silty clay on dark gray clay.	
½ 71 ₩ <u>71</u>	73	2	Clay, firm, dark gray.	2 to 3' of firm clay.
73	79	6	Sand, fine to coarse.	
O 79	86	7	Sand, fine to coarse, with gravel, also clay streaks.	"Heavy chatter" Added 2 sacks Revert
× 86	89	3	Sand, as above.	"Drills soft but has streaks of heavy chatter
89	99	10	Sand, fine to coarse, with 5% gr 1 1/4" to 1/2", also light gray, cream,	
ı			and light poddich busing clause of the first took at the annual	ret c

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BEGUN 2/1/84 FINISHED 2/1/84 LOGGED BYW.P.StilsonDRILLED BY Holland Well Company DRILL HOLE 0-76
GRADE
ELEV. 31,7' TOTAL DEPTH 114' LOCATION Stauffer Grid Coordinates S. 2309.5 E. 2097.

Bits: 9-3/4" Drag Bit 0 to 10';

Foo	TAGE	THICK-	2 TO CO 1.D. Mud. Domisons Revert	
FROM	ТО	NESS	LITHOLOGY	REMARKS
99	112	13	Sand, fine to coarse (with some buff to light yellowish silty sand) mixed with clay streaks.	
112	114	2	Clay, stiff, about 1' yellow-brown to orange silty clay on top of stiff blud clay.	
T.D.	114	·	In stiff bluish clay.	
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3 10 -				
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APPENDIX XXVIII
WATER-LEVEL DATA

BÂRIUE EO RETKES LADINHSET ONIFERNIONE NEETESW



Stauffer Chemical Company 1391 S. 49th St. Richmond, CA 94804 Tel. (415) 231-1300 TELEX 36-1-12

October 30, 1987

Mr. Mark Taylor Camp Dresser & McKee, Inc. 2100 Riveredge Parkway Suite 400 Atlanta, Georgia 30328

Dear Mr. Taylor:

The scheduled program to obtain additional water level and pumping rate data from the Cold Creek-LeMoyne site, as set forth in my September 28. 1987 letter to Ms. Van Duzee (EPA, copy to Mark Taylor), was essentially terminated on the morning of October 7th and rescheduled for October 13th through 15th, 1987. Ms. Van Duzee was on site to oversee data collection on October 14th and 15th.

Groundwater level and pumping rate data obtained between October 6 and 16. 1987 are presented according to the CDM Reference number and in the same format used in my July 31. 1987 letter, in which the data from last November's program were presented. The groundwater level data are presented in Attachment 1, information pertaining to the amount of water pumped from Stauffer's water supply wells is presented in Attachment 2, and information pertaining to the amount of water pumped from Stauffer's Groundwater Intercept System wells is presented in Attachment 3.

In an attempt to minimize the problems encountered during the data collection period in November 1986, personnel at both plants were contacted during mid and late September and made aware of the program scheduled for the week of October 5th, 1987. The Cold Creek plant indicated that they would supply their water needs from well CC-12 (as they usually do) and would not schedule any operations requiring above "normal" amounts of water. Additionally, it was agreed that the LeMoyne plant would 1) shut down wells LM-5 and LM-10 on or about September 28th, and that these wells would remain off until the program was completed; 2) not schedule any maintenance work requiring the shutdown of any operating well; and 3) not schedule operations requiring above "normal" amounts of water.

If the program had been completed by October 9, 1987, as originally scheduled, we would have been successful in obtaining a set of water level data from just about as stable of a pumping rate and pattern as is likely to occur at the Cold Creek-LeMoyne site. However, the water supply header from well LM-6 broke between 1500 and 1600 hours on Friday, October 9th. This resulted in the plant shutting down LM-6 so the water line could be repaired and turning on wells LM-5 and LM-10 to provide sufficient water to keep the various plants operating.

Mr. Mark Taylor October 30, 1987 Page 2

With well LM-6 shut down for almost 70 hours, the approximately 18-foot deep cone of depression would have almost totally recovered; and, water levels in the vicinity of wells LM-5 and LM-10 would be declining rapidly from Friday afternoon until Monday. Then, by Monday afternoon, the 12th, the wells were returned to their pre-existing pumping pattern and water level trends in the respective areas underwent rapid reversals. As a result of the above, groundwater levels for wells completed in the lower confined aquifer in the northwestern part of the property (0-8, 0-27, CC-11, LM-5, & LM-10) were lower by one to several feet on October 13th than they were on October 7th.

The intercept well system at LeMoyne has been operating at a fairly steady rate over the past several weeks. The center well (IW-2) is now pumping about 325 gpm, which is only 72 percent of the designated operating rate. This decrease in pumping rate results in the elevation of the water surface in the center well being a few tenths of a foot higher than the elevation of the water surface at either of the two end wells; and does not appear to significantly reduce the effectiveness of the intercept system.

The average daily pumping rate from well CC-12 at the Cold Creek Plant has ranged from a high of 235 gpm on September 28, 1987 to a low of 108 gpm on October 8, 1987, and the overall average daily pumping rate was 168 gpm for the 23-day period (9/28/87-10/20/87).

Mr. John Stewart, the Environmental Services Manager at Courtauld's plant informed me that their water use has remained at a fairly steady rate of about 7.5 million gallons per day for the period from September 28th to October 19th, 1987. They are using ten wells to obtain this volume of water. Nine wells are process water supply wells producing about 550 gpm each, and the tenth well is their drinking water supply well (No. 16) which may be producing 200 to 250 gpm. From October 9-12, they were using wells nos. 3, 4, 5, 7, 9, 10, 11, 13, 16, and 17; and from October 12-15 they were using well nos. 3, 4, 5, 7, 9, 11, 13, 14, 16, and 17. They use their well no.14 as a "swing well", which they put on line when one of the regularly used wells are taken off line for treatment.

We feel certain that the groundwater elevation data resulting from the October 1987 field effort (see Attachments) provide a more adequate data base than do those data obtained in November 1986; and furthermore, that these new data will be of significant benefit to anyone attempting to develop and calibrate a groundwater flow model for the area of the Stauffer plant sites.

3 10 00537-

Mr. Mark Taylor October 30, 1987 Page 3

If you have any questions regarding the attached data or comments, please do not hesitate to contact me.

Very truly yours,

William P. Stilson Senior Geologist

WPS:16

cc: Ms. Ellen Van Duzee Site Project Manager Environmental Protection Agency 345 Courtland Street Atlanta, GA 30365

> Mr. Joe Downey Alabama Dept. of Environmental Management 1751 Federal Drive Montgomery, AL 36130

Mr. Kurt Batsel
Camp Dresser & McKee, Inc.
2100 Riveredge Drive
Suite 400
Atlanta, GA 30328

- L. Erickson
- D. Flack
- R. Halstead
- T. Sayers
- D. Smith

ATTACHMENT I

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GROUNDWATER LEVEL MEASUREMENTS FOR WELLS IN SECTION V Based on Depth to Water Measurements Made on October 6-16, 1987

CDM Reference No. Section V	Stauffer Well Description	Date	Time	Elevation Top of Casing or Measuring Point	Depth to Water	Elevation of Groundwater	Comment	10 005
None	LM-2 Process Water Supply	10-06-87 10-07-87 10-08-87 10-13-87 10-14-87 10-15-87 10-16-87	1610 1145 1227 1040 1359 1645	46.70	77.53* 79.84* 76.14* 79.14* 74.98*	-30.83 -33.14 -29.44 -32.44 -28.28	pumping pumping pumping pumping pumping	ω Ω1
None	LM-5 Process Water Supply	10-06-87 10-07-87 10-08-87 10-09-87 10-12-87 10-13-87 10-14-87 10-15-87	1540 1155 1220 1600 0930 1148 1051 1735 1345	44.09	50.23* 51.62* 51.39*	 - 6.14 - 7.53 - 7.30	well off well off well off well turned on well turned off well off well off well off well off	
65	LM-6 Process Water Supply	10-06-87 10-07-87 10-08-87 10-09-87 10-12-87 10-13-87 10-14-87 10-15-87	1550 1130 1234 1600 1200 0953 1104 1658 1325	44.18	67.68* 68.83* 67.68* 71.87* 68.60*	-23.50 -24.65 -23.50 -27.69 -24.42	pumping pumping pumping well turned off well turned on pumping pumping pumping pumping	
6 6	LM-7 Drinking Water and Process Water Supply	10-06-87 10-07-87 10-08-87 10-13-87 10-14-87 10-15-87	1532 1211 1218 1109 1042	40.98	58.20* 60.51* 60.51*	 -17.22 -19.53 -19.53	bendajue bendajud bendajud bendajud bendajud	

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GROUNDWATER LEVEL MEASUREMENTS FOR WELLS IN SECTION V Based on Depth to Water Measurements Made on October 6-16, 1987

CDM Reference No. Section V	Stauf Wel Descr		Date 	Time	Elevation Top of Casing or Measuring Point	Depth to Water	Elevation of Groundwater	Comment	10 00
67	I M-10	Drinking Water and	10-07-87	1222	43.91	43.98	- 0.07	well off	(M (G)
07	L11 10	Process Water Supply	10-08-87	1203	43.71	43.70	- 0.07	well off	9
		Trocess nacer suppry	10-09-87	1620				well turned on	
			10-12-87	1615				well turned off	
			10-13-87	1120		50.00	- 6.09	well off	
			10-14-87	1027		47.94	4.03	well off	
			10-16-87	1301				well off	
68	CC-11	Standby Only Drinking	10-07-87	1631	39.61	45.00	- 5.39	well off	
*		Water and Process Water Supply	10-13-87	1546		47.00	- 7.39	well off	
69	CC-12	Drinking Water and	10-07-87	1622	33.97			pumping	
		Process Water Supply	10-13-87	1615				pumpjng	
70	1W-1	Intercept Well	10-06-87	1352	36.42	55.05	-18.36	pumping	
			10-07-87	1027		55.16	-18.74	pumping	
			10-08-87	1246		55.10	-18.68	pumping	
			10-13-87	1003		55.38	-18.96	pumping	
			10-14-87	1121		55.41	-18.99	punping	
			10-15-87	0858		55.48	-19.06	pumping	
			10-16-87	1330				pumping	
71	IW-2	Intercept Well	10-06-87	1359	37.06	54.94	-17.88	թսաթ i ո ց	
			10-07-87	1034		55.03	-17.97	pumping	
			10-08-87	1255		55.13	-18.07	pumping	
			10-13-87	1010		55.20	-18.14	pumping	
			10-14-87	1130		55.27	-18.21	. pumping	
			10-15-87	0926		55.30	-18.24	pumping	
•			10-16-87	1332				brinbjuð	

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GROUNDWATER LEVEL MEASUREMENTS FOR WELLS IN SECTION V Based on Depth to Water Measurements Made on October 6-16, 1987

		•				-	
CDM Reference No. Section V	Stauffer Well Description	Date 	Time	Elevation Top of Casing or Measuring Point	Depth to Water	Elevation of Groundwater	Comment
72	IW-3 Intercept N	dell 10-06-87	1406	36.01	54.34	-18.33	pumpIng
12	IM-3 Incercebe	10-03-67 10-07-87	1040	30.01	54.47	-18.46	
		10-08-87	1303		54.59	-18.58	brimbiling —
		10-13-87	1015		54.58	-18.57	pumping pumping
		10-14-87	1135		54.65	-18.64	pumping
		10-15-87	0938		54.70	-18.69	pumping pumping
		10-16-87	1334		54.70	10.07	pumping
		10 10 57	1334			. 	prantiting
		Based on Depth to Wa	iter Measi	urements Made on	October 1	3-15, 1987	
73	#0-17	10-13-87	1611	38.95	27.38	+11.57	
74	"0- 5	10-13-87	1551	34.92	23.17	+11.75	
75	0-8	10-13-87	1520	34.73	42.39	- 7.67	completed in confined
76	0-16 (CCM-9)	10-13-87	1536	31.40	31.77	- 0.37	aquifer
77	CCM-6	10-13-87	1528	28.04	30.07	- 2.03	
78	0-14 (CCH-8)	10-13-87	1328	31.80	33.90	- 2.10	
79	0-26 (CCM-11)	10-13-87	1556	46.85	36.91	+ 9.94	
80	0-27	10-13-87	1601	45.82	52.68	- 6.86	completed in confined
81	0-25	10-13-87	1126	47.24	55.21	- 7.97	aquifer
82	CCH-7	10-13-87	1515	33.23	37.27	- 4.04	
83	0-21 (CCM-10)	10-13-87	1334	30.26	35.98	- 5.72	
84	0-18	10-13-87	1134	45.27	53.12	- 7.85	
85	0-80 (HM-1)	10-13-87	1136	44.08	52.03	- 7.95	
86	1M-3	10-13-87	1447	45.59	52.88	- 7.29	
. 87	CCM-3	10-13-87	1500	32.79	39.50	- 6.71	
88	0-6	10-14-87	1411	42.96	54.96	-11.85	
							top of casing on well
89	0-78	10-14-87	~				broken off - substitute 0-77 for 0-78
None	0-77	10-14-87	1406	45.17	52.25	-14.08	
90	0-35	10-14-87	1731	39.35	52.45	-13.10	
91	0-33	10-14-87	1739	37.35	51.14	- 13.79	
92	0~31	10-15-87	0953	38.42	53.46	- 15.04	
60	Δ 44	10 14 07	174,9	115 2A	AH 24	13.00	

GROUNDWATER LEVEL MEASUREMENTS FOR WELLS IN SECTION V Based on Depth to Water Measurements Made on October 13-15, 1987

CDM Reference No. Section V	Stauffer Well Description	Date	Time	Elevation Top of Casing or Measuring Point	Depth to Water	Elevation of Groundwater	Comment	3 10 (
								0
94	#0-46	10-14-87	1649	39.07	50.30	-11.23		(J1
95	0-38	10-14-87	1719	32.53	46.54	-14.01		4
96	0-42	10-14-87	1642	40.56	51.74	-11.18		
97	0-48	10-14-87	1535	32.45	45.48	-,13.03		
98	0-74	10-14-87	1547	34.19	45.84	-11.65		
99	0-66	10-14-87	1528	36.83	49.92	-13.09		
100	0-62	10-14-87	1451	35.04	46.47	-11.43		
101	0-56	10-14-87	1442	30.44*	41.09	-10.65		
102	0-59	10-14-87	1430	38.13	47.72	- 9.59		
103	0-53	10-14-87	1419	35.41*	44.21	- 8.80		
104	0-70	10-14-87	1557	30.71	39.53	- 8.82		
105	0-68	10-14-87	1623	26.76	34.79	- 8.03		
106	0-50 (CCM-12)	10-13-87	1343	41.34	44.45	- 3.11		
107	0-51	10-13-87	1356	47.27	50.70	- 3.43		
108	0-52	10-13-87	1351	47.51	52.62	- 5.11		
109	0-23	10-14-87	1232	13.48	15.22	- 1.47		
110	0-24	10-14-87	1200	13.90	14.12	- 0.22		
111	NM- I	10-14-87	1225	46.91	52.64	- 5.73		
None	0-22	10-14-87	1215	34.83	34.66	- 3.40		
112	NM-2	10-14-87	1210	24.91	27.20	- 2.29		
	CNAM-30	10-15-87	1038	41.26	55.20	-13.94		
114	CNAH-25	10-15-87	1054	43.18	57.38	-14.20		
115	CNAM-32	10-15-87	1108	37.99	52. 9 4	-14.95		
116	CNAM-22	10-15-87	1149	41.77	55.84	-14.07		
117	CNAM-34	10-15-87	1134	45.00	59.70	-14.70		
118	CNAM-23	10-15-87	1221	36.34	49.69	-13.35		
	(Courtaulds Well MW-l) CNAM-l	10-15-87	1202	37.25	50.40	-13.15		
	(Courtaulds Well MW-2) CNAM-2	10-15-87	1212	45.45	58.46	-13.01		
	(Courtaulds Well MW-3)	10-15-87		48.81			did not make	Measur (more

ATTACHMENT 2

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GROUNDWATER PUMPING RATES AT COLD CREEK - LEMOYNE, ALABAMA FOR WATER SUPPLY WELLS AND PROCESS WATER WELLS IN SECTION V (CDM REFERENCE) Based on Information Obtained During Week of October 5-9 and 12-16, 1987

CDM Reference No. Section V	Stauffer Well Description	Date	Time	Discharge Pressure (psig)	<u>Estimated</u> Flow Rate (gpm)	Comments
(None)	LM-2 (LeMoyne Plant	10-06-87	1610	61.0(T)	580	(T) test gauge (7)
	process water well)			59.5	600	4.
		10-07-87	. 1145	· 56.5(T)	640	2
	•			56.5	640	
		10-08-87	1227	52.5	680	
		10-13-87	1040	62.5(T)	550	
				61.0	580	
		10-14-87	1359	58.0	620	
		10-15-87	1645	62.5	550	
		10-16-87	1345	55.0	660	
(None)	LM-5 (LeMoyne Plant	10-06-87	1540	0	0	well off
	process water well)	10-07-87	1155	0	0	well off
		10-08-87	1220	0	0	well off
		10-09-87	1600			well turned on
		10-09-87	1720			well turned off
		10-10-87	0720			well turned on
		10-10-87	1130		~ ~ ~	well turned off
		10-10-87	1415		 -	well turned on
		10-12-87	0930			well turned off
		10-13-87	0944	0	0	well off
		10-13-87	1148	0	0	well off
		10-14-87	1051	0	0	well off
		10-15-87	1735	0	0	well off
		10-16-87	1316	0	0	well off
65	LM-6 (LeMoyne Plant	10-06-87	1550	67.0(T)	430	
	process water well)	10-07-87	1130	63.5(T)	480	
		10-08-87	1234	63.0	470	
		10-13-87	0953	68.0(T)	410	
		10-14-87	1104	59.5(T)	520	
				61.2(?)	490	(?) gauge
		10-15-87	8	65.0(1) 60.0(?)	460	Lionable
		10 10 00	4 - 2	00.0(;)	520	

ATTACHMENT 2

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GROUNDWATER PUMPING RATES AT COLD CREEK - LEMOYNE, ALABAMA FOR WATER SUPPLY WELLS AND PROCESS WATER WELLS IN SECTION V (CDM REFERENCE) Based on Information Obtained During Week of October 5-9 and 12-16, 1987

CDM Reference No. Section V	Stauffer Well Description	Date	Time	Discharge Pressure (psig)	Estimated Flow Rate (gpm)	Comments (7)
66	LM-7 (LeMoyne Plant	10-06-87	1532	broken gauge	520	5 min. average flow
	Process Water Well			J	(440-560)	3
		10-07-87	1211	65-74(T)	370	10 min. average flow
		10-08-87	1218	broken gauge	400	10 min. average flow
		10-13-87	1109	74.0(N)	380	(N) installed new gaug
				72.0(T)		10 min. average flow
		10-14-87	1042	72.0(N)	340	10 min. average flow
		10-16-87	1257	66-75(N)	360	5 min. average flow
	•				(180~440)	
67	LM-10 (LeMoyne Drinking	10-07-87	1222	0	0	well off
	and Process Water Well)	10-08-87	1203	0	0	well off
		10-09-87	1620			well turned on
		10-12-87	1615	0	0	well turned off
		10-13-87	1120	0	0	well off
	•	10-14-87	1027	0	0	well off
		10-16-87	1301	0	, 0	well off
68	CC-11 (Cold Creek Drinking	10-07-87	1631	0	0	well off
	and Process Water Well)	10-13-87	1546	0	0	well off
69	CC-12 (Cold Creek Drinking	10-07-87	1622	76.0	130	10 min. average flow
	and Process Water Well)	10-13-87	1615	61.0	180	10 min. average flow

ATTACHMENT 3

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GROUNDWATER PUMPING RATES AT COLD CREEK -LEMOYNE ALABAMA FOR GROUNDWATER INTERCEPT SYSTEM WELLS (AT LEMOYNE) IN SECTION V (CDM REFERENCE) Based on Information Obtained During Week of October 5-9 and 12-16, 1987

CDM Reference No. Section V	Stauffer Well Description	Date	Time	Discharge Pressure (psig)	Estimated Flow Rate (gpm)	Comments (7)
70	IW-1 intercept Well No. 1	10-06-87	1352	96	330	4
70	(east)	10-07-87	1027	94	375	by difference
	(6656)	10-08-87	1246	96	325(?)	by meter(?) meter
		.0 00 0.	1240	,,	323(1)	not working proper le
	·				375	by difference
		10-13-87	1003	96	375	by difference
		10-14-87	1121	96	355	2, 2,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
		10-15-87	0858	96	330	by meter
		10-16-87	1330	96	360	by meter
- 71	IW-2 Intercept Well No. 2	10-06-87	1359	58	310	
•	(center)	10-07-87	1034	58	325	
-	· · ·	10-08-87	1255	58	325	
		10-13-87	1010	58	325	
		10-14-87	1130	58	325	
		10-15-87	0926	57.5	320	
		10-16-87	1332	56	320	
72	IW-3 Intercept Well No. 3	10-06-87	1406	59	500	
	(west)	10-07-87	1040	59	500	•
		10-08-87	1303	60	510	
		10-13-87	1015	59	500	
		10-14-87	1135	59	520	
		10-15-87	0938	58	520	
		10-16-87	1334	58.5	520	
None	Total Flow IW-1, 2, & 3	10-06-87	1421		1180	4 min. average flow
	at Totalizing Flow Meter	10-07-87	1048		1200	3 min. average flour
	_	10-08-87	1309		1200	4 min. average flow
		10-13-87	1024		1200	5 min. average flow
		10-14-87	150 15		1200	10 min. Prage flow
		10~15-87	4 15		1 (80	10 min - Prage flow
		10 16 07	10		toma	Free in the second second

APPENDIX XXIX
ECKHARDT SURVEY INFORMATION
REGARDING THE LEMOYNE LANDFILL

INSTRUCTIONS

WASTE DISPOSAL SITE SURVEY .

PURPOSE

WHO IS INCLUDED IN THE SURVEY

TIME PERIOD TO BE COVERED FORMS: A: GENERAL FACILITY INFO FORMS: A: GENERAL FACILITY INFORMATION
B: DISPOSAL SITE INFORMATION
C: HAULER INFORMATION
D: SUPPLEMENTAL HAULER INFORMATION
COMPLETING THE FORM
WHO TO CALL WITH QUESTIONS

This Maste Disposal Site Survey is being conducted by the Subcommittee on Oversight and Investigations of the Committee on Interstate and Foreign Commerce, U. S. House of Representatives. The purpose of the survey is to begin to identify the location of sites in the United States used for the disposal of chemical plant process waste materials since 1950. The fifty largest chemical companies and their subsidiaries or affiliates are included in this first national survey. Information gathered will assist the Congress in addressing the problems posed by active as well as inactive or abandoned waste disposal sites. The information will also be useful to the U.S. Fauti-manager. The information will also be useful to the U. S. Environmental Protection Agency in effectively implementing the Resource Conservation and Recovery Act of 1976.

WHO IS INCLUDED IN THE SURVEY

The survey is based on the experiences of the 50 largest chemical The survey is based on the experiences of the 30 largest chemical companies in the United States. A complete set of instruments is to be filled out for every facility or plant in the United States owned, operated or leased by one of these companies or any of their subsidiary or affiliated companies.

companies in the out for every facility or plant in the United
States owned, operated or leased by one of these companies or
any of their subsidiary or affiliated companies.

It is recommended that the corporate headquarters, to whom the
survey instruments are sent, send a complete set of instruments
and the instruction manual to the plant manager or supervisor of
each of the company's plants or facilities. The plant manager or
or supervisor should then complete the forms using whatever
records or employee knowledge he or she may have at his or her.
disposal. It is further requested that completed forms from each
facility be returned to the corporate headquarters for finel
collation before returning them to the Subcommittee.

TIME PERIOD TO BE COVERED

It is the Subcommittee's intention to collect information on
waste disposal sites used since 1950 or since the time a
particular facility began operation if after 1950). Some facilities
may not have formal records on waste disposal dating back to:
1950. The Subcommittee requests that every effort is made to
records do not exist. Most important is the identification (by
name and location) of all sites used for the disposal of process
wastes from a facility since 1950. It is very likely that
employees with some tenure at a facility will know where
wastes were disposed of; thus the knowledge employees may have
of waste disposal practices by the facility should be explored
along with record searches.

FORMS

The instrument package consists of 4 separate forms as described below:

information on the total amount of process waste generated by a facility in 1978 and the methods used other than sale oy a ractifity in 1978 and the methods used other than sale of the disposal of these wastes. This information will provide a general picture of the facility's current operations. The form also requests information on the provided of sites used since 1950 for the disposal of process. wastes and the hauling of process wastes from the facility. The answers to these questions will indicate whether or not
(and in what numbers) Forms B, C and D need to be completed.

FORM B: DISPOSAL SITE INFORMATION: This form is to be completed for every disposal site used by the facility since 1950 for the disposal of the facility's process awastes. The property on which the facility is located may also have been used for waste disposal; if so, one Form

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"B" should be used for this "on site" disposal. The form elicits information on the name, location and ownership of the site, the dates the site was used by the facility, the amount and content of the process waste disposed at the site from the facility, the current status of the site as well as the types of disposal methods used at the site.

In all instances a facility should seek the answers to each of the questions... (NOTE: Form "B", consists of 2 pages.)

FORM C: HAULER INFORMATION: This form asks a facility to list the names and addresses of all firms or independent contractors (including the company and its affiliates and subsidiaries) who since 1050 accounts.

contractors (including the company and its affiliates and subsidiaries) who since 1950 removed process waste materials from the facility. Information on the years used is also requested.

FORM D: SUPPLEMENTAL HAULER INFORMATION: Some process wastes may have been hauled from a facility and taken to a location unknown to the facility. For every firm or contractor who has taken waste in this manner from a facility, Form D elicits information on the content and amount of wastes

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...

has taken waste in this manner from a facility, Form D elicits information on the content and amount of wastes hauled and the dates the hauler was used.

COMPLETING THE FORMS

The information requested on the forms is largely numerical in nature. Block spaces have been provided for this information. Respondents are requested to write (or type) responses clearly within these spaces. In any instance in which the response has fewer digits than the number of spaces provided, the response should be right justified. For example, if a facility generated 21,292 tons of process waste during 1978, the response on Form A, Question 3 would be recorded as follows:

hundred tons

hundred tons

hundred tons firm/contractor, three different categories (gallons, tons and cubic yards) have been provided. One or all of these categories may be used, depending upon which is most convenient for a facility. In no instance, however, should the amount of waste be double counted (i.e. a given amount should not be recorded as both gallons and tons).

All non-numerical responses (eg. names and addresses) should be written legibly or typed in the spaces provided. If there is a need to clarify responses to any questions, clarification should be provided on the back of the form or on an appended page.

WHO TO CALL WITH QUESTIONS

The Subcommissions.

The Subcommittee will hold a private briefing for all companies asked to participate in this survey on Friday, April 27, 1979 at 3:00 p.m. in 2123 Rayburn House Office Building in order to answer any questions or concerns. Companies should hold all-initial questions for this private briefing. Following this date, all questions should be directed to anne Cohn, Survey Coordinator at (202) 225-4231 or Mark Raabe, Staff Director, at (202) 225-4441.

3450<u>5</u> Albama Name of Person Completing Form: PLANT MANAGER Phone Number: (205) 1. Year Facility Opened ... Primary SIC Code Estimate the total amounts of process wastes (excluding wastes sold for use) generated by this facility during 1978: thousand gallons ... hundred tons thousand cubic yards Estimate (in whole percents) how these process wastes generated in 1978 were disposed of: in landfill .. in pit/pond/lagoon . incinerated reprocessed/recycled evaporated other (Specify____ What is the total number of known sites (including disposal on the property where this facility is located as one site) that have been used for the disposal of process wastes from this facility since 1950?..... DO-PLETE ONE FORM "B" FOR EACH OF THE SITES Have any of the process wastes generated at this facility been hauled (removed) from this facility for disposal? (Yes=1; no=2) IF YES, COMPLETE FORM "C" Do you know the disposal site locations of all of the process waste hauled from your facility since 1950? (Yes=1; no=2) IF NU, COMPLETE ONE FORM "ID" FOR EACH FIRM OR CONTRACTOR Specify the earliest year represented by information from employee browledge supplied on this and other forms

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LITTLY NEW LE NIGHER FLANT (TO	<u>o)</u>	•	3
teller ARGUNIST DUNIP SITE	_		و سارم
Components (or characteristics) of process widesposed at site: (1-present in waste; 2-mot 9-dun't know)	waste from this facility present in waste;	y	74. 4
FILL IN EVERY BLOCK SPACE	er e		
Acid solutions, with pH<3. pickling liquor netal plating waste circuit etchings inorganic acid manufacture organic acid manufacture Base solutions, with pH>10 caustic soda manufacture nylon and similar polymer generation scrubber residual Heavy metals & truce metals (bonded organics arsenic, selenium, antimony mercury iron, manganese, magnesium zinc, cadatum, copper, chromium (trivale chromium (hexavalent) lead Radioactive residuals & residuals for UF6 re lathanide series elements and rare earth phosphate slag thorium radium other alpha, beta & gamma emitters Organics pesticides & intermediates herbicides & intermediates herbicides & intermediates halogenated aliphatics halogenated aliphatics halogenated aliphatics solvents polar (except water) carbontetrachloride trichloroethylene other solvents nonpolar solvents halogenated aliphatic solvents halogenated aromatic oils and oil sludges esters and ethers alcohols ketenes & aldehydes dioxins Inorganics salts mercaptans Misc. thurmaceutical wastes	mt) cycling salts		11) 12) 13) 14) 15) 16) 17) 18) 19) 22) 22) 22) 22) 22) 22) 23) 24) 25) 27) 28) 33) 33) 33) 33) 40) 41) 44) 44) 45) 55) 56) 57) 58) 59) 51) 55) 56) 57) 58) 59) 59) 50) 51) 52) 53) 54) 55) 56) 57) 58) 58) 58) 58) 58) 58) 58) 58